1061

Synthesis of 2,3,6,7-Tetrahydro- and 2,3,4,5,6,7-Hexahydro-1*H*-1,4-diazepines *via* a Tandem Michael-type Addition–Intramolecular Aza-Wittig Sequence

Aziza Benalil,^a Andrea Guerin,^b Bertrand Carboni^{a,*} and Michel Vaultier^a

^a Groupe de Physicochimie Structurale, URA 704, Université de Rennes I, Campus de Beaulieu, Avenue du Général Leclerc, 35042 Rennes Cédex, France ^b Department of Chemistry, University of Sheffield, Sheffield S3 7HF, UK

Intramolecular aza-Wittig reaction of azides derived from 1,2-amino azides and α , β -unsaturated ketones leads to 2,3,6,7-tetrahydro-1*H*-1,4-diazepines. Reduction of these compounds with lithium aluminium hydride affords the corresponding saturated heterocycles.

2,3,6,7-Tetrahydro-1*H*-1,4-diazepines have most commonly been prepared by the addition of 1,2-diamines to conjugated enones.¹ As has been shown by monitoring the reactions by ¹³C NMR spectroscopy, complex mixtures are frequently formed (Scheme 1) and the isolated product often depends on the reaction conditions, the identity of the enone and the workup techniques.²

Another limitation of this approach is that a mixture of regioisomers results in the case of an unsymmetrical 1,2-diamine 1c.1f.t (Scheme 2).

We have recently demonstrated the utility of 1,*n*-amino azides as synthetic equivalents of 1,*n*-diamines.⁴ The simultaneous presence of a free and a potential amino group prevents the formation of the polyalkylation products usually observed, for example, in the reaction of amines with alkyl halides. Furthermore, it enables complete control of the regioselectivity^{4a} (Scheme 3).

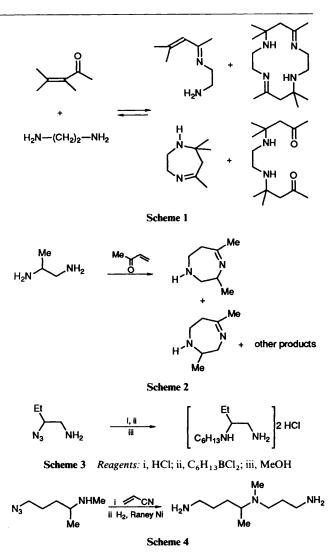
In addition to this, it is also possible to perform reactions chemioselectively on the free amino moiety. For example, Michael-type addition of 4-azido-1,N-dimethylbutylamine to acrylonitrile gave a single product which was converted into the corresponding spermidine analogue^{4b} (Scheme 4).

In continuation of this work, we now report a new route to 2,3,6,7-tetrahydro- and 2,3,4,5,6,7-hexahydro-1*H*-1,4-diazepines.

Results and Discussion

This approach is based on the following tandem sequence: Michael-type addition of the 1,2-amino azide 1 to the α,β unsaturated ketone 2, then intramolecular aza-Wittig cyclisation of the adduct 3. Reduction of 5 with lithium aluminium hydride affords the corresponding saturated heterocycle 6 (Scheme 5).

Synthesis of 2,3,6,7-Tetrahydro-1H-1,4-diazepines.—The starting 1,2-amino azides were prepared according to reported procedures.^{4a,4b} They can be obtained bearing a wide variety of substituents \mathbb{R}^1 , \mathbb{R}^2 and \mathbb{R} . For this study, azides $1\mathbf{a}$ -d and ketones $2\mathbf{a}$ -c were chosen as model substrates. Michael-type additions⁵ were carried out in diethyl ether at room temperature, except for $2\mathbf{c}$, where no solvent was used. The adducts 3 were isolated in good yields by distillation for $3\mathbf{a}$ and $3\mathbf{c}$ -f, and by recrystallization of the hydrochloride for $3\mathbf{b}$ (Table 1).



Treatment of azides 3 with an equimolar amount of triphenylphosphine in anhydrous diethyl ether led to the corresponding 2,3,6,7-tetrahydro-1*H*-1,4-diazepines 5. The formation of these cyclization products can be assumed to proceed *via* the expected iminophosphorane intermediates 4 which were not isolated. This is an additional example of the use of the intramolecular aza-Wittig reaction for the synthesis of nitrogen heterocycles.⁶ Compounds 5 were isolated by bulb-to-bulb distillation in good to moderate yields (Table 2).

Although the presence of 5a and 5b (R = H) was detected in

[†] The reaction of propane-1,2-diamine and methyl vinyl ketone gave a complex mixture of products. Distillation afforded the two regioisomeric 2,3,6,7-tetrahydro-1*H*-1,4-diazepines in a 20% yield.³

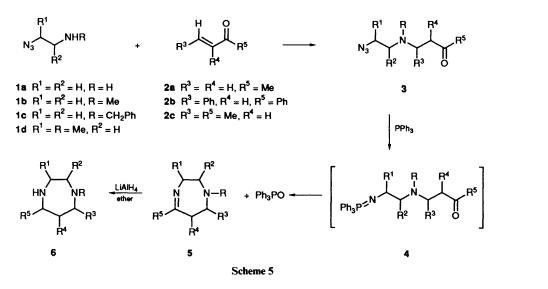


 Table 1
 Synthesis of the azides 3

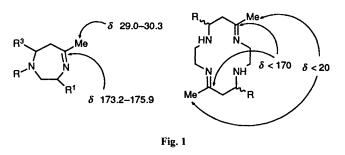
Compound	R ¹	R ²	R	R ³	R⁴	R ⁵	Yield (%)"
3a	н	Н	н	н	н	Me	70
3b	н	н	н	Ph	Н	Ph	68 <i>^b</i>
3c	н	н	Me	Н	н	Me	83
3b 3c 3d 3e 3f	Н	Н	Me	Me	н	Me	79
3e	н	н	CH ₂ Ph	н	н	Me	70
3f	Me	Н	Me	Н	Н	Me	80

" Isolated yields. b Isolated as its hydrochloride.

Table 2 Synthesis of the 2,3,6,7-tetrahydro-1H-1,4-diazepines 5

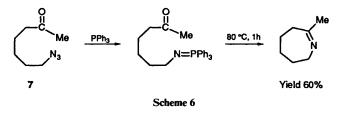
Compound	R ¹	R ²	R	R ³	R⁴	R ⁵	Yield (%)"
5c	Н	н	Me	н	Н	Me	84
5d	н	н	Me	Me	н	Me	62
5d 5e	Н	н	PhCH ₂	Н	н	Me	72
5f	Me	Н	Me	Н	Н	Me	70

" Isolated yields



the ¹H and ¹³C NMR spectra of the crude mixtures, we were unable to isolate them in reasonable yields. The 2,3,6,7tetrahydro-1*H*-1,4-diazepine structure of **5** was supported by spectroscopic data. The observation of the molecular ion in the mass spectra and the presence of signals at δ 29.0–30.3 (C–CH₃) and 173.2–175.9 (C=N–) in the ¹³C NMR spectra, which correspond closely to those given in the literature for related compounds,^{2,7} allows the exclusion of a dimeric structure (Fig. 1).

It is also worth noting that the reaction of ω -keto azide 7 with triphenylphosphine actually gives an iminophosphorane at room temperature, but that it is necessary to increase the reaction temperature to 80 °C in order to observe an intramolecular cyclisation⁸ (Scheme 6).



Synthesis of 2,3,4,5,6,7-Hexahydro-1H-1,4-diazepines 6.— Various reagents (NaBH₄, H₂ in the presence of PtO₂) have been used to reduce 2,3,6,7-tetrahydro-1H-1,4-diazepines.⁹ In our experience, low and irreproducible yields were obtained. The reduction with a solution of LiAlH₄ in diethyl ether gave the best results and the saturated heterocycles 6 were isolated by distillation (after hydrolysis) (Table 3). For 6d a mixture of two diastereoisomers (60:40) was obtained. Compounds 6 were characterized by mass spectroscopy, elemental analysis, ¹H and ¹³C NMR spectroscopy.

To conclude, we have reported an efficient new route to 2,3,6,7-tetrahydro- and 2,3,4,5,6,7-hexahydro-1*H*-1,4-diazepines **5** and **6** that complements those existing in the literature. This approach involves a Michael-type addition followed by an intramolecular aza-Wittig cyclisation. Compounds **5** and **6** are, in general, isolated in good yields and, in the case of an unsymmetrical derivative, the use of a 1,2-amino azide facilitates the production of a single regioisomer. 1,2-Amino azides are therefore valuable synthetic equivalents of the corresponding alkane-1,2-diamines and should prove to be useful building blocks in the construction of other heterocyclic structures.

Experimental

¹H and ¹³C NMR spectra were recorded on a Bruker WP 80 CW and a Bruker AM 300 spectrometer (75.5 MHz for ¹³C). Chemical shifts are reported in δ (ppm) and coupling constants (J) are given in Hz. Mass spectra were measured at 70 eV on a Varian MAT 311 Spectrometer (Centre Régional de Mesures Physiques de l'Ouest). Elemental analyses were performed by the Laboratoire Central d'Analyses du CNRS, Lyon. M.p.s were determined with a Kofler apparatus and are uncorrected. Reagent grade reagents and solvents were used as received from chemical suppliers. The starting 1,2-amino azides were prepared according to reported procedures.^{4a.4b}

CAUTION: Because of their potentially explosive properties, all experiments (especially distillations) involving azides were

Table 3 Synthesis of the 2,3,4,5,6,7-hexahydro-1H-1,4-diazepines 6

Compound	R ¹	R ²	R	R ³	R⁴	R ⁵	Yield (%)"
6c	н	н	Me	н	н	Me	63
6d	н	н	Me	Me	Н	Me	88 <i>*</i>
6e	Н	н	PhCH ₂	Н	Н	Me	91

" Isolated yield. b Mixture of diastereoisomers.

carried out with the appropriate protection under a well ventilated hood.

General Procedure for the Preparation of Azides 3.—A solution of 1,2-amino azide 1 (0.01 mol) in dry diethyl ether (10 cm³) was added dropwise to a solution of the α , β -unsaturated ketone 2 (0.01 mol) in the same solvent (10 cm³) at 0 °C. For 1a and 2a, the addition was done very slowly with a syringe pump.

The reaction mixture was stirred at room temperature for 8 h. Diethyl ether was removed under reduced pressure. The residual material was filtered on a short column of silica gel, eluting with diethyl ether-pentane (1:1) to give an oil which was purified by bulb-to-bulb distillation. For **3b**, slow addition of a 1 mol dm⁻³ HCl solution in diethyl ether (10 cm³) to the crude mixture gave the corresponding hydrochloride which was collected by filtration and recrystallised from ethanol.

4-[N-(2-*Azidoethyl*)*amino*]*butan*-2-*one* **3a**. 70%; b.p. 80–85 °C (0.01 mmHg); $\delta_{H}(300 \text{ MHz}; \text{CDCl}_{3})$ 1.79 (1 H, s), 2.17 (3 H, s), 2.65 (t, J 6.3, 2 H), 2.80 (t, J 5.7, 2 H), 2.85 (t, J 6.3, 2 H) and 3.40 (t, J 5.7, 2 H); $\delta_{C}(75 \text{ MHz}, \text{CDCl}_{3})$ 30.1, 43.6, 43.9, 48.5, 51.2 and 208.1; $m/z \text{ M}^{*+}$ was not observed (Found: [M – CH₂-N₃·]⁺, 100.0766. C₅H₁₀NO requires 100.0762).

3-[N-(2-Azidoethyl)amino]-1,3-diphenylpropan-1-one hydrochloride **3b**-HCl. 68%; m.p. 160–162 °C (ethanol); $\delta_{H}(300$ MHz; CDCl₃) 2.86–2.97 (2 H, m), 3.70–3.80 (1 H, m), 3.82–3.92 (1 H, m), 4.03 (dd, J 7.8 and 17.8, 1 H), 4.24 (dd, J 5.7 and 17.8, 1 H), 4.86 (dd, J 5.7 and 7.8, 1 H), 7.37–7.50 (8 H, m), 7.72–7.76 (1 H, m), 7.82–7.88 (1 H, m) and 9.60–10.80 (2 H, br s); $\delta_{C}(75$ MHz; CDCl₃) 42.7, 44.4, 47.0, 59.7, 128.3, 128.8, 128.9, 129.6, 129.9, 133.6, 134.0, 135.7 and 196.2 (Found: C, 61.8, H, 5.8; N, 16.4. C_{1.7}H₁₉ClN₄O requires C, 61.7; H, 5.7; N, 16.9%).

4-[N-(2-Azidoethyl)-N-methylamino]butan-2-one 3c. 83%; b.p. 44–60 °C (0.03 mmHg); $\delta_{\rm H}(300 \text{ MHz}; \text{ CDCl}_3)$ 2.18 (s, 3 H), 2.26 (s, 3 H), 2.55–2.64 (m, 4 H), 2.68–2.74 (m, 2 H) and 3.82 (t, J 6.0, 2 H); $\delta_{\rm C}(75 \text{ MHz}; \text{ CDCl}_3)$ 30.2, 41.6, 42.0, 48.7, 52.0, 56.5 and 207.8 [Found: C, 39.3; H, 4.3; N, 24.4. C₇H₁₄N₄O, C₆H₃N₃O₇ (picrate) requires C, 39.1; H, 4.3; N, 24.6%].

4-[N-(2-Azidoethyl)-N-methylamino]pentan-2-one **3d**. 79%; b.p. 60–65 °C (0.01 mmHg); $\delta_{\rm H}(300 \text{ MHz}; \text{ CDCl}_3)$ 0.99 (d, J 6.6, 3 H), 2.17 (s, 3 H), 2.28 (s, 3 H), 2.36 (dd, J 7.5 and 15.5, 1 H), 2.60 (t, J 6.3, 2 H), 2.65 (dd, J 6.5 and 15.5, 1 H), 3.23–3.53 (m, 1 H) and 3.26 (t, J 6.3, 2 H); $\delta_{\rm C}(75 \text{ MHz}; \text{ CDCl}_3)$ 14.0, 30.3, 36.6, 47.9, 49.3, 52.8, 55.2 and 207.9; $m/z \text{ M}^{++}$ was not observed (Found: M – CH₂N₃•]⁺, 128.1070. C₇H₁₄NO requires 128.1075).

4-[N-(2-Azidoethyl)-N-benzylamino]butan-2-one **3e**. 70%; b.p. 105–110 °C (0.1 mmHg); $\delta_{H}(300 \text{ MHz}; \text{CDCl}_{3})$ 1.91 (3 H, s), 2.50 (t, J 7.0, 2 H), 2.57 (t, J 6.0, 2 H), 2.73 (t, J 7.0, 2 H), 3.15 (t, J 6.0, 2 H), 3.52 (2 H, s) and 7.15–7.26 (5 H, m); $\delta_{C}(75 \text{ MHz}; \text{CDCl}_{3})$ 30.2, 41.7, 48.9, 49.1, 53.4, 59.2, 127.2, 128.4, 128.8, 138.8 and 208.0; $m/z \text{ M}^{*+}$ was not observed (Found: [M – CH₂-N₃·]⁺, 218.1414. C₁₃H₁₈N₂O requires 218.1419).

4-([N-(2-Azidopropyl)-N-methylamino]butan-2-one **3f**. 80%; b.p. 50–55 °C (0.01 mmHg); $\delta_{\rm H}$ (300 MHz; CDCl₃) 1.17 (d, J 6.6, 3 H), 2.17 (3 H, s), 2.26 (3 H, s), 2.31 (dd, J 4.7 and 13.1, 1 H), 2.46 (dd, J 8.6 and 13.1, 1 H), 2.58–2.80 (4 H, m) and 3.56–3.67 (1 H, m); [Found: C, 40.9; H, 4.7; N, 23.6. C₈H₁₆N₄O, C₆H₃N₃O₇ (picrate) requires C, 40.7; H, 4.6; N, 23.7%]. General Procedure for the Cyclisation of Azides 3.—Triphenylphosphine (1.315 g; 5 mmol) was added in small portions under nitrogen to a stirred solution of the azide 3 (5 mmol) in dry diethyl ether (10 cm³). The mixture was stirred at room temperature for 8 h. After addition of dry pentane (20 cm³), triphenylphosphine oxide was filtered off. The solvents were removed under reduced pressure and the residual material was triturated with diethyl ether-pentane (5 cm³-15 cm³). Filtration and evaporation afforded a colourless oil which was purified by bulb-to-bulb distillation.

2,3,6,7-*Tetrahydro-1,5-dimethyl*-1H-1,4-*diazepine* **5c**. B.p. 20–25 °C (0.01 mmHg); $\delta_{H}(300 \text{ MHz}; \text{CDCl}_{3})$ 1.85 (3 H, s), 2.11 (3 H, s), 2.09–2.18 (2 H, m), 2.20–2.27 (2 H, m), 2.28–2.36 (2 H, m) and 3.40–83.48 (2 H, m); $\delta_{C}(75 \text{ MHz}; \text{CDCl}_{3})$ 29.1, 35.0, 47.0, 51.7, 52.6, 56.1 and 175.4; (Found: M^{*+}, 126.1164. C₇H₁₄N₂ requires 126.1157).

2,3,6,7-*Tetrahydro*-1,5,7-*trimethyl*-1H-1,4-*diazepine* **5d**. B.p. 35–40 °C (0.01 mmHg); $\delta_{\rm H}(300$ MHz; CDCl₃) 1.04 (d, *J* 6.7, 3 H), 2.06 (3 H, s), 2.35 (3 H, s), 2.36–2.52 (2 H, m), 2.55–2.72 (2 H, m), 2.74–2.83 (1 H, m) and 3.56–3.76 (2 H, m); $\delta_{\rm C}(75$ MHz; CDCl₃) 14.8, 30.3, 41.7, 43.4, 51.4 (2 C), 54.1 and 174.5 (Found: M⁺⁺, 140.1319. C₈H₁₆N₂ requires 140.1313).

2,3,6,7-*Tetrahydro*-1,3,5-*trimethyl*-1H-1,4-*diazepine* **5f**. B.p. 40–45 °C (0.1 mmHg); $\delta_{\rm H}(80$ MHz; CDCl₃) 1.30 (d, J 6.9, 3 H), 1.75–2.95 (6 H, m), 2.08 (3 H, s), 2.30 (3 H, s) and 3.33–3.85 (1 H, m); $\delta_{\rm C}(20$ MHz; CDCl₃) 23.2, 29.3, 35.3, 47.2, 52.3, 55.9, 63.0 and 173.2.

General Procedure for the Reduction of 5.—To a cooled and stirred solution of the compound 5 (2 mmol) in dry diethyl ether (10 cm³) was added a 1 mol dm⁻³ diethyl ether solution (2 cm³) of LiAlH₄ at a rate so as to maintain the temperature at 0–5 °C. The mixture was then stirred at room temperature for 4 h. After the successive addition of water (75 mm³), 20% aq. NaOH (60 mm³) and water (190 mm³),¹⁰ the reaction mixture was filtered and the residue washed with diethyl ether (2 × 10 cm³). The filtrate was dried over K₂CO₃. Evaporation of the solvent left a colourless oil which was purified by bulb-to-bulb distillation.

2,3,4,5,6,7-*Hexahydro*-1,5-*dimethyl*-1H-1,4-*diazepine* **6c**. B.p. 20–25 °C (0.01 mmHg); $\delta_{H}(300 \text{ MHz}; \text{CDCl}_{3})$ 1.08 (d, J 6.4, 3 H), 1.44 (dddd, J 3.4, 8.4, 8.4 and 14.7, 1 H), 1.86 (dddd, J 3.4, 4.4, 7.9 and 14.8, 1 H), 1.83 (1 H, br s), 2.37 (3 H, s), 2.48–2.69 (4 H, m), 2.85 (ddd, J 4.7, 5.7 and 6.7, 1 H) and 2.95–3.10 (2 H, m), $\delta_{H}(75 \text{ MHz}; \text{CDCl}_{3})$ 23.9, 37.1, 46.7, 47.3, 52.8, 55.2 and 61.6 [Found: C, 38.8; H, 3.9; N, 19.1. C₇H₁₆N₂, C₁₂H₆N₆O₁₄ (dipicrate) requires C, 38.9; H, 3.8; N, 19.1%] (Found: M^{*+}, 128.1308. C₇H₁₆N₂ requires 128.1313).

2,3,4,5,6,7-*Hexahydro*-1,5,7-*trimethyl*-1H-1,4-*diazepine* 6d. B.p. 35–40 °C (0.01 mmHg); mixture of two diastereoisomers A and B (60:40), $\delta_{H}(300 \text{ MHz}; \text{ CDCl}_{3})$ 1.03–1.13 (6 H, m), 1.47–1.72 (2 H, m), 2.17 (1 H, br s), 2.38 (3 H, s) and 2.60–3.12 (6 H, m); $\delta_{C}(75 \text{ MHz}; \text{ CDCl}_{3})$; isomer A (major) 20.7, 24.3, 41.8, 45.0, 45.5, 52.7, 56.7 and 58.9. Isomer B (minor) 19.5, 24.3, 40.6, 41.2, 48.1, 50.4, 56.3 and 58.3 [Found: C, 40.0; H, 4.0; N, 18.3. $C_8H_{18}N_2$, $C_{12}H_6N_6O_4$ (dipicrate) requires C, 40.0; H, 4.0; N, 4.0; N, 18.6%].

1-Benzyl-2,3,4,5,6,7-hexahydro-5-methyl-1H-1,4-diazepine **6e**. B.p. 70–75 °C (0.01 mmHg); $\delta_{\rm H}$ (300 MHz; CDCl₃) 1.08 (d, *J* 6.5, 3 H), 1.50 (dddd, *J* 3.7, 8.6, 8.6 and 14.5, 1 H), 1.63 (1 H, bs), 1.80 (dddd, *J* 3.8, 3.8, 7.8 and 14.4, 1 H), 2.52–2.64 (4 H, m), 2.68– 2.84 (1 H, m), 2.92–3.07 (2 H, m), 3.62 (2 H, s) and 7.18–7.36 (5 H, m); $\delta_{\rm C}$ (75 MHz; CDCl₃) 23.9, 37.7, 47.3, 52.8, 53.1, 58.9,

Acknowledgements

The authors thank the ERASMUS programme for financial support to A. G.

References

- (a) J. Elguero, E. Gonzalez and R. Jacquier, Bull. Soc. Chim. Fr., 1969, 2054; (b) K. Hideg and D. Lloyd, J. Chem. Soc. C, 1971, 3441; (c) A. Bonvicini and G. Cantatore, Chim. Ind. (Milan), 1972, 54, 974; (d) K. Samula and E. Jurkowka-Kowalczyk, Rocz. Chem., 1974, 48, 2287; (e) S. P. Kasprzyk and S. Szymanski, Pol. J. Chem., 1979, 53, 525; (f) S. P. Kasprzyk and R. A. Kolinski, Pol. J. Chem., 1984, 58, 721.
- 2 D. Lloyd, W. Scheibelein and K. Hideg, J. Chem. Res. 1981, (S), 61; (M), 838 and references therein.
- 3 A. Benalil, unpublished results.
- 4 (a) A. Benalil, B. Carboni and M. Vaultier, *Tetrahedron*, 1991, 47, 8177; (b) B. Carboni, A. Benalil and M. Vaultier, *J. Org. Chem.*, in the press.

- 5 M. S. Gibson, in *The Chemistry of the Amino Group*, ed. S. Patai, Interscience Publishers, London, 1968, pp. 61-65.
- 6 Y. Gololobov and L. F. Kasukhin, Tetrahedron, 1992, 48, 1353; S. Eguchi, Y. Matsushita and K. Yamashita, Org. Prep. Proced. Int., 1992, 24, 209. For recent results, see also: J. Kurita, T. Iwata, S. Yasuike and T. Tsuchiya, J. Chem. Soc., Chem. Commun., 1992, 81; G. A. McCort and J. C. Pascal, Tetrahedron Lett., 1992, 33, 4443; P. Molina, A. Arques, A. Alias, M. V. Vinader, M. C. Fores-Fores and F. H. Cano, Tetrahedron, 1992, 48, 3091; S. Eguchi, K. Yamashita and Y. Matsushita, Synlett., 1992, 295.
- 7 O. H. Hankovsky, K. Hideg, D. Lloyd and H. McNab, J. Chem. Soc., Perkin Trans. 1, 1979, 1345.
- 8 M. Vaultier, P. H. Lambert and R. Carrié, Bull. Soc. Chim. Fr., 1986, 83.
- 9 G. Cantatore and A. Bonvinici, *Chim. Ind.* (*Milan*), 1972, 54, 980; S.
 P. Kasprzyk and R. A. Kolinski, *Bull. Pol. Acad. Sci. Chem.*, 1984, 32, 151; W. Heinrich and W. Heigel, *Ger. Pat.*, 1,047,785/1958 (*Chem. Abstr.*, 1961, 55, 4552e).
- 10 N. G. Gaylord, *Reduction with complex metal hydrides*, Interscience Publishers, New York, 1956, p. 1011.

Paper 2/06620C Received 14th December 1992 Accepted 28th January 1993