Rhodium(I)-Catalyzed Intramolecular Carbocyclic Ring Fusion: A New Approach to Medium-Sized-Ring Ketones

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Intramolecular carbocyclic ring enlargement reactions in organic synthesis consist of three primary types: (1) incorporation of a side chain into the ring, (2) bridge cleavage in bicyclic compounds, and (3) ring cleavage with simultaneous bond formation between two side chains (pericyclic reactions).¹ A novel synthetic method would be the fusion of two small rings into a single larger one. This transformation might be accomplished in either an intermolecular or an intramolecular manner through the ability of transition metal complexes to insert into strained rings (eq 1).²⁻⁴ Described here is the successful development of the intramolecular variant of the strategy for the synthesis of seven- and eight-membered-ring dienones.^{5,6}

Previous work has shown that cyclobutenones react with certain transition metal reagents and catalysts to give reactive η^2 - and η^4 -vinylketene complexes.^{7,8} Building on this knowledge, the search for a ring fusion catalyst was begun with the 4-cyclopropyl-2-cyclobutenone 1a. Thermolysis of 1a in C_6D_6 led to slow decomposition without formation of any detectable cycloheptadienone, a result consistent with the findings of Herndon using a related system.⁹ However, in the presence of catalytic quantities of rhodium(I) complexes, cycloheptadienone 2a was formed in a few hours at 60 °C. Of the catalysts investigated, $RhCl(PPh_3)_3$ gave the highest yield (Table I, entry 1). The other 4-cyclopropyl-2-cyclobutenones (1b-e) showed better reactivity with $[Rh(\mu-$ Cl)(1,5-cyclooctadiene)]₂ (entries 2-5), and most required reaction temperatures of 90-120 °C.10 At these elevated temperatures, partial isomerization of the initially formed 2,4dienones 2 to the 3,5-dienones 3 occurred. This 1,5-hydrogen shift is a known thermal reaction,¹¹ and heating 2a to 90 °C for 16 h in the absence of catalyst gave a greater than 3:1 ratio of **3a** to **2a** (93% mass recovery). No catalysis of the 1,5-hydrogen shift by rhodium was observed at lower temperatures.

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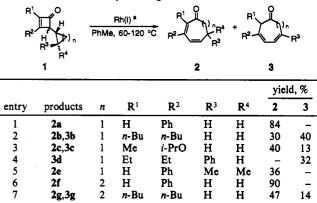
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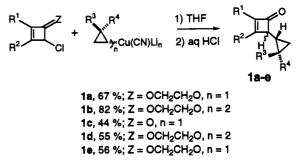
(8) Huffman, M. A.; Liebeskind, L. S. J. Am. Chem. Soc. 1991, 113, 2771.
(9) Tumer, S. U.; Herndon, J. W.; McMullen, L. A. J. Am. Chem. Soc. 1992, 114, 8394.

Table I. Rhodium-Catalyzed Ring Fusion



^a Rh(I) catalyst = 5 mol % RhCl(PPh₃)₃, entries, 1, 6, 7; 2.5 mol % [Rh(µ-Cl)(cod)]₂, entries 2-5.

Scheme I^a



^a See Table I for substituents.

Substitution on the cyclopropyl group led only to products from cleavage of the less substituted cyclopropane ring bond (entries 4, 5), although yields were lower in these cases. Similar results were obtained starting from pure diastereomers or from mixtures. Furthermore, in reactions monitored by ¹H NMR, each isolated diastereomer of 1d equilibrated to a 1.4:1 mixture of diastereomers by the time 50% reaction had occurred.

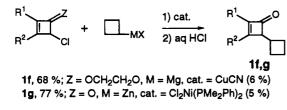
4-Cyclobutyl-2-cyclobutenones also undergo ring fusion to give cyclooctadienones (entries 6, 7). $[Rh(\mu-Cl)(cod)]_2$ is a poor catalyst for these reactions, indicating that donor ligands such as PPh₃ may be required in order to open the cyclobutane ring. The cyclooctadienone products also undergo double-bond migration at elevated temperatures (entry 7).

4-Cyclopropyl-2-cyclobutenones **1a**—e were made by the coupling of cyclopropyl copper reagents with 4-chlorocyclobutenones, protected in most cases as the ethylene acetal (Scheme I).

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⁽¹⁰⁾ All new compounds were spectroscopically characterized and furnished adequate analytical or high-resolution mass-spectral data. In a representative experiment, 1b (299 mg, 1.36 mmol) was dissolved in dry toluene (14 mL) under argon. [Rh(μ -Cl)(cod)]₂ (17 mg, 0.034 mmol) was added against an argon flow, and the mixture was heated to reflux for 24 h. Radial chromatography onsilica with hexanes/ethyl acetate (40:1) eluted two product bands. The first gave 3b as a colorless oil (119 mg, 0.54 mmol, 40%): 'H NMR (360 MHz, CDCl₃) δ 6.16 (dd, J = 10.6, 5.0 Hz, 1 H), 5.90 (d, J = 5.0 Hz, 1 H), 5.61 (dt, J = 10.6, 5.3 Hz, 1 H), 3.10 (m, 3 H), 2.10 (q, J = 6.7 Hz, 2 H), 1.85 (m, 1 H), 1.64 (m, 1 H), 1.41–1.15 (m, 8 H), 0.87 (t, J = 7.0 Hz, 3 H), 0.86 (t, J = 7.0 Hz, 3 H); 13 C NMR (75 MHz, CDCl₃) δ 207.7, 141.3, 128.6, 123.0, 122.9, 58.0, 44.9, 36.6, 30.4, 30.1, 27.3, 22.7, 22.4, 13.9, 13.8; IR (CH₂Cl₂) 2961 (s), 2933 (s), 2875, 2863, 1703 (s), 1607, 1468, 1380 cm⁻¹. Anal. Calcd for C1₃H₂₄O: C, 81.76; H, 10.98. Found: C, 81.52; H, 10.93. The second band gave 2b as a colorless oil (90 mg, 0.41 mmol, 30%): 'H NMR (360 MHz, CDCl₃) δ 6.21 (dt, J = 11.2, 6.1 Hz, 1 H), 3.66 (m, 8 H), 0.20 (m, 6 H); ¹³C NMR (75 MHz, CDCl₃) δ 200.4.1 (m3.4, 13.4, 13.4, 13.4, 13.2, 13.4, 13.2, 13.4, 13.2, 13.4, 13.4, 13.4, 13.4, 13.4, 13.4, 13.4, 13.4, 13.4, 14.4,



^a See Table I for substituents.

Cyclobutylcyclobutenones were made by copper-catalyzed coupling of cyclobutylmagnesium chloride with an acetal-protected 4-chlorocyclobutenone, or by nickel-catalyzed coupling of the unprotected chlorocyclobutenone with a cyclobutylzinc reagent (Scheme II). These synthetically interesting secondary aliphaticsecondary allylic cross couplings will be dealt with in more detail in the future.

Together the coupling and fusion reactions represent a fundamentally new strategy for medium-sized-ring synthesis. The

full scope of catalyzed ring fusion and the mechanistic questions that it presents remain to be explored.

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Supplementary Material Available: Experimental details for the syntheses of 1a-g, 2a-c, e-g, and 3a-d,g (13 pages). Ordering information is given on any current masthead page.