

PII: S0040-4039(96)02371-4

An Enantiocontrolled Synthesis of Pyrrolizidines, (-)-Platynecine and (-)-Hadinecine

Sung Ho Kang ,* Geun Tae Kim and Yong Sang Yoo

Department of Chemistry, Korea Advanced Institute of Science and Technology, Taejon 305-701, Korea

Abstract : Trisubstituted allylic alcohols 13 and 14 have been converted into a single isomeric trans-oxazoline 16 via an intramolecular iodoamidation of the corresponding trichloro-acetimidates, which have been elaborated into (-)-platynecine 1 and (-)-hadinecine 2 via a common intermediate pyrrolizidine 3. © 1997, Elsevier Science Ltd. All rights reserved.

The pyrrolizidine alkaloids are a class of the most prevalent naturally occurring compounds in flowering plant families.¹ The alkaloids are composed of necine bases and carboxylic acids by ester linkages, which display a variety of biological activities such as antitumor, carcinogenic, hepatotoxic, antiinflammatory, hypotensive, local anesthetic and antispasmodic property.² Necine bases, having the 1-azabicyclo[3. 3. 0]octane ring system as a common structural feature, comprise (+)-retronecine,³ (-)-platynecine,^{3b, 3d, 7c} (-)-hadinecine,⁴ (+)-heliotridine,^{3d, 3f, 3g, 5} (-)-turneforcidine,^{3d} (+)-hastanecine^{3d, 5b, 6, 7a} and so forth.⁷ Their intriguing molecular structures and their potentially valuable pharmacological properties led us to be interested in their synthesis. In this paper we discolse our enantioselective synthetic route to (-)-platynecine 1 and (-)-hadinecine 2.

On the basis of the retrosynthetic analysis toward 1 and 2, the bicyclic pyrrolizidine with an exocyclic double bond 3 was chosen as a common key synthetic intermediate. Since 3 could be derived from a vicinal *threo* amino hydroxy derivative, we planned to secure the corresponding amino alcohol by employing an intramolecular iodoamidation⁸ of trichloroacetimidate generated from allylic alcohol 4.



A 7 : 1 inseparable mixture of benzylidenes 5 and 6 from D-malic acid⁹ was oxidized under Swern conditions¹⁰ and the resulting aldehydes were olefinated using phosphonate 7,¹¹ which was prepared in 89% yield by Arbuzov reaction of α -bromo- γ -butyrolactone in triethylphosphite at 120°C (Scheme 1). The desired conjugated lactones 8, $[\alpha]_D^{21}$ -35.9 (*c* 1.0, CHCl₃) and 9, $[\alpha]_D^{21}$ -10.2 (*c* 1.0, CHCl₃) were produced in a ratio of 1.3 to 1 along with lactones 10 from 5-membered benzylidene 6 in 90% combined overall yield. After the mixture of 8-10 was reduced with DIBAL followed by sodium borohydride, the resulting diols were acetylated and then the remaining



<u>Reagents</u>: i. BH₃·SMe₂, B(OMe)₃, THF, 0~20°C. ii. PhCHO, *p*-TsOH, PhMe, Dean-Stark trap, 140°C. iii. Swern ox. iv. 7, DBU, LiCl, MeCN, 0°C. v. DIBAL, CH_2Cl_2 , -78°C, then MeOH, NaBH₄, 0°C. vi. Ac₂O, DMAP, Et₃N, CH_2Cl_2 , 0°C. vii. AcOH-H₂O (4:1), 20°C. viii. NaIO₄, MeOH-H₂O (4:1), 20°C. ix. TBSCl, imidazole, DMF, -40°C.

benzylidene groups were hydrolyzed in aqueous acetic acid to yield a mixture of 1,3- and 1,2-diols. For the facile purification, 1,2-diols were removed by treatment of the mixture with sodium periodate to afford the desired 1,3-diols **11**, $[\alpha]_D^{17}$ -16.0 (*c* 1.0, MeOH) and **12**, $[\alpha]_D^{20}$ -12.6 (*c* 1.0, MeOH) in 70% overall yield from the lactones. The regioselective silylation of **11** and **12** with *t*-butyldimethylsilyl chloride (TBSCI) furnished monosilyl ethers **13**, $[\alpha]_D^{18}$ -14.4 (*c* 1.0, CHCl₃) and **14**, $[\alpha]_D^{16}$ -13.5 (*c* 1.0, CHCl₃) in 93% yield along with 4% of disilyl ethers.

After chromatographic separation of **13** and **14**, each was subjected to trichloroacetonitrile in the presence of DBU and then the generated allylic trichloroacetimidate was cyclized using iodine in the presence of potassium carbonate to give a 1:1 mixture of *trans*-oxazolines **15** in 76% and 89% yield from **13** and **14**, respectively (Scheme 2). It is worth noting that the identical *trans*-oxazolines **15** were formed from **13** and **14**, and that any appreciable amount of *cis*-oxazoline could not be isolated. Interestingly, it was ascertained that the isomeric positions of **15** were the carbons adjacent to iodine but not to nitrogen. More conveniently, the mixture of **13** and **14** was converted into **15** in 82% yield under the described iodoamidation conditions. When **15** reacted with zinc in the presence of ammonium chloride in aqueous *t*-butanol for the reductive elimination of its β -iodo acetate group,¹² its trichloromethyl group was concomitantly reduced to methyl group to provide a single olefinic *trans*-oxazoline **16**, $[\alpha]_D^{16} + 81.1$ (*c* 1.0, CHCl₃). Compound **16** was completely deprotected with methanolic HCl and the subsequent double cyclization of the resultant amino diol was effected using carbon tetrachloride and triphenylphosphine in the presence of triethylamine



<u>Reagents</u>: i. Cl_3CCN , DBU, MeCN, 0°C. ii. I_2 , K_2CO_3 , MeCN, 0~20°C. iii. Zn, NH₄Cl, *t*-BuOH-H₂O (4:1), 0~20°C. iv. 6N HCl-MeOH (1:5), 20°C. v. CCl₄, Ph₃P, Et₃N, DMF, 20°C. vi. BH₃-THF, THF-CH₂Cl₂(2:1), 0°C, then basic H₂O₂, EtOH, 50°C. vii. 6N HCl-MeOH (1:1), 50°C. viii. OsO₄, NMO, acetone-H₂O (4:1), 0°C.

in DMF¹³ to produce the bicyclic pyrrolizine 3,⁴ $[\alpha]_D^{16}$ -96.5 (*c* 0.52, CHCl₃) in 51% overall yield from **15**. Treatment of **3** with borane•THF followed by basic hydrogen peroxide furnished platynecine-borane complex, which was decomplexed with hot methanolic HCl to afford (-)-platynecine **1**, mp 147-148°C, $[\alpha]_D^{18}$ -60.3 (*c* 0.30, CHCl₃) in 79% yield. On the other hand, dihydroxylation of **3** with a catalytic amount of osmium tetroxide in the presence of N-methyl-morpholine N-oxide (NMO)^{4, 14} gave (-)-hadinecine **2**, $[\alpha]_D^{18}$ -67.8 (*c* 0.39, MeOH) in 80% yield.

For the identification of the synthetic materials, (-)-platynecine 1 and the olefinic pyrrolizidine 3 were prepared from the commercially available crotaline by the known procedures.^{4, 15} While 1 and 3 from our synthesis were identical with those from crotaline in all aspects,¹⁴ the spectroscopic data of our synthetic (-)-hadinecine 2 were quite different from those in the literature.^{16, 17} When the synthetic (-)-hadinecine 2 was converted into (-)-hadinecine•HCl salt, mp 154-156°C, $[\alpha]_D^{18}$ -69.1 (*c* 0.48, MeOH), its spectroscopic data matched with the reported.¹

To sum up, we have established an efficient synthetic route to pyrrolizidine alkaloids exemplified by (-)-platynecine and (-)-hadinecine, which has culminated in the enantiospecific intramolecular amidation of trichloroacetimidates from trisubstituted allylic alcohols **13** and **14**.

Acknowledgement : This work was supported by the Organic Chemistry Reserch Center sponsored by the Korea Science and Engineering Foundation.

References and Notes

- a) Robins, D. J. Prog. Chem. Org. Nat. Prod. 1982, 41, 115~203. b) Robins, D. J. Nat. Prod. Rep. 1984, 1, 235~243 ; 1985, 2, 213~220 ; 1986, 3, 297~305 ; 1987, 4, 577~590.
- 2. a) Mattocks, A. R. in Chemistry and Toxicology of Pyrrolizidine Alkaloids, Acedemic Press,

London, U. K. 1986. b) Rizk, A. M. in *Naturally Occurring Pyrrolizidine Alkaloids*, CRC Press, Boca Raton, 1991. c) Robins, D. J. Nat. Prod. Rep. 1992, 9, 313~321; 1994, 11, 613~619.

- a) Rüeger, H.; Benn, M. Heterocycles 1982, 19, 23~25. b) Rüeger, H.; Benn, M. Heterocycles 1983, 20, 1331~1334. c) Buchanan, J. G.; Singh, G.; Wightman, R. H. J. Chem. Soc., Chem. Commun. 1984, 1299~1300. d) Chamberlin, A. R.; Chung, J. Y. L. J. Org. Chem. 1985, 50, 4425~4431. e) Buchanan, J. G.; Jigajinni, V. B.; Singh, G.; Wightman, R. H. J. Chem. Soc., Perkin Trans. 1, 1987, 2377~2384. f) Kametani, T.; Yukawa, H.; Honda, T. J. Chem. Soc., Chem. Commun. 1988, 685~687. g) Kametani, T.; Yukawa, H.; Honda, T. J. Chem. Soc., Perkin Trans. 1, 1990, 571~577.
- 4. Hanselmann, R.; Benn, M. Tetrahedron Lett. 1993, 34, 3511~3514.
- a) Chamberlin, A. R.; Chung, J. Y. L. J. Am. Chem. Soc. 1983, 105, 3653~3656. b) Choi, J. -K.; Hart, D. J. Tetrahedron 1985, 41, 3959~3971. c) Kano, S.; Yuasa, Y.; Shibuya, S. Heterocycles 1988, 27, 253~256 d) Dener, J. M.; Hart, D. J. Tetrahedron 1988, 44, 7037~7046.
- a) Danishefsky, S.; McKee, R.; Singh, R. K. J. Am. Chem. Soc. 1977, 99, 7711~7713. b) Hart, D. J.; Yang, T.-K. J. Chem. Soc., Chem. Commun. 1983, 135~136.
- a) Mulzer, J.; Scharp, M. Synthesis 1993, 615~622. b) Robins, D. J. Nat. Prod. Rep. 1995, 12, 413~418. c) Hashimura, K.; Tomita, S.; Hiroya, K.; Ogasawara, K. J. Chem. Soc., Chem. Commun. 1995, 2291~2292. d) Denmark, S. E.; Thorarensen, A.; Middleton, D. S. J. Am. Chem. Soc. 1996, 118, 8266~8277.
- 8. Kang, S. H.; Kim, G. T. Tetrahedron Lett. 1995, 36, 5049~5052.
- 9. Herradon, B. Tetrahedron: Asymmetry 1991, 2, 191~194.
- 10. Mancuso, A. J.; Swern, D. Synthesis 1981, 165~185.
- 11. Blanchette, M. A.; Choy, W.; Davis, J. T.; Essenfeld, A. P.; Masamune, S.; Roush, W. R.; Sakai, T. Tetrahedron Lett. 1984, 25, 2183~2186.
- 12. Larock, R. C. in Comprehensive Organic Trasformation, VCH, New York, 1989, 138.
- a) Appel, R. Angew. Chem. Int. Ed. Engl. 1975, 14, 801~811. b) Stoilova, V.; Trifonov, L. S.; Orahovats, A. S. Synthesis 1979, 105~106 c) Johnson, C. R.; Golebiowski. A.; Sundram, H.; Miller, M. W.; Dwaihy, R. L. Tetrahedron Lett. 1995, 36, 653~654.
- 14. VanRheenen, V.; Kelly, R. C.; Cha, D. Y. Tetrahedron Lett. 1976, 1973~1976.
- The physical data of 3 have not been reported in literatures. For the physical data of 1, see
 a) Adams, R.; Van Duuren, B. L. J. Am. Chem. Soc. 1954, 76, 6379~6383 b) Rodgers, M.; Robins, D. J. J. Chem. Soc., Perkin Trans. 1, 1989, 2437~2439.
- 16. Were, O.; Benn, M.; Munavu, R. M. J. Nat. Prod. 1991, 54, 491~499.
- 17. All new compounds showed satisfactory spectral data. Spectroscopic data for 2 and 3: 2:¹H NMR (300MHz, $Py-d_s$) δ 1.94-2.19(3H, m), 2.65(1H, td, *J* 10.6, 12.0Hz), 3.08(1H, br t, *J* 9.7Hz), 3.19-3.30(1H, m), 3.72-3.78(1H, m), 3.94(1H, dt, *J* 7.0, 10.7Hz), 4.09(1H, br s), 4.41(1H, d, *J* 11.0Hz), 4.53(1H, d, *J* 11.0Hz) and 4.74(1H, br s); ¹³C NMR (75.5MHz, Py-d₅) δ 35.9, 37.2, 54.6, 54.9, 65.7, 70.0, 80.4 and 82.3 ; IR v_{max} /cm⁻¹(neat) 3337, 2942, 1418, 1318, 1174, 1117, 1030 and 832. 3: ¹H NMR (300MHz, CDCl₃) δ 1.84-2.07(2H, m), 2.47-2.55(2H, m), 2.60-2.68(1H, m), 2.77(1H, ddd, *J* 6.5, 10.9, 9.8Hz), 3.03-3.16(2H, m), 3.88-3.92(1H, m), 4.11-4.15(1H, m), 4.86(1H, td, *J* 2.1, 1.8Hz) and 5.16(1H, td, *J* 1.9, 1.9Hz) ; ¹³C NMR (75.5MHz, CDCl₃) δ 34.6, 35.6, 52.9, 54.9, 72.1, 73.9, 107.8 and 148.0 ; IR v_{max} /cm⁻¹(neat) 3292, 3076, 2915, 1663, 1434,1332, 1175, 1128, 1093, 1016, 886 and 833.