ENANTIOSELECTIVE SYNTHESIS OF THE SPIROKETAL UNIT OF MILBEMYCINS AND 22,23-DIHYDROAVERMECTINS

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<u>Summary</u>: The crystalline spiroketal sub-unit 2 has been synthesized in ten steps from the 2S,3R-diol 2 using a remote partial stereocontrol for the aldolisation step.

The unique biological activity of milbertycins and avermeetins as well as their challenging structural features (1) led us to develop a program towards their synthesis (2). Numerous routes to the spiroketal moiety of these molecules have been reported (3). Our approach of the spiroketal system of milbertycins and 22,23-dihydroavermeetins (b series) involved a stepwise condensation at both ends of 2,4-pentanedione 1 as shown on the Scheme (2a). This letter presents an enantioselective synthesis of the subunit 2 using a remote stereocontrol for the crucial aldolisation reaction. Previous work had shown the possibility of controlling the chiral center at C-21 (4) by use of a suitable synthon affording the correct stereochemistry at C-24 and C-25. Control of the C-17 center of 2 was next investigated.

Considering that the lithium cation of the kinetic enolate of $\underline{5}$ is likely to be coordinated with both oxygen atoms of the ketal function, it can be reasonably expected that the diastereofacial approach on the enolate might depend on the relative bulk of the R¹O- group and tetrahydropyranyl unit (O-C21 (to) C25) of $\underline{5}$ (5). A series of racemic ketals $\underline{5b}$ to $\underline{5e}$ with R¹ groups of increasing bulkiness was therefore tested in the aldolisation reaction with aldehydes <u>6a</u> to <u>6d</u> (see Table). The yields were always good (75-90%). The triphenylethylether <u>5e</u> led to an improved diastereoselectivity with aldehydes <u>6a</u>, <u>b</u> and <u>c</u> (entries 4,6,9) but unfortunately not with <u>6d</u> (entries 12 to 14). The lack of stereoselectivity in the latter case might involve the bulk or the coordinating power of the sulfonyl group of <u>6d</u>. Moreover a rapid aldol equilibration seemed to take place with this very reactive aldehyde (entries 12,14).

The asymmetric synthesis of 2 was therefore carried out as shown on the scheme using the phenylthicaldehyde <u>6c</u> as electrophile for the aldolisation reaction with the optically active triphenylethylketal <u>5e</u> (entry 9).



Scheme a) LDA-HMPT, 4.5 eq. -78°C then 3, -78°C to 20°C b) THF/H2O/acetic acid, 20°C c) 5b to 5c, see footnote table d) HCl, CHCl3 e) m-CPBA, CH2Cl2, 0°C f) LiAlH4, benzene, 70

The known (2S,3R)-diol 2 (11) was first converted into the trimethylsilyloxybromide 2, $[\alpha]_D^{20} = -39^\circ$ (c 1.5, CHCl₃) in three steps according to reference (2a) : tosylation, bromination and silylation, in 85% overall yield. Alkylation of the dianion of 2,4-pentanedione 1 with 3 afforded the crude diketone 4 $(Y = OSiMe_3, enol-ketone mixture)$ which was readily hydrolysed to give after chromatography the pure hemiketal $\underline{5a}$ which crystalized on drying, m.p. = 52-54°C, $[\alpha]_D^{20} = +122°$ (c 2.0, CHCl₃), 75% yield. Protection of 5a with 2,2,2-triphenylethanol (12) gave the crystalline ketal 5e in 85% yield m.p. = 142-145°C, $[\alpha]_{D}^{20}$ = +122° (c 2.0, CHCl₃). Condensation of this ketal with the phenylthioaldehyde $\underline{6c}$ gave a mixture of $\underline{7a}$ and $\underline{7b}$ ($\underline{7a}/\underline{7b}$ = $\underline{72}/28$ from HPLC analysis of an aliquot, see table). Direct acidic work-up of the reaction mixture followed by oxidation of the thioether function of crude <u>8a</u> ($R^2 = p$ -S-Tol) and chromatography gave, after crystallization, the pure spirotetrahydropyrone 8b in 55% yield from 5e, m.p. = 134-136°C, $[\alpha]_D^{20} = +48^\circ$ (c 2.2, CHCl₃). Final reduction of the C-19 carbonyl function of <u>8b</u> with LiAlH4 (2a) in benzene furnished, after chromatography, the expected spiroketal 2 in 68% yield, m.p. = 141-142°C, $[\alpha]_{D}^{20} = +47^{\circ}$ (c 2.0, CHCl₃) together with 17% of the C-19 epimeric spiroketal <u>10</u>, m.p. = 129-131°C, $[\alpha]_D^{20} = +45^\circ$ (c 2.0, CHCl₃).

The optically active spiroketal 9 has been thus prepared from the readily available (2S,3R)-diol 2 in ten steps in a c.a. 20% overall yield.

Entry	Aldehyde	Ketal	Diastereoselectivity ^[b]	
Entry 1 2 3 4 5 6 7 8 9 10 11 12 13	Aldenyde 5a 5a 5b 5b 5b 5c 5c 5c 5c 5c 5c 5c 5c 5c 5c 5c 5c 5c	Ketal 5b 5c 5d 5b 5b 5b 5b 5b 5c 5b 5c 5c 5c	Diastereose 7a 47 56 60 71 47 66 45 61 72 64 43 44 55	ECTIVITY (0) <u>7b</u> 53 44 40 29 53 34 55 39 28 36 57 56 45
14	<u>6d</u> [d]	<u>5e</u>	53	47

<u>Table</u> Condensation of aldehydes 6 with enolates of $5^{[a]}$

[a] Unless otherwise stated the aldehyde in THF at -78°C was added to the preformed lithium enolate of 5 (1.2 eq LDA, THF, -78°C, 30 min). After 1h at -78°C the reaction mixture was hydrolysed with saturated aqueous NH4Cl and extracted with ether.

[b] Deduced from averaged HPLC peak area (silica-gel : isooctane/isopropylether and isooctane/ethyl acetate mixtures). Assignement of the configuration at C-17 was made after acidic cyclisation of <u>7</u>, the natural isomer <u>7a</u> giving one single product <u>8</u> whereas <u>7b</u> gave at least two bicyclic products (13).
[c] The aldehyde was treated with one equivalent of ZnCl₂ before addition to lithium enolate.

[d] The reaction mixture was quenched after 30 seconds.

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