

Assignment of ^{13}C -Signals from the *meso*-Carbons by Syntheses of ^{13}C -Protoporphyrin-IX Dimethyl Esters

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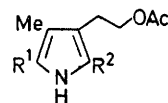
Summary $[\beta\text{-}^{13}\text{C}]$ -, $[\gamma\text{-}^{13}\text{C}]$ -, and $[\delta\text{-}^{13}\text{C}]$ -Protoporphyrin-IX dimethyl esters have been synthesised by unambiguous routes from ^{13}C -formaldehyde to allow assignment of the four n.m.r. ^{13}C -signals from the *meso*-carbons; the dimethyl ester of diacetyldeuteroporphyrin-IX has been similarly studied.

It was essential for our biosynthetic studies¹ to assign the four n.m.r. ^{13}C -signals near δ 97 arising² from the *meso*-carbons (α , β , γ , δ) of protoporphyrin-IX dimethyl ester (10). This has been achieved by the following syntheses³ of specifically ^{13}C -labelled porphyrins which, for the β -, and δ -labelled samples, were based on Fischer's pyrromethene route^{3a} using Johnson's *a,c*-biladiene method⁴.

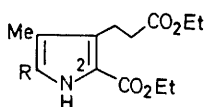
The 5-methyl of pyrrole^{3d} (1) was converted into carboxy and decarboxylation gave the α -free pyrrole (2) which was reductively methylated⁵ with ^{13}C -formaldehyde (90 atom% in all experiments) yielding $[5\text{-Me-}^{13}\text{C}]$ -(1). Similarly, pyrrole⁶ (5) afforded the $[5\text{-Me-}^{13}\text{C}]$ pyrrole (6). The former was converted⁷ into the ^{13}C pyrromethene (8) [69% from pyrroles (3) and (4)] and the latter into the ^{13}C pyrromethene (9) (55%) which was known⁴ in unlabelled form. Reaction of the ^{13}C pyrromethene (8) with unlabelled (9) in $\text{AcOH-CHCl}_3\text{-SnCl}_4$ followed by methanolic HBr gave the *a,c*-biladiene system. Cyclisation of this total material in $\text{Me}_2\text{SO-pyridine}$ ⁴ and treatment of the products with acidic methanol gave the known diol^{3d} (11) in 69% yield from the pyrromethenes. Diol (11) was converted essentially by Jackson and Kenner's halide-elimination method^{2d} into $[\beta\text{-}^{13}\text{C}]$ protoporphyrin-IX dimethyl ester (10). The foregoing sequence affords the porphyrin (10) in 42% yield overall from the dipyrrolic units. $[\delta\text{-}^{13}\text{C}]$ Protoporphyrin-IX dimethyl ester (10) was similarly synthesised from ^{13}C -pyrromethene (9) and unlabelled (8).

The route to $[\gamma\text{-}^{13}\text{C}]$ protoporphyrin-IX dimethyl ester (10) made use of the ring *c*-ring *D* symmetry by condensation of ^{13}C -formaldehyde with the *a,c*-biladiene (17); this approach has not been widely used⁸ and proof that it yields one pure isomer was first obtained by synthesis in this way⁹ of coproporphyrin-II tetramethyl ester. The route to the diene (17) started with the pyrromethane⁷ (18) synthesised from the pyrroles (15) and (16) each being prepared from (14); diborane reduction¹⁰ of (18) and acetylation gave (19). The pyrromethane (20) obtained from (19) by deprotection and decarboxylation condensed with the aldehyde (7) to yield the unstable (17). Acid-catalysed ring closure using ^{13}C -formaldehyde followed by methanolysis gave the $[\gamma\text{-}^{13}\text{C}]$ diol (11) [7% overall yield from (19)]; this was converted as before into $[\gamma\text{-}^{13}\text{C}]$ protoporphyrin-IX dimethyl ester (10).

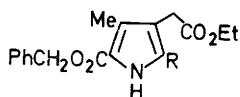
The three labelled samples of (10) were diluted with (10) from natural sources to give about 5 atom% ^{13}C . The ^{13}C -spectra then showed in each case one strongly enhanced signal from the ^{13}C -enriched *meso*-site and three small signals from the three other *meso*-carbons. The



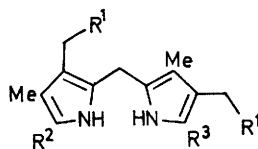
- (1) $\text{R}^1 = \text{PhCH}_2\text{O}_2\text{C}$, $\text{R}^2 = \text{Me}$
- (2) $\text{R}^1 = \text{PhCH}_2\text{O}_2\text{C}$, $\text{R}^2 = \text{H}$
- (3) $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{CHO}$
- (4) $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{Me}$



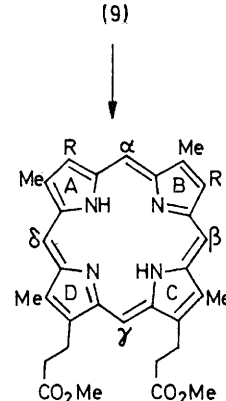
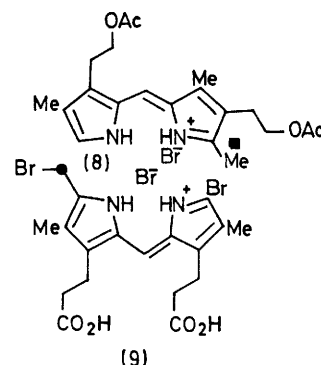
- (5) $\text{R} = \text{H}$
- (6) $\text{R} = \text{Me}$
- (7) $\text{R} = \text{CHO}, \text{CO}_2\text{H}$ in place of $2\text{-CO}_2\text{Et}$



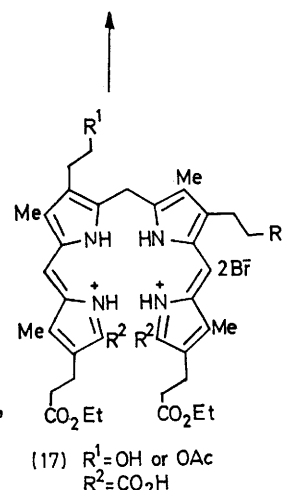
- (14) $\text{R} = \text{Me}$
- (15) $\text{R} = \text{CH}_2\text{OAc}$
- (16) $\text{R} = \text{CO}_2\text{Bu}^t$



- (18) $\text{R}^1 = \text{CO}_2\text{Et}$, $\text{R}^2 = \text{PhCH}_2\text{O}_2\text{C}$, $\text{R}^3 = \text{CO}_2\text{Bu}^t$
- (19) $\text{R}^1 = \text{CH}_2\text{OAc}$, $\text{R}^2 = \text{PhCH}_2\text{O}_2\text{C}$, $\text{R}^3 = \text{CO}_2\text{Bu}^t$
- (20) $\text{R}^1 = \text{CH}_2\text{OAc}$, $\text{R}^2 = \text{R}^3 = \text{H}$



- (10) $\text{R} = \text{CH}=\text{CH}_2$
- (11) $\text{R} = \text{CH}_2\text{CH}_2\text{OH}$
- (12) $\text{R} = \text{CH}(\text{OH})\text{Me}$
- (13) $\text{R} = \text{COMe}$



results allow the following unambiguous assignments:[†] α -*meso* at δ 97.7, β -*meso* at 97.1, δ -*meso* at 96.7, and γ -*meso* at 95.8.

[†] At 0.01–0.02M in CDCl_3 ; the precise chemical shifts are significantly affected by concentration.

Each ^{13}C -labelled sample of (10) was converted¹¹ into haematoporphyrin dimethyl ester (12) and this then oxidised with Jones' reagent to afford the diketone (13). The ^{13}C -spectra so obtained allowed the well spread ^{13}C -signals¹² from the *meso*-carbons to be assigned: α -*meso* at δ 102.3, β -*meso* at 99.9, δ -*meso* at 97.3, and γ -*meso* at 95.4.

The foregoing results are of key importance for current¹ and future research on the biosynthesis of the porphyrin macrocycle.

We thank the Nuffield Foundation and the S.R.C for financial support.

(Received, 19th April 1973; Com. 560.)

¹ A. R. Battersby, E. Hunt, and E. McDonald, following Communication.

² A. R. Battersby, E. McDonald, J. Moron, and J. Feeney, *J.C.S. Chem. Comm.*, 1972, 920.

³ For earlier syntheses of protoporphyrin-IX see (a) H. Fischer and K. Zeile, *Annalen*, 1929, **468**, 114 and refs. therein; (b) R. P. Evstigneeva, V. N. Guryshv, A. F. Mironov, and G. Ya. Volodarskaya, *Zhur. obshchei Khim.*, 1969, **39**, 2558; (c) R. Grigg, A. W. Johnson, and M. Roche, *J. Chem. Soc. (C)*, 1970, 1928; (d) R. P. Carr, A. H. Jackson, G. W. Kenner, and G. S. Sach, *J. Chem. Soc.*, 1971, 487; (e) A. M. d'A. Rocha Gonsalves, G. W. Kenner, and K. M. Smith, *Chem. Comm.*, 1971, 1304.

⁴ P. Bamfield, R. Grigg, A. W. Johnson, and R. W. Kenyon, *J. Chem. Soc.*, 1968, 1259; P. Bamfield, R. L. N. Harris, A. W. Johnson, I. T. Kay, and K. W. Shelton, *ibid.*, 1966, 1436.

⁵ M. W. Roomi and S. F. MacDonald, *Canad. J. Chem.*, 1970, **48**, 139.

⁶ F. Morsingh and S. F. MacDonald, *J. Amer. Chem. Soc.*, 1960, **82**, 4377.

⁷ All dipyrrolic units reported were synthesised by standard pyrrole chemistry (See R. L. N. Harris, A. W. Johnson, and I. T. Kay, *Quart. Rev.*, 1966, **20**, 211; K. M. Smith, *ibid.*, 1971, **25**, 31) from the indicated monopyrroles; confirmatory analytical and spectroscopic data were obtained for each new substance.

⁸ See A. W. Johnson and I. T. Kay, *J. Chem. Soc.*, 1965, 1620.

⁹ Details in our full paper.

¹⁰ K. M. Biswas and A. H. Jackson, *Tetrahedron*, 1968, **24**, 1145.

¹¹ H. Fischer and H. Orth, 'Die Chemie des Pyrrols', Akad. Verlagsgesellschaft, Leipzig, 1937, Vol. 2, Part I, p. 421.

¹² For spectrum of (13) at natural abundance see D. Doddrell and W. S. Caughey, *J. Amer. Chem. Soc.*, 1972, **94**, 2510.