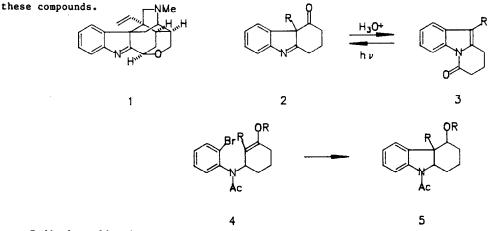
INTRAMOLECULAR RADICAL CYCLIZATION REACTIONS. AN APPROACH TO THE INDOLE ALKALOIDS

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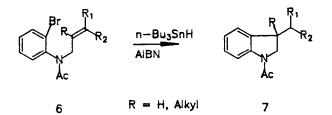
Abstract: The intramolecular radical cyclization reaction of N-allyl substituted derivatives of o-bromoacetanilide provides a short and effective route to a variety of dihydroindole systems.

In connection with our work on the synthesis of the Chinese medicinal alkaloid koumine $(\underline{1})^1$ we required a synthesis of the 4-keto-ll-alkyltetrahydrocarbazolenine ring system 2. Previous attempts to prepare this system via acid catalyzed Fischer Indole synthesis were unsuccessful.² Under the conditions of the Fischer Indole reaction, the desired product undergoes acid catalyzed ring opening to a carboxylic acid and subsequent lactam formation $(\underline{2} \cdot \underline{3})$. Interestingly, the lactam $\underline{3}$ is convertible back to the desired carbazolenine $\underline{2}$ by photochemical means although in poor yield.³ Consequently, we have initiated an investigation of the radical mediated arylation reaction $(\underline{4} + \underline{5})$ as a means for preparation of



Radical cyclization reactions have received a significant amount of attention in recent years.⁴ The literature is abundant in examples of intramolecular radical induced cyclizations for the preparation of carbocycles⁵ and heterocycles⁶, often with good

stereochemical control. Free radical arylation reactions are less widely published but show considerable promise for the preparation of benzofuran and dihydrobenzofuran systems.⁷ Relatively little attention, however, has been given to preparation of the indole and dihydroindole ring systems by this method.⁸ We present here a fast and efficient method for the preparation of dihydroindoles 7 from the corresponding allyl amides <u>6</u>.



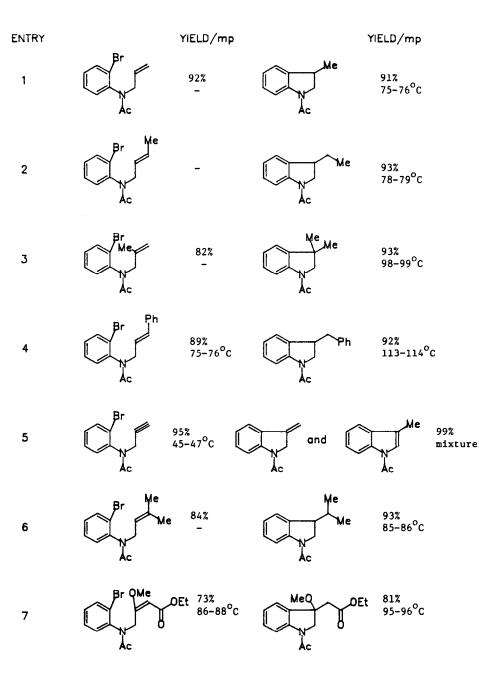
All cyclizations were carried out in refluxing benzene with 1.2 molar equivalents of n-Bu₃SnH and AIBN as catalyst. Workup simply entailed evaporation of the solvent. Facile separation of the tin byproducts was achieved by flash chromatography to afford cyclized products 7 in 81-99% yield ⁹ (Table 1). The following general procedure is given for the preparation of the allyl amides <u>6</u> and dihydroindoles <u>7</u>. o-Bromoacetanilide was prepared by the usual procedure from o-bromoaniline¹⁰, pyridine and acetic anhydride.

<u>Preparation of Allyl Amides 6</u>. o-Bromoacetanilide (10 mmol) was added to a stirred suspension of KH (30%, 1.33 g) in dry THF (75 ml) under nitrogen at -78 $^{\circ}$ C. Stirring was continued until a clear solution was obtained. The allyl halide (10 mmol) was then added and stirring was continued at -78 $^{\circ}$ C for 30 min and at room temperature for 8 h. The reaction mixture was poured into water (250 ml) and extracted with ethyl acetate (3 x 50 ml). The combined organic extracts were washed with brine, dried over anhydrous magnesium sulfate and evaporated to afford crude <u>6</u>. The products were chromatographed on silica gel (hexane: ethyl acetate, 2:1) to provide allyl amides <u>6</u> in 73-95% yield).

<u>Preparation of Dihydroindoles 7</u>. To a refluxing solution of allyl amide <u>6</u> (1 mmol) and AIBN (10 mg) in dry benzene (80 ml) was added a solution of $n-Bu_3SnH$ (1.2 mmol) in dry benzene (20 ml) ove. 30 min. The resulting mixture was heated at reflux for 1 h and the solvent was removed at reduced pressure. Chromatography on silica gel (hexane: ethyl aetate, 2:1) provided dihydroindoles <u>7</u> in 81-99% yield.

Thus far the range of products available by the radical cyclization procedure include 3monosubstituted and 3,3-disubstituted dihydroindoles (entry 1-4 and 6-7), 3-alkoxy-3-alkyl





dihydroindoles (entry 7) and dihydroindole systems with an exomethylene substituent at the 3 position (entry 5). In conclusion the radical cyclization reaction of N-allyl substituted obromoacetanilide derivatives provides a short and effective means for the preparation of a variety of dihydroindole systems.

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