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Synthesis of Simple Oxetanes Carrying Reactive 2-Substituents

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Ring-expansion of substituted epoxides using dimethyloxosulphonium methylide provides a convenient route to oxetanes carrying reactive 2substituents that are capable of further modification.

Whereas several methods are available for the synthesis of oxetanes, 1-9 no general method exists for the preparation of simple examples carrying a reactive substituent such as hydroxymethyl in the 2-position. Although various substituted 2hydroxymethyloxetanes have been synthesised from 3,4epoxyalcohols, 10,11 the method is unsuitable for the synthesis of 2-hydroxymethyloxetane (2a) itself, since it proceeds in poor yield, and the product is heavily contaminated with 3hydroxytetrahydrofuran.

We now report a much improved route to 2-hydroxymethyloxetane (2a) and related compounds via the ring-expansion of substituted epoxides 1b-f using dimethyloxosulphonium methvlide as methylene transfer agent (Scheme). Okuma et al. 12 have also ring-expanded epoxides, but only to synthesise oxetanes bearing aromatic or cycloalkyl substituents.

1, 2	R
а	CH,OH
b	CH ₂ OCH(CH ₃)OC ₂ H ₅
c	$CH_2OCH_2CH = CH_2$
d	$CH_2OC_6H_5$
e	$CH_2CH_2CH = CH_2$
f	$CH(OC_2H_5)_2$

Thus, 2,3-epoxypropan-1-ol (1a) was first protected by interaction with ethoxyethene, and the resulting ether 1b was stirred for 3d at 50°C with dimethyloxosulphonium methylide (prepared

from trimethyloxosulphonium iodide and potassium tertbutoxide in tert-butyl alcohol). This gave the corresponding oxetane ether 2b, which was readily deprotected in acid to give 2hydroxymethyloxetane (2a).

Corresponding ring-expansions occurred when the readily available epoxides 1c-e were similarly treated with dimethyloxosulphonium methylide. The results are shown in the Table.

Epoxide 1f was prepared from 2,3-dihydroxypropanal diethyl acetal13 via cyclisation of its monotosyl derivative, since both the published methods available 14,15 for 1f failed in our hands.

The oxetane substituents may be modified whilst preserving the integrity of the oxetane ring. Thus, for example, 2-hydroxymethyloxetane (2a) was readily converted via its tosylate 2g into the thioether 2h by interaction of the latter with 4-chlorobenzenethiol.

The unsaturated side-chains in oxetanes 2c and 2e could also be modified. For example, both gave epoxides (2i and 2j, respectively) on treatment with m-chloroperoxybenzoic acid (MCPBA) and dibromides (2k and 2l, respectively) on treatment with bromine in the presence of potassium carbonate. Oxetane-2-carbaldehyde could not be isolated from the acetal 2f, but the latter did give oxetane-2-carbaldehyde 2,4-dinitrophenylhydra-

Table. Oxetanes 2a-f Prepared

Product	Yield (%)	b.p. (°C)/Torr	Molecular Formula ^a	IR (film) v (cm ⁻¹)	1 H-NMR (CDCl ₃ /TMS) δ , J (Hz)
2a	74	94-98/15	$C_4H_8O_2$	3400, 980, 920	2.7 (m, 2H); 3.8 (m, 3H); 4.4-5.0 (m, 3H)
2 b	70	55-58/0.4	(88.1) C ₈ H ₁₆ O ₃ (160.2)	700,	1.2 (m, 6H); 2.7 (m, 2H); 3.6 (m, 4H); 4.7 (m, 4H)
2c	65	52-58/1	$C_7H_{12}O_2$ (128.2)	1640, 980, 920	2.7 (m, 2H); 3.6 (d, 2H, <i>J</i> = 5); 4.0 (d, 2H, <i>J</i> = 5); 4.7 (m, 3H); 5.2 (m, 2H); 5.9 (m, 1H) 2.7 (m, 2H); 3.9 (m, 2H); 4.7 (m, 2H); 5.0 (m, 1H); 7.1 (m, 5H)
2d	83	116118/0.8 ^b		3060, 1600, 1580, 970, 950	
2e	56	$62-64/20^{\rm b}$		3080, 1640, 980,	1.6-2.4 (m, 4H); 2.7 (m, 2H); 4.3-5.2 (m, 5H); 6.0 (m, 1H)
2f	59	88-90/10 ^b			1.3 (m, 6H); 2.7 (m, 2H); 3.7 (m, 4H); 4.6 (m, 4H)

Satisfactory microanalyses obtained: $C \pm 0.30$, $H \pm 0.30$; except for **2e**: H + 0.50.

Product isolated by flash chromatography prior to distillation. Elution solvents: 2d: ether/pentane (2:3); 2e: ether/pentane (1:4); and 2f: ether/pentane (1:3).

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zone (2m) on stirring overnight with 2,4-dinitrophenylhydrazine in methanol and concentrated sulphuric acid. Attempts to oxidise the acetal to ethyl oxetane-2-carboxylate using either peroxyacetic acid or MCPBA were unsuccessful.

2,3-Epoxy-1-(1-ethoxyethoxy)propane (1b):

To a magnetically stirred solution of 2,3-epoxypropanol (40.0 g, 0.54 mol) in ethyl vinyl ether (200 mL) is added TsOH (1 g) portionwise, keeping the temperature below 40 °C. The mixture is stirred for 3 h and sat. aq. NaHCO₃ (100 mL) is then added. The organic layer is separated, dried and evaporated under reduced pressure. Distillation of the residue gives 1b as a colourless liquid; yield: 72.6 g (92%); b.p. 152-154 °C.

 $C_7H_{14}O_3$ calc. C 57.58 H 9.66 (146.2) found 57.77 9.55 IR (film): v = 1350, 1250 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 1.3 (m, 6 H); 2.7 (m, 2 H); 3.1 (m, 1 H); 3.6 (m, 4 H); 4.7 (q, 1 H, J = 6 Hz).

2,3-Epoxy-1,1-diethoxypropane (1f):

To a magnetically stirred mixture of TsCl (22.0 g, 0.115 mol) in dry Py (15 mL) is added dropwise a solution of 2,3-dihydroxypropanal diethyl acetal (17.6 g, 0.107 mol) in dry Py (15 mL) keeping the temperature below 0 °C. The mixture is stirred for 4 h then kept in a refrigerator at 2 °C overnight. Ice/water (100 mL) is then added, and the mixture is extracted with CH_2Cl_2 (3 × 100 mL). The extract is washed successively with 10 % aq. H_2SO_4 (200 mL) and sat. brine (3 × 100 mL), then dried (MgSO₄) and evaporated under reduced pressure to give the monotosylate; yield: 22.1 g (65 %).

A solution of the monotosylate (5.00 g, 15.7 mmol) in dry ether (50 mL) is added dropwise to a magnetically stirred solution of KOBu-t (1.94 g, 17.3 mmol) in dry ether (50 mL) keeping the temperature at 0°C . The mixture is stirred at 0°C for 6 h, then at room temperature overnight. The mixture is filtered and the solid washed with dry ether $(3 \times 30 \text{ mL})$. The combined filtrate and washings is dried (MgSO₄), and the solvent is carefully evaporated. Distillation of the residue gives 1f as a colourless liquid; yield: 1.10 g (48 %); b.p. $58-62^{\circ}\text{C}/10$ Torr (Lit. 14,15 b.p. $60-64^{\circ}\text{C}/13$ Torr).

Ring-Expansion of Epoxide 1b to Oxetane 2b; Typical Procedure:

A mixture of KOBu-t (61.5 g, 0.548 mol) and trimethyloxosulphonium iodide (120.6 g, 0.548 mol) in dry t-BuOH (700 mL) is stirred magnetically at 50 °C for 1 h. A solution of 2,3-epoxy-1-(1-ethoxyethoxy) propane (1b; 40.0 g, 0.274 mol) in dry t-BuOH (100 mL) is then added dropwise over 30 min at 50 °C and stirring is then continued at the same temperature for 3d. The solvent is carefully evaporated under reduced pressure and water (200 mL) is added to the residual suspension. The mixture is extracted with pentane (4×150 mL), and the combined extracts are dried (MgSO₄) and evaporated under reduced pressure. Distillation of the residue (after column chromatographic separation in the cases of 2d-f) gives 2-(1-ethoxyethoxy)methyloxetane 2b as a colourless liquid; yield: 30.06 g (70%) (see Table).

2-Hydroxymethyloxetane (2a):

A solution of **2b** (10.0 g, 62.5 mmol) in 10% aq. HOAc (50 mL) is magnetically stirred overnight. The reaction mixture is neutralised with solid NaHCO₃ (7.5 g), and the mixture is distilled until the distillation temperature reaches 100°C. The residual solution is cooled, and after addition of an excess of solid K_2CO_3 , is extracted with ether (3×100 mL). Evaporation of the dried (MgSO₄) extract and distillation of the residue gives **2a** as a colourless liquid; yield 4.07 g (74%) (see Table).

2-(Tosyloxymethyl)oxetane (2g):

To a magnetically stirred solution of TsCl (7.15 g. 37.5 mmol) in Py (10 mL) is added a solution of **2a** (3.00 g, 34.1 mmol) in Py (5 mL) keeping the reaction temperature below 0 °C. The mixture is stirred for a further 1 h at 0 °C and is then left in a refrigerator at 2 °C overnight. Ice/water (75 mL) is added, and the mixture is extracted with ether (3 × 100 mL). The combined extract is washed successively with 10 % $\rm H_2SO_4$ (3 × 100 mL) and sat. brine (3 × 100 mL), then dried (MgSO₄) and evaporated. The residue is flash chromatographed on silica gel. Elution with ether/pentane (4:1) gives **2g** as a colourless oil, which gradually gives a white solid; yield: 6.02 g (73 %); m.p. 58–59 °C.

C₁₁H₁₄O₄S calc. C 54.53 H 5.82 (242.3) found 54.30 5.87

IR (CHBr₃): v = 1600, 1180, 1155, 980, 920 cm⁻¹. ¹H-NMR (CDCl₃): $\delta = 2.4$ (s, 3 H); 2.7 (m, 2 H); 4.2 (m, 2 H); 4.5 (m, 2 H); 4.9 (m, 1 H); 7.3–8.8 (m, 4 H).

2-(4-Chlorophenylthiomethyl)oxetane (2h):

To a magnetically stirred solution of NaOMe (0.45 g, 8.33 mmol) in MeOH (10 mL) is added 4-chlorobenzenethiol (1.20 g, 8.31 mmol). The mixture is stirred for 30 min and 2-(tosyloxymethyl)oxetane (2g; 2.00 g, 8.27 mmol) is added, stirring then being continued overnight. The reaction mixture is filtered, and the solid washed with MeOH $(2 \times 5 \text{ mL})$. The combined filtrate and washings are evaporated under reduced pressure, and the residue is flash chromatographed on silica gel. Elution with ether/pentane (1:9) gives 2h as a colourless oil; yield: 1.31 g (74%).

C₁₀H₁₁ClOS calc. C 55.94 H 5.16 (214.7) found 56.08 5.33 IR (film): v = 3050, 1580, 980, 940 cm⁻¹. ¹H-NMR (CDCl₃/TMS): $\delta = 2.3-2.9$ (m, 2 H); 3.1 (m, 2 H); 4.4-5.1 (m, 3 H); 7.1-7.3 (m, 4 H).

2-(2,3-Epoxypropoxymethyl)oxetane (2i):

To a magnetically stirred solution of 2-(allyloxymethyl)oxetane (2c; $1.00 \, \mathrm{g}$, $7.8 \, \mathrm{mmol}$) in CH₂Cl₂ ($100 \, \mathrm{mL}$) is added at room temperature dropwise a solution of MCPBA ($1.80 \, \mathrm{g}$, $10.4 \, \mathrm{mmol}$) in CH₂Cl₂ ($20 \, \mathrm{mL}$) over 5 min. Stirring is continued for 2 d. The solution is washed successively with $10 \, \%$ aq. sodium metabisulphite ($75 \, \mathrm{mL}$), sat. aq. NaHCO₃ ($3 \times 50 \, \mathrm{mL}$) and sat. brine ($2 \times 50 \, \mathrm{mL}$), then dried (MgSO₄) and evaporated under reduced pressure. The residue is flash chromatographed on silica gel. Elution with ether/pentane (4:1) gives 2i as a colourless oil; yield: $0.74 \, \mathrm{g}$ ($65 \, \%$).

 $C_7H_{12}O_3$ calc. C 58.30 H 8.39 (144.2) found 58.33 8.56 IR (film): v = 1350, 1250, 980, 920 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 2.2–2.8 (m, 4H); 3.1–3.4 (m, 1H); 3.5–3.9 (m, 4H); 4.3–4.7 (m, 2H); 4.8–5.1 (m, 1 H).

2-(3,4-Epoxybutyl)oxetane (2i):

Using a similar procedure, 2-(3-butenyl)oxetane (**2e**; 0.60 g, 5.36 mmol) gives **2j** as a colourless oil; yield: 0.43 g (63%); b.p. 70 - 72°C/1 Torr, after elution with ether/pentane (9:1) from a silica gel column.

C₇H₁₂O₂ calc. C 65.0 H 9.44 (128.2) found 65.28 9.57

IR (film): v = 1350, 1250, 980, 920 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 1.4–2.0 (m, 4H); 2.2–2.4 (m, 2H); 2.5–3.1 (m, 3H); 4.3–5.0 (m, 3H).

2-(2,3-Dibromopropoxymethyl)oxetane (2k):

To a magnetically stirred mixture of 2-(allyloxymethyl)oxetane (2c; $1.00\,\mathrm{g}$, $7.8\,\mathrm{mmol}$) and anhydrous $\mathrm{K}_2\mathrm{CO}_3$ (0.60 g) in dry $\mathrm{CH}_2\mathrm{Cl}_2$ (50 mL) is added dropwise a solution of bromine (1.27 g, $7.95\,\mathrm{mmol}$) in dry $\mathrm{CH}_2\mathrm{Cl}_2$ (41 mL) keeping the temperature at $-78\,^\circ\mathrm{C}$. Stirring is continued at $-78\,^\circ\mathrm{C}$ for 15 min, then 20 % aq. sodium metabisulphite (25 mL) is added. The organic layer is separated, and the aqueous phase is extracted with $\mathrm{CH}_2\mathrm{Cl}_2$ (2 × 50 mL). The combined organic solution is washed successively with sat. aq. NaHCO₃ (3 × 50 mL) and sat. brine (3 × 50 mL), then dried (MgSO₄) and evaporated under reduced pressure. The residue is flash chromatographed. Elution with ether/pentane (1:3) gives 2k as a colourless oil; yield: 0.74 g (34 %); b.p. $118-120\,^\circ\mathrm{C}/0.8\,\mathrm{Torr}$.

 $C_7H_{12}Br_2O_2$ calc. C 29.20 H 4.20 (288.0) found 29.70 4.55 IR (film): v = 980, 920 cm⁻¹. 1H -NMR (CDCl₃/TMS): $\delta = 2.2 - 2.7$ (m, 2H); 3.4 - 3.9 (m, 6H); 3.9 - 4.1 (m, 1 H); 4.2 - 4.5 (m, 2 H); 4.6 - 4.9 (m, 1 H).

2-(3,4-Dibromobutyl)oxetane (21):

Using a similar procedure, 2-(3-butenyl)oxetane (2e; 0.60 g, 5.36 mmol) gives 2l as a colourless oil; yield: 0.76 g (52%); b.p. 98-100°C/0.9 Torr. after elution with ether/pentane (1:8) from a silica gel column.

 $C_7H_{12}Br_2O$ calc. C 30.91 H 4.45 (272.0) found 31.13 4.55 IR (film): $v = 980, 920 \text{ cm}^{-1}$.

¹H-NMR (CDCl₃/TMS): δ = 1.9–2.6 (m, 6 H); 3.3 (m, 2 H); 3.5–4.2 (m, 3 H); 4.5 (m, 1 H).

Oxetane-2-carbaldehyde 2,4-dinitrophenylhydrazone (2m):

To a solution of 2,4-dinitrophenylhydrazine (0.25 g, 1.26 mmol) in MeOH (5 mL) and conc. H_2SO_4 (0.5 mL) is added oxetane-2-carbaldehyde diethyl acetal (2f; 0.10 g, 0.63 mmol) in MeOH (1 mL), and the mixture is stirred overnight. The mixture is filtered and crystallisation of the residue from acetone gives 2m as orange needles; yield: 0.12 g (72 %); m.p. 234-235 °C.

C₁₀H₁₀N₄O₅ calc. C 45.12 H 3.79 N 21.05 (266.2) found 44.82 3.98 21.29

IR (Nujol): $v = 3250, 1600, 1580, 980, 920 \text{ cm}^{-1}$.

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- Lucas, K., Weyerstahl, P., Marschall, H., Nerdel, F. Chem. Ber. 1971, 104, 3607.
- (2) Daremon, C., Rambaud, R. Bull. Soc. Chim. Fr. 1971, 1, 294.
- (3) Pri-bar, I., Pearlman, P.S., Stille, J.K. J. Org. Chem. 1983, 48, 4629.
- (4) Welch, S.C., Rao, A.S.C.P., Lyon, J.T., Assercq, J.M. J. Am. Chem. Soc. 1983, 105, 252.
- (5) Ongoka, P., Mauze, B., Miginiac, L. J. Organomet. Chem. 1985, 284, 139.
- (6) Delmond, B., Pommier, J.C., Valade, J. J. Organomet. Chem. 1973, 47, 337.
- (7) Portnyagin, Y.M., Pak, N.E. Zh. Org. Khim. 1971, 7, 1629.
- (8) Kuznetsov, N.V., Krasavtzev, I.I. Ukr. Khim. Zh. 1978, 44, 744.
- (9) Shimizu, M., Kuwajima, I. J. Org. Chem. 1980, 45, 4063.
- (10) Bats, J.P., Moulines, J., Picard, P., Leclerq, D. Tetrahedron 1982, 38, 2139.
- (11) Masamune, T., Sato, S., Abiko, A., Ono, M., Murai, A. Bull. Chem. Soc. Jpn. 1980, 53, 2895.
- (12) Okuma, K., Tanaka, Y., Kaji, S., Ohta, H. J. Org. Chem. 1983, 48, 5133.
- (13) Witzemann, E.J., Evans, W.L., Hass, H., Schroeder, E.F. Org. Synth. Coll. Vol. 2 1943, 307.
- (14) Weisblat, D.I., Magerlein, B.J., Myers, D.R., Hanze, A.R., Fairburn, E.I., Rolfson, S.T. J. Am. Chem. Soc. 1953, 75, 5893.
- (15) Williams, P.H., Payne, G.B., Sullivan, W.J., Van Ess, P.R. J. Am. Chem. Soc. 1960, 82, 4883.