The Intramolecular Nitrile Oxide Cycloaddition Route to Forskolin

Pier Giovanni Baraldi,ª Achille Barco,*b Simonetta Benetti,b Gian Piero Pollini,*a Eleonora Polo,b and Daniele Simonia

- a Dipartimento di Scienze Farmaceutiche, Via Scandiana 21, Università di Ferrara, l-44100, Italy
- ^b Dipartimento Chimico, Via L. Borsari 46, Università di Ferrara, I-44100, Italy

A strategy for the construction of the AB ring system of forskolin based on [3 + 2] nitrile oxide intramolecular cycloaddition is reported.

Forskolin (1), the major diterpenoid isolated from the Indian plant Coleus forskohlii, 1 is a challenging synthetic target owing to its unusual structure with the presence of eight asymmetric centres, the high degree of functionalization and, last but not least, the important biological properties.² Recently three conceptually similar approaches to a functionalized AB ring system of (1), based on the use of an intramolecular Diels-Alder reaction have been reported,3-5 prompting us to describe our preliminary results in this area. Our strategy involves the utilization of intramolecular nitrile oxide cycloaddition, a reaction employed in the synthesis of natural products only comparatively recently,6 but continuously growing in both popularity and practicability.⁷

Commercially available α -damascone (2) underwent tetramethylguanidine-catalysed nitromethane addition to afford a 95% yield of the Michael adduct (3).† Treatment of (3) under Mukaiyama conditions8 for generating a nitrile oxide furnished a 20% yield of (4) [m.p. 95—97 °C (from light petroleum)], in which some of the required stereochemistry has been introduced. However, the cis-ring junction is clearly not required in the final molecule, but epimerization to the trans-decalin could be envisaged by taking advantage of the suitably located carbonyl group. The stereochemistry of the methyl group at C-8 is controlled by a sterically less congested transition state, which minimizes methyl-methyl interaction.⁷ It poses no problem since further elaboration will require the

with concomitant epimerization to give (5) [m.p. 73-74 °C (from n-pentane)] in 55% yield. Removal of the acetal moiety by the usual aqueous acid treatment led quantitatively to the trans-fused ketone (6) [m.p. 117—118 °C (from Et₂O-light petroleum, 1:1)]. Alteration of the molecule to produce a more favourable geometry for the cycloaddition step was

formation of a double bond between this centre and C-7. Acetalization of (4) under standard conditions proceeded

[†] All new compounds gave satisfactory spectral and analytical data. Yields are not optimized.

achieved by reducing (3) to the alcohol (7) (AlH₃; Et₂O; 0 °C). The corresponding acetyl derivative (8), obtained quantitatively by a standard procedure (Ac₂O; pyridine; room temp.; 6 h), underwent intramolecular cycloaddition on the trisubstituted double bond (PhNCO; Et₃N; 80 °C; 48 h) to afford a 90% yield of (9) as a mixture of diastereoisomers. Saponification (K₂CO₃; MeOH-H₂O, 3:1; reflux; 4 h) gave the alcohol (10), which, without purification, was transformed by Moffatt oxidation into an inseparable mixture (1:1) of ketones (4) and (6), which was directly acetalized to produce (5) in a less straightforward but more convenient pathway.

Unmasking of the heterocyclic nucleus to reveal the β-hydroxy ketone by known hydrogenolytic methods⁹ preserving the protective group (W-2 Raney nickel, acetic acid or boric acid) led surprisingly to the isolation of the stable β-hydroxy imine intermediate (11) (m.p. 99—100 °C‡) which, however, can be hydrolysed to the crucial intermediate (12) [m.p. 103—104 °C (from n-pentane)] by stirring for 12 h in the presence of SiO₂. In conclusion a highly functionalized AB ring system along the route to forskolin (1) has been synthesized, and its elaboration to the target compound is currently under active investigation.

We acknowledge Boehringer Mannheim GmbH for support of this work, and Firmenich, Milano, for a generous gift of α -damascone.

Received, 6th February 1986; Com. 172

References

- 1 S. V. Bhat, B. S. Bajwa, H. Dornauer, N. J. de Souza, and H. W. Fehlhaber, Tetrahedron Lett., 1977, 1669.
- 2 K. B. Seamon, Annu. Rep., Med. Chem., 1984, 19, 293.
- 3 P. R. Jenkins, K. A. Menear, P. Barraclough, and M. S. Nobbs, J. Chem. Soc., Chem. Commun., 1984, 1423.
- 4 K. C. Nicolaou and W. S. Li, J. Chem. Soc., Chem. Commun., 1985, 421.
- 5 F. E. Ziegler, B. H. Jaynes, and M. T. Saindane, Tetrahedron Lett., 1985, 3307.
- 6 G. Desimoni, G. Tacconi, A. Barco, and G. P. Pollini, 'Natural Products Synthesis through Pericyclic Reactions,' A.C.S. Monograph, 180, Washington, 1983.
- A. P. Kozikowski, Acc. Chem. Res., 1984, 17, 410.
- 8 T. Mukaiyama and T. Hoshino, J. Am. Chem. Soc., 1960, 82, 5339.
- 9 D. P. Curran, J. Am. Chem. Soc., 1983, 105, 5826.

[‡] To the best of our knowledge this is first example of isolation of the labile β-hydroxy imine intermediate during the transformation of Δ^2 -isoxazolines to β -hydroxy ketones.