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PALLADIUM CATALYZED REDUCTIVE CYCLIZATION REACTION IN ALKALOID SYNTHESIS - AN ENANTIOSELECTIVE TOTAL SYNTHETIC ROUTE TO (+)-PUMILIOTOXIN C

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Abstract: Beginning with the chirally homogeneous cyclohexenol 3, an enantioselective total synthetic route to (+)-pumiliotoxin C (1) has been described. Palladium catalyzed reductive cyclization reaction was employed to prepare a key component in the synthesis. Copyright © 1996 Elsevier Science Ltd

Owing to their intriguing pharmacological properties and unique structures, pumiliotoxin C (1) and gephyrotoxin (2) have proved to be challenges in organic synthesis.¹ Both arrow poison frog toxins (1) and (2) possess a *cis*-decahydroquinoline ring system with side chain substituents at the C-2 and C-5 positions (the C-2 side chain of 2 being attached at nitrogen). Pumiliotoxin C (1) has served as prototype for the synthesis of *cis*-decahydroquinoline alkaloids and has frequently been used to illustrate new methods for construction of decahydroquinoline ring system.



Pumiliotoxin C (1)



Gephyrotoxin (2)

Much effort has been devoted to the synthesis of the *cis*-decahydroquinoline alkaloids,² however, there remains conspicuous needs of general procedures to prepare both simple and complex congeners and flexible ones to synthesize topographical relatives.³ We herein embark upon a program to develop a unified strategy for the synthesis of the *cis*-decahydroquinoline alkaloids employing palladium catalyzed reductive cyclization reaction as the key step. The target chosen to explore this approach was (+)-pumiliotoxin C (1).

The chirally homogeneous cyclohexenol 3,⁴ $[\alpha]_D^{27}$ -5.0 (c 0.45, CHCl₃), prepared from (*R*)-(-)pantolactone, was used as the starting material. The inversion of hydroxyl group in 3 under carefully controlled Mitsunobu reaction conditions⁵ (*p*-nitrobenzoic acid, Ph₃P, DEAD, THF, -30 °C \rightarrow rt; LAH, THF, 80% overall) resulted in the formation of the alcohol 4, $[\alpha]_D^{27}$ -82.8 (c 0.15, CHCl₃).⁶ Thermal Claisen rearrangement of 4 with triethyl orthoacetate was next conducted in the presence of *o*-nitrophenol at 160 °C for 2 h to furnish the corresponding ester (78%), which was reduced with LAH to afford the alcohol 5⁶ in 91% yield. The compound 5 was converted efficiently into the bromide (CBr₄, Ph₃P, 86%), which was then replaced with lithium trimethylsilylacetylide in THF-HMPA (-78 °C \rightarrow rt, 87%). The use of low temperature and of HMPA as cosolvent is critical to the success of this coupling process. Exposure of the resulting TMSacetylene to 1N methanolic sodium hydroxide gave rise to the key enyne 6, $[\alpha]_D^{22}$ +108.3 (c 0.31, CHCl₃), (92%).⁶

With 6 in hand, the pivotal palladium catalyzed reductive cyclization reaction⁷ for the construction of the *cis*-decahydroquinoline ring system of (+)-pumiliotoxin C (1) was examined. As a result of testing, the cyclization of 6 in 1,2-dichloroethane (DCE) in the presence of (dba)₃Pd₂•CHCl₃ (2.5 mol %), *N*, *N'*-bis(benzylidene)ethylenediamine (BBEDA) (5.0 mol %), polymethylhydrosiloxane (PMHS) (10 eq.), and acetic acid (1 eq.) proceeded quite nicely to provide the *exo*-olefin 7⁶, $[\alpha]_D^{24}$ +101.7 (c 0.43, CHCl₃), in 61% yield.

With the efficient synthesis of the *cis*-hydrindane derivative 7, the stage was now set for the completion of the synthesis. The benzyl ether 7 was transformed to the keto alcohol 8⁶ by the standard procedure using sodium in liquid ammonia (88%), follwed by ozonolysis (O₃, MeOH, -78 °C; Me₂S, 65%). After treatment of 8 with 1,1'-thiocarbonyldiimidazole and 4-dimethylaminopyridine (DMAP), the corresponding thioimidazolide, obtained in 95% yield, was allowed to react with tributyltin hydride in the presence of 2,2'-azobisisobutyronitrile (AIBN) in benzene under reflux, giving the ketone (+)-9 (79%),⁶ $[\alpha]_D^{23}$ +80.7 (c 0.64, CHCl₃), which displays the same spectra with those provided by Mehta in a total synthesis of (±)-pumiliotoxin C (1).^{2e, 8} Finally, Beckmann rearrangement of 9 to 10 was conducted, in 61% yield, under standard conditions.^{2e} The analytical properties (¹H NMR, IR, MS) of (+)-10⁶ were identical in all respects to those reported^{2c} with the exception of the optical rotation, which was opposite in sign. Since the amide 10 has been transformed previously into pumiliotoxin C (1), ^{2c, 2e} the present enantioselective synthesis of 10 gives rise to the formal total synthetic route to (+)-pumiliotoxin C (1). Our methodology based upon

palladium catalyzed reductive cyclization reaction is believed to be an efficient tool in the synthesis of other complex *cis*-decahydroquinoline alkaloids, such as gephyrotoxin (2) and lepadin A.⁹



Reagents and conditions: (a) *p*-nitrobenzoic acid, Ph₃P, DEAD, THF, -30 °C \rightarrow rt. (b) LAH, THF. (c) MeC(OEt)₃, *p*-nitrophenol (cat.), 160 °C. (d) LAH, THF. (e) CBr₄, Ph₃P, CH₂Cl₂. (f) LiC=CTMS, HMPA, THF, -78 °C \rightarrow rt. (g) 1N NaOH, MeOH. (h) (dba)₃Pd₂•CHCl₃ (2.5 mol %), BBEDA (5.0 mol %), PMHS (10 eq.), AcOH (1 eq.), DCE. (i) Na, Liq. NH₃, THF, -78 °C; NH₄Cl. (j) O₃, MeOH, -78 °C; Me₂S. (k) 1,1'-thiocarbonyldiimidazole, DMAP, CH₂Cl₂, reflux. (l) ⁿBu₃SnH, AIBN, C₆H₆, reflux. (m) NH₂OH•HCl, NaOAc, MeOH; TSCI, NaOH, aq. THF.

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References and Notes

- Pumiliotoxin C: (Isolation) Daly, J. W.; Tokuyama, T.; Habermehl, G.; Karle, I. L.; Witkop, B. Liebigs Ann. Chem. 1969, 729, 198-204. (Physiological property) (a) In Alkaloids: Chemical and Biological Perspectives; Pelletier, S. W., Ed.; Wiley: New York, 1986; Vol. 4, p 162. (b) In The Alkaloids; Cordell, G. A., Ed.; Academic Press: San Diego, 1993; Vol. 43, p 185. Gephyrotoxin: (Isolation) Daly, J. W.; Witkop, B; Tokuyama, T.; Nishikawa, T.; Karle, I. L.; Helv. Chim. Acta 1977, 60, 1128-1140. (Physiological property) Mensah-Dwumah, M.; Daly, J. W. Toxicon. 1968, 16, 189-194.
- For chiral syntheses of natural (-)-pumiliotoxin C: (a) Oppolzer, W.; Flaskamp, E. Helv. Chim. Acta 1977, 60, 204-207. (b) Bonin, M.; Royer, J.; Grielson, D. S.; Husson, H.-P. Tetrahedron Lett. 1986, 27, 1569-1572. (c) Murahashi, S.; Sasao, S.; Saito, E.; Naota, T. J. Org. Chem. 1992, 57, 2521-2523. Idem Tetrahedron 1993, 49, 8805-8826. (d) Comins, D. L.; Dehghani, A. J. Chem. Soc. Chem. Comm. 1993, 1838-1839. (e) Naruse, M.; Aoyagi, S.; Kibayashi, C. Tetrahedron Lett. 1994, 35, 9213-9216. The unnatural (+)-enantiomer: (f) Oppolzer, W.; Flaskamp, E. Helv. Chim. Acta 1977, 60, 204-207. (g) Schultz, A. G.; McCloskey, P. J.; Court, J. J. Am. Chem. Soc. 1987, 109, 6493-6502. For recent syntheses of (±)-pumiliotoxin C: (h) Comins, D. L.; Dehghani, A. Tetrahedron Lett. 1991, 32, 5697-5700. (i) Braudi, A.; Cordero, F. M.; Goti, A; Guarua, A. Tetrahedron Lett. 1992, 33, 6697-6700. (j) Polniaszek, R. P.; Dillard, L. W. J. Org. Chem. 1992, 57, 4103-4110. (k) Meyers, A. I.; Milot, G. J. Am. Chem. Soc. 1993, 115, 6652-6660. (l) Mehta, G.; Praveen, M. J. Org. Chem. 1995, 60, 279-280.
- 3. For a review: Inubushi, Y.; Ibuka, T. Heterocycles 1977, 8, 633-660.
- 4. Toyota, M.; Matsuura, M.; Fukumoto, K. Heterocycles 1995, 41, 661-664.
- 5. Martin, S. F.; Dodge, J. A. Tetrahedron Lett. 1991, 32, 3017-3020.
- 6. New compounds were characterized by combustion or high resolution mass spectroscopy, infrared spectroscopy, and 300 MHz NMR spectroscopy.
- 7. Trost, B. M.; Rise, F. J. Am. Chem. Soc. 1987, 109, 3161-3163. For a recent review: Trost, B. M. Angew. Chem. Int. Ed. Engl. 1995, 34, 259-281.
- 8. Direct comparison of the spectral data of synthetic (+)-9 and (+)-10 with those synthesized using the established procedure¹⁰ showed them to be completely superimposable. A more detailed discussion of this point will appear in our full account of this work.
- 9. Stefan, B. Tetrahedron 1991, 47, 8729-8732.
- (a) El-Abbady, A. M.; El-Ashry, M.; Doss, S. H. Can. J. Chem. 1969, 47, 1483-1486. (b) Maruoka, K.; Miyazaki, T.; Ando, M.; Matsumura, Y.; Sakane, S.; Hattori, K.; Yamamoto, H. J. Am. Chem. Soc. 1983, 105, 2831-2843.

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