

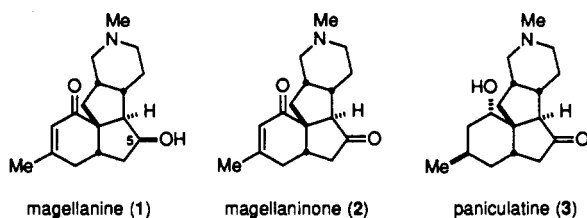
First Total Synthesis of *Lycopodium* Alkaloids of the Magellanane Group. Enantioselective Total Syntheses of (-)-Magellanine and (+)-Magellaninone

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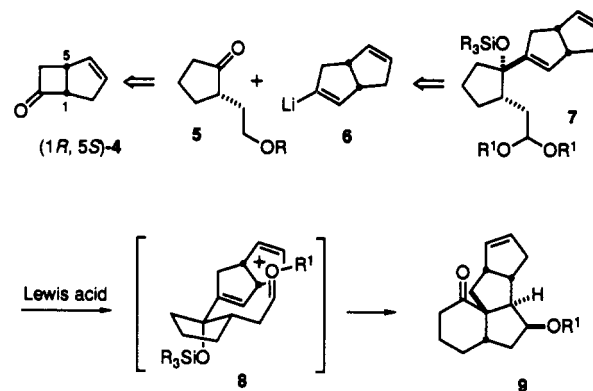
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A strikingly diverse array of polycyclic alkaloids are found in club moss (genus *Lycopodium*), and these structures have long served to stimulate innovations in organic synthesis.^{1,2} Some years ago investigations of *Lycopodium paniculatum* and *Lycopodium magellanicum* revealed the presence of three new alkaloids having a unique tetracyclic skeleton: magellanine (1), magellaninone (2), and lycopanulatin (paniculatin) (3).^{3,4} The structures of 2 and 3 were secured by X-ray crystallography,³ while absolute configurations were assigned by optical methods.⁴ In this communication we report the first total syntheses of *Lycopodium* alkaloids having the magellanane skeleton.⁵⁻⁷ These total syntheses highlight the power of pinacol-terminated cationic cyclizations for assembling angularly-fused polycyclics.^{6c,8}



The angularly-fused carbotetracycle **9** was envisaged as the immediate precursor of the magellanane skeleton (Scheme I). The heart of our plan was the formation of this late intermediate by Prins-pinacol rearrangement of the dienyl acetal **7**. Central to the design of this strategy was the expectation that the desired stereochemical outcome would result if Prins cyclization took place from the convex face of the cis-fused bicyclooctadiene fragment as illustrated in cyclization conformer **8**.^{6c,8} The readily available enantiopure (1*R*,5*S*)-bicyclo[3.2.0]heptenone **4**⁹ would

Scheme I



serve as the common precursor of **5** and **6**, the direct progenitors of the cyclization substrate **7**.

The *cis*-bicyclo[3.3.0]octadienyl iodide **12** was prepared from **4** as summarized in Scheme II. Treatment of (+)-**4** with [bis(methylthio)methyl]lithium, followed by reaction of the resulting alcohol with Cu(OTf)₂·C₆H₆, as described by Cohen,¹⁰ produced the ring-expanded α -sulfonyl ketone **10** in 70% yield and >10:1 regioselectivity.^{11,12} The methylthio substituent in **10** was subsequently exploited to introduce selectively the required second unsaturation in the bicyclo[3.3.0]octane fragment. Sequential treatment of **10** with Li-NH₃, Me₃SiCl, MeLi, and *N*-phenyltriflamide¹³ provided vinyl triflate **11** in 49% yield. This intermediate was then treated in turn with Pd(Ph₃P)₄ and hexamethylditin¹⁴ and then *N*-iodosuccinimide¹⁵ to afford iodide **12** in good yield. This overall sequence allowed vinyl iodide **12** to be prepared in enantiopure fashion in 27% overall yield from (+)-**4**.

Addition of the vinyl lithium reagent **6** derived from **12** to the (*S*)-cyclopentanone **5**¹⁵⁻¹⁷ was plagued by the propensity of cyclopentanone **5** to enolize. Enolization was minimized when this addition was carried out in Et₂O at -110 °C, conditions that afforded diol **13** and its stereoisomer (ds = 8:1) in 71% yield after desilylation. Conversion of this mixture to the bis(triethylsilyl) ethers followed by selective Swern oxidation of the primary silyl ethers^{17,18} provided aldehydes **14**, which were converted to the dimethyl acetals **15** (an 8:1 mixture of stereoisomers, 72% overall yield from **13**) by treatment with trimethyl orthoformate and pyridinium *p*-toluenesulfonate in CH₂Cl₂.¹⁹ The critical rearrangement of **15** was effected by exposure to this intermediate to 1.1 equiv of SnCl₄ in CH₂Cl₂ (-78 → -20 °C) to give the tetracyclic ethers **16** and **17** in a 2:1 ratio and 57% yield, together

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(1) MacLean, D. B. *Alkaloids* (Academic Press) **1985**, 26, 241 and earlier reviews in this series.

(2) Blumenkopf, T. A.; Heathcock, C. H. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W., Ed.; Wiley: New York, 1985; Vol. 3, pp 185-240.

(3) (a) Castillo, M.; Morales, G.; Loyola, L. A.; Singh, I.; Calvo, C.; Holland, H. L.; MacLean, D. B. *Can. J. Chem.* **1975**, 53, 2513. (b) Castillo, M.; Loyola, L. A.; Morales, G.; Singh, I.; Calvo, C.; Holland, H. L.; MacLean, D. B. *Can. J. Chem.* **1976**, 54, 2893. (c) Castillo, M.; Morales, G.; Loyola, L. A.; Singh, I.; Calvo, C.; Holland, H. L.; MacLean, D. B. *Can. J. Chem.* **1976**, 54, 2900.

(4) Loyola, L. A.; Morales, G.; Castillo, M. *Phytochemistry* **1979**, 18, 1721.

(5) Approaches to these alkaloids⁶ and the synthesis of the tetracyclic magellanane skeleton⁷ have been described.

(6) (a) St. Laurent, D. R.; Paquette, L. A. *J. Org. Chem.* **1986**, 51, 3861. (b) Mehta, G.; Rao, K. S. *J. Chem. Soc., Chem. Commun.* **1987**, 1578. (c) Hirst, G. C.; Howard, P. N.; Overman, L. E. *J. Am. Chem. Soc.* **1989**, 111, 1514. (d) Crimmins, M. T.; Watson, P. S. *Tetrahedron Lett.* **1993**, 34, 199.

(7) Mehta, G.; Reddy, M. S. *Tetrahedron Lett.* **1990**, 31, 2039.

(8) For brief reviews, see: (a) Overman, L. E. *Acc. Chem. Res.* **1992**, 25, 352. (b) Overman, L. E. In *Selectivities in Lewis Acid-Promoted Reactions*; NATO ASSI Series 289; Schinzer, D., Ed.; Kluwer Academic: Dordrecht, The Netherlands, 1989; pp 1-20.

(9) Collington, E. W.; Wallis, C. J.; Waterhouse, I. *Tetrahedron Lett.* **1983**, 24, 3125 and footnote 15 therein.

(10) Abraham, W. D.; Bhupathy, M.; Cohen, T. *Tetrahedron Lett.* **1987**, 28, 2203. Cohen, T.; Kuhn, D.; Falck, J. R. *J. Am. Chem. Soc.* **1975**, 97, 4749.

(11) All intermediates were fully characterized by ¹H and ¹³C NMR, IR, and MS analysis. The elemental composition of analytical samples of new compounds was confirmed by combustion analysis or high-resolution mass spectrometry. All yields refer to isolated, purified products.

(12) The abbreviations used for standard reagents can be found in the following: *J. Org. Chem.* **1992**, 57, 14A.

(13) McMurry, J. E.; Scott, W. J. *Tetrahedron Lett.* **1983**, 24, 979.

(14) Wulff, W. D.; Peterson, G. A.; Bauta, W. E.; Chan, K.-S.; Faron, K. L.; Gilbertson, S. R.; Kaesler, R. W.; Yang, D. C.; Murray, C. K. *J. Org. Chem.* **1986**, 51, 277.

(15) Seyferth, D. J. *Am. Chem. Soc.* **1957**, 79, 2133.

(16) Prepared from (1*R*,5*S*)-**4** by sequential treatment with (a) H₂, Pd-C, (b) *m*-CPBA, TFA, CH₂Cl₂, (c) LiAlH₄, (d) *tert*-butyldiphenylsilyl chloride and pyridine, and (e) the Swern reagent.¹⁷

(17) Mancuso, A. J.; Huang, S. L.; Swern, D. *J. Org. Chem.* **1978**, 43, 2480.

(18) Mahrowald, R.; Schick, H.; Vasil'eva, L. L.; Pivnitsky, K. K.; Weber, G.; Schwartz, S. J. *Prakt. Chem.* **1990**, 332, 169.

(19) Miyashita, M.; Yoshikoshi, A.; Grieco, P. A. *J. Org. Chem.* **1977**, 42, 3772.

