

Pyrazin-2(1*H*)-ones from 3,4-Disubstituted 4-Aminoisoxazol-5(4*H*)-ones

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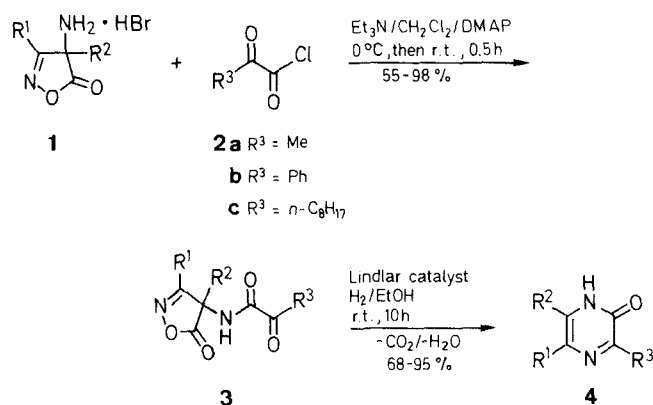
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A new synthesis for 3,5,6-trisubstituted pyrazin-2(1*H*)-ones from 3,4-disubstituted 4-(phenyloxalylamino)- and 4-(alkyloxalylamino)-isoxazol-5(4*H*)-ones is reported.

We have previously reported on the synthesis of 3,4-disubstituted 4-aminoisoxazol-5(4*H*)-ones **1** from isoxazol-5(4*H*)-ones and on their use in a new synthesis, affording imidazoles in high yield.¹

We report now a new and easy synthesis of trisubstituted pyrazin-2(1*H*)-ones **4** starting from 4-aminoisoxazol-5(4*H*)-ones **1** (Scheme). The amines **1** are first transformed into the corresponding α -oxo acid amides **3** by reaction with the corresponding oxo acid chlorides **2**.² The hitherto unknown 2-oxodecanoyl chloride (**2c**) was prepared according to a known method,² and used directly as raw material in the next step without purification due to its instability (Table 1). The amides **3** are reduced with hydrogen in the presence of Lindlar catalyst in ethanol at room temperature. Hydrogenolysis of the N–O bond, followed by decarboxylation and intramolecular cyclization gives pyrazinones **4** in good yield (Table 2). Hydrogenation with palladium on charcoal or with Raney-nickel is faster but always gives lower yields. Spectroscopic data of the new compounds are reported in Table 3.

This synthesis is a direct and convenient method for the preparation of trisubstituted pyrazin-2(1*H*)-ones **4**, as an alternative to the previously reported methods from α -amino acid amides and 1,2-dicarbonyl compounds.³



1	R ¹	R ²	3, 4	R ¹	R ²	R ³	3, 4	R ¹	R ²	R ³
a	Me	<i>i</i> -Pr	aa	Me	<i>i</i> -Pr	Me	ca	Ph	Bn	Me
b	Me	Bn	ab	Me	<i>i</i> -Pr	Ph	cb	Ph	Bn	Ph
c	Ph	Bn	ac	Me	<i>i</i> -Pr	<i>n</i> -C ₈ H ₁₇	cc	Ph	Bn	<i>n</i> -C ₈ H ₁₇
d	Ph	Et	ba	Me	Bn	Me	da	Ph	Et	Me
			bb	Me	Bn	Ph	db	Ph	Et	Ph
			bc	Me	Bn	<i>n</i> -C ₈ H ₁₇				

Scheme

Melting points are determined on a Büchi apparatus and are uncorrected. IR spectra are recorded on a Perkin-Elmer 298 instrument, in Nujol mull for solids and liquid film for oils. ¹H-NMR are

Table 1. 3,4-Disubstituted 4-(Phenyloxalylamino)- and 4-(Alkyloxalylamino)isoxazol-5(4*H*)-ones **3** Prepared

Prod- uct	Yield (%)	mp (°C) (solvent) ^a	Molecular Formula ^b
3aa	87	117–118 (Et ₂ O)	C ₁₀ H ₁₄ N ₂ O ₄ (226.2)
3ab	98	116–117 (<i>i</i> -Pr ₂ O)	C ₁₅ H ₁₆ N ₂ O ₄ (288.3)
3ac	55	60–61 (Hx)	C ₁₇ H ₂₈ N ₂ O ₄ (324.4)
3ba	80	139–140 (Et ₂ O)	C ₁₄ H ₁₄ N ₂ O ₄ (274.3)
3bb	89	159–160 ^c (CH ₂ Cl ₂ /Hx)	C ₁₉ H ₁₆ N ₂ O ₄ (336.3)
3bc	77	103–104 (Et ₂ O/Hx)	C ₂₁ H ₂₈ N ₂ O ₄ (372.4)
3ca	88	142–143 (CH ₂ Cl ₂ /Hx)	C ₁₉ H ₁₆ N ₂ O ₄ (336.3)
3cb	88	156–157 ^c (CH ₂ Cl ₂ /Hx)	C ₂₄ H ₁₈ N ₂ O ₄ (398.4)
3cc	57	oil	C ₂₆ H ₃₀ N ₂ O ₄ (434.5)
3da	82	96–97 (Et ₂ O/Hx)	C ₁₄ H ₁₄ N ₂ O ₄ (274.3)
3db	93	153–154 ^c (CH ₂ Cl ₂ /Hx)	C ₁₉ H ₁₆ N ₂ O ₄ (336.3)

^a Hx = hexane.

^b Satisfactory microanalyses obtained: C \pm 0.12, H \pm 0.1, N \pm 0.12.

^c With decomposition.

Table 2. Pyrazin-2(1*H*)-ones **4** Prepared

Prod- uct	Yield (%)	mp (°C) (solvent) ^a	Molecular Formula ^b
4aa	90	140–142 (Hx)	C ₉ H ₁₄ N ₂ O (166.2)
4ab	81	187–188 (CH ₂ Cl ₂ /Hx)	C ₁₄ H ₁₆ N ₂ O (228.3)
4ac	78	57–59 (Hx)	C ₁₆ H ₂₈ N ₂ O (264.4)
4ba	77	156–157 (CH ₂ Cl ₂ /Hx)	C ₁₃ H ₁₄ N ₂ O (214.3)
4bb	95	203–204 (CH ₂ Cl ₂ /Hx)	C ₁₈ H ₁₆ N ₂ O (276.3)
4bc	68	114–115 (Et ₂ O/Hx)	C ₂₀ H ₂₈ N ₂ O (312.4)
4ca	81	193–194 (CH ₂ Cl ₂ /Hx)	C ₁₈ H ₁₆ N ₂ O (276.3)
4cb	92	258–260 (CH ₂ Cl ₂ /Hx)	C ₂₃ H ₁₈ N ₂ O (338.4)
4cc	82	109–111 (Et ₂ O/Hx)	C ₂₅ H ₃₀ N ₂ O (374.5)
4da	77	182–183 (CH ₂ Cl ₂ /Hx)	C ₁₃ H ₁₄ N ₂ O (214.3)
4db	80	250–252 (CH ₂ Cl ₂ /Et ₂ O)	C ₁₈ H ₁₆ N ₂ O (276.3)

^a Hx = hexane.

^b Satisfactory microanalyses obtained: C \pm 0.12, H \pm 0.1, N \pm 0.12.

recorded on a Varian EM-390 or on a Bruker AC 300 spectrometer with TMS as internal standard in CDCl₃ if not stated otherwise. Lindlar catalyst (5% Pd–CaCO₃, poisoned with lead) obtained from Fluka. Column chromatography is performed on Merck Kieselgel 60, 0.063–0.2 mm. Evaporation is carried out under vacuum in a rotary evaporator.

2-Oxodecanoyl Chloride (**2c**):

A mixture of 2-oxodecanoic acid⁴ (3.72 g, 20 mmol) and Cl₂CHOMe (2.7 mL, 30 mmol) is heated at an oil bath temperature of 50°C for 30 min. Evaporation at 5 Torr gives the chloride **2c** which is used directly in the next step without purification.

4-(Alkyloxalylamino)- or 4-(Phenyloxalylamino)isoxazol-5(4*H*)-ones **3**; General Procedure:

The 4-aminoisoxazol-5(4*H*)-one hydrobromide **1** (4 mmol) is dissolved in CH₂Cl₂ (50 mL) and then Et₃N (1.4 mL, 10 mmol) and DMAP (50 mg, 0.41 mmol) are added. The solution is cooled to 0–5°C and the acid chloride **2** (5 mmol) in CH₂Cl₂ (10 mL) is added under stirring. After 30 min at r.t., the mixture is washed with H₂O (30 mL) and dil. aq HCl (4%, 30 mL). The organic layer is dried (Na₂SO₄), filtered and evaporated. The residue is purified by silica gel column chromatography, eluent hexane/CH₂Cl₂ (1 : 1).

Table 3. Spectral Data of Compounds 3, 4

Product	IR (Nujol or film) (cm ⁻¹)	¹ H-NMR (CDCl ₃ /TMS) δ, J (Hz)
3aa	3380, 1798, 1724, 1685	1.05 (d, 3H, J = 7.5), 1.20 (d, 3H, J = 7.5), 2.02 (s, 3H), 2.26 (m, 1H), 2.50 (s, 3H), 7.45 (br s, 1H) ^a
3ab	3318, 1784, 1681, 1665	1.05 (d, 3H, J = 7), 1.22 (d, 3H, J = 7), 2.07 (s, 3H), 2.23 (m, 1H), 7.41 (m, 3H), 7.76 (br s, 1H), ^a 8.31 (m, 2H)
3ac	3365, 1793, 1722, 1680	0.86 (t, 3H, J = 7), 1.02 (d, 3H, J = 7.2), 1.15 (d, 3H, J = 7.2), 1.24 (m, 8H), 1.57 (m, 4H), 1.94 (s, 3H), 2.17 (m, 1H), 2.84 (t, 2H, J = 7), 7.32 (br s, 1H) ^a
3ba	3360, 1796, 1728, 1695	2.06 (s, 3H), 2.48 (s, 3H), 3.28 (m, 2H), 7.15 (m, 2H), 7.31 (m, 3H), 7.53 (br s, 1H) ^a
3bb	3318, 1798, 1681, 1664	2.12 (s, 3H), 3.33 (m, 2H), 7.10–7.60 (m, 8H), 7.75 (br s, 1H), ^a 8.33 (m, 2H)
3bc	3310, 1780, 1729, 1678	0.90 (m, 3H), 1.32 (m, 10H), 1.55 (m, 2H), 2.05 (s, 3H), 2.85 (t, 2H, J = 7), 3.26 (s, 2H), 7.25 (m, 5H), 7.66 (br s, 1H) ^a
3ca	3280, 1804, 1729, 1690	2.44 (s, 3H), 3.46 (m, 2H), 6.96 (m, 2H), 7.22 (m, 3H), 7.50 (m, 3H), 7.75 (m, 3H) ^b
3cb	3263, 1805, 1692, 1660	3.45 (d, 1H, J = 12), 3.52 (d, 1H, J = 12), 6.98 (m, 2H), 7.26 (m, 3H), 7.51 (m, 6H), 7.82 (m, 2H), 8.01 (br s, 1H), ^a 8.33 (m, 2H)
3cc	3300, 1800, 1725, 1680	0.86 (m, 3H), 1.31 (m, 10H), 1.51 (m, 2H), 2.82 (t, 2H, J = 7), 3.43 (d, 1H, J = 13), 3.56 (d, 1H, J = 13), 6.90 (m, 2H), 7.21 (m, 3H), 7.45 (m, 3H), 7.81 (m, 2H), 8.16 (br s, 1H) ^a
3da	3325, 1813, 1741, 1686	0.97 (t, 3H, J = 7.5), 2.26 (q, 2H, J = 7.5), 2.44 (s, 3H), 7.52 (m, 3H), 7.59 (br s, 1H) ^a 7.75 (m, 2H)
3db	3240, 1797, 1687, 1649	0.97 (t, 3H, J = 7.5), 2.25 (q, 2H, J = 7.5), 7.44 (m, 5H), 7.78 (m, 3H), 8.05 (br s, 1H) ^a 8.26 (m, 2H)
4aa	1647	1.35 (d, 6H, J = 7.5), 2.33 (s, 3H), 2.45 (s, 3H), 3.05 (m, 1H), 12.50 (br s, 1H) ^a
4ab	1642	1.39 (d, 6H, J = 7), 2.42 (s, 3H), 3.11 (m, 1H), 7.38 (m, 3H), 8.42 (m, 2H), 12.60 (br s, 1H) ^a
4ac	1653	0.86 (t, 3H, J = 7.5), 1.25 (m, 6H), 1.31 (m, 8H), 1.63 (m, 4H), 2.28 (s, 3H), 2.75 (m, 2H), 3.01 (m, 1H), 11.20 (br s, 1H) ^a
4ba	1647	2.37 (s, 3H), 2.46 (s, 3H), 3.95 (s, 2H), 7.28 (m, 5H), 13.05 (br s, 1H) ^a
4bb	1634	2.5 (s, 3H), 4.02 (s, 2H), 7.30 (m, 8H), 8.35 (m, 2H), 13.50 (br s, 1H) ^a
4bc	1649	0.86 (m, 3H), 1.30 (m, 10H), 1.65 (m, 2H), 2.38 (s, 3H), 2.75 (t, 2H, J = 7), 3.91 (s, 2H), 7.28 (m, 5H), 12.00 (br s, 1H) ^a
4ca	1667	2.49 (s, 3H), 4.01 (s, 2H), 7.20 (m, 5H), 7.42 (m, 5H), 12.02 (br s, 1H) ^a
4cb	1650	DMSO- <i>d</i> ₆ : 4.01 (s, 2H), 7.06 (m, 2H), 7.24 (m, 3H), 7.42 (m, 8H), 8.34 (m, 2H), 12.78 (br s, 1H) ^a
4cc	1661	0.86 (m, 3H), 1.25 (m, 10H), 1.69 (m, 2H), 2.75 (t, 2H, J = 7), 4.01 (s, 2H), 7.12 (m, 2H), 7.26 (m, 3H), 7.42 (m, 5H), 12.02 (br s, 1H) ^a
4da	1655	1.36 (t, 3H, J = 8), 2.55 (s, 3H), 2.73 (q, 2H, J = 8), 7.43 (m, 5H), 7.91 (br s, 1H) ^a
4db	1651	1.44 (t, 3H, J = 7.5), 2.82 (q, 2H, J = 7.5), 7.44 (m, 5H), 7.57 (m, 3H), 8.49 (m, 2H), 13.10 (br s, 1H) ^a

^a Exchange with D₂O.^b 2H after D₂O.**3,5,6-Trisubstituted Pyrazin-2(1H)-ones 4; General Procedure:**

The α-oxo acid amide 3 (2 mmol) is dissolved in EtOH (40 mL), Lindlar catalyst (120 mg) is added and the mixture is hydrogenated under atmospheric pressure at r. t. After 10 h the catalyst is filtered off, the solvent evaporated and the residue crystallized from the reported solvent (Table 2).

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