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**A NEW ACCESS TO "ISO"-BETA-IONONE FROM  
DELTA-PYRONENE**

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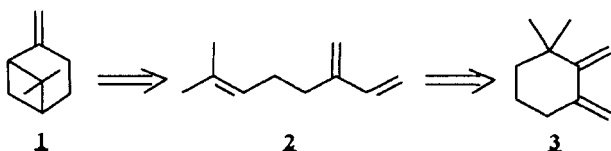
**Abstract :** New uses of delta-pyronene as an intermediate to terpenic synthesis have been studied. In particular, we proposed a new way to "iso"-beta-ionone from delta-pyronene.

We recently reported<sup>1</sup> that delta-pyronene **3** could easily be obtained from myrcene **2**, an available raw material from beta-pinene **1**, which is itself a constituent of turpentine (Scheme 1).

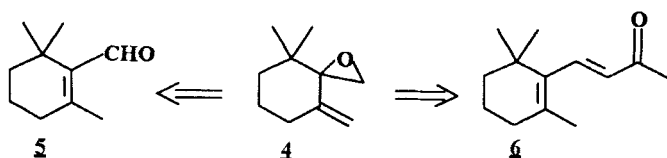
Preliminary work<sup>2</sup> concerning the reactivity of delta-pyronene **3** showed that, via its mono-epoxide **4**, this compound is an interesting intermediate in the synthesis of beta-cyclocitral **5** and beta-ionone **6** (Scheme 2).

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Scheme 1



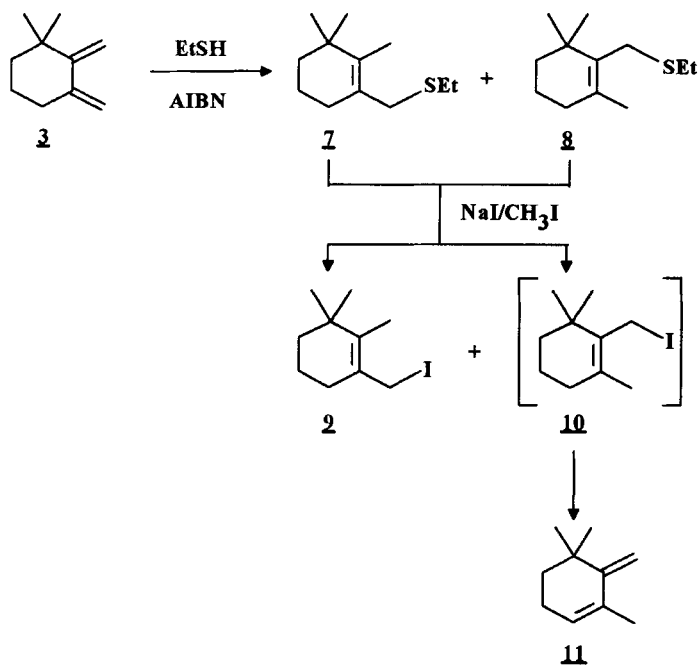
Scheme 2

The latter two are key compounds of industrial interest for the manufacture of perfumes, vitamin A and other retinoids<sup>3-6</sup>.

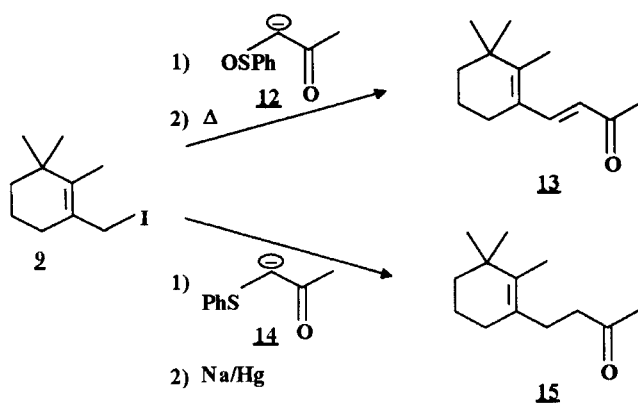
In this paper, we propose other potential applications for delta-pyrone in order to obtain new terpenic compounds with a cyclogeranyl skeleton.

The reaction of delta-pyrone **3** with ethanethiol<sup>7</sup> in the presence of AIBN thus produces a 60/40 mixture of allylic sulfides **7** and **8**, resulting from 1,4-addition reactions on the dienic system. The yield of this reaction is 95% (Scheme 3).

Treatment of the major allylic sulfide **7** with NaI/CH<sub>3</sub>I in an acetonitrile solution<sup>8</sup> quantitatively gives the iodide derivative **9** with 80% yield of isolated product after silica gel chromatography. This iodide derivative **9** may be used in alkylation reactions with C<sub>3</sub> units in order to obtain new cyclogeranyl derivatives with a C<sub>13</sub> skeleton (Scheme 4).



Scheme 3



Scheme 4

Homologation with the anion of keto-sulfoxide **12** followed by a 50°C elevation of the temperature led to the  $\alpha,\beta$ -unsaturated ketone **13**, isomer of the beta-ionone **6** with a 72% yield. The major interest of this synthetic route is the possible direct obtention of "iso"-beta-ionone **13** from delta-pyrone (with an overall yield of 35%) without isolating either the allylic sulfide **7** or the iodide derivative **9**. We have noticed indeed that the other iodide isomer **10**, expected from substitution of allylic sulfide **8**, underwent an elimination reaction leading to gamma-pyrone **11**. This hydrocarbon compound was easily extracted from the crude reaction mixture by evaporation under reduced pressure.

In a similar alkylation reaction, we realised the homologation of iodide derivative **9** with the anion of keto-sulfide **14**, followed by a desulfurization of the intermediate adduct with an Na/Hg amalgam. We obtained the "iso"-dihydrobeta-ionone **15**, isomer of the dihydrobeta-ionone with an overall yield of 32%.

Regarding the reactivities of delta-pyrone derivatives for the preparation of new terpenic compounds with a cyclogeranyl skeleton further work is in progress.

### **Experimental section**

The keto-sulfide **14** is prepared from chloracetone by the method described by S. Warren<sup>9</sup>. Its oxidation with sodium metaperiodate in MeOH/H<sub>2</sub>O solution yields quantitatively the keto-sulfoxide **12**.

**Addition of ethanethiol to  $\delta$ -pyrone.** A mixture of 4.08 g (30 mmoles) of  $\delta$ -pyrone, 500 mg of AIBN, and 4.4 ml (60 mmoles) of ethanethiol was heated, under N<sub>2</sub>, at 50°C for 5 h. After cooling, the mixture was diluted

with ether, washed with 1 N NaOH solution and water and dried over  $\text{MgSO}_4$ . Removal of solvent gave 5.645 g (95%) of a 60/40 mixture of sulfides **7** and **8**.

**7** :  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 250 MHz)  $\delta$  1.00 (s,6H), 1.25 (t,3H), 1.65 (s,3H), 2.47 (q,2H), 3.14 (s,2H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 62.5 MHz)  $\delta$  13.4 (q), 15.0 (q), 19.4 (t), 25.6 (t), 28.0 (2q), 29.7 (t), 34.9 (s), 35.1 (t), 39.5 (t), 126.3 (s), 137.2 (s).

**8** :  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 250 MHz)  $\delta$  1.06 (s,6H), 1.28 (t,3H), 1.73 (s,3H), 2.55 (q,2H), 3.20 (s,2H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 62.5 MHz)  $\delta$  14.8 (q), 19.3 (t), 20.2 (q), 27.7 (t), 28.7 (2q), 30.6 (t), 32.8 (t), 34.6 (s), 39.5 (t), 131.9 (s), 133.3 (s).

**Conversion of sulfide 7 to iodide derivative 9.** 523 mg (2.6 mmoles) of sulfide **7** was stirred with 4 ml of methyl iodide and 530 mg of sodium iodide in 3 ml of acetonitrile for 3 h. After removal of solvent, the mixture was filtered and washed with  $\text{CCl}_4$ . The solvent was evaporated and the residue was chromatographed on silica gel with petroleum ether to give 550 mg (80%) of iodide derivative **9**.

**9** :  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 250 MHz)  $\delta$  1.06 (s,6H), 1.75 (s,3H), 2.31 (t, 2H), 3.50 (s, 2H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 62.5 MHz)  $\delta$  15.2 (q), 19.6 (t), 28.1 (2q), 32.5 (t), 35.5 (s), 39.5 (t), 58.7 (t), 124.6 (s), 142.7 (s).

**Synthesis of "iso"-beta-ionone 13.** 1.346 g (7.4 mmoles) of keto-sulfoxide **12** in 8 ml dry THF was added dropwise to 150 mg (6 mmoles) sodium hydride suspended in 23 ml THF by vigorous stirring at room temperature, under  $\text{N}_2$ . There was a rapid evolution of hydrogen to give the yellow anion. 1.573 g (6 mmoles) of iodide derivative **9** was syringed in and stirring

continued for 4 h, ammonium chloride solution added and the aqueous layer extracted with chloroform. The organic fractions were dried ( $\text{MgSO}_4$ ) and evaporated. The crude product was then dissolved in 25 ml  $\text{CCl}_4$  and heated to reflux for 4 h. The reaction mixture was concentrated in vacuo and chromatographed on  $\text{Al}_2\text{O}_3$  (6%  $\text{H}_2\text{O}$ ) to yield 830 mg (72%) of "iso"beta-ionone **13**.

**13** : IR  $\nu$  ( $\text{cm}^{-1}$ ) 1660 ( $\text{C}=\text{O}$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 250 MHz)  $\delta$  1.00 (s, 6H), 1.84 (s, 3H), 2.22 (s, 3H), 6.03 (m, 1H), 7.63 (m, 1H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 62.5 MHz)  $\delta$  14.2 (q), 18.7 (t), 26.2 (t), 27.5 (q), 27.7 (2q), 36.3 (s), 38.7 (t), 125.0 (d), 127.2 (s), 142.4(d), 151.3(s), 199.1(s).

**Synthesis of "iso"-dihydrobeta-ionone 15.** In a similar manner we realised the alkylation of keto-sulfide **14** (800 mg; 4.31 mmoles) with the iodide derivative **9** (1.016g; 4.81 mmoles) to afford 850 mg (73%) of alkylated product. This compound was then added, under  $\text{N}_2$ , to a mixture of 4.5 g pulverized 6% sodium amalgam and 2.976 g anhydrous disodium hydrogen phosphate in 60 ml dry methanol at  $0^\circ\text{C}$ . The solution was stirred for 1 h, poured into an ammonium chloride solution and extracted with chloroform. After usual work-up the crude product was chromatographed on  $\text{Al}_2\text{O}_3$  (6%  $\text{H}_2\text{O}$ ) to give 437 mg (80%) of pure "iso"-dihydrobeta-ionone **15**.

**15** : IR  $\nu$  ( $\text{cm}^{-1}$ ) 1713 ( $\text{C}=\text{O}$ );  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 250 MHz)  $\delta$  0.87 (s, 6H), 1.56 (s, 3H), 2.13 (s, 3H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 62.5 MHz)  $\delta$  13.0 (q), 19.5 (t), 28.0 (2q), 28.3 (t), 29.9 (q), 30.4 (t), 34.4 (s), 39.5 (t), 42.2 (t), 128.3 (s), 134.3 (s), 209.1 (s).

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