

Synthesis of Hexakis(alkylgermesquioxane)s from Alkyl(chloro)ethoxygermanes and Their Formation Mechanism

Masato Nanjo, Takaomi Sasage, and Kunio Mochida*

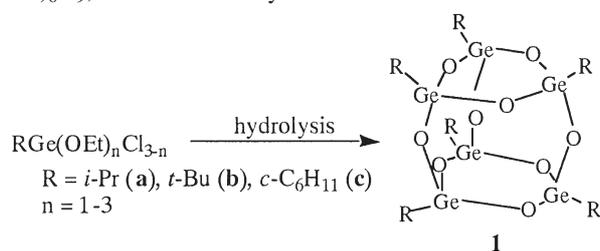
Department of Chemistry, Faculty of Science, Gakushuin University, 1-5-1 Mejiro, Toshima-ku, Tokyo 171-8588

(Received August 8, 2002; CL-020669)

Alkyl(chloro)ethoxygermanes were hydrolyzed with water to give 1,3,5-trialkyl-1,3,5-trichlorocyclotrigermoxanes. Hydrolysis of 1,3,5-trichlorocyclotrigermoxanes gave 5,7-dichloro-1,3,5,7,9,11-hexaalkyltricyclo[7.3.1.1^{3,7}]-hexagermoxanes. The tricyclic *anti*-form ladder hexagermoxanes reacted with water to afford hexakis(alkylgermesquioxane)s. These cage and ladder germoxanes were identified by spectroscopic and X-ray diffraction methods.

Organosilsesquioxanes with the formula (RSiO_{1.5})_n containing Si-O bonds as a main chain have been of considerable interest because of their unusual structures and properties, and as new materials.¹ Matsumoto, Unno, and co-workers,² and other groups³⁻⁷ reported the synthesis of cage, ladder, and sheet-like oligomers and polymers of (RSiO_{1.5})_n and established their structures by X-ray analysis. They are found to be organometallic analogues of silicate anions, (Si_nO_{2.5n})ⁿ⁻. The modified silicate can be used as pre-ceramic building blocks,⁸ organolithic macromolecular materials,⁹ heterogeneous silica-supported metal catalyst,¹⁰ potential photoresists,¹¹⁻¹³ and so on. While the organosilsesquioxanes have been amply investigated, there have been few reports on organogermesquioxanes.¹⁴ We describe herein the synthesis of hexakis(alkylgermesquioxane)s by hydrolysis of alkyl(chloro)ethoxygermanes and cyclic germoxanes as intermediates of the cage germoxanes, and determination of these structures by X-ray diffraction methods.

The alkyl(chloro)ethoxygermanes, RGe(OEt)_nCl_{3-n} (R = *i*-Pr, *n* = 0; R = *t*-Bu, *n* = 0-3; R = *c*-C₆H₁₁, *n* = 0) reacted with aqueous NaOH solution in xylene at 130-140 °C for 3 h. The concentration of the reaction mixture by evaporation of xylene followed by crystallization from pentane gave colorless crystals with a composition of hexakis(alkylgermesquioxane)s, (RGe)₆O₉, **1a-c** in 60-98% yields.



The germesquioxane **1a-c** was fully established by spectroscopic and X-ray diffraction methods. As a typical example, a molecular structure of hexakis(*isopropylgermesquioxane*), (*i*-PrGe)₆O₉, **1a** is shown in Figure 1.¹⁵ The average Ge-O bond length is 1.755 Å in six- and eight-membered rings. The average O-Ge-O and Ge-O-Ge bond angles are 108° and 130.5°, respectively. All these values are within the normal range,

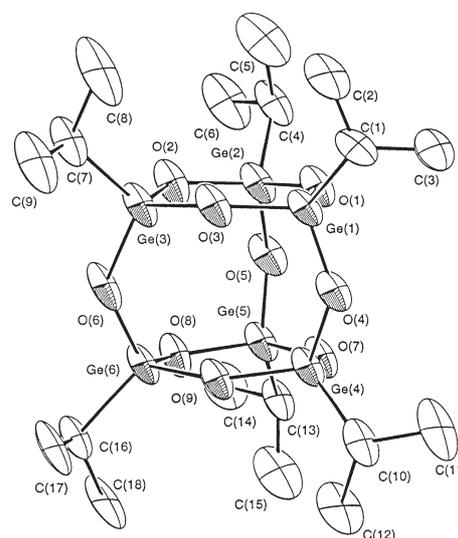
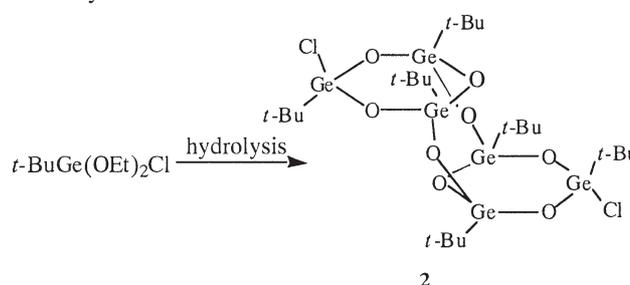


Figure 1. An ORTEP representation of the structure of **1a** (hydrogen atoms are omitted for clarity). Selected bond length (Å) and angles (°): Ge1-O4 1.746(4), Ge1-O3 1.758(4), Ge1-O1 1.763(4), Ge1-C1 1.916(5), O4-Ge1-O3 107.95(18), O4-Ge1-O1 108.27(17), O3-Ge1-O1 107.26(17), O4-Ge1-C1 111.2(2).

showing that this molecule is strain-free.

The *tert*-butyl(chloro)diethoxygermane, *t*-BuGe(OEt)₂Cl prepared by *tert*-butyltrichlorogermane, *t*-BuGeCl₃ and ethanol in the presence of pyridine at room temperature for 2 weeks, was carefully hydrolyzed with water in ethanol at 5 °C for 6 h. The reaction mixture was concentrated by evaporation of ethanol. The residue was stirred in benzene at 5 °C and the generated solids were filtered off. The concentration of the organic layer by evaporation of benzene followed by recrystallization from hexane gave only 5,7-dichloro-1,3,5,7,9-hexa-*tert*-butyltricyclo[7.3.1.1^{3,7}]octagermoxane, (*t*-BuGe)₆O₈Cl₂ **2** in 29% isolated yield.



The ¹H NMR spectrum of **2** displayed two *tert*-butyl signals at 1.29 and 1.32 ppm in 1 : 2 ratio. The ¹³C NMR showed four signals at 25.9, 26.5, 30.5, and 34.7 ppm. These NMR data

indicated that germoxane **2** was a symmetrical structure. The structure was established by X-ray crystallography.¹⁶ The molecular structure is shown in Figure 2 as an *anti*-form ladder germoxane. The average Ge–O bond length is 1.76 Å in six- and eight-membered rings. The average O–Ge–O bond angle is 107°; the average Ge–O–Ge is 126.5°. The structural parameters of **2** (e.g., Ge–O bond lengths, Ge–O–Ge angles, O–Ge–O angles) are similar to those for **1**.^{14,15}

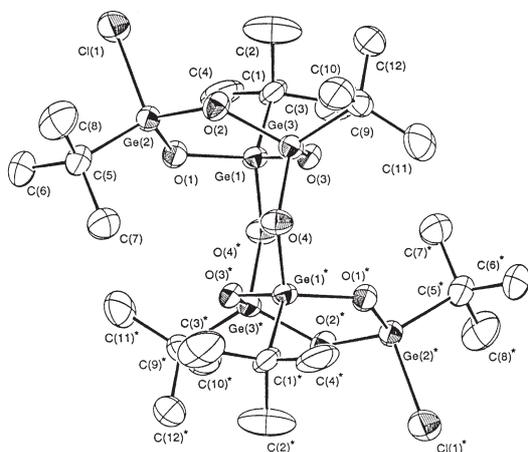
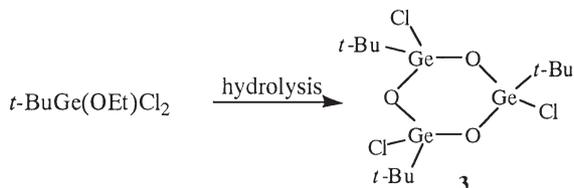


Figure 2. An ORTEP representation of the structure of **2** (hydrogen atoms are omitted for clarity). Selected bond length (Å) and angles (°): Ge1–O4^{#1} 1.751(3), Ge1–O3 1.765(3), Ge1–O1 1.780(3), Ge1–C1 1.947(4), Ge2–O1 1.746(3), Ge2–O2 1.755(3), Ge2–C5 1.946(4), Ge2–Cl1 2.1739(12), O4^{#1}–Ge1–O3 107.39(13), O4^{#1}–Ge1–O1 106.43(13), O3–Ge1–O1 107.68(13), O4^{#1}–Ge1–C1 109.80(15), O3–Ge1–C1 112.05(16), O1–Ge1–C1 113.18(16).

After hydrolysis of **2** with aqueous NaOH in xylene at 140 °C for 3 h, the cage germoxanesquioxane **1** was formed in 18% isolated yield together with polygermoxane. The formation of **1** suggests that the germoxane **2** is clearly an intermediate for the formation of **1**.

The *tert*-butyl(dichloro)ethoxygermane, *t*-BuGe(OEt)Cl₂, prepared by *t*-BuGeCl₃ and ethanol at room temperature for 2 weeks, was treated with water at 5 °C for 3 h to give **3** as a sole product. The product **3** was isolated by distillation in 31% yield. ¹H NMR spectrum of **3** displayed two *tert*-butyl signals at 1.29 and 1.33 ppm in 1 : 2 ratio. ¹³C NMR of **3** showed four signals at 25.2, 25.5, 35.4, and 39.6 ppm. The fragment peak (M⁺-*t*-Bu) with *m/z* 487 was observed. The NMR (¹H and ¹³C{¹H}) and GC-MS spectra of **3** disclosed it to be 1,3,5-tri-*tert*-butyl-1,3,5-trichlorotriggermoxane. After hydrolysis of **3** for an additional 33 h, the *anti*-form ladder germoxane **2** was formed.



A reasonable mechanism is that two cyclic germoxanes **3** having probably *cis,trans*-1,3,5-tri-*tert*-butyl-1,3,5-trichlorocyclo-triggermoxane geometry join co-facially to form the *anti*-form

ladder germoxane **2**, which undergoes isomerization to give the *syn*-type isomer. Dehydration of the *syn*-form germoxane gives the cage germoxane **1**.

In summary, we synthesized cage hexakis(alkylgermasesquioxane), (RGe)₆O₉, (R = *i*-Pr, *t*-Bu, *c*-C₆H₁₁) from hydrolysis of RGe(OEt)_nCl_{3-n} (*n* = 0–3), and determined its crystal structure. Cyclic and *anti*-form ladder germoxanes as intermediates of the cage germoxanes by careful hydrolysis of RGe(OEt)_nCl_{3-n} were examined.

The authors thank Asai Germanium Research Institute Co., Ltd., for providing us trichlorogermane.

References and Notes

- a) R. H. Baney, M. Itoh, A. Sakakibara, and T. Suzuki, *Chem. Rev.*, **95**, 1409 (1995). b) F. Feher, in "Silicon, Germanium, Tin, and Lead Compounds. Metal Alkoxides, Diketonates and Carboxylates," ed. by B. Arkles, Gelest, Tullytown (1998), pp 43–59 and references cited therein.
- a) M. Unno, B. A. Shamsul, H. Saito, and H. Matsumoto, *Organometallics*, **15**, 2413 (1996). b) M. Unno, K. Takada, and H. Matsumoto, *Chem. Lett.*, **1998**, 489. c) M. Unno, B. A. Shamsul, M. Arai, K. Takada, and H. Matsumoto, *Appl. Organomet. Chem.*, **13**, 1 (1991). d) M. Unno, A. Suto, K. Takada, and H. Matsumoto, *Bull. Chem. Soc. Jpn.*, **73**, 215 (2000). e) M. Unno, A. Suto, and H. Matsumoto, *J. Am. Chem. Soc.*, **124**, 1574 (2002).
- U. Dittmar, B. J. Hendan, U. Floerke, and H. C. Marsmann, *J. Organomet. Chem.*, **489**, 185 (1995) and references cited therein.
- a) F. J. Feher, K. Rahimian, T. A. Budzichowski, and Z. W. Ziller, *Organometallics*, **14**, 3920 (1995) and references cited therein. b) F. J. Feher, D. A. Newman, and J. F. Walzer, *J. Am. Chem. Soc.*, **111**, 1741 (1989).
- a) Y. I. Smolin, *Kristallografiya*, **15**, 31 (1970). b) Y. I. Smolin, Y. F. Shepelev, A. S. Ershov, D. Hoebbel, and W. Wieker, *Kristallografiya*, **29**, 712 (1984). c) D. Hoebbel, A. Engelhardt, A. Samoson, K. Ujjaszsy, and Y. I. Smolin, *Z. Anorg. Allg. Chem.*, **552**, 236 (1987).
- J. F. Brown and L. H. Vogt, *J. Am. Chem. Soc.*, **87**, 4313 (1965).
- H. Behbehani, B. J. Brisdon, M. F. Mahon, and K. Molloy, *J. Organomet. Chem.*, **469**, 19 (1994).
- V. W. Day, W. G. Klempner, V. V. Mainz, and M. Miller, *J. Am. Chem. Soc.*, **107**, 8262 (1985).
- P. A. Agaskar, *J. Am. Chem. Soc.*, **111**, 6858 (1989).
- F. J. Feher, T. A. Budzichowski, K. Rahimian, and J. W. Ziller, *J. Am. Chem. Soc.*, **114**, 3859 (1992) and references cited therein.
- H. Tosaka, K. Yamaguchi, M. Orihara, N. Ide, and K. Otaki, Jpn. Kokai Tokkyo Koho 87-24268 (1987); *Chem. Abstr.*, **107**, 124570 (1987).
- M. Uchida, Jpn. Kokai Tokkyo Koho 87-15560 (1987); *Chem. Abstr.*, **107**, 124563 (1987).
- R. Beer, H. Burgy, G. Calzaferri, and I. Kamber, *J. Electron Spectrosc. Relat. Phenom.*, **44**, 121 (1987).
- H. Puff, S. Franken, and W. Schuh, *J. Organomet. Chem.*, **256**, 23 (1983).
- Crystal data for **1a**: C₁₈H₄₂Ge₆O₉; fw = 838.06; crystal size 0.40 × 0.20 × 0.20 mm³; triclinic, space group P $\bar{1}$, Z = 2, a = 11.9140(11) Å, b = 12.5490(11) Å, c = 13.2720(15) Å, α = 63.855(4)°, β = 64.205(5)°, γ = 89.764(5)°; V = 1559.3(3) Å³, D_{calcd} = 1.784 g/cm³; Goodness of fit = 1.048, R = 0.0388 (R_{all} = 0.0479 for 4280 reflections), R_w = 0.1243 for 3522 reflections with I > 2σ(I). Crystallographic data for **1a** have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC-192754.
- Crystal data for **2**: C₂₄H₅₄Cl₂Ge₆O₈; fw = 977.11; crystal size 0.40 × 0.30 × 0.30 mm³; triclinic, space group P $\bar{1}$, Z = 1, a = 10.2730(10) Å, b = 10.8580(10) Å, c = 10.9450(10) Å, α = 65.663(5)°, β = 71.909(5)°, γ = 63.960(5)°; V = 986.84(16) Å³, D_{calcd} = 1.644 g/cm³; Goodness of fit = 1.133, R = 0.0354 (R_{all} = 0.031 for 2604 reflections), R_w = 0.1188 for 2469 reflections with I > 2σ(I). CCDC-192755.