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PRACTICAL CHIRAL ROUTE TO MUSCARINE AND ITS THREE DIASTEREOMERS

Yukio OHSHIBA, Takehiko YOSHIMITSU, and Kunio OGASAWARA* Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980-77, Japan

> A practical route to all possible stereoisomers of the muscarine alkaloids in optically pure forms has been developed starting from a readily accessible chiral starting material.

> **KEY WORDS** chiral synthesis; muscarine; *epi*-muscarine; *allo*muscarine; epiallo-muscarine

The interesting physiological activity on the parasympathetic nervous system of (+)-(2S,3R,5S)muscarine (1), isolated from Amanita muscaria, has generated much interest over the years. 1) This has greatly stimulated chiral construction of muscarine (1) itself as well as three other naturally occurring diastereoisomers, (+)-(2S,3S,5S)-epi-muscarine (2), (+)-(2S,3R,5R)-allo-muscarine (3), and (-)-

Chart 1

(2S,3S,5R)-epiallo-muscarine (4). Although a number of chiral syntheses have been performed, 1) there has been no flexible method enabling us to produce all four diastereomers based on the same methodology. We report here a convenient method for the practical preparation of muscarine (1) and its three diastereomers in optically pure forms by employing cis-dihydroxylation reaction on the olefin

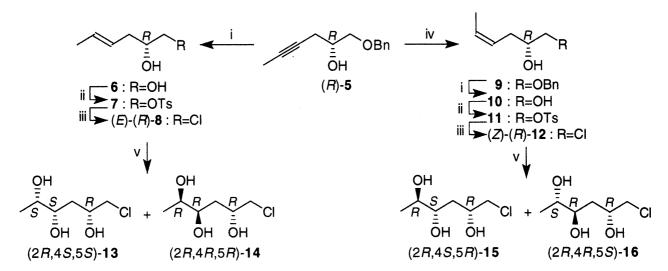


Chart 2. Reagents and conditions: i) Na, liq. NH₃ (84% for 6; 75% for 10). ii) p-TsCl, pyridine (90% for 7; 85% for 11). iii) LiCl, DMF (80% for 8; 96% for 12). iv) H₂, Lindlar cat., AcOEt (91%), v) see Table 1.

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substrates bearing chiral secondary hydroxyl functionality. The present procedure necessitates diastereomeric separation; however, all diastereomers of the muscarine alkaloids may be obtained in both enantiomeric forms in optically pure form by employing the same methodology.

Chart 3

Table 1. cis-Dihydroxylation of (E)-(R)-8 and (Z)-(R)-12

	Product (%)			et (%)				Product (%)	
Entry	Substrate	Oxidant	13	14	Entry	Substrate	Oxidant	15	16
1	8	OsO ₄ a)	25c)	38c)	4	12	OsO ₄	38	60
2	8	AD-mix- $\alpha^{b)}$	54	36	5	12	AD-mix-α	26	65
3	8	AD-mix- β b)	6c)	67 ^{c)}	6	12	AD-mix-β	33	62

- a) OsO₄ (1 mol%), K₃Fe(CN)₆ (3 eq), and K₂CO₃ (3 eq) in 50% aqueous tert-BuOH (10 ml/mmol of substrate).
- b) AD-mix reagent (1.4 g/mmol of substrate) was used in 50% aqueous tert-BuOH (10 ml/mmol of substrate).
- c) Yield was that of the corresponding tetrahydrofuran (18 from 13; 19 from 14) after the base-catalyzed cyclization.

The synthesis started with isomeric olefins E-8 and Z-12, which were prepared from the common acetylenic alcohol²⁾ 5 originating from chiral O-benzylglycidol. Thus, (R)-1-benzyloxy-4-hexyne-2-ol [(R)-5], obtained in 66% yield from (S)-O-benzylglycidol in two steps, 2) was transformed into either the E-olefinic chlorohydrin (E)-(R)-8 by sequential Birch reduction, mono-tosylation, and substitution or the Z-olefinic chlorohydrin (Z)-(R)-12 by sequential cis-hydrogenation, Birch reduction, mono-tosylation, and substitution, both in satisfactory overall yields. When they were subjected to three dihydroxylation conditions, both afforded the corresponding two diastereomers in good total yields, though in not high diastereoselectivity (**Table 1**). Noteworthy was the stereochemical difference between AD-mix- α^{4}) and AD-mix- β^{4}) in the reaction with (E)-(R)-8 where diastereoselection was exhibited only by the latter oxidant giving (2R,4R,5R)-14 as the major product, as expected by the empirical rule.⁴⁾ On the contrary, the former oxidant did not exhibited high diastereoselection, though the expected (2R,4S,5S)-13 was the major product. These may be taken in the former combination to be the "mismatched pairing⁵)" and the latter to be the "matched pairing⁵)." As expected,⁴) the (Z)-olefin (Z)-(R)-12 did not exhibit high diastereoselection; it furnished the mixture of (2R,4S,5R)-15 and (2R,4R,5S)-16 in about 1:2 ratio in an excellent total yield under all dihydroxylation conditions (Entries 4~6). The observed overwhelmed β -face selectivity (except in Entry 2) was presumed to be mostly due to the steric effect of the hydroxyl group present in 8 and 12.

Very fortunately, the product mixtures from both 8 and 12 were readily (or after a base-catalyzed cyclization) separable by column chromatography to give the corresponding tetrahydrofurans, 18~21, via the transient epoxide intermediates 17 on exposure of the separated glycol, 13~16, to aqueous base, respectively. Thus, (2R,4S,5S)-13 was treated with potassium carbonate (5 eq) in 50% aqueous tert-

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Chart 4

butanol at room temperature to afford the tetrahydrofuran (2S,3S,5S)-18, $[\alpha]_D^{31}$ +47.7° $(c=0.90, \text{CHCl}_3)$ [lit.6): $[\alpha]_D^{25}$ +51.6° (c 0.50, CHCl₃)], in 65% yield. Quite similarly, (2R,4R,5R)-14 gave (2R,3R,5S)-19, $[\alpha]_D^{30}$ -4.22° $(c=0.96, \text{CHCl}_3)$, in 61% yield, which then was treated with iodine in the presence of triphenylphosphine and imidazole⁷⁾ to give the iodide 22, $[\alpha]_D^{25}$ -42.2° $(c=1.1, \text{CHCl}_3)$ [lit.7): -33.12° $(c=0.88, \text{CHCl}_3)$: 81% ee], in 68% yield. (2R,4S,5R)-Triol (15) afforded the iodide 23, $[\alpha]_D^{29}$ -14.7° $(c=0.49, \text{CHCl}_3)$ [lit.8): $[\alpha]_D$ -10.64° $(c=1.3, \text{CHCl}_3)$: 81% ee], in 47% overall yield *via* (2R,3S,5S)-alcohol 20, $[\alpha]_D^{28}$ +45.0° $(c=0.50, \text{CHCl}_3)$, while (2R,4R,5S)-triol (16) afforded (2S,3R,5S)-21, $[\alpha]_D^{27}$ -6.44° $(c=0.42, \text{CHCl}_3)$ [lit.2): $[\alpha]_D^{22}$ -6.2° $(c=0.5, \text{CHCl}_3)$], in 69% yield.

Since natural (+)-epi-muscarine (2) from the (2S,3S,5S)- 18^6 and natural (+)-muscarine (1) from the (2S,3R,5S)- 21^2 have been obtained and since unnatural (-)-epiallo-muscarine (ent-4) from the (2R,3R,5S)-iodide⁸ (22) and unnatural (+)-allo-muscarine (ent-3) from the (2R,3S,5S)-iodide⁶ (23) have been obtained, the present procedure constitutes formal syntheses of all diastereomers of both the natural and the unnatural forms of the muscarine alkaloids using an appropriate chiral O-benzylglycidol.³⁾

REFERENCES AND NOTES

- Pertinent reviews: a) Wang P. -C., Joullie M. M., The Alkaloids, ed. Brossi A., Academic Press, New York, 1984, Vol. 23, pp. 327-380; b) Antkowiak R., Antkowiak W. Z., *ibid.*, ed Brossi A., Academic Press, New York, 1991, Vol. 40, pp. 189-340.
- 2) Takano S., Iwabuchi Y., Ogasawara K., J. Chem. Soc., Chem. Commun., 1989, 1371.
- 3) A review of enantiocontrolled natural product syntheses using chiral *O*-benzylglycidol: Takano S., Ogasawara K., *J. Syn. Org. Chem. Jpn.*, **47**, 813 (1989).
- 4) A pertinent review: Kolb H. C., VanNieuwenhze M. S., Sharpless K. B., *Chem. Rev.*, **94**, 2483 (1994).
- 5) Masamune S., Choy W., Petersen J., Sita L. R., Angew. Chem. Int. Ed. Engl., 24, 1 (1985).
- 6) Bandzouzi A., Chapleur Y., J. Chem. Soc., Perkin Trans. 1, 1987, 661.
- 7) Garegg P. J., Samuelsson B., J. Chem. Soc., Chem. Commun., 1979, 978.
- 8) De Amici M., De Micheli C., Molteni G., Pitre D., Carrea G., Riva S., Spezia S., Zetta L., J. Org. Chem., 56, 67 (1991).

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