

Hydrazines in the Synthesis of *N*-Substituted 1,5,3-Dithiazocan-3-amines Catalyzed by Ti and Cu Compounds

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Abstract—Efficient preparation method was developed for *N*-aryl(benzyl, alkyl)-1,5,3-dithiazocan-3-amines consisting in the transamination of 3-*tert*-butyl-1,5,3-dithiazocane with aryl(benzyl)hydrazines, and also in the reaction of *N*¹,*N*¹,*N*⁷,*N*⁷-tetramethyl-2,6-dithiaheptane-1,7-diamine with aryl(benzyl, alkyl)hydrazines in the presence of catalytic amounts of Ti and Cu compounds.

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Recently [1] we developed a selective method of the synthesis of *N*-aryl-1,5,3-dithiazocanes capable of exhibiting fungicidal activity [2] consisting in the transamination of 3-*tert*-butyl-1,5,3-dithiazocane with primary arylamines in the presence of Sm and Co complexes.

In extension of the previous research and also aiming at the efficient procedure of the synthesis of *N*-substituted 1,5,3-dithiazocan-3-amines we studied the transamination of 3-*tert*-butyl-1,5,3-dithiazocane by substituted hydrazines in the presence of catalysts based on transition metals [1]. Among the tested in this reaction catalysts based on salts and complexes of Fe, Cu, V, Pd, Co, Zr, Hf, Ti Cp_2TiCl_2 exhibited the highest activity. The initial hydrazines involved into the reaction were phenylhydrazine, 4-nitrophenylhydrazine, 2,4-dinitrophenylhydrazine, 4-methylphenylhydrazine, benzylhydrazine, methylhy-

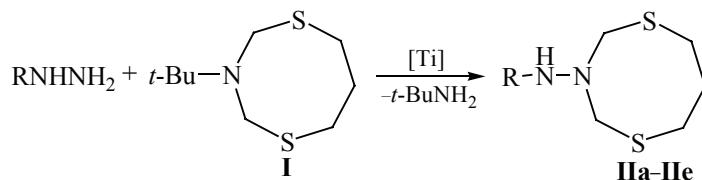
drazine, and *tert*-butylhydrazine.

In the presence of catalytic quantities of Cp_2TiCl_2 the reaction of aryl(benzyl)hydrazines with 3-*tert*-butyl-1,5,3-dithiazocane (**I**) [molar ratio aryl(benzyl)-hydrazine-(**I**)- Cp_2TiCl_2 1 : 1 : 0.05, solvent acetonitrile, 20°C, 0.5 h] proceeds selectively to afford *N*-aryl(benzyl)-1,5,3-dithiazocan-3-amines **IIa–IIe** in 68–80% yields (Scheme 1).

Without catalyst the yield of compounds **IIa–IIe** was less than 40% after 24–48 h; without catalyst prevailing formation of oligomeric compounds was observed, and the yield of amines **II** did not exceed 30%.

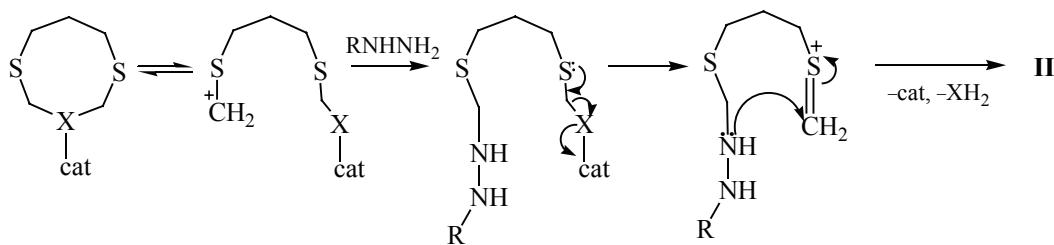
It is presumable that the catalytic transamination of 3-*tert*-butyl-1,5,3-dithiazocane (**I**) by the substituted hydrazines includes a stage of the heterocycle opening under the action of the catalyst, the nucleophilic addition

Scheme 1.

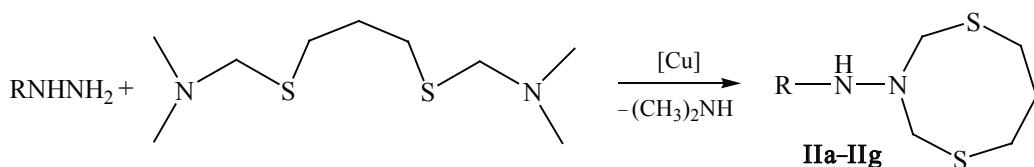


R = Ph (**a**), 4-NO₂C₆H₄ (**b**), 2,4-NO₂C₆H₃ (**c**), 4-CH₃C₆H₄ (**d**), C₆H₅CH₂ (**e**).

Scheme 2.



Scheme 3.



R = Ph (**a**), 4-NO₂C₆H₄ (**b**), 2,4-NO₂C₆H₃ (**c**), 4-CH₃C₆H₄ (**d**), C₆H₅CH₂ (**e**), CH₃ (**f**), (CH₃)₃C (**g**).

of the hydrazine to the carbocation followed by intramolecular cyclization with the formation of *N*-substituted 1,5,3-dithiazocan-3-amine **II** (Scheme 2).

We failed to bring into the reaction under the chosen conditions the *N*-alkylhydrazines (methyl-, *tert*-butylhydrazine) for the preparation of the corresponding *N*-alkyl-substituted 1,5,3-dithiazocan-3-amines.

In order to develop the preparation procedure for *N*-alkyl-substituted 1,5,3-dithiazocan-3-amines we carried out under the catalysis with the complexes of transition metals the reaction of cyclothiomethylation of alkylhydrazines with *N¹,N¹,N⁷,N⁷*-tetramethyl-2,6-dithiaheptane-1,7-diamine [3] similarly to the reaction of the cyclothiomethylation of carboxylic acids hydrazides [4]. In the reaction of *N¹,N¹,N⁷,N⁷*-tetramethyl-2,6-dithiaheptane-1,7-diamine with methylhydrazine without catalyst (60°C, CHCl₃, 1 h) the yield of *N*-methyl-1,5,3-dithiazocan-3-amine (**IIf**) did not exceed 15%, and in the presence of 5 mol% of CuCl₂ the yield of amine **IIf** attained 74%. Under the developed conditions (60°C, CHCl₃, 5 mol% CuCl₂, 1 h) from *N¹,N¹,N⁷,N⁷*-tetramethyl-2,6-dithiaheptane-1,7-diamine and *tert*-butylhydrazine we obtained *N*-*tert*-butyl-1,5,3-dithiazocan-3-amine (**IIg**) in ~70% yield.

In the subsequent experiments we applied the developed method of alkylhydrazines cyclothiomethylation with the use of *N¹,N¹,N⁷,N⁷*-tetramethyl-2,6-dithiaheptane-1,7-diamine to the synthesis of *N*-aryl(benzyl)-1,5,3-dithiazocanes **II** and we obtained them in the yields of

79–85% (Scheme 3).

The structure of *N*-aryl(benzyl, alkyl)-1,5,3-dithiazocan-3-amines was confirmed by ¹³C and ¹H NMR spectra and the other physicochemical methods.

EXPERIMENTAL

The reaction progress was monitored by TLC on Silufol W-254 plates, eluent petroleum ether–EtOAc–CHCl₃, 5 : 1 : 1. The reaction products were analyzed by HPLC on a chromatograph Altex-330 (USA) equipped with a UV detector at the wavelength 340 nm. ¹³C and ¹H NMR spectra were registered on a spectrometer Bruker Avance 400 (100.62 and 400.13 MHz respectively), solvent CDCl₃. GC-MS measurements were carried out on instruments Finnigan-4021 (glass capillary column 5000 × 0.25 mm, stationary phase HP-5, carrier gas helium, ramp from 50 to 300°C at a rate 5 deg/min, vaporizer temperature 280°C, ion source temperature 250°C, 70 eV) and Shimadzu QP-2010 Plus (capillary column Supelco PTE-5 30 m × 0.25 mm). For column chromatography silica gel KSK (100–200 μm) was utilized.

***N*-Aryl(benzyl)-1,5,3-dithiazocan-3-amines.** *a.* In a Schlenk vessel was placed 10 mmol of 3-*tert*-butyl-1,5,3-dithiazocinane [1] and 0.5 mmol of catalyst Cp₂TiCl₂, the mixture was stirred for 15–30 min, 5 ml of acetonitrile and 10 mmol of an appropriate aryl(benzyl)hydrazine, was added, the reaction mixture was stirred for 3 h at 20°C.

The reaction products were passed through a thin bed of SiO_2 , the solvent was distilled off on the rotary evaporator.

b. In a Schlenk vessel was placed 10 mmol of N^1,N^1,N^7,N^7 -tetramethyl-2,6-dithiaheptane-1,7-diamine, 10 mmol of aryl(benzyl, alkyl)hydrazine, 5 ml of chloroform, 0.5 mmol of catalyst CuCl_2 , the mixture was stirred for 1 h at 60°C . The reaction products were passed through a thin bed of SiO_2 , the solvent was distilled off on the rotary evaporator. Amines **II** are brown oily fluids, well soluble in the most organic solvents.

***N*-Phenyl-1,5,3-dithiazocan-3-amine (**IIa**).** Yield 71% (*a*), 79% (*b*), R_f 0.55, oily fluids. ^1H NMR spectrum, δ , ppm: 1.87 s (2H, CH_2), 2.80 s (2H, CH_2), 4.51 s (2H, CH_2), 6.80–7.33 m (5H, CH). ^{13}C NMR spectrum, δ , ppm: 29.01 (C⁷), 35.07 (C^{6,8}), 59.74 (C^{4,2}), 119.20 (C^{10,14}), 120.28 (C¹²), 125.75 (C^{11,13}), 145.64 (C⁹). Mass spectrum, m/z (I_{rel} , %): 240 [M]⁺ (60); 147 [$\text{C}_4\text{H}_7\text{N}_2\text{S}_2$]⁺ (50), 120 [$\text{C}_3\text{H}_6\text{NS}_2$]⁺ (100), 106 [$\text{C}_6\text{H}_6\text{N}_2$]⁺ (15), 91 [$\text{C}_6\text{H}_5\text{N}$]⁺ (35), 77 [C_6H_5]⁺ (23), 75 [$\text{CH}_3\text{N}_2\text{S}$]⁺ (25), 60 [CH_2NS]⁺ (25), 51 [CNCN]⁺ (50), 46 [CH_2S]⁺ (10). M_{calc} 240.39.

***N*-(4-Nitrophenyl)-1,5,3-dithiazocan-3-amine (**IIb**).** Yield 68% (*a*), 83% (*b*), mp 198–200°C. ^1H NMR spectrum, δ , ppm: 1.74 s (2H, CH_2), 2.55 s (2H, CH_2), 4.83 s (2H, CH_2), 6.54 d (2H, CH, J 4 Hz), 7.02 d (2H, CH, J 4 Hz). ^{13}C NMR spectrum, δ , ppm: 28.92 (C⁷), 33.81 (C^{6,8}), 58.66 (C^{4,2}), 118.19 (C^{10,14}), 122.59 (C^{11,13}), 140.09 (C¹²), 147.71 (C⁹). Mass spectrum, m/z (I_{rel} , %): 285 [M]⁺ (40), 198 [$\text{C}_7\text{H}_8\text{N}_3\text{O}_2\text{S}$]⁺ (15), 226 [$\text{C}_9\text{H}_{12}\text{N}_3\text{O}_2\text{S}$]⁺ (100), 272 [$\text{C}_{10}\text{H}_{14}\text{N}_3\text{O}_2\text{S}_2$]⁺ (20), 92 [$\text{CH}_2\text{SCH}_2\text{S}$]⁺ (20), 78 [C_6H_6]⁺ (15), 46 [CH_2S]⁺ (65). M_{calc} 285.38.

***N*-(2,4-Dinitrophenyl)-1,5,3-dithiazocan-3-amine (**IIc**).** Yield 75% (*a*), 80% (*b*), mp 203–205°C. ^1H NMR spectrum, δ , ppm: 1.91 s (2H, CH_2), 2.70 s (2H, CH_2), 4.56 s (2H, CH_2), 6.90–7.13 m (3H, CH). ^{13}C NMR spectrum, δ , ppm: 27.50 (C⁷), 36.00 (C^{6,8}), 60.10 (C^{2,4}), 113.41 (C¹⁴), 119.24 (C¹¹), 123.13 (C¹²), 139.69 (C¹⁰), 145.00 (C⁹). Mass spectrum, m/z (I_{rel} , %): 330 [M]⁺ (40), 287 [$\text{C}_{10}\text{H}_{13}\text{N}_4\text{O}_4\text{S}$]⁺ (40), 272 [$\text{C}_9\text{H}_{11}\text{N}_4\text{O}_4\text{S}$]⁺ (20), 244 [$\text{C}_7\text{H}_7\text{N}_4\text{O}_4\text{S}$]⁺ (50), 92 [$\text{CH}_2\text{SCH}_2\text{S}$]⁺ (10), 46 [CH_2S]⁺ (100). M_{calc} 330.39.

***N*-(4-Methylphenyl)-1,5,3-dithiazocan-3-amine (**IID**).** Yield 78% (*a*), 85% (*b*), R_f 0.67, oily fluids. ^1H NMR spectrum, δ , ppm: 1.50 s (3H, CH_3), 2.00 s (2H, CH_2), 2.95 s (2H, CH_2), 4.69 s (2H, CH_2), 6.97–7.35 m (4H, CH). ^{13}C NMR spectrum, δ , ppm: 19.23 (C¹⁵), 29.32 (C⁷), 32.92 (C^{6,8}), 59.43 (C^{2,4}), 112.41 (C^{10,14}), 119.15

(C^{11,13}), 145.59 (C⁹). Mass spectrum, m/z (I_{rel} , %): 254 [M]⁺ (100), 209 [$\text{C}_{11}\text{H}_{17}\text{N}_2\text{S}$]⁺ (50), 152 [$\text{C}_8\text{H}_{10}\text{NS}$]⁺ (30), 121 [$\text{C}_7\text{H}_9\text{N}_2$]⁺ (70), 92 [$\text{CH}_2\text{SCH}_2\text{S}$]⁺ (20), 46 [CH_2S]⁺ (50). M_{calc} 254.41.

***N*-Benzyl-1,5,3-dithiazocan-3-amine (**IIe**).** Yield 80% (*a*), 81% (*b*), R_f 0.43, oily fluids. ^1H NMR spectrum, δ , ppm: 1.93 s (2H, CH_2), 2.68 s (2H, CH_2), 4.55 s (2H, CH_2), 4.69 s (2H, CH_2), 6.88–7.55 m (5H, CH). ^{13}C NMR spectrum, δ , ppm: 29.13 (C⁷), 33.69 (C^{6,8}), 59.41 (C^{2,4}), 69.20 (C⁹), 125.27 (C¹³), 127.92 (C^{11,15}), 129.30 (C^{12,14}), 143.80 (C¹⁰). Mass spectrum, m/z (I_{rel} , %): 254 [M]⁺ (100), 182 [$\text{C}_9\text{H}_{14}\text{N}_2\text{S}$]⁺ (40), 135 [$\text{C}_8\text{H}_{12}\text{N}_2$]⁺ (80), 92 [$\text{CH}_2\text{SCH}_2\text{S}$]⁺ (15), 46 [CH_2S]⁺ (100). M_{calc} 254.42.

***N*-Methyl-1,5,3-dithiazocan-3-amine (**IIIf**).** Yield 74% (*b*), R_f 0.47, oily fluids. ^1H NMR spectrum, δ , ppm: 1.81 s (2H, CH_2), 4.35 s (3H, CH_3), 2.43 s (2H, CH_2), 4.01 s (2H, CH_2). ^{13}C NMR spectrum, δ , ppm: 29.03 (C⁷), 37.85 (C^{6,8}), 59.56 (C^{2,4}), 45.24 (C⁹). Mass spectrum, m/z (I_{rel} , %): 178 [M]⁺ (70), 233 [$\text{C}_5\text{H}_{13}\text{N}_2\text{S}$]⁺ (20), 105 [$\text{C}_3\text{H}_9\text{N}_2\text{S}$]⁺ (50), 45 [CH_5N_2]⁺ (70). M_{calc} 178.32.

***N*-tert-Butyl-1,5,3-dithiazocan-3-amine (**IIg**).** Yield 74% (*b*), R_f 0.53, oily fluids. ^1H NMR spectrum, δ , ppm: 1.13 s (9H, CH_3), 1.77 s (2H, CH_2), 2.85 s (2H, CH_2), 4.15 s (2H, CH_2). ^{13}C NMR spectrum, δ , ppm: 28.32 (C⁷), 35.44 (C^{6,8}), 50.35 (C^{11–13}), 58.46 (C^{2,4}), 45.24 (C¹⁰). Mass spectrum, m/z (I_{rel} , %): 220 [M]⁺ (50), 161 [$\text{C}_7\text{H}_{17}\text{N}_2\text{S}$]⁺ (50), 133 [$\text{C}_5\text{H}_{13}\text{N}_2\text{S}$]⁺ (70), 87 [$\text{C}_4\text{H}_{11}\text{N}_2$]⁺ (100). M_{calc} 220.40.

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