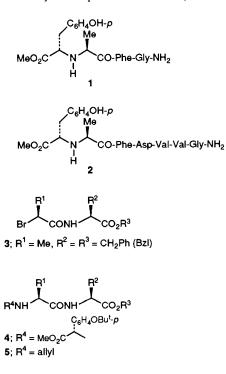
## 2-Bromoamides as Synthons for Pseudopeptides containing Aminodicarboxy Units

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The monoalkylating and enantioselective behaviour of a chiral 2-bromoamide allows the synthesis of pseudopeptides in which a dipeptide component is changed into an aminodicarboxy moiety, with overall retention of configuration.

The well known role of peptides in biology has prompted a wealth of synthetic and structure-activity studies of partially altered structures, *e.g.* pseudopeptides where the CONH link is replaced by NHCO,  $CH_2NH$ ,  $CH_2S$ ,  $CH_2CO$ , CSNH, *etc.*<sup>1</sup> We considered it interesting to provide a simple access to pseudopeptides where the NH<sub>2</sub> group of an amino acid or peptide provides the NH link of a built-in aminodicarboxy unit. Accordingly, we looked again at the early Fisher peptide synthesis where a 2-bromoacyl amino acid was allowed to react with ammonia,<sup>2a</sup> as well as the early synthesis of 'imino acids.'<sup>2b</sup> Aminodicarboxylic acids (imino acids, opines, alines, *etc.*), extensively studied in connection with some sea organisms, are obtained by nucleophilic substitution (with inversion



of configuration); $^{2b,3a,b}$  related compounds have been obtained upon amination-reduction (as racemates).<sup>4,5</sup> Following our recent demonstration that a 2-bromopropanamide reacts slowly with inversion of configuration with representative amines, but faster, and with retention of configuration, with the same amines in the presence of silver oxide,<sup>6</sup> we have now found that the free amino group of an amino acid ester substitutes the bromine of a chiral 2-bromoacylamino acid ester in the presence of Ag<sub>2</sub>O, with relevant retention of configuration.

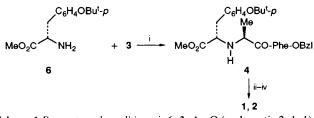
We believe that, under the present conditions, Ag<sub>2</sub>O behaves as a 'coupling agent' and promotes a neighbouring group mechanism, with the observed reaction rates and stereochemistry. The mechanism operating in the homogeneous phase<sup>7a</sup> as well as the role of Ag<sub>2</sub>O are under active investigation.<sup>7b</sup> We report as examples, the synthesis of: (a) two pseudopeptides 1 and 2<sup>†</sup> where the original Tyr<sup>1</sup>-D-Ala<sup>2</sup> moiety of the opioid tetrapeptide dermorphin<sup>8</sup> and heptapeptide deltorphin-C<sup>9</sup> have been changed into a related aminodicarboxy unit; and (b) a modified peptide 5 carrying a substituted *N*-terminal allylamino group.

<sup>†</sup> All new compounds gave satisfactory analytical and spectral data. Selected data for 1: m.p. 179–181 °C;  $[\alpha]_D^{20}$  + 7.6 (*c* 1, DMF); *K'* 5.54 (capacity factor) was determined using Vydac C<sub>18</sub> with a gradient consisting of two mobile phases: B = 60% acetonitrile in 0.1% CF<sub>3</sub>CO<sub>2</sub>H; A = 10% acetonitrile in 0.1% CF<sub>3</sub>CO<sub>2</sub>H. A 25 min linear gradient was run from 0% B to 50% B in 25 min, flow rate 1.0 ml min<sup>-1</sup>,  $\lambda$  = 220 nm; FAB-MS (MH<sup>+</sup>) *m*/z 471; amino acid analysis: Phe 0.97, Gly 1.0 [PITC methodology; peptides (50–1000 pmol) were hydrolysed in 200 µl 6 mol 1<sup>-1</sup> HCl containing 1% phenol for 1 h at 150 °C].

Selected data for 2: m.p.  $151-153 \,^{\circ}$ C;  $[\alpha]_D^{20} + 10.8 (c 1, DMF); K' 8.28; FAB-MS (MH<sup>+</sup>) <math>m/z$  784; amino acid analysis: Phe 1.02, Asp 0.97, Val 1.89, Gly 1.0.

Intermediate 4. <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>):  $\delta$  1.12 (d, 3H), 1.32 (s, 9H), 2.69 (d, 2H), 2.81–3.16 (m, 3H), 3.34–3.49 (m, 1H), 3.6 (s, 3H), 4.69–4.8 (m, 1H), 5.17–5.22 (m, 2H) and 6.8–7.37 (m, 15H); yield 80%,  $R_f$  0.43 (n-hexane–AcOEt, 4:1 v/v).

*N*-Allylamine **5**. <sup>1</sup>H NMR:  $\delta$  1.19 (d, 3H), 1.25 (br s, 1H), 2.98–3.22 (m, 5H), 4.86–5.2 (m, 4H), 5.6–5.85 (m, 1H), 7.02–7.36 (m, 10H) and 7.66 (d, 1H); yield 89%,  $R_{\rm f}$  0.2 (n-hexane–AcOEt, 1:1 v/v).



Scheme 1 Reagents and conditions: i,  $6:3: Ag_2O$  (molar ratio 2:1:1), room temp., 3 h, toluene, 75%; ii, H<sub>2</sub>, Pd/C, EtOH, room temp., 1 h, 95%; iii, N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDC), Et<sub>3</sub>N, hydroxybenzotriazole (HOBt), dimethylformamide (DMF), H-Gly-NH2·HBr or H-Asp(OBu<sup>t</sup>)-Val-Val-Gly-NH<sub>2</sub>,<sup>13</sup> -5 °C to room temp., 16 h, 71%; iv, CF<sub>3</sub>CO<sub>2</sub>H (95%), 0 °C, 1 h, 100%. Compounds indicated in the text as 1', 2' etc. bear the opposite configuration of the alanine unit.

(S,S)-2-Bromopropanoyl-Phe-OBzl 3 was prepared by acylating Phe-OBzl with (S)-2-bromopropanoyl chloride. The latter was obtained, in turn, from L-alanine, via the diazonium salt and the 2-bromoacid, with overall retention of configuration.<sup>10</sup> The diastereoisomeric mixture of **3** and (R)-2-bromopropanoyl-Phe-OBzl 3' was obtained, in turn, starting from commercial (R,S)-2-bromopropanoyl bromide (Fluka).

The monoalkylating and enantioselective behaviour of a chiral non-racemic 2-bromoamide synthon becomes apparent when the mixture 3, 3', or pure 3 is independently allowed to react with  $Tyr(Bu^t)$ -OMe 6, in the presence of  $Ag_2O$  (Scheme 1). Whereas two diastereoisomeric products, *i.e.* (S,S,S)-4 and (S,R,S)-4' were obtained starting from the diastereoisomeric mixture 3, 3', only one diastereoisomer was obtained starting from diastereoisomerically pure 3, as confirmed by careful HPLC screening. We assign the (S, S, S)-configuration to the diastereoisomer 4 arising from 3, by assuming that the Ag<sub>2</sub>O-promoted substitution of bromine occurs with retention of configuration. We believe that traces of the undesired diastereoisomer are due to impurities in the reagent rather than to a leak in the mechanism of the substitution at the Ca-Br bond.6.7a.b

Intermediate 4, including an aminodicarboxylic unit, was debenzylated at the C-terminal benzyl ester function and condensed with the C-terminal part of dermorphin tetrapeptide or deltorphin-C, with no protection of the novel secon-dary amino group. The resulting 1 and 2 have the final (S,S)-configuration of the new moieties, as shown. The diastereoisomeric 1', 2' with the related (S,R)-configurations were also obtained, starting from the diastereoisomeric mixture 4, 4'; preparative HPLC allowed the separation of the final products.

The N-allylamine 5 was obtained upon reaction of allyl-

amine with 3 or 3, 3'; no HPLC separation could be achieved, in this case, for the diastereoisomeric mixture.

The four compounds 1, 2; 1', 2' were tested in quantitative opioid binding assays: 1' and 2' proved better<sup>11</sup> than compounds containing the D-Tyr1-D-Ala2 unit.12

In conclusion, in pseudopeptides such as 1 and 2, the secondary amino function results from the amino acid reacting as a nucleophile with the bromoamide, whereas the peptide linkages result from routine peptide synthesis. The scope and limitations of our findings are under investigation.

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