

РП: S0040-4039(96)01036-2

Efficient Synthesis of a Carbocyclic Core Moiety with the Stereochemistry of the C-1027 Chromophore

Itaru Sato, Yuri Akahori, Kyo-ichiro Iida, and Masahiro Hirama*

Department of Chemistry, Graduate School of Science, Tohoku University, Sendai, 980-77, Japan

Abstract: A stereoselective and concise route to the bicyclo[7.3.0]dodecadiyne core moiety (26) of the C-1027 chromophore (1) through highly efficient LiN(TMS)₂/CeCl₃-mediated cyclization of alkynyl aldehyde 24 has been established. Copyright © 1996 Elsevier Science Ltd

We recently reported the efficient cyclization of the conformationally non-rigid precursor 3 to the highly strained nine-membered cyclic diyne 4, a model compound for the C-1027 chromophore (1),¹ using LiN(TMS)₂/CeCl₃.^{2,3} However, the reaction of diastereomer 5 gave 6 as a mixture of stereoisomers in low yield (Scheme 1).² Very recently, the configuration of the C-1027 chromophore (1) has been determined, as shown below.⁴ Therefore, it would be useful in synthetic studies of 1 and related molecules such as kedarcidin (2)⁵ to know whether the yield and stereoselectivity of the intramolecular acetylide additions of diastereomers 7 and 9 are affected by the relative configurations of C4, C9 and C13.



C-1027 chromophore (1)

Kedarcidin chromophore (2)

Our results are shown in Scheme 2. Both 7 and 9 cyclized in better yield than 5 and in a highly diastereoselective manner. Thus, only diastereomer 5 was not an appropriate substrate for this cyclization reaction. Diastereomer 8, which possesses the same stereochemistry as $1,^{4b}$ has been synthesized effectively from 7. However, 7 had been prepared from minor product 13 of Grignard addition⁶ (Scheme 3), and the C12-C1 double bond was constructed by shifting from the C11-C12 position.² Therefore, we developed a stereoselective route to an intermediate corresponding to 13. After considerable preliminary experiments,⁷ we found that acyclic fragment 18 was the most suitable for obtaining the desired stereoisomer selectively through 1,2-chelation control in the addition reaction of the propargyl Grignard reagent (Scheme 4). Ketone 18 was



Scheme 1. Differences in efficiency and stereoselectivity between the intramolecular acetylide additions of diastereomers 3 and $5.^2$



Scheme 2. The efficient intramolecular acetylide addition reactions of diastereomers 7 and 9.

synthesized from L-(+)-dimethyl tartrate 14. While the reaction of 18 with the propargyl Grignard reagent in an ether solvent always gave a 1:1 mixture of 19⁸ and 20, regardless of the presence or absence of MgBr₂•OEt₂, stereoselectivity was improved to 2.5:1 in CH₂Cl₂/Et₂O (4:1).⁹ In the presence of ZnBr₂ (3.9 eq), high stereoselectivity favoring 19 (13:1) was achieved.¹⁰ Cyclopentene derivative 22^{11,12} was designed as a segment to couple with alkyne 21. This coupling was conducted under standard Hagihara-Sonogashira conditions¹³ to give 23. The TES group on the alkyne carbon of 23 was removed selectively,¹⁴ and the nitrile



Scheme 3. Addition reaction of the propargyl Grignard reagent to ketone 11.

group was then reduced to aldehyde 24, still as a diastereometric mixture. Treatment of 24 with LiN(TMS)₂/ CeCl₃ gave the desired nine-membered diyne 26^{15} (55% yield), which is stable at room temperature for a few days.² To clarify the efficiency of this cyclization, pure 24 was prepared through reduction-separationoxidation sequence. The cyclization of pure 24 afforded 26 as a single isomer in 84% yield. Thus, 24 showed very high efficiency in the cyclization. On the other hand, 5 which gave low efficiency in the cyclization exhibited NOE (3.1%) between an acetonide methyl and C12 vinyl proton. These observations suggest that there would exist unfavorable steric interactions between the acetonide group and the cyclopentene mojety in the transition state for the cyclization of 5.

In summary, we have confirmed that the LiN(TMS)₂/CeCl₃-mediated intramolecular acetylide addition to aldehyde is a general method for constructing a highly strained nine-membered diyne system^{2,16} and have established a stereoselective and concise route to the key intermediate (26) for the total synthesis of the C-1027 chromophore (1).



Scheme 4. Reagents and conditions: (a) p-MeOPhCH(OMe)₂, cat. TsOH, 100°C. (b) LiAlH₄, THF, 0°C, 82% (2 steps). (c) NaH, THF, 0°C, then TBSCl, 100%. (d) LiAlH₄, AlCl₃, Et₂O, -50°C, 80%. (e) NaIO₄, 60% aq. THF. (f) Ethynyltrimethylsilane, BuLi, THF, -78°C, 81% (2 steps). (g) Dess-Martin periodinane, CH₂Cl₂. (h) HCCCH₂MgBr, ZaBr₂, CH₂Cl₂/Et₂O (2:1), 82% (2 steps) (19:20=13:1). (i) BuLi, THF, -70°C, then TESCl, 74%. (j) Bu4NBr, THF, 1M aq. NaOH, 73%. (k) 22, cat. PdCl₂(Ph₃)₂, cat. CuI, Et₂NH, 50°C, 47% from 22. (l) AgNO₃, THF, i-PrOH, H₂O, then 2,6-lutidine, 59%. (m) DIBAL, CH₂Cl₂, -50°C, 74%. (n) LiN(TMS)₂, CeCl₃, THF, -30°C to r.t., 55% for a 5:1 diastereometric mixture of 24. (o) DIBAL, CH₂Cl₂, -50°C and separated from an isomer, 59%. (p) Dess-Martin periodinane, 83%. (g) LiN(TMS)₂, CeCl₃, THF, -30°C to r.t., 84% for pure 24.

Reference and Notes

- Minami, Y.; Yoshida, K.; Azuma, R.; Saeki, M.; Otani, T. Tetrahedron Lett. 1993, 34, 2633-2636; Yoshida, K.; Minami, Y.; Azuma, R.; Saeki, M.; Otani, T. Tetrahedron Lett. 1993, 34, 2637-2640; Yoshida, K.; Minami, Y.; Otani, T.; Tada, Y.; Hirama, M. Tetrahedron Lett. 1994, 35, 5253-5256.
- 2. Iida, K.; Hirama, M. J. Am. Chem. Soc. 1994, 116, 10310-10311.
- Iida, K.; Hirama, M. J. Am. Chem. Soc. 1995, 117, 8875-8876. For reviews of related syntheses, see: Nicolaou, K. C.; Dai, W.-M. Angew. Chem. Int. Ed. Engl. 1991, 30, 1387-1416; Hirama, M. Synthesis and Chemistry of Neocarzinostatin Analogs. In Recent Progress in the Chemical Synthesis of Antibiotics and Related Microbial Products, Vol 2; Lukacs, G. Ed.; Springer-Verlag: Berlin, 1993; pp. 293-329; Hirama M. J. Syn. Org. Chem. Jpn. 1994, 52, 980-991.
- (a) Iida, K.; Ishii, T.; Hirama, M.; Otani, T.; Minami, Y.; Yoshida, K. Tetrahedron Lett. 1993, 34, 4079-4082; (b) Iida, K.; Fukuda, S.; Tanaka, T.; Hirama, M.; Imajo, S.; Ishiguro, M.; Yoshida, K.; Otani, T. Tetrahedron Lett. in press.
- Leet, J. E.; Schroeder, D. R.; Langley, D. R.; Colson, K. L.; Huang, S.; Klohr, S. E.; Lee, M. S.; Golik, J.; Hofstead, S. J.; Doyle, T. W.; Matson, J. A. J. Am. Chem. Soc. 1993, 115, 8432-8443 and references therein.
- For related addition reactions, see: Nakatani, K.; Arai, K.; Hirayama, N.; Matsuda, F.; Terashima, S. Tetrahedron, 1992, 48, 633-650; Chattopadhyay, A.; Mamdapur, V. R. J. Org. Chem. 1995, 60, 585-587; Mukaiyama, T.; Suzuki, K.; Yamada, T.; Tabusa, F. Tetrahedron, 1990, 46, 265-276.
- For instance, addition reactions of lithium or magnesium trimethylsilylacetylide to 27 in the presence of CeCl₃ gave 13 (R=TES) as a major product in ratios of 1.5:1 and 2.5:1, respectively.⁶ The reaction did not occur without CeCl₃.
- 27 -TES
- 19: colorless oil; [α]_D³⁰ +12 (c 1.24, CHCl₃); IR (neat) v 3464, 3314, 2958, 2862, 2172, 1613, 1589, 1516, 1303, 1251 cm⁻¹; ¹H NMR (200MHz, CDCl₃) δ 0.10 (6H, s), 0.20 (9H, s), 0.95 (9H, s), 2.04 (1H, dd, J=2.5, 2.5Hz), 2.65 (1H, dd, J=16.5, 2.5Hz), 2.79 (1H, dd, J=16.5, 2.5Hz), 3.68 (1H, dd, J=4.1, 3.0Hz), 3.80 (3H, s), 4.03 (1H, dd, J=10.7, 3.0Hz), 4.25 (1H, dd, J=10.7, 4.1Hz), 4.25 (1H, s), 4.61 (1H, d, J=11.0Hz), 4.76 (1H, d, J=11.0Hz), 6.88 (2H, AA'BB'), 7.30 (2H, AA'BB'); Anal.
- Calcd. for C₂₅H₄₀O₄Si₂: C, 65.17; H, 8.75. Found: C, 64.98; H, 8.50.
 9. Evans, D. A.; Barrow, J. C.; Leighton, J. L.; Robichaud, A. J.; Sefkow, M. J. Am. Chem. Soc. 1994, 116, 12111-12112; Keck, G. E.; Andrus, M. B.; Romer, D. R. J. Org. Chem. 1991, 56, 417-420.
- 10. Mead, K. T. Tetrahedron Lett. 1987, 28, 1019-1022.
- Synthesized as a 5:1 diastereomeric mixture from (1R,4S)-1-acetoxy-4-hydroxycyclopent-2-ene [Hirama, M.; Gomibuchi, T.; Fujiwara, K.; Sugiura, Y.; Uesugi, M. J. Am. Chem. Soc. 1991, 113, 9851-9853; Deardorff, D. R.; Windham, C. Q.; Craney, C. L. Org. Synth. 1995, 73, 25-35] according to a recently developed method.¹²
- 12. Iguchi, S.; Toyama, K.; Hirama, M. manuscript in preparation.
- 13. Sonogashira, K.; Tohda, Y.; Hagihara, N. Tetrahedron Lett. 1975, 50, 4467-4470.
- 14. Schmidt, H. M.; Arens, J. F. Rec. Trav. Chim. 1967, 86, 1138-1142.
- 26: pale yellow oil; [α]_D²³-57 (c 0.48, CH₂Cl₂); IR (neat) v 3452(br), 2958, 2934, 2886, 2860, 2860, 1615, 1516, 1473 cm⁻¹; ¹H NMR (600MHz, CD₂Cl₂) δ 0.08 (3H, s, TBS), 0.09 (6H, s, TBS), 0.09 (3H, s, TBS), 0.15 (3H, s, TBS), 0.15 (3H, s, TBS), 0.70 (6H, q, J=8.0Hz, TES), 0.89 (9H, s, TBS), 0.89 (9H, s, TBS), 0.92 (9H, s, TBS), 0.94 (9H, t, J=8.0Hz, TES), 1.40 (1H, d, J=8.6Hz, C⁸-OH), 1.87 (1H, dd, J=13.8, 4.2Hz, H¹⁰), 2.29 (1H, dd, J=16.5, 1.2Hz, H⁵), 2.73 (1H, dd, J=13.8, 7.8Hz, H¹⁰), 2.89 (1H, d, J=16.5Hz, H⁵), 3.54 (1H, dd, J=7.9, 2.0Hz, H¹³), 3.77 (3H, s, MPM), 3.84 (1H, dd, J=10.9, 7.9Hz, H¹⁴), 3.97 (1H, br d, J=8.6Hz, H⁸), 4.21 (1H, dd, J=10.9, 2.0Hz, H¹⁴), 4.57 (1H, d, J=2.3Hz, MPM), 4.72 (1H, ddd, J=7.8, 4.2, 2.3Hz, H¹¹), 4.80 (1H, d, J=10.8Hz, MPM), 6.09 (1H, d, J=2.3Hz, H¹²) 6.84 (2H, AA'XX', MPM), 7.27 (2H, AA'XX', MPM); ¹³C NMR (150MHz, CD₂Cl₂) ppm -4.8, -4.7, -4.2, -4.2, 6.5, 7.5, 14.7, 18.7, 18.9, 19.0, 26.4, 26.4, 26.6, 33.0 (C⁵), 47.4 (C¹⁰), 56.0, 66.4 (C¹⁴), 69.6 (C⁸), 74.5 (C¹¹), 75.2, 80.4 (C⁴), 86.4 (C¹³), 89.8 (C³ or C⁶), 91.7 (C⁷), 92.1 (C²), 92.7 (C⁹), 98.8 (C³ or C⁶), 114.4, 129.1 (C¹), 130.5, 131.8, 142.8 (C¹²), 160.0.
- Myers, A. G.; Harrington, P. M.; Kuo, E. Y. J. Am. Chem. Soc. 1991, 113, 694-695. For tenmembered ring formation, see: Nishikawa, T.; Isobe, M.; Goto, T. Synlett, 1991, 393-395.

(Received in Japan 24 April 1996; revised 28 May 1996; accepted 29 May 1996)