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# Resin glycosides from Porana duclouxii

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### Resin glycosides from Porana duclouxii

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A new intact resin glycoside (3) and two glycosidic acids (1 and 2), all having a common trisaccharide moiety and (11*S*)-hydroxytetradecanoic acid or (3*S*,11*S*)-dihydroxytetradecanoic acid as the aglycone, were obtained from the roots of *Porana duclouxii*. Their structures were elucidated by spectroscopic analyses and chemical correlations. These compounds represent the first examples of resin glycosides from the genus *Porana*.

Keywords: Porana duclouxii; resin glycoside; triglycoside

#### 1. Introduction

The genus Porana (Fam. Convolvulaceae) consists of about 20 species occurring in subtropical and tropical areas worldwide [1]. Some plants of this genus, such as Porana racemosa and Porana mairei, are recorded in Chinese folk medicine for the treatment of cough, nameless swelling, overworked pain, and severe fever [2]. Previous phytochemical studies on this genus led to the isolation of ecdysteroids, coumarins, sterols, and flavanoids [3-6]. Resin glycosides are known to be characteristic of constituents in Convolvulaceae, which have been obtained from more than 34 species belonging to six genera (Ipomoea, Merremia, Convolvulus, Operculina, Cuscuta, and Calystegia) of this family [7]. However, they have not been reported from any *Porana* species so far.

*Porana duclouxii*, a Chinese endemic species, is distributed in Hubei, Yunnan and Sichuan provinces of Chinese mainland [1]. In our phytochemical study on *P. duclouxii*, a crude resin glycoside fraction was obtained. This paper describes the isolation and characterization of an intact resin glycoside and two glycosidic acids from the fraction, which were trivially named poranaside A (3), poranic acid A (1), and poranic acid B (2), respectively.

#### 2. Results and discussion

Poranic acid A (1) was obtained as a white powder, with m.p. 175-178°C. Its molecular formula, C32H58O16, was determined from NMR (<sup>1</sup>H and <sup>13</sup>C) and ESI-MS data and confirmed by HR-ESI-MS. The negative ion ESI-MS of 1 exhibited an  $[M-H]^-$  ion peak at m/z 697 as well as a series of fragment ions at m/z 551 [697–146 (C<sub>6</sub>H<sub>10</sub>O<sub>4</sub>)]<sup>-</sup>, 389 [551-162 (C<sub>6</sub>H<sub>10</sub>O<sub>5</sub>)]<sup>-</sup>, and 243 [389-146  $(C_6H_{10}O_4)$ ]<sup>-</sup>. The <sup>1</sup>H NMR signals of 1 displayed recognizable signals for three anomeric protons [ $\delta_{\rm H}$  6.33 (br s); 5.87 (d, J = 7.0 Hz; 4.87 (d, J = 7.5 Hz)]. The <sup>13</sup>C NMR spectrum gave 32 signals, of which 18 carbons were assignable to a trisaccharide moiety and 14 carbons to the aglycone

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moiety. All the above evidence indicated that 1 was a triglycoside with a hydroxytetradecanoic acid as the aglycone. The <sup>1</sup>H and <sup>13</sup>C NMR signals of 1 were carefully assigned with the aid of H–H COSY, HSQC, and HMBC experiments (Table 1). Comparing the <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts of 1 and the related triglycosides in the literatures [8,9], this compound was very similar to cuscutic acid A2 [8], except the inner Dfucose in cuscutic acid A2 was substituted by a D-quinovose in **1**. By complete acidic hydrolysis, **1** furnished 11-hydroxytetradecanoic acid, D-quinovose, D-glucose, and L-rhamnose. The absolute configuration of 11-hydroxytetradecanoic acid was

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR spectral data of compounds **1**, **2**, and **3**.

	1		2		3	
Position	$\delta_{ m H}$	$\delta_{\rm C}$	δн	$\delta_{\rm C}$	$\delta_{ m H}$	$\delta_{\rm C}$
Qui-1	4.87 (d, 7.5)	102.1	4.87 (d, 7.5)	102.1	4.81 (d, 7.4)	102.3
2	4.34 (dd, 7.5, 8.8)	79.6	4.34 (dd, 7.5, 8.5)	79.5	$4.40^{\alpha}$	78.8
3	4.42 (dd, 9.0, 8.8)	79.0	4.41 (dd, 9.0, 8.5)	79.0	4.32 (dd, 9.0, 8.5)	79.3
4	3.59 (dd, 9.0, 9.0)	76.7	3.59 (dd, 9.0, 9.0)	76.7	3.59 (dd, 9.0, 9.0)	76.6
5	3.70 (m)	72.2	3.70 (m)	72.2	3.72 (m)	72.3
6	1.55 (d, 6.0)	18.4	1.55 (d, 6.1)	18.4	1.61 (d, 6.0)	18.4
Glc-1	5.87 (d, 7.0)	101.7	5.87 (d, 7.2)	101.7	5.97 (d, 7.2)	101.2
2	4.27 (dd, 7.2, 8.9)	78.6	4.27 (dd, 7.2, 8.9)	78.6	$4.24^{\alpha}$	77.8
3	$4.25^{\alpha}$	78.9	$4.25^{\alpha}$	78.9	4.23 <sup>α</sup>	79.2
4	4.10 (t, 8.9)	72.2	4.10 (t, 8.9)	72.2	4.07 (t, 9.0)	72.2
5	3.86 (m)	77.5	3.86 (m)	77.5	$3.85^{\alpha}$	77.7
6	4.46 (dd, 11.4, 2.6) 4.26 (11.4, 6.7)	62.9	4.43 (dd, 9.2, 2.3) 4.26 $^{\alpha}$	62.9	4.43 (dd, 9.2, 2.3 $4.27^{\alpha}$	62.8
Rha-1	6.33 (br s)	101.8	6.34 (br s)	101.9	6.41 (br s)	101.4
2	4.76 (br s)	72.1	4.74 (br s)	72.1	4.70 (br s)	72.2
3	4.69 (dd, 9.2, 3.2)	72.4	4.66 (dd, 9.2, 3.3)	72.4	4.71 (dd, 9.2, 3.3)	70.0
4	4.32 (dd, 8.9, 3.3)	74.1	4.32 (dd, 8.9, 3.3)	74.1	5.94 (t, 9.4)	75.6
5	5.00 (m)	69.4	5.00 (m)	69.4	5.10 (m)	66.9
6	1.80 (d, 6.1)	18.6	1.80 (d, 6.2)	18.6	1.63 (d, 6.3)	18.3
Ag-1		177.1		175.3		172.7
2	2.48 (t, 7.4)	35.2	2.87 (br d, 6.0)	43.8	2.70 (dd, 6.4, 3.1)	43.2
3	1.73 (m)	25.6	4.53 (m)	68.4	4.53 (m)	68.0
4	1.34 <sup>α</sup>	29.6	1.70 (m)	37.9	1.64 - 1.70 (m)	37.9
5-9	1.69 (m)	25.1	1.68 (m)	25.2	1.69 (m)	25.2
	$1.20 - 1.40^{\alpha}$	29.5	1.54 (m)	26.1	1.63 (m)	26.0
	$1.20 - 1.40^{\alpha}$	29.8	$1.26 - 1.47^{\alpha}$	30.0	$1.26 - 1.44^{\alpha}$	29.9
	$1.20 - 1.40^{\alpha}$	29.9	$1.26 - 1.47^{\alpha}$	30.0	$1.26 - 1.44^{\alpha}$	30.0
	$1.20 - 1.40^{\alpha}$	30.1	$1.26 - 1.47^{\alpha}$	30.3	$1.26 - 1.44^{\alpha}$	30.3
10	1.65 - 1.80 (m)	34.6	1.63 - 1.77 (m)	34.6	1.65 - 1.74 (m)	34.6
11	3.88 (m)	80.2	3.88 (m)	80.1	3.88 (m)	80.6
12	1.77 (m)	37.3	1.75 (m)	37.3	1.78 (m)	37.3
	1.60 (m)	0110	1.58 (m)	0110	1.60 (m)	0110
13	1.57 (m)	18.6	1.65 (m)	18.8	1.53 (m)	18.5
14	0.90(t, 7.1)	14.2	0.90(t, 7.1)	14.2	0.91 (t. 6.1)	14.2
0Me	0.90 (0, 7.1)	11.2	0.90 (0, 7.1)	11.2	3.61 (s)	51.1
ang_1					5.01 (5)	167.7
2						128.4
-3					5.88 (m)	120.4
1					2.00 (m)	16.0
т 2_Ме					2.07 (uu, 7.2, 1.4) 2 00 (d. 1.4)	20.0
2-1010					2.00 (u, 1.4)	20.9

<sup> $\alpha$ </sup> Signals are overlapping, **S** in ppm, *J* in Hz.

determined as 11S by comparison of specific rotation of methyl ester with those obtained from Mexican jalap roots [10]. The absolute configurations of sugars were determined by direct comparison of specific optical rotation with authentic samples and confirmed by GC-MS with their TMSi-derivatives, respectively. The connectivities between the hydroxyl fatty acid aglycone and oligosaccharide, and among monosaccharides in 1 were supported by its HMBC spectrum. The long-range correlations were observed as follows: H-1 ( $\delta_{\rm H}$  4.87) of Qui with C-11 ( $\delta_{\rm C}$  80.2) of the aglycone (11Sconvolvulinolic acid), H-1 ( $\delta_{\rm H}$  5.87) of Glc with C-2 ( $\delta_C$  79.6) of Qui, H-1 ( $\delta_H$  6.33) of Rha with C-2 ( $\delta_{\rm C}$  78.6) of Glc. These data confirmed that the linkage of 1 was Rha- $(1 \rightarrow 2)$ -Glc- $(1 \rightarrow 2)$ -Qui- $(1 \rightarrow 11)$ -aglycone. Furthermore, the component sugar moieties were determined to be  $\beta$ -quinopyranosyl,  $\beta$ -glucopyranosyl, and  $\alpha$ -rhamnopyranosyl, respectively, by vicinal <sup>1</sup>H-<sup>1</sup>H coupling constants (Table 1). Accordingly, 1 was characterized as (11S)-hydroxytetradecanoic acid 11-O-a-L-rhamnopyranosyl- $(1 \rightarrow 2)$ -*O*-β-D-glucopyranosyl- $(1 \rightarrow 2)$ -β-D-quinovopyranoside (Figure 1).

The molecular formula of poranic acid B (2), with m.p.  $180-183^{\circ}$ C, was determined as  $C_{32}H_{58}O_{17}$ , by the negative ESI-MS, showing the pseudo-molecular ion peak at m/z 713 [M–H]<sup>-</sup>, along with fragment peaks at m/z 567 and 405, all of which were 16 mass units (OH) more than those corresponding ions of 1. The <sup>1</sup>H and

<sup>13</sup>C NMR spectra of **2** were superimposable on those of **1** except for signals due to the aglycone moiety. Moreover, the HMBC and <sup>13</sup>C NMR spectra demonstrated that the oligosaccharide moieties of **1** and **2** were identical. On mild acid hydrolysis, compound **2** afforded D-quinovose, D-glucose, L-rhamnose, and (3*S*, 11*S*)-dihydroxytetradecanoic acid (ipurolic acid) [11,12]. Thus, **2** was characterized as (3*S*,11*S*)-dihydroxytetradecanoic acid 11-*O*-α-L-rhamnopyranosyl-(1 → 2)-*O*-β-D-glucopyranosyl-(1 → 2)-β-D-quinovopyranoside (Figure 1).

Poranaside A (3) had a molecular formula of C<sub>38</sub>H<sub>66</sub>O<sub>18</sub>, as determined from a quasi-molecular ion at m/z 811.4316  $[M + H]^+$  in the positive ion HR-ESI-MS. Its <sup>1</sup>H and <sup>13</sup>C NMR spectra (Table 1) displayed characteristic signals for a glycosidic acid core closely similar to poranic acid B (2), as well as additional resonances indicating the presence of a methoxyl group  $[\delta_{\rm H} 3.61 (3 {\rm H}, {\rm s})/\delta_{\rm C} 51.1]$  and a short organic acid moiety (angeloyl moiety). The signals readily distinguished for an olefinic methine at  $\delta_{\rm H}$  5.88 (m), an olefinic tertiary methyl group at  $\delta_{\rm H}$  2.00 (d,  $J = 1.4 \,\rm{Hz}$ ), and an olefinic secondary methyl group at  $\delta_H$  2.07 (dd, J = 7.2, 1.4 Hz) in the <sup>1</sup>H NMR spectrum, and five correspounding carbon resonances at  $\delta_{\rm C}$  167.7, 137.3, 128.4, 20.9, and 16.0 in the <sup>13</sup>C NMR spectrum (Table 1) which were due to the angeloyl moiety [13]. Compound 3 was then subjected to saponification with 5% KOH and the products fractionated into organic and glycosidic acid



Figure 1. Structures of 1, 2, and 3.

fractions. The former examined by GC-MS gave a peak corresponding to angelic acid, with EI-MS ions at m/z 100 [M]<sup>+</sup>(100), 85 (29), 72 (0.8), 55 (86), 53 (16), and 39 (25). The glycosidic acid fraction furnished poranic acid B (2), which was confirmed by <sup>1</sup>H and <sup>13</sup>C NMR and HPLC. The long-range correlations were observed from  $OCH_3$  ( $\delta_{\rm H}$ 3.61) to the aglycone carbon ( $\delta_{\rm C}$  172.7), and from H-4 ( $\delta_{\rm H}$  5.94) of Rha to the angeloyl carbon ( $\delta_{\rm C}$  167.7) in the HMBC spectrum. The HMBC information indicated that the aglycone was a methyl ester, while the angeloyl moiety was located at C-4 of Rha. Comparing the NMR spectral data of 3 with its glycosidic acid core (2), the signal at 4-H of Rha was shifted toward downfield by 1.62 ppm, and the C-1 of aglycone was shifted toward upfield by 2.6 ppm (Table 1), which further confirmed the linkage of 3. Therefore, the structure of poranaside A(3)was characterized as (3S,11S)-dihydroxytetradecanoic acid 11-O-(4-O-angeloyl)- $\alpha$ -Lrhamnopyranosyl- $(1 \rightarrow 2)$ -O- $\beta$ -D-glucopyranosyl- $(1 \rightarrow 2)$ - $\beta$ -D-quinovopyranoside, methyl ester (Figure 1).

#### 3. Experimental

#### 3.1 General experimental procedures

Optical rotations were obtained on WZZ-2B automatic polarimeter (Precision Instrument Co., Shanghai, China); melting point was measured using a X-4 micromelting point apparatus (Cany Precision Instruments Co., Shanghai, China). The <sup>1</sup>H, <sup>13</sup>C, and 2D NMR spectra were recorded on a Bruker DRX-400 instrument (Bruker BioSpin GmbH Company, Rheinstetten, Germany) using TMS as an internal standard. EI-MS and HR-ESI-MS data were obtained on an API QSTAR mass spectrometer (Applied Biosystem/MSD Sciex, Concord, ON, Canada). Preparative HPLC was performed with a Waters 1525 Binary HPLC pump and a Waters 2414 refractive index detector using a YMC-Pack ODS-A column (5  $\mu$ m, 250 mm × 20 mm). For column chromatography (CC), silica gel 60 (200–300 mesh, Qingdao Marine Chemical Ltd, Qingdao, China), Develosil ODS (50 μm, Nomura Chemical Co. Ltd., Osaka, Japan), and Sephadex LH-20 (GE Healthcare, Uppsala, Sweden) were used. Analytical GC was carried out on a Shimadzu GCMS-QP2000 Plus apparatus (Shimadzu Corporation, Tokyo, Japan) with an ionization energy of 70 eV.

#### 3.2 Plant material

The roots of *P. duclouxii* were collected in Shennongjia, Hubei, China, in February 2012, and identified by Prof. Zhang Daigui (Key Laboratory of Plant Resources Conservation and Utilization, Jishou University). A voucher specimen (zdg3431) has been deposited at Jishou University.

#### 3.3 Extraction and isolation

The powdered air-dried roots of P. duclouxii (150 g) were extracted with methanol thrice in an ultrasonic bath, each for 48 h at room temperature, and the extract was concentrated under reduced pressure to give 80 g of residue. The residue was suspended in water and then extracted with EtOAc. A lot of emulsion layer occurred between water and EtOAc. The emulsion layer was concentrated under reduced pressure and subjected to Sephadex LH-20 column chromatography to give a crude resin glycoside fraction (R, 5.0 g). A part of R (2.0 g) was further subjected to ODS column chromatography, using MeOH/H<sub>2</sub>O mixtures of decreasing polarities (80:20 to 95:5) to give four subfractions  $(R_1-R_4)$ .  $R_1$  (150 mg) was separated by preparative HPLC using MeOH/H<sub>2</sub>O (8:2, v/v) with a flow rate of 5 ml/min. Each peak was collected by manual recycling, to afford poranaside A (3) (30 mg,  $t_{\rm R}$  45 min). However, other efforts to isolate individual components from these sub-fractions were unsuccessful.

The resin glycoside fraction (2.0 g) was treated with 5% KOH at 85°C for 4 h. The reaction mixture was acidified to pH 4.0

and extracted with Et<sub>2</sub>O (50 ml × 3). The aqueous layer was further extracted with *n*-BuOH (3 × 100 ml), and then the *n*-BuOH layer was concentrated *in vacuo* to yield a residue (540 mg), which was subjected to Sephadex LH-20 CC using MeOH to afford a glycosidic acid fraction (300 mg). A portion (100 mg) of this fraction was subjected to preparative HPLC using MeOH/H<sub>2</sub>O (7:3) with a flow rate of 5 ml/ min to obtain poranic acids A (1) (28 mg,  $t_R$ 35 min), B (2) (36 mg,  $t_R$  60 min).

#### 3.3.1 Poranic acid A(1)

White powder, m.p.  $175-178^{\circ}C$ ,  $[\alpha]_{D}^{20}$ -24.0 (*c* 0.3, MeOH); For <sup>1</sup>H NMR (400 MHz, C<sub>5</sub>D<sub>5</sub>N) and <sup>13</sup>C NMR (100 MHz, C<sub>5</sub>D<sub>5</sub>N) spectroscopic data, see Table 1; positive ion ESI-MS *m/z*: 721 [M + Na]<sup>+</sup>; negative ESI-MS *m/z*: 697 [M-H]<sup>-</sup>, 551, 389, 243; HR-ESI-MS *m/z*: 697.3646 [M-H]<sup>-</sup> (calcd for C<sub>32</sub>H<sub>57</sub>O<sub>16</sub>, 697.3652).

#### 3.3.2 Poranic acid B(2)

White powder, m.p.  $180-183^{\circ}$ C,  $[\alpha]_{D}^{20}$  – 20.0 (*c* 0.3, MeOH); For <sup>1</sup>H NMR (400 MHz, C<sub>5</sub>D<sub>5</sub>N) and <sup>13</sup>C NMR (100 MHz, C<sub>5</sub>D<sub>5</sub>N) spectroscopic data, see Table 1; positive ion ESI-MS *m/z*: 737 [M + Na]<sup>+</sup>; negative ESI-MS *m/z*: 713 [M-H]<sup>-</sup>, 567, 405; HR-ESI-MS *m/z*: 713.3592 [M-H]<sup>-</sup> (calcd for C<sub>32</sub>H<sub>57</sub>O<sub>17</sub>, 713.3601).

#### 3.3.3 Poranaside A (3)

Colorless syrup,  $[\alpha]_D^{20} - 14.0$  (*c* 0.1, MeOH); For <sup>1</sup>H NMR (400 MHz, C<sub>5</sub>D<sub>5</sub>N) and <sup>13</sup>C NMR (100 MHz, C<sub>5</sub>D<sub>5</sub>N) spectroscopic data, see Table 1; positive ion ESI-MS *m*/*z*: 811 [M + H]<sup>+</sup>, 833 [M + Na]<sup>+</sup>; negative ESI-MS *m*/*z*: 845 [M + Cl]<sup>-</sup>, 809 [M-H]<sup>-</sup>; HR-ESI-MS *m*/*z*: 811.4316 [M + H]<sup>+</sup> (calcd for C<sub>38</sub>H<sub>67</sub>O<sub>18</sub>, 811.4322).

#### 3.4 Acid hydrolysis of 1 and 2

Compounds 1 and 2 (13 and 15 mg, respectively) were separately dissolved in 5% H<sub>2</sub>SO<sub>4</sub> (2 ml) and heated at 95°C for 1 h. The reaction mixture was diluted with H<sub>2</sub>O (2 ml) and then extracted with Et<sub>2</sub>O. The Et<sub>2</sub>O layer was dried (anhydr. Na<sub>2</sub>SO<sub>4</sub>) and concentrated to afford a hydroxy fatty acid fraction. The hydroxy fatty acid fraction was treated with CH<sub>2</sub>N<sub>2</sub> to yield corresponding methyl ester **1a** (4.0 mg, 11*S*-hydroxytetradecanoate) and **2a** (4.5 mg, 3*S*,11*S*-dihydroxytetradecano-ate), respectively.

Compound **1a**: m.p.  $26-28^{\circ}$ C;  $[\alpha]_{D}^{20} + 1.3$ (*c* 0.3, CHCl<sub>3</sub>); <sup>13</sup>C NMR (in CDCl<sub>3</sub>, 100 MHz)  $\delta$ : 14.1 (C-14), 18.8, 25.4, 24.7, 25.6, 29.1, 29.2, 29.4, 29.5, 29.6, 34.1, 37.5, 39.7, 51.6 (*O*CH<sub>3</sub>), 71.7 (C-11), and 174.3 (C-1). Its physical and spectral properties were identical to that of (11*S*)hydroxytetradecanoate (convolvulinic acid methyl ester) [10].

Compound **2a**: m.p.  $67-70^{\circ}$ C;  $[\alpha]_{D}^{20} + 11.5$  (*c* 0.5, CHCl<sub>3</sub>); <sup>13</sup>C NMR (in CDCl<sub>3</sub>, 100 MHz)  $\delta$ : 14.1 (C-14), 18.8, 25.4, 25.6, 29.4, 29.5, 29.6, 36.5, 37.5, 39.7, 41.1, 51.8 (*O*CH<sub>3</sub>), 68.0 (C-3), 71.7 (C-11), and 173.4 (C-1). Its physical and spectral properties were identical to that of (3*S*,11*S*)-dihydroxytetradecanoate (ipurolic acid methyl ester) [11].

The aqueous layer was neutralized with 5% NaOH and desalted by passage through a Sephadex LH-20 column using MeOH to afford a mixture of sugars (6 mg), from which D-glucose ( $R_{\rm f}$  0.3), D-quinovose ( $R_{\rm f}$  0.51), and L-rhamnose ( $R_{\rm f}$  0.52) were detected by TLC (CHCl<sub>3</sub>:MeOH:H<sub>2</sub>O, 6:4:1) [11]. An aliquot of the mixture was separated by HPLC, using the Thermo Hypersil GOLD Amino column 250 mm × 4.6 mm column, condition: 78% CH<sub>3</sub>CN: H<sub>2</sub>O, flow rate 0.5 ml/min] to give L-rhamnose [ $t_{\rm R}$  14.0 min, [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 6.8 (c 0.5, water)], D-quinovose [ $t_{\rm R}$  14.2 min, [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 40.0 (c 0.3, water)], and D-glucose

[ $t_R$  20.1 min,  $[\alpha]_D^{20} + 90.7$  (*c* 0.4, water)]. Another aliquot of the mixture (1.0 mg) was derivatized with Sigma Sil-A for 35 min at 70°C and analyzed by GC–MS [HP-5MS column (30 m × 0.25 mm, 0.25 µm); injection temperature: 250.0°C; column flow: 1.33 ml/min; ion source temperature: 200.0°C; interface temperature: 220.0°C], showing  $t_R$  of 9.11, 10.56, and 12.17 min, identical to those of authentic L-rha, D-qui, and D-glc (J&K Chemical<sup>®</sup>, J&K Scientific Ltd, Beijing, China) derivatives prepared in a similar way, respectively.

#### 3.5 Alkaline hydrolysis of 3

Compound 3 (20 mg) was treated with 5%KOH (3 ml) at 85°C for 4 h. The reaction mixture was acidified to pH 4.0 and extracted with Et<sub>2</sub>O ( $3 \times 10$  ml). The Et<sub>2</sub>O layer was dried (anhydr. MgSO<sub>4</sub>) and concentrated to a small volume (about 0.2 ml) to afford a short chain organic acid fraction. The aqueous layer was further extracted with *n*-BuOH ( $3 \times 20$  ml). The *n*-BuOH solution was concentrated in vacuo to yield a residue which was subjected to Sephadex LH-20 CC using MeOH to afford a glycosidic acid 2 (15 mg). The short chain organic acid fraction obtained from 3 (1.0 mg) was examined by GC-MS using a Shimadzu GCMS-QP2000 Plus apparatus, equipped with Rxi®-5 ms fused silica  $(30 \text{ m} \times 0.25 \text{ mm})$ capillary column 0.25 µm). The carrier gas was helium. Column temperature was initially 50°C and increased to 160°C at 20°C/min and to 220°C at 5°C/min, to give one predominant peak which was identified to be angelic acid ( $t_{\rm R}$  3.7 min): EI-MS: m/z 100 [M]<sup>+</sup>(100), 85 (29), 72 (0.8), 55 (86), 53 (16), and 39 (25).

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