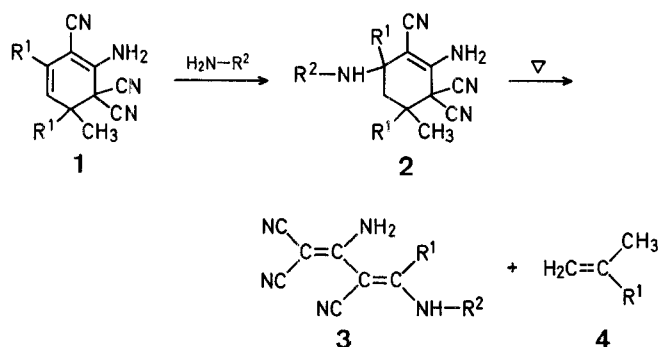


Syntheses with Unsaturated Nitriles; IV¹. Synthesis of 4-Alkyl-2-amino-4-(substituted amino)-1,1,3-tricyano-1,3-butadienes via Selective Thermal Decomposition of Amino Adducts

Maciej ADAMCZYK, Maria MOKROSZ

Department of Organic Chemistry, Jagellonian University, PL-30-060 Kraków, Poland

We have previously described a simple preparation of ylidenemalononitrile dimers² and selective transformation of their alkoxy adducts to monoketoesters³. As a continuation of our study on the applications of ylidenemalononitrile dimers in organic synthesis, we now report on the synthesis of a new type of 1,3-diene **3**.



The key step in this synthesis involves elaboration of the amino adducts **2** of ylidenemalononitrile dimers **1**. We have found that primary amines add smoothly to the dimers **1** in aqueous ethanol or dioxan solution, however, in some cases, addition of a more polar solvent i.e. dimethylformamide was required. The formation of amino adducts **2** takes place at room temperature and gives good yields.

Those adducts (**2a**, **b**) which possess two unsubstituted amino groups also react with phenyl isocyanate or phenyl isothiocyanate and we have found that only the amino group at the 5 position undergoes substitution forming *N,N'*-disubstituted derivatives of urea or thiourea (**2k**, **2l**), even with an excess of the isocyanate or isothiocyanate.

Adducts **2** may be considered as pseudo-adducts of Diels-Alder cycloaddition⁴ and we have found that they undergo a retro-Diels-Alder process⁵ under mild conditions. These reactions were carried out in a refluxing, non-polar solvent such as xylene or decalin, depending on the temperature required. All the dienes **3** are solids in contrast to the gaseous second component **4** of the decomposition reaction. This fact ensures that the reaction is clean and easy to perform and that the products can be isolated in satisfactory purity.

Analysis of the mass spectra of adducts **2** shows that the main fragmentation pattern is the retro-Diels-Alder process. In some cases we could not detect the molecular ions even when the temperatures of the inlet system and the ion source, and ionizing potential were decreased. In these spectra, the base ions correspond to the molecular ions of the main products of the thermal decomposition described above.

4-Amino-4,6-dialkyl-4-methyl-6-(substituted amino)-1,3,3-tricyanocyclohexenes **2**; General Procedure:

A mixture of the dimer **1** (0.1 mol), solvent (10 ml; ethanol for **2a**, **c**, **e**, **g**, **h**; dioxan for **2f**, **i**; 2:1 dioxan/dimethylformamide for **2b**, **d**; dimethylformamide for **2j**), and concentrated aqueous ammonia (10 ml)

¹² P. D. Bartlett, L. H. Knox, *Org. Synth. Coll. Vol. V*, 196 (1973).

Table 1. 4-Amino-4,6-dialkyl-6-(substituted amino)-1,3,3-tricyanocyclohexenes **2**

Product No.	R ¹	R ²	Yield [%]	m.p. [°C]	Molecular formula ^a or Lit. m.p. [°C]	M.S. <i>m/e</i> (M ⁺)	¹ H-N.M.R. (solvent) δ [ppm]
2a	CH ₃	H	90	> 300°	> 300° ^b	229	(DMF): 6.67 (s, 2H); 6.30 (s, 2H); 3.65 (s, 3H); 1.50 (s, 2H); 1.70 (s, 3H); 1.22 (s, 3H)
2b	C ₂ H ₅	H	91	> 300°	C ₁₄ H ₁₉ N ₅ (257.3)	—	(DMF): 6.60 (s, 2H); 6.20 (s, 2H); 3.6 (m, 2H); 1.9–0.8 (m, 13H)
2c	CH ₃	NH ₂	95	> 300°	C ₁₂ H ₁₆ N ₆ (244.3)	244	insoluble
2d	C ₂ H ₅	NH ₂	89	> 300°	C ₁₄ H ₂₀ N ₆ (272.4)	—	[(CD ₃) ₂ CO]: 2.95 (s, 3H); 2.42 (s, 2H); 1.6 (m, 6H); 1.25 (s, 3H); 0.95 (t, 6H)
2e	CH ₃	CH ₃	90	231°	C ₁₃ H ₁₇ N ₅ (243.3)	243	[(CD ₃) ₂ CO]: 3.07 (s, 3H); 2.87 (s, 3H); 1.72 (d, 2H); 1.57 (s, 3H); 1.20 (s, 6H)
2f	C ₂ H ₅	CH ₃	91	118°	C ₁₅ H ₂₁ N ₅ (271.4)	—	(DMF): 6.47 (s, 2H); 4.35 (br s, 1H); 2.75 (s, 3H); 2.0–0.5 (m, 15H)
2g	CH ₃	C ₂ H ₅	86	132–134°	C ₁₄ H ₁₉ N ₅ (257.3)	257	(DMF): 3.1 (br s, 3H); 1.9–1.4 (m, 5H); 1.3–0.8 (m, 11H)
2h	C ₂ H ₅	C ₂ H ₅	82	132°	C ₁₆ H ₂₃ N ₅ (285.4)	285	(DMF): 8.00 (s, 1H); 6.82 (s, 2H); 4.17 (g, 2H); 2.0–0.8 (m, 18H)
2i	CH ₃	C ₆ H ₅ —CH ₂	66	143–144°	C ₁₉ H ₂₁ N ₅ (319.4)	—	(DMSO): 7.25 (s, 5H); 6.9 (m, 1H); 6.60 (s, 2H); 4.2 (m, 2H); 1.42 (s, 3H); 1.02 (s, 6H)
2j	C ₂ H ₅	C ₆ H ₅ —CH ₂	64	260°	C ₂₁ H ₂₅ N ₅ (347.4)	—	(DMSO): 7.17 (s, 5H); 5.55 (d, 1H); 2.35 (s, 2H); 1.17 (s, 3H); 0.9 (t, 6H)
2k	CH ₃	C ₆ H ₅ —NH—CO	76	> 300°	C ₁₉ H ₂₀ N ₆ O (348.4)	—	[(CD ₃) ₂ CO]: 7.8–7.0 (m, 5H); 6.50 (s, 1H); 3.57 (s, 1H); 2.92 (s, 2H); 1.85 (s, 2H); 1.67 (s, 3H); 1.27 (s, 6H)
2l	C ₂ H ₅	C ₆ H ₅ —NH—CS	65	148–149°	C ₂₁ H ₂₄ N ₆ S (392.4)	—	(DMSO): 11.95 (br s, 1H); 10.75 (br s, 1H); 7.8–7.0 (m, 7H); 3.57 (s, 2H); 1.8–0.8 (m, 13H)

^a Satisfactory microanalyses obtained: C \pm 0.38, H \pm 0.47, N \pm 0.40.**Table 2.** 4-Alkyl-2-amino-4-(substituted amino)-1,1,3-tricyano-1,3-butadienes **3**

Product No.	R ¹	R ²	Yield [%]	m.p. [°C]	Molecular formula ^a or Lit. m.p. [°C]	M.S. <i>m/e</i> (M ⁺)	¹ H-N.M.R. (DMSO/DSS) δ [ppm]
3a	CH ₃	H	95	> 300°	> 300° ^b	173	7.10 (s, 2H); 6.95 (s, 2H); 2.27 (s, 3H)
3b	C ₂ H ₅	H	93	298–300°	C ₉ H ₉ N ₅ (187.2)	187	7.25 (s, 2H); 7.07 (s, 2H); 2.60 (s, 2H); 1.22 (t, 3H)
3c	CH ₃	NH ₂	89	> 300°	C ₈ H ₈ N ₆ (188.2)	188	12.0 (br s, 1H); 7.15 (s, 2H); 5.67 (s, 2H); 2.50 (s, 3H)
3d	C ₂ H ₅	NH ₂	87	> 300°	C ₉ H ₁₀ N ₆ (202.2)	202	7.30 (s, 2H); 3.9 (br s, 3H); 2.85 (q, 2H); 1.30 (t, 3H)
3e	CH ₃	CH ₃	95	248–300°	C ₉ H ₉ N ₅ (187.2)	187	7.30 (q, 1H); 6.96 (s, 2H); 2.90 (d, 3H); 2.42 (s, 3H)
3f	C ₂ H ₅	CH ₃	95	210–211°	C ₁₀ H ₁₁ N ₅ (201.2)	201	7.35 (br s, 1H); 6.95 (s, 2H); 2.95 (d, 3H); 2.60 (s, 2H); 1.22 (t, 3H)
3g	CH ₃	C ₂ H ₅	92	228°	C ₁₀ H ₁₁ N ₅ (201.2)	201	7.30 (t, 1H); 6.92 (s, 2H); 3.50 (s, 2H); 2.45 (s, 3H); 1.12 (t, 3H)
3h	C ₂ H ₅	C ₂ H ₅	90	132°	C ₁₁ H ₁₃ N ₅ (215.2)	215	7.55 (s, 2H); 7.00 (s, 1H); 4.50 (q, 2H); 2.80 (s, 2H); 1.5–1.0 (m, 6H)
3i	CH ₃	C ₆ H ₅ —CH ₂	98	208°	C ₁₅ H ₁₃ N ₅ (263.2)	263	7.92 (t, 1H); 7.30 (s, 5H); 7.02 (s, 2H); 4.60 (d, 2H); 2.35 (s, 3H)
3j	C ₂ H ₅	C ₆ H ₅ —CH ₂	94	285°	C ₁₆ H ₁₅ N ₅ (277.3)	—	7.9 (br s, 5H); 7.15 (br s, 3H); 2.70 (s, 2H); 1.20 (t, 3H)

^a Satisfactory microanalyses obtained: C \pm 0.37, H \pm 0.31, N \pm 0.36.

or 50% aqueous amine solution (10 ml) is stirred at room temperature for 2–3 h. The product is then filtered [for products **2f–j**, addition of water (20 ml) to the mother liquid is necessary to precipitate the adduct] and washed with cooled 70% aqueous ethanol (Table 1).

Reaction of 2a, b with Phenyl Isocyanate or Isothiocyanate:

The adduct **2a, b** (4 mmol), phenyl isothiocyanate or phenyl isocyanate (0.5 ml), and dioxan (10 ml) are boiled for 15 min. The mixture is

then allowed to cool and water (20 ml) is added. The separated oil is treated with ethanol to give the crystalline product **2k, l** (Table 1).

4-Alkyl-2-amino-4-(substituted amino)-1,1,3-tricyano-1,3-butadienes 3; General Procedure:

A mixture of adduct **2** (2 mmol) and xylene (10 ml; for **2b, d, e, f, g, h, i, j**) or decalin (10 ml; for **2a, c**) is refluxed for 1 h. The mixture is then allowed to cool and the product is filtered off. In the case of **3b** and

3f, a few drops of hexane are added to precipitate the product. Samples for analysis are crystallized from 2:1 dimethylformamide/water.

Received: April 6, 1981

¹ For Part III, see: M. Adamczyk, M. Mokrosz, *Chem. Scripta*, in press.

² J. Mirek, M. Adamczyk, M. Mokrosz, *Synthesis* **1980**, 296.

³ M. Adamczyk, M. Mirek, M. Mokrosz, *Synthesis* **1980**, 916.

⁴ J. L. Ripoll, *Tetrahedron* **34**, 19 (1978).

⁵ H. von Brachel, U. Bahr, in Houben-Weyl, *Methoden der Organischen Chemie*, 4th Edn., E. Müller, Ed., Vol. V/1c, Georg Thieme Verlag, Stuttgart, 1970, p. 782.

⁶ J. K. Williams, *J. Org. Chem.* **28**, 1054 (1963).

0039-7881/81/1032-0804 \$ 03.00

© 1981 Georg Thieme Verlag · Stuttgart · New York