Preparation of Enantioenriched Tetrahydropyridines by Iminium Ion-Vinylsilane Cyclizations

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Abstract. Tetrahydropyridines can be prepared from (S)-amino acids in high enantiomeric purity as outlined in equation 1.

Electrophilic cyclization reactions of iminium ions (Mannich cyclizations) constitute some of the most important methods for preparing nitrogen heterocycles. A continuing theme in the evolution of this chemistry has been the development of selective methods for initiating and terminating the cyclization event,^{1,2} In recent years vinylsilanes have emerged as extremely valuable π -nucleophiles in Mannich-type cyclizations.^{2,3} A wide variety of nitrogen heterocycles containing endocyclic or exocyclic unsaturation have been prepared by iminium ion-vinylsilane cyclizations. Among the more important ring systems assembled in this way is the 1,2,5,6-tetrahydropyridine ring, a heterocyclic unit found in a variety of alkaloids and other pharmacologically active agents.⁴ In this Letter we report, contrary to the conclusions of an earlier study,⁵ that substituted 1,2,5,6-tetrahydropyridines can be prepared in high enantiomeric purity by Mannich cyclizations of amino acid-derived 4-silyl-3-alkenylamines (eq 1). Differing methods of iminium ion generation and their effect on the stereochemical outcome of these iminium ion-vinylsilane cyclizations are also described.



Enantiopure (S)-N-benzyl-5-(trimethylsilyl)-4Z-pentenyl-2-amine (6) was prepared from Lalanine along lines outlined earlier.⁵ Amine 6 of excellent enantiomeric purity (>96% ee) could be obtained from reduction of benzamide 3 or imine 4, or more conveniently by N-benzylation of 1 to afford 5 followed by desulfonylation with Na(Hg).⁶ The enantiomeric purity of 6 was best assessed by converting it to the *N*-tosyl derivative 5 followed by HPLC analysis using a Chiracel OD column.⁷ Pentenylamine 6, prepared by the preferred $1 \rightarrow 5 \rightarrow 6$ sequence, was >98% enantiomerically pure.



We initially examined the formation of tetrahydropyridines 10 and 11 (eq 2). Three methods for forming the iminium ion intermediates were examined: direct acid promoted reaction of 6 with paraformaldehyde or formatin, oxidation of α -silylmethylamine 7,¹ and silver promoted decyanation of cyanoalkylamines 8 and 9. As summarized in Table 1, all precursors afforded enantioenriched tetrahydropyridine products 10 or 11.⁸ As reported earlier, only the *trans*-2,6-disubstituted stereoisomer 11 is produced from 9.⁵ The reactions summarized in entries 1-5 produced significant amounts of 1-benzyl-1,2,5,6-tetrahydropyridine (12), which required careful chromatography to separate it from 10. Monitoring the reaction summarized in entry 4 by GLC demonstrated that 10 and *N*-benzyl-4-(trimethylsilyl)-3*E*-pentenylamine (13) formed within hours, while 12 was produced more slowly, presumably from cyclization of 13. Tetrahydropyridine 12 was not detected (GLC analysis) in reactions in which water or formaldehyde were not present (e.g., Table 1, entries 6-8).



Results of related cyclizations of imine 4 and α -ethoxycarbamate 16⁹ are summarized in eqs 3 and 4. As previously reported, both reactions afford a ~1:1 mixture of stereoisomeric 2,6-disubstituted tetrahydropyridine products in crude yields of 50-70%.⁵ After separation by flash chromatography or HPLC, analytically pure 14 and 15 (and 17 and 18) were isolated in combined yields of 33-37%.

Scheme I

Entry	Cyclization Conditions ^a	Amine 10 or 11				12
		Cpd	Yield, %	[α]p ^b	ee, % ^c	Yield, % ^d
1	(CH ₂ O) _n , p-TsOH·H ₂ O, MeCN, 60 °C ^e	10	67-69	+33-36	64-67	(4-10)
2	(CH2O)n, p-TsOH, 4Å MS, MeCN, 60 °Ce	10	38-48	+50-54	97-98	15
3	(CH ₂ O) _n , TFA, MeCN, 70 °C	10	40-45	+56	>99	21
4	formalin, TFA, THF-H2O (1:3), 23 °C ^e	10	44-50	+57-62	98-99	(15-27)
5	formalin, TFA, THF-H ₂ O (1:3), 70 °C	10	44	+60	>99	9
6	7, (Bu4N)2Ce(NO3)6, MeCN, 23 °C	10	61	+56	>90 ^f	
7	8, AgBF4, MeCN, 100 °C	10	33-34	+50-51	92- 9 4	
8	9, AgBF4, MeCN, 100 °Ce	11	45-55	+136-147	79-83	

Table 1. Iminium Ion-Vinylsilane Cyclizations of 6 - 9.

^{*a*}The substrate for entries 1-5 was 6; 1.1-1.3 equiv of acid was employed. ^{*b*}In CHCl₃, c = 1.0; except entry 4 (c 0.4) and entries 2 and 7 (c 0.25). ^{*c*}By HPLC analysis (Chiracel OD) of 10 or capillary GLC analysis (J&W Cyclodex-B) of 11. ^{*d*}GLC yields are in parentheses. ^{*c*}Results of 2-3 identical experiments. ^{*f*}IH NMR analysis of the MPTA derivative of 2-methylpiperidine, prepared from 10, showed only a single diastereomer.

Enantiomeric purity of 14 and 15 was best determined by HPLC analysis (Chiracel OJ) of the carbamate derivatives 17 and 18. As shown in eqs 3 and 4, the tetrahydropyridine epimers were formed in identical enantiomeric purities: 85% for cyclization of the iminium ion derivative at 60 $^{\circ}$ C and 95% for cyclization of the more reactive *N*-ethoxycarbonyl iminium ion derivative at 0 $^{\circ}$ C.



The origin of the partial racemization seen in some of the cyclizations reported here is undoubtedly complex. As we have discussed, racemization results from a combination of aza-Cope equilibrations and iminium ion stereomutations.^{5a} It is not surprising that the formation of 10 in high enantiopurity (entries 2-5) is accompanied by significant formation of the "des-methyl" tetrahydropyridine 12, since conversion of 19 to 20 and eventually to 13 would prevent 19 from further rearrangements leading to racemization (Scheme II).^{5a} However, further mechanistic discussion must be deferred to a future full account of this work.



In summary, enantioenriched 6-substituted and 2,6-disubstituted tetrahydropyridines can be prepared by iminium ion-vinylsilane cyclizations of the L-alanine-derived silylpentenylamine 6. Cyclization is more rapid than racemization with both iminium ion (N-benzyl and N-H) and acyl iminium ion (N-ethoxycarbonyl) intermediates. Under optimum conditions, tetrahydropyridine products of 85-99% ee can be obtained. The extent of racemization in cyclizations of the N-benzyl formaldiminium ion to produce 10 varies somewhat with reaction conditions, although further investigations will be required to fully understand these effects. Lastly, oxidative cyclization of the α -silylmethyl precursor 7 allows 10 to be formed in good yield at room temperature and under neutral conditions.

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References and Notes

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- 7. Amine 6 yellows readily and its optical rotation is somewhat unreliable in gauging enantiomeric purity.
- 8. Contrary to the earlier report and consistent with the results of entry 7, reexamination of a sample of 11 prepared earlier⁵ showed that it was enantioenriched $[\alpha]_D + 106$ (c 0.9, CHCl₃).
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