Intramolecular Nucleophilic Addition of Silylenol Ether to Photosensitized Electron Transfer (PET) Generated Arene Radical Cations: A Novel *Non-Reagent* Based Carboannulation Reaction:

Ganesh Pandey*, A. Krishna, K. Girija and M. Karthikeyan

Division of Organic Chemistry (Synthesis) National Chemical Laboratory[#] Pune - 411 008 INDIA

Abstract: A New carboannulation strategy by intramolecular addition of silylenol ether to PET generated arene radical cation is reported.

Carboannulation processes are among the most important reactions in organic synthesis¹. Among the strategies reported for aromatic annulation reactions, the Friedel-Crafts type electrophilic substitution reactions² and electrocyclisations of *o*-quinodimethanes³ have been the most common. More recently, addition of radical to arene rings⁴ and Pd(II) catalysed diene reaction with aryl halides⁵ have also been reported. Considering the importance of carbocyclization reactions, formulating a new strategy, incorporating latent functionality, is always desirable. In this context our attention was drawn towards the possible utilization of intramolecular nucleophilic substitution of aromatic rings with proximate carbon nucleophiles involving arene radical cations⁶⁻⁸ have been reported, their utilization as intermediates in carboannulation reactions have remained unexplored. Our spectacular success⁹ in promoting the intramolecular nucleophilic addition of -OH and -NH₂ groups to PET generated arene radical cation Scheme-I

SCHEME-I

$$R \xrightarrow{h \vartheta}_{CH_{3}CN : H_{2}O} [1]^{*} \xrightarrow{DCN}_{PET Rxn.} \left[R \xrightarrow{f \vartheta}_{Nu} \right] \xrightarrow{} R \xrightarrow{f \vartheta}_{Nu}$$

$$1 \xrightarrow{\underline{2}} \xrightarrow{\underline{3}}$$

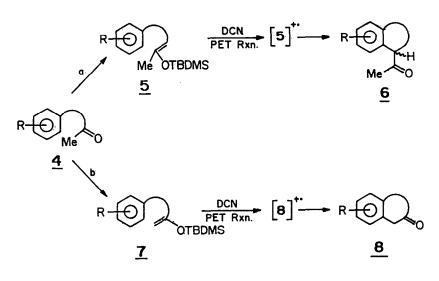
DCN = 1,4-dicyanonaphthalene

led us to envision addition of proximate silyl enol ether¹⁰ to arene radical cation for carbocyclisation purposes. We are pleased to disclose our preliminary result in this communication.

[#] NCL Communication No: 5834

From starting compound 4, two different types of silvl end ether ($\underline{5}$ and $\underline{7}$) can be obtained by following the reported¹¹ experimental procedures for the synthesis of thermodynamic and kinetic silvl end ethers, respectively. Therefore, cyclisations of these ethers are expected to produce two different types of carbocyclic compounds varying in ring sizes from the same ketone as shown in Scheme - II.

SCHEME-II



a = TBDMS·Cl, ImH, DMF, Δ , 48h b = LDA, THF, -78°C, TBDMS·Cl, 3h

Silylenol ethers of type 5 are prepared in about 83 % yield by heating a mixture of 4 (5 mmol), TBDMSCl (6 mmol) and imidazole (12 mmol) in DMF (10 ml) for 48h followed by the usual workup, while those of type 7 are synthesized quantitatively (98 %), by the reaction of TBDMSCl (6 mmol) on the lithium enolate of type 4 (5 mmol) generated by the reaction of LDA at -78°C. A typical photochemical reaction procedure consist of the irradiation of a mixture of silylenol ethers (2 mmol) and DCN 0.06 g (0.34 mmol) in 500 ml of CH₃CN:H₂O (4:1) for 3-4 h through pyrex filtered light (>280 nm all light absorbed by enolether only) using 450-W Hanovia lamp without removing the dissolved oxygen. Removal of the solvent and chromatographic purification of the reaction mixture gave carbocyclic products **19-24** (65-74 %). DCN was recovered quantitatively (98 %) at the end of the reaction¹². During the irradiation of silylenol ethers of type 7, minor quantity (~10%) of starting ketones 4 are also formed which has been shown to be formed by the thermal reversion of 7 by adequate control experiment. Generality of the strategy have been shown by studying a number of carboannulation reactions as depicted in **Table - I**.

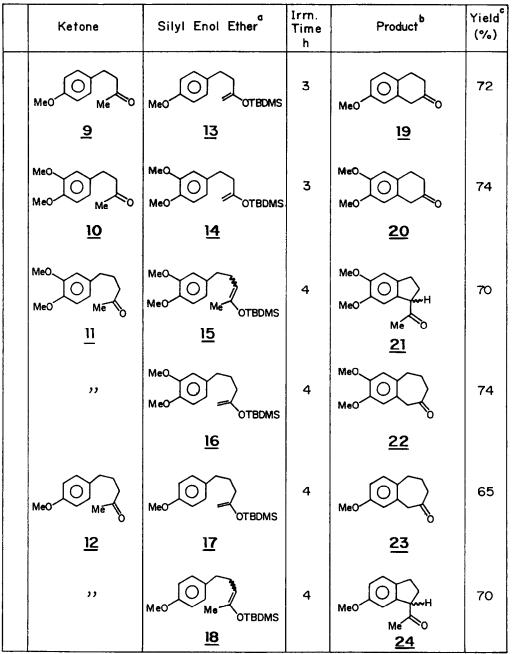


TABLE-I : Photoinduced Electron Transfer Promoted Carboannulation Reactions

a) Silyl enol ethers <u>15</u> and <u>18</u> are the mixture of E and Z isomers and have been used as such (b) Characterised by ¹H NMR, ¹³C NMR and mass spectral data. (c) yields not optimised.

The regiospecificity of these cyclizations are in accordance with the calculated electron densities (Huckel or MNDO) at different carbons of the HOMO of the arene radical cations as observed earlier¹².

In conclusion, we have demonstrated a novel carboannulation strategy which is flexible enough to incorporate amenable functionality. The attractive feature of this strategy lies in its ability to alter the ring size of carboannulated product from a single starting compound.

Acknowledgement: KG and MK are thankful to CSIR, New Delhi for the award of research fellowships.

REFERENCES

- a) Trost, B. M. Chem. Soc. Rev., 1982, 11, 141. (b) Posner, G. H. Chem. Rev., 1986, 86, 831. (c)
 Piers, E. Pure Appl. Chem., 1988, 60,107.
- (2) a) Taylor, S. K.; Blankespoor, C. L.; Harvey, S. M.; Richardson, L. J. J. Org. Chem., 1988, 53, 3309. (b) Manas, A. R. B.; Smith, R. A. J. Tetrahedron 1987, 43, 1847. (c) Trost, B. M. J. Am. Chem. Soc. 1979, 101, 257. (d) Birch, A. J.; Rao, G. S. R. S. Aust. J. Chem., 1970, 23, 547. (e) Posner, G. H.; Chapdelaine, M. J; Lentz, C. M. J. Org. Chem., 1979, 44, 3661. (f) Tamura, Y.; Choi, H. D.; Shindo, H.; Uenishi, J.; Ishibashi, H. Tetrahedron Lett. 1981, 22, 81. (g) Rigby, J. H.; Kotnis, A.; Kramer, J. Tetrahedron Lett. 1983, 24, 2939.
- (3) Shishido, K.; Yamashita, A.; Hiroya, K; Fukumoto, K, *Tetrahedron* **1989**, *45*, 5791. and references cited therein.
- (4) Larock, R. C.; Fried, C.A. J. Am. Chem. Soc., 1990, 112, 5882.
- (5) Citterio, A. J. Org. Chem., 1989, 54, 2703
- (6) a) Cornelisse, R. C., Lodder, G.; Havinga, E. Rev. Chem. Intermed., 1978, 2, 231. (b) Cornelisse, J.; Havinga, E. Chem. Rev., 1975, 75, 353.
- (7) Yoshida, K., "Electro oxidation in Organic Chemistry", Wiley & Sons, Inc.: New York, 1984.
- (8) a) Minisci, F., Top. Curr. Chem., 1976, 62, 1. (b) Minisci. F. Synthesis 1973, 1
- (9) a) Pandey, G.; Krishna, A.; Rao, M. J. *Tetrahedron Lett.* 1986, 27, 4075. (b) Pandey, G; Krishna, A. J. Org. Chem., 1988, 53, 2364. (c) Pandey, G.; Krishna, A; Bhalerao, U.T. *Tetrahedron Lett.* 1989, 30, 1867. (d) Pandey, G.; Sridhar, M.; Bhalerao, U. T. *Tetrahedron Lett.* 1990, 31, 5373.
- Silyl enol ethers are intermediates that have been frequently used for C-C bond formation reactions. For review see: a) Mukaiyama, T. Angew. Chem. Int. Ed. Engl., 1977, 16, 817., Org. React. 1982, 28, 203. (b) Rasmusseu, J. K. Synthesis 1977, 91. (c) Brownbridge, P. Synthesis 1983, 1 and 85.
- (11) a) House, H.O; Czuba, L. J; Gall, M.; Olmstead, H. D. J. Org. Chem., 1969, 39, 2324. (b) Fleming,
 I; Paterson, I. Synthesis 1979, 736.
- (12) For the details of DCN regeneration mechanism, see: Pandey, G. Top. Curr. Chem., 1993, 168, 000.

(Received in UK 19 July 1993; accepted 13 August 1993)