A. E. Zelenin, N. D. Chkanikov,

UDC 542.91:547.484.23:547.551.1

A. F. Kolomiets, and A. V. Fokin

Saloutin et al. [1] have reported that methyl trifluoropyruvate reacts with aniline to form the N-alkylation product. We have found that carrying out this reaction in benzene at 120°C leads to 3-hydroxy-3-trifluoromethylbenzo[b]pyrrolidin-2-one (I).

The structure of pyrrolidone (I) was supported by 13 C NMR spectral data, mass spectral data, and elemental analysis. The 13 C NMR spectrum was taken on a Bruker WP-200SY spectrometer at 50.31 MHz. The chemical shifts were determined for acetone solutions relative to TMS as the internal standard.

 $\frac{3\text{-Hydroxy-3-trifluoromethylbenzo[b]pyrrolidin-2-one (I).}{3.12\text{ g methyl trifluoropyruvate in 8 ml anhydrous benzene}}$ A solution of 1.40 g aniline and 3.12 g methyl trifluoropyruvate in 8 ml anhydrous benzene was heated in a glass ampul for 6 h at 120°C. The ampul was opened, and the crystals formed were filtered off and washed with hexane. The mother liquor was evaporated to yield additional product. The total yield of white crystalline (I), mp 215-217°C, was 2.0 g (61.44%), R_f 0.25 (3:1 CCl₄-acetone). $^{13}\text{C-}^{14}\text{H}^{14}$ NMR spectrum (\$\delta\$, ppm): 76.01 (\$C^3\$, \$^2\$J_{C-F}\$ = 29.06 Hz), 111.32 (\$C^7\$), 123.45 (\$C^5\$), 124.60 (\$CF_3\$, \$^1\$J_{C-F}\$ = 285.80 Hz), 125.33 (\$C^9\$), 126.77 (\$C^6\$), 132.26 (\$C^4\$), 143.78 (\$C^8\$), 172.51 (\$C^2\$). Found: C 40.66; H 2.80; N 6.38; F 26.19%. Calculated for \$C_9\$H_6\$NO_2\$F_3: C 49.76; H 2.76; N 6.45; F 26.26%, \$M^{\dagger}\$ 217.

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A. N. Nesmeyanov Institute of Heteroorganic Compounds, Academy of Sciences of the USSR, Moscow. Translated from Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya, No. 4, pp. 955-956, April, 1985. Original article submitted December 5, 1984.