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## Corrigendum

## Corrigendum to "Lessons learned and lessons to be learned for developing homogeneous transition metal complexes catalyzed reduction of N<sub>2</sub> to ammonia" [J. Organomet. Chem. 752 (2014) 44–58]



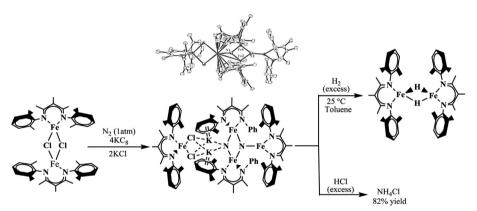
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This paper is a corrigendum to the previously published paper (J. Organomet. Chem. 752 (2014) 44–58): "Lessons learned and lessons to be learned for developing homogeneous transition metal complexes catalyzed reduction of N<sub>2</sub> to ammonia".

The authors regret to inform that the incorrect Scheme 8 has been published in the original article; therefore we would like to publish the correct Scheme 8.

The correct Scheme 8 is shown below:



Scheme 8. Functionalization of dinitrogen into ammonia using iron catalyst and HCl.

Relevant to the Scheme 8, reaction with excess HCl indeed gives ammonium chloride in 82% yield, however does not give the hydride complex. Nevertheless the bis-nitride complex does react with  $H_2$  to give the hydride complex, but no ammonia is produced in this reaction. Therefore in conclusion, the nitride compound does react with  $H_2$  to give hydride, but not ammonia. The nitride compound reacts with acid to give ammonia, but not hydride complex.

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