## SHORT REPORTS

# STEREOSPECIFIC SYNTHESIS AND ABSOLUTE CONFIGURATION OF (+)-RHODODENDROL\*

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**Key Word Index**—*Rhododendron maximum; Acer nikoense;* (+)-rhododendrol; (-)-rhododendrol; absolute configuration; enzymatic reduction; baker's yeast.

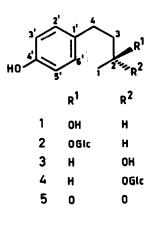
Abstract—The stereospecific synthesis of (+)-rhododendrol, a constituent of *Rhododendron maximum* and *Acer* nikoense, by enzymatic reduction of 4-(4'-hydroxyphenyl)-2-butanone has shown that the absolute configuration of the molecule is S.

## INTRODUCTION

(+)-Rhododendrol [(+)-4-(4'-hydroxyphenyl)-2-butanol] (1), mp 81–83° (CHCl<sub>3</sub>),  $[\alpha]_{\rm D}$  + 17.1° (EtOH; c 2.0), was isolated [1] for the first time from Rhododendron maximum. Later, it was isolated [2] along with its glucoside (2) from Acer nikoense and its absolute configuration was proposed to be S on the basis of the empirical rule on the relationship between optical rotation and chirality. There is no other report on the study on (+)-rhododendrol to determine its absolute configuration. It is interesting that its enantiomer, (-)-rhododendrol (3), and its glucoside (4) have been isolated [3] from R. chrysanthum and that these compounds have been reported [4] from different species of different genera. The absolute configuration of (-)-rhododendrol was proposed [5] as S by a biosynthetic method and more recently X-ray crystallographic analysis suggested [4] its absolute configuration to be R. The synthesis of 3 and  $(\pm)$ -rhododendrol has been achieved [6] but 1 has not yet been synthesized. Here we report the stereospecific synthesis of 1 by an enzymatic process which provides an easy method to determine its absolute configuration as S.

## **RESULTS AND DISCUSSION**

The stereospecific synthesis of 1 was achieved from 4-(4'-hydroxyphenyl)-2-butanone (5). The latter was synthesized by a reported method [6]. Chemical reduction of 5 with sodium and ethanol [6] or sodium borohydride [7] afforded the corresponding racemic alcohol. Earlier attempts [7] to resolve the racemic alcohol into its optically active forms were unsuccessful. However, it is now well documented [8–12] that the enzymatic reduction of an unsymmetrical carbonyl compound proceeds



stereospecifically to an optically active alcohol and that a 4-aryl-2-butanone can be reduced with baker's yeast to afford in good yield and high optical purity the corresponding S-alcohol. We, therefore, treated 5 with baker's yeast under the conditions reported earlier for the reduction of a 4-aryl-2-butanone [12]. The product was characterized as 4-(4'hydroxyphenyl)-2-butanol from its spectral data (Experimental) and it showed an optical rotation of  $+13.6^{\circ}$  (EtOH; c 0.4135). Hence it may be concluded that 1 possesses S configuration. This conclusion is supported by the result obtained from the X-ray crystallographic analysis [4] of 3 which confirmed the absolute configuration of the molecule as R. The enzymatic process discussed here offers a useful means for the determination of the absolute configuration of arylbutanols analogues of 1.

### EXPERIMENTAL

Mps: uncorr; IR: KBr; <sup>1</sup>H NMR: 200 MHz, CDCl<sub>3</sub> with TMS as int. standard; CC: silica gel (B.D.H.,

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100–200 mesh); TLC: silica gel G. The TLC spots were visualized by exposure of the plates to  $I_2$  vapour. Baker's yeast (*Saccharomyces cerevisiae*, Type I, Sigma) was used for enzymatic reduction.

Synthesis of (+)-rhododendrol. Baker's yeast (3 g) was added to a vigorously stirred soln of sucrose (1.5 g) in tap  $H_2O$  (200 ml). The suspension was stirred for 0.5 hr at 30°. Compound 5 [6] (70 mg) was added and the stirring was continued. Three portions of fermenting baker's yeast [1.5 g in a soln of sucrose (750 mg) in tap water (50 ml)] were added during 72 hr and the suspension was stirred for another 48 hr at room temp. The mixt. was worked up by first adding celite and filtering through a sintered glass funnel. The filtrate was satd with NaCl and extracted with EtOAc ( $4 \times 100$  ml). The extract was dried, concd and purified by CC over silica gel to produce 1, 48 mg, mp 80-81° (CHCl<sub>3</sub>),  $[\alpha]_{D}^{25}$  + 13.6° (EtOH; c 0.4135), IR  $v_{max}^{nujol}$ cm<sup>-1</sup>: 3330, 1595, 1500; <sup>1</sup>H NMR:  $\delta$  7.04 (2H, d, J = 8.0 Hz, H-3' and H-5'), 6.72 (2H, d, J = 8.0 Hz, H-2' and H-6'),  $3.78(1H, m, H-2), 2.64(1H, t, J = 7.0 Hz, H_2-4), 1.79-1.58$ (2H, m, H<sub>2</sub>-3), 1.20 (3H, d, J = 8.0 Hz, Me); MS m/z (rel. int.): 166 [M] + (32), 148 (18), 133 (95), 107 (100).

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