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A New Approach to Indolizidine Alkaloids: Asymmetric Formal Total Synthesis of (-)-Swainsonine

Wei-Shan Zhou^{**}, Wen-Ge Xie^b, Zhi-Hui Lu^{*}, Xin-Fu Pan^b

^a Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Lu, Shanghai 200032, China; ^bDepartment of Chemistry, Lanzhou University, Lanzhou 730000, China.

Abstract: A concise, noncarbohydrate-based approach to (-)-Swainsonine (1) has been achieved by utilizing the kinetic resolution of the α -furfuryl amide 4 and the Sharpless ADH reaction as key steps.

Polyhydroxylated indolizidine alkaloids, typified by Swainsonine (1),¹ Castanospermine (2),² Lentiginosine $(3)^3$ and their derivatives are of considerable importance due to their potent activities as inhibitors of glycosidase and glycoprotein processing⁴. These compounds have also exhibited interesting activities in anticancer, antiviral, antiretroviral and immunoregulatory.⁵ Consequently, much attention has been devoted to the synthesis of Swainsonine (1) over the past decade.⁶⁻¹⁸ Most of the previous methodologies utilized carbohydrates as starting material.⁶⁻¹³ Others used R-glutamic acid¹⁴, D-tartaric acid,¹⁵ D-malic acid,¹⁶ and D-iso ascorbic acid¹⁷ as the chiral precursors. However, to the best of our knowledge, only one approach to the target compound 1 was reported¹⁸, starting from a racemic allylic alcohol derivative instead of the above mentioned chiral pool.



Notwithstanding this plethora of methods, interest in the synthesis of Swainsonine and its analogues remains undiminished. Development of general methods which could have flexibility for the construction of these compounds and analogues continues to be important to probe structure-activity relationship. We have previously developed an efficient method for the kinetic resolution of α -furfuryl amide by using the modified Sharpless asymmetric epoxidation reagent.¹⁹ This reaction afforded two versatile chiral building blocks, both of them are very suitable to be used for elaboration of the skeleton of many types of alkaloids.²⁰ As part of a program designed to develop a new general strategy for the enantioselective synthesis of biologically active

alkaloids and explore the use of the reaction in alkaloid synthesis, we undertook a synthesis of (-)-Swainsonine (1), utilizing the α -furfurylamide 4 as the starting material.



Scheme: Reagents and conditions: a. $Ti(O'Pr)_4$, D-(-)-DIPT, TBHP, SiO₂, CaH₂, CH₂Cl₂, 25 °C, 2 days; b. HC(OEt)₃, BF₃.Et₂O, 4A Ms, Et₂O, r.t.; c. NaBH₄, MeOH, -40- -30 °C; d. BnBr, NaH, Bu₄N⁺I' (Cat.), THF; e. NaBH₄, HCO₂H, 0°C; f. OsO₄ (Cat.), NMMO, DHQN-CLB, trace CH₃SO₂NH₂, acetone-H₂O, ultrasonication; g. *p*-TsOH, ¹BuOH, reflux; h. Na/naphthalene, DME, -60 °C; i. Ph₃P, CCl₄, Et₃N, DMF; j. CH₃C(OCH₃)₂CH₃, *p*-TsOH, CH₂Cl₂.

The synthesis of (-)-Swainsonine (1) is depicted in Scheme. Kinetic resolution of α -furfuryl amide 4 under the reported procedure¹⁹ yielded the (2S,6S)-dihydropyridinone 5 in 42% yield.²¹ Preliminary attempts to reach 9 in two steps by directly exposure 5 to a solution of sodium borohydride in formic acid²² followed by benzylation of the resulting alcohol was unsuccessful. The reduction of 5 gave a complex mixture. Therefore, we circumvented this problem by first treatment of 5 with triethyl orthoformate in Et₂O in the presence of a catalytic amount of BF₃.Et₂O to give 6 in 92% yield. Next, reduction of 6 with sodium borohydride in methanol at -40 °C to -30 °C afforded solely the alcohol 7 in 88% yield, with the desired sense of stereochemistry.²³ Subsequent benzylation of the alcohol 7, followed by reduction with sodium borohydride in formic acid at 0 °C furnished 9 in 80% yield.

Having the nivotal intermediate 9 in hand we next tried to convert 9 into the desired diol 10a. Thus, the Sharpless asymmetric dihydroxylation reagent(DHON-CLB as chiral ligand)²⁴ was tried out on 9, however, no reaction occurred. Fortunately, we eventually found that performance of the reaction in an ultrasonic cleaner, proceeded smoothly to form a separable mixture of the desired diol $10a^{21}$ and its epimer 10b in a ratio of 10:1 respectively in 80% combined yield. The stereochemical assignments for these products were based on the Sharpless' empirical rule and the major isomer 10a was judged to have the desired 7S.8R configuration. Further confirmation of this assignment was provided by transformation of 10a to a known compound, vide infra. In contrast, the use of DHOD-CLB instead of DHON-CLB as ligand resulted in an opposite and somewhat lower diastereoselectivity (10a:10b 1:4). The inherent diastereoselectivity of the olefin 9 was 2.5:1 in favor of 10a as observed from dihydroxylation with OsO.-NMMO. Removal of the MOM group in 10a by treatment with p-TsOH gave the triol 11 in 90% yield. Deprotection of 11 by sodium naphthalide and without purification direct treatment of the crude product with Ph₂P. CCl₄. Et₃N in DMF underwent cyclization to afford the 8-benzyloxy Swainsonine (12) 21 in 50% overall yield from 11. Attempts to obtain Swainsonine by debenzylation of 11 was not successful, mainly due to the unfeasibility of isolation. To fulfill a formal synthesis of the target molecule, the diol 12 was converted into the known acetonide $13^{16.21}$ by treatment with dimethoxypropane in the presence of a catalytic amount of p-TsOH in 94% yield, which would deliver 1 by sequential hydrogenolysis and acidic hydrolysis, according to the results of C Kibayashi.¹⁶

In summary, we have developed an efficient method for preparing polyhydroxylated indolizidine alkaloids by employing the kinetic resolution of α -furfuryl amide. The synthesis of the other structure related polyhydroxylated indolizidine alkaloid, castanospermine (2) is currently under investigation.

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- 21. The data of some typical intermediate are listed below: 5. colorless oil; $\left[\alpha\right]_{0}^{20} = -9.0^{\circ}$ (c 1.0, MeOH); ¹H-NMR (400 MHz, CDCl₃): 7.71-7.28 (each d, J=J'=7 Hz, each 2H), 6.91 (dd, J=4.54 Hz, J'=5.77 Hz, 1H), 5.88 (d, J= 5.77Hz, 1H), 5.68 (d, J=4.57 Hz, 1H), 5.16 (d, J= 8.54 Hz, 1H), 4.69 (d, J=1.98 Hz, 2H), 4.44 (dd, J=6.26Hz, J=7.50Hz, 1H), 4.23 (m, 1H), 3.97, 3.69 (each m, each 1H), 3.42 (s, 3H), 2.40 (s, 3H); FAB-MS(m/z); 368 (M⁺+1, 5%), 350 (M⁺+1-H₂O, 100%); Anal. Calcd. for $C_{17}H_{21}NO_6S$: C 55.57, H 5.76, N 3.81; Found: C 55.32, H 5.92, N 3.98; **10a**. crystals; m.p. 140-142 °C; $[\alpha]_D^{20} = -70.5^\circ$ (c 1.0, MeOH); ¹H-NMR (400 MHz, CDCl₃); 7.58, 7.16 (each d, J=J'=8.14 Hz, each 2H), 7.39-7.28 (m, 5H), 4.69 (s. 2H), 4.51 (d. J=2.15 Hz, 2H), 4.31 (m, 1H), 4.02 (m, 1H), 4.28 (m, 1H), 3.93 (m, 1H), 3.85, 3.58 (each m, 2H), 3.81 (m, 1H), 3.40 (s, 3H), 3.33 (m, 1H), 2.39 (s, 3H), 2.04, 1.69 (each m, each 1H), 1.51, 1.05 (each m, each 1H); FAB-MS m/z: 480 (M⁺+1, 5%), 448 (M⁺+1-CH₃OH, 88%), 344 (100%); Anal. Calcd. for C24H33NO7S: C 60.11, H 6.94, N 2.92; Found: C 60.27, H 7.16, N 2.88.; 12. colorless oil; $[\alpha]_{D}^{20} = -79.4^{\circ}$ (c 1.0, MeOH); ¹H NMR (400 MHz, CD₃OD): 7.35-7.28 (*m*, 5H), 4.69 (*s*, 2H), 4.32, 4.02 (each m, each 1H), 4.29 (m, 1H), 3.94 (m, 1H), 3.85, 3.59 (each m, each 1H), 3.81 (m, 1H), 3.08 (m, 1H), 2.05, 1.69 (each m, each 1H), 1.51, 1.05 (each m, each 1H), FAB-MS m/z: 264 (M⁺+1, 10%), 246 (M⁺+1-H₂O, 8%), 91 (100%); Anal. Calcd. for C₁₅H₂₁NO₃: C 68.40, H 8.04, N 5.34; Found: C 68.00, H 8.37, N 5.09; 13. colorless oil; $[\alpha]_D^{20}$ -64.2° (c 0.5, CHCl₃) {Lit.¹⁶ colorless oil; $[\alpha]_D^{26}$ -58.9° (c 0.27, CHCl₃); ¹H-NMR (400 MHz, CDCl₃): 7.37-7.21 (m, 5H), 4.68 (dd, J=3.2, J'=7.2 Hz, 1H), 4.61 (s, 2H), 4.39 (m, 1H), 3.51 (m, 1H), 3.12 (m, 1H), 2.44 (dd, J=3.2 Hz, J'=7.2 Hz, 1H), 2.32 (t, J=8.3, J'=10.2 Hz, 1H), 2.08 (dbr, J=14.6 Hz, 1H), 1.94 (m, 1H), 1.68 (m, 2H), 1.53 (s, 4H), 1.35 (s, 3H), 1.25 (m, 1H); FABMS m/z: 304 (M⁺+1, 3%), 289 (M⁺+1-CH₃, 7%), 214 (100%); Anal. Calcd. for C₁₈H₂₅NO₃: C 71.26, H 8.31, N 4.62; Found: C 70.88, H 8.52, N 4.77.
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