Bis-Functionalization of 1,3-Dienes through 1,4-Conjugate Addition of Amphiphilic Bis-π-Allyl and Related Palladium Intermediates

T. V. Baiju,^{a,b} Ajesh Vijayan,^{a,b} Nayana Joseph,^b Preethanuj Preethalayam,^b K. V. Radhakrishnan,^{*a,b} E. Suresh,^c Yoshinori Yamamoto^{*d}

- ^a Academy of Scientific and Innovative Research (AcSIR), New Delhi 110001, India
- ^b Organic Chemistry Section, National Institute for Interdisciplinary Science and Technology (CSIR), Trivandrum 695019, India Fax +91(471)2491712; E-mail: radhu2005@gmail.com
- ° Central Salt and Marine Chemicals Research Institute, CSIR-CSMCRI, Bhavnagar 364021, Gujarat, India
- ^d WPI-AIMR (Advanced Institute for Materials Research), Tohoku University, Katahira 2-1-1, Aobaku, Sendai 980-8577, Japan E-mail: yoshi@mail.tains.tohoku.ac.jp

Received: 25.09.2013; Accepted after revision: 25.10.2013

Abstract: Palladium-catalyzed three-component coupling of allylstannane, allyl chloride and a functionalized diene is described. Regioselective 1,4-functionalization of the Michael acceptor 1,3-diene is accomplished by the amphiphilic bis- π -allylpalladium complex. To the best of our knowledge, this is the first time a functionalized 1,3-butadiene has been used as a Michael acceptor. The scope of the present strategy is further extended to 1,4-allylation–oxyallylation of functionalized dienes.

Key words: palladium, allylation, allyl complexes, coupling, Michael addition

Strategies for 1,4-functionalization of 1,3-dienes through classical methods include cycloaddition reactions involving singlet oxygen,¹ and heterodienophiles such as nitroso compounds² and azodicarboxylates.³ Cycloadducts are further transformed into target molecules through suitable synthetic manipulations. Pioneering work on palladiumcatalyzed 1,4-functionalization of 1,3-dienes was reported by Backvall and co-workers,⁴ and transition-metal-catalyzed transformations of 1,3-dienes have attracted the attention of a number of organic chemists.⁵ Synthetic transformations of 1,3-dienes using nickel,⁶ palladium,⁷ iron,⁸ and rhodium⁹ have been reported, and Hilt et al. have utilized the cobalt-catalyzed 1,4-hydrovinylation of 1,3-dienes for the synthesis of functionalized 1,4-dienes.¹⁰ RajanBabu and co-workers have carried out detailed investigations on asymmetric hydrovinylation of linear and cyclic 1,3-dienes.¹¹ Very recently, Sigman and co-workers achieved palladium-catalyzed 1,4-addition across the commodity chemical 1,3-butadiene to afford skipped polyene products.¹²



Figure 1 π -Allyl and bis- π -allyl palladium complexes

Bis- π -allylpalladium (Figure 1) and related intermediates show amphiphilic reactivity on reaction with activated olefins¹³ and benzynes.¹⁴ Inter- and intramolecular reactions of aldehydes, imines and activated olefins with bis- π -allylpalladium complexes have paved the way toward the synthesis of a number of highly functionalized organic molecules.^{13,15} Recently we have reported the bis-functionalization of isatylidenes by using the bis- π -allylpalladium complex as a facile route toward spiro-indol-2ones.¹⁶ With the highly conjugated heptafulvene, the bis- π -allylpalladium complex undergoes 1,8-conjugate addition leading to bis-functionalized cycloheptatriene derivatives.¹⁷ A palladium-catalyzed deconjugative allylation reaction of 1,3-diene was reported for the first time by Sato and co-workers.¹⁸ Cheng et al. reported the reaction



Scheme 1 Reports by Sato et al.¹⁸ (top) and Cheng et al.¹⁹ (bottom)

SYNLETT 2014, 25, 0359–0364 Advanced online publication: 02.12.2013 DOI: 10.1055/s-0033-1340171; Art ID: ST-2013-B0914-L © Georg Thieme Verlag Stuttgart · New York

LETTER

of a bis- π -allylpalladium complex generated from allyl chloride and allenylstannane with 1,3-diene, which afforded 1,2-addition products with exclusive regio- and chemoselectivity (Scheme 1).¹⁹

As part of our continuing interest in utilizing the amphiphilic nature of the bis- π -allylpalladium complex in organic synthesis, we undertook an investigation of its reactivity with highly functionalized 1,3-butadienes.²⁰ Our preliminary experiments involved the reaction of 1,3-butadiene derivative **1a** with allyl chloride (**2**) and allyltributylstannane (**3**) in the presence of [PdCl₂(PPh₃)₂] in THF at room temperature (Table 1, entry 1).²¹ The reaction afforded 1,4-bis-allylated product **4a** in 78% yield, the structure of which was established on the basis of a range of spectroscopic techniques.²²

 Table 1
 Palladium-Catalyzed Bis-Allylation of 1,3-Butadiene^a

The reaction was then optimized to establish the best catalytic conditions (see the Supporting Information). Based on these studies, the optimal conditions for the transformation was found to be diene (1.0 equiv.), allyl chloride (2.0 equiv), allyltributylstannane (2.0 equiv), and 5 mol% [PdCl₂(PPh₃)₂] in THF (2 mL) at room temperature.

The substrate scope for the bis-allylation strategy was investigated by utilizing various functionalized 1,3-butadienes. It should be noted that a variety of the highly functionalized 1,3-butadienes derived from substituted benzylidine malononitriles (1b-d, 1g and 1h) and heteroaryl malononitriles (1e and 1f) can be used in this approach (Table 1). The use of highly substituted 1,3-butadienes makes this method potentially valuable for the synthesis of a number of biologically important targets.





 Table 1
 Palladium-Catalyzed Bis-Allylation of 1,3-Butadiene^a (continued)

^a Reaction conditions: 1,3-diene **1** (1.0 equiv), allyl chloride **2** (2.0 equiv), allyl tributylstannane **3** (2.0 equiv), [PdCl₂(PPh₃)₂] (5 mol%), THF (2 mL), r.t., 8 h.

A plausible mechanistic pathway for bis-allylation is illustrated in Scheme 2. The initial event involves oxidative addition of allyl chloride **2** to a palladium(0) species to produce the η^3 -allylpalladium intermediate **A**. This intermediate undergoes ligand exchange with allyltributylstannane **3** to generate bis- η^3 -allylpalladium intermediate **B**, which subsequently undergoes nucleophilic 1,4-addition with diene **1** to form intermediate **C**, which, on reductive elimination, forms 1,4-bis-allylated product **4**.

To demonstrate the use of related π -allyl palladium intermediates, we initiated our investigations with the allylation-oxyallylation reaction of **1a** and diallylcarbonate **5** by the use of [Pd(PPh₃)₄] as the catalyst and THF as the solvent. To our delight, the desired allylated-oxyallylated product **6a** was obtained in 71% yield (Table 2, entry 1). The structure of the allylated-oxyallylated product was unambiguously confirmed by single-crystal X-ray analysis of **6b** (from the reaction of **1b** and **5**; Figure 2).²³ The reaction presumably proceeds through oxidative addition of palladium to the diallyl carbonate, followed by loss of CO_2 to give intermediate **D**. Nucleophilic 1,4-addition followed by reductive elimination results in the formation of product **6**.

To develop conditions that were suitable for this transformation, we surveyed a variety of palladium catalysts and solvents (see the Supporting Information) and found that the optimal conditions for this reaction were: a mixture of 1,3-diene/diallyl carbonate (1:2) with 5 mol% [Pd(PPh₃)₄] in THF (2 mL) with a reaction time of 8 hours.





Figure 2 Single-crystal X-ray structure of 6b

Scheme 2 Mechanistic rationale of palladium-catalyzed bis-allylation of 1,3-butadiene

A detailed study to expand the 1,4-allylation–oxyallylation strategy to other 1,3-dienes (**1b–f**) was undertaken, the results of which demonstrated that a wide range of substitution patterns are tolerated (Table 2).

In conclusion, we have developed a simple and efficient strategy for the bis-functionalization of 1,3-butadiene de-

Table 2 Palladium-Catalyzed Allylation–Oxyallylation Reaction of 1,3-Dienes^a

rivatives by a palladium-catalyzed three-component coupling reaction. To the best of our knowledge, this is the first report on a palladium-catalyzed 1,4-conjugate addition reaction of 1,3-butadiene derivatives via amphiphilic bis- π -allylpalladium and related complexes. Further synthetic manipulations of the synthesized trienes and investigations on the scope of other related palladium intermediates are under way and will be reported in due course.





Table 2 Palladium-Catalyzed Allylation–Oxyallylation Reaction of 1,3-Dienes^a

^a Reaction conditions: 1,3-diene 1 (1.0 equiv.), diallyl carbonate 5 (2.0 equiv.), [Pd(PPh₃)₄] (5 mol%), THF (2 mL), r.t., 8 h

Acknowledgment

T.V.B., A.V., N.J., and P.P. thank CSIR, New Delhi and UGC for research fellowships. Financial assistance from the Department of Science and Technology (DST No: SR/S1/OC-78/2009) and the Council of Scientific and Industrial Research, New Delhi [12th FYP project on catalysts for speciality chemicals (CSC-0125)] are greatly acknowledged. The authors also thank Ms. Saumini Mathew, Mr. Saran P. Raveendran, and Ms. S. Viji of CSIR-NIIST for recording NMR and mass spectra, respectively.

Supporting Information for this article is available online at http://www.thieme-connect.com/ejournals/toc/synlett.

References and Notes

- (a) Balci, M. Chem. Rev. 1981, 81, 91. (b) Davis, K. M.; Carpenter, B. K. J. Org. Chem. 1996, 61, 4617. (c) Gollnick, K.; Griesbeck, A. Tetrahedron Lett. 1983, 24, 3303.
 (d) Aubry, J.-M.; Mandard-Cazin, B.; Rougee, M.; Bensasson, R. V. J. Am. Chem. Soc. 1995, 117, 9159.
 (e) Mehta, G.; Uma, R. J. Org. Chem. 2000, 65, 1685.
- (2) (a) Belleau, B.; Au-Young, Y.-K. J. Am. Chem. Soc. 1963, 85, 64. (b) Streith, J.; Augelmann, G.; Fritz, H.; Strub, H. Tetrahedron Lett. 1982, 23, 1909. (c) Keck, G. E.; Fleming, S. A. Tetrahedron Lett. 1978, 4763.
- (3) (a) Diels, O.; Blom, J. H.; Knoll, W. *Justus Liebigs Ann. Chem.* **1925**, *443*, 242. (b) Cookson, R. C.; Gilani, S. S. H.; Stevens, I. D. R. *Tetrahedron Lett.* **1962**, 615. (c) Giuliano, R. M.; Jordan, A. D.; Gauthier, A. D.; Hoogsteen, K. *J. Org. Chem.* **1993**, *58*, 4979.
- (4) (a) Verboom, R. C.; Perssonand, B. A.; Bäckvall, J. E.
 J. Org. Chem. 2009, 69, 3102. (b) Aranyos, A.; Szabo, K. J.;

Bäckvall, J. E. J. Org. Chem. 1998, 63, 2523. (c) Bäckvall,
J. E. Pure Appl. Chem. 1996, 68, 535. (d) Bäckvall, J. E.;
Nyström, J. E.; Nordberg, R. E. J. Am. Chem. Soc. 1985,
107, 3676. (e) Bäckvall, J. E.; Byström, S. E.; Nordberg,
R. E. J. Org. Chem. 1984, 49, 4619. (f) Bäckvall, J. E. Pure
Appl. Chem. 1983, 55, 1669.

- (5) (a) He, Z.; Yi, C. S.; Donaldson, W. A. Org. Lett. 2003, 5, 1567. (b) Gerdin, M.; Moberg, C. Adv. Synth. Catal. 2005, 347, 749. (c) Kliman, L. T.; Mlynarski, S. N.; Ferris, G. E.; Morken, J. P. Angew. Chem. Int. Ed. 2012, 51, 521. (d) Mizutani, K.; Shinokubo, H.; Oshima, K. Org. Lett. 2003, 5, 3959. (e) Sasaki, Y.; Zhong, C.; Sawamura, M.; Ito, H. J. Am. Chem. Soc. 2010, 132, 1226.
- (6) (a) Ely, R. J.; Morken, J. P. J. Am. Chem. Soc. 2010, 132, 2534. (b) Shirakura, M.; Suginome, M. J. Am. Chem. Soc. 2008, 130, 5410. (c) Gerdin, M.; Moberg, C. Org. Lett. 2006, 14, 2929. (d) Saha, B.; RajanBabu, T. V. Org. Lett. 2006, 8, 4657. (e) Takimoto, M.; Mori, M. J. Am. Chem. Soc. 2001, 123, 2895. (f) Kimura, M.; Ezoe, A.; Shibata, K.; Tamaru, Y. J. Am. Chem. Soc. 1998, 120, 4033. (g) Buono, G.; Siv, C.; Peiffer, G.; Triantaphylides, C.; Denis, P.; Mortreux, A.; Petit, F. J. Org. Chem. 1985, 50, 1781.
- (7) (a) Takaya, J.; Sasano, K.; Iwasawa, N. Org. Lett. 2011, 13, 1698. (b) Rozhkov, R. V.; Larock, R. C. J. Org. Chem. 2010, 75, 4131. (c) Liao, L.; Sigman, M. S. J. Am. Chem. Soc. 2010, 132, 10209.
- (8) (a) Wu, J. Y.; Stanzl, B. N.; Ritter, T. J. Am. Chem. Soc. 2010, 132, 13214. (b) Wu, J. Y.; Moreau, B.; Ritter, T. J. Am. Chem. Soc. 2009, 131, 12915.
- (9) Watkins, A. L.; Landis, C. R. Org. Lett. 2011, 13, 164.
- (10) (a) Arndt, M.; Dindaroglu, M.; Schmalz, H. G.; Hilt, G. Org. Lett. 2011, 13, 6236. (b) Arndt, M.; Reinhold, A.; Hilt, G. J. Org. Chem. 2010, 75, 5203. (c) Hilt, G.; du Mesnil, F. X.; Luers, S. Angew. Chem. Int. Ed. 2001, 40, 387.
- (11) (a) Page, J. P.; RajanBabu, T. V. J. Am. Chem. Soc. 2012, 134, 6556. (b) Sharma, R. K.; RajanBabu, T. V. J. Am. Chem. Soc. 2010, 132, 3295. (c) Saha, B.; Smith, C. R.; RajanBabu, T. V. J. Am. Chem. Soc. 2008, 130, 9000.
 (d) Zhang, A.; RajanBabu, T. V. J. Am. Chem. Soc. 2006, 128, 54. (e) RajanBabu, T. V. Chem. Rev. 2003, 103, 2845.
- (12) McCammant, M. S.; Liao, L.; Sigman, M. S. J. Am. Chem. Soc. 2013, 135, 4167.
- (13) (a) Patil, N. T.; Yamamoto, Y. *Synlett* 2007, 1994; and references cited therein. (b) Nakamura, H.; Sekido, M.; Ito, M.; Yamamoto, Y. *J. Am. Chem. Soc.* 1998, *120*, 6838.
 (c) Sekido, M.; Aoyagi, K.; Nakamura, H.; Kabuto, C.; Yamamoto, Y. *J. Org. Chem.* 2001, *66*, 7142.
- (14) Yoshikawa, E.; Radhakrishnan, K. V.; Yamamoto, Y. *Tetrahedron Lett.* **2000**, *41*, 729.
- (15) (a) Lu, S.; Jin, T.; Bao, M.; Asiri, A. M.; Yamamoto, Y. *Tetrahedron Lett.* **2012**, *53*, 1210. (b) Nakamura, H.;

- Shimizu, K. *Tetrahedron Lett.* 2011, *52*, 426. (c) Lu, S.; Xu,
 Z.; Bao, M.; Yamamoto, Y. *Angew. Chem. Int. Ed.* 2008, *47*,
 4366. (d) Wallner, O. A.; Szabo, K. J. *Org. Lett.* 2002, *4*,
 1563. (e) Pichierri, F.; Yamamoto, Y. J. *Org. Chem.* 2007, *72*, 861. (f) Nakamura, H.; Shibata, H.; Yamamoto, Y. *Tetrahedron Lett.* 2000, *41*, 2911. (g) Nakamura, H.;
 Ohtaka, M.; Yamamoto, Y. *Tetrahedron Lett.* 2002, *43*,
 7631. (h) Ohtaka, M.; Nakamura, H.; Yamamoto, Y. *Tetrahedron Lett.* 2004, *45*, 7339. (i) Bao, M.; Nakamura,
 H.; Yamamoto, Y. *J. Am. Chem. Soc.* 2001, *123*, 759.
 (j) Nakamura, H.; Asao, N.; Yamamoto, Y. *J. Chem. Soc., Chem. Commun.* 1995, 1273.
- (16) George, S. C.; John, J.; Anas, S.; John, J.; Yamamoto, Y.; Suresh, E.; Radhakrishnan, K. V. *Eur. J. Org. Chem.* 2010, 5489.
- (17) George, S. C.; Thulasi, S.; Anas, S.; Radhakrishnan, K. V.; Yamamoto, Y. Org. Lett. 2011, 13, 4984.
- (18) Sato, Y.; Oonishi, Y.; Mori, M. J. Org. Chem. 2003, 68, 9858.
- (19) (a) Jaganmohan, M.; Shanmugasundaram, M.; Cheng, C.-H. J. Org. Chem. 2004, 69, 4053. (b) Jaganmohan, M.; Shanmugasundaram, M.; Cheng, C.-H. Org. Lett. 2003, 5, 881.
- (20) Nair, V.; Remadevi, B.; Vidya, N.; Menon, R. S.; Abhilash, N.; Rath, N. P. *Tetrahedron Lett.* **2004**, *45*, 3203.
- (21) Typical Procedure (Compound 4a): To a degassed solution of [PdCl₂(PPh₃)₂] (4.4 mg, 0.0064 mmol) in anhydrous THF (2 mL) in a Schlenk tube, allyltributyl-stannane 3 (85.2 mg, 0.25 mmol) was added followed by allyl chloride 2 (19.6 mg, 0.25 mmol). To this, 1a (42.02 mg, 0.12 mmol) was added (in THF) and the mixture was stirred at room temperature for 8 h. After the completion of the reaction (as evident by TLC), the solvent was removed under reduced pressure and the residue was purified by using silica gel (100–200 mesh) column chromatography (EtOAc–hexane, 12%) to afford 4a (41.1 mg, 78%).
- (22) **Spectral Data of 4a:** $R_f = 0.46$ (EtOAc–hexane, 4:6); IR (neat): 3079, 2955, 2919, 2850, 2313, 2246, 1734, 1604, 1510, 1461, 1376, 1290, 1248, 1177, 1118, 1032 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): $\delta = 7.26-7.24$ (m, 1 H), 7.08 (d, J = 8.5 Hz, 1 H), 6.98–6.94 (m, 2 H), 5.94–5.86 (m, 1 H), 5.56–5.47 (m, 1 H), 5.42–5.35 (m, 2 H), 5.05–4.98 (m, 2 H), 3.89 (s, 3 H), 3.84 (s, 3 H), 3.72 (s, 3 H), 3.18–3.15 (m, 1 H), 2.87–2.75 (m, 2 H), 2.63–2.58 (m, 1 H), 2.27–2.21 (m, 1 H); ¹³C NMR (125 MHz, CDCl₃): $\delta = 171.0$, 166.0, 160.4, 137.4, 136.6, 134.0, 131.1, 130.1, 128.7, 126.0, 123.5, 118.1, 114.5, 114.3, 113.7 (2C), 55.3, 52.6, 52.5, 48.2, 42.7, 42.5, 33.8; HRMS (ESI): m/z [M + Na]⁺ calcd for C₂₃H₂₄N₂NaO₅: 431.15829; found: 431.15646. (23) CCDC-933875.

Copyright of Synlett is the property of Georg Thieme Verlag Stuttgart and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.