

0957-4166(94)00139-1

Gold(I)-Catalyzed Asymmetric Aldol Reactions of Fluorinated Benzaldehydes with an α-Isocyanoacetamide

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Abstract: The use of N,N-dimethyl- α -isocyanoacetamide instead of methyl α -isocyanoacetate in the gold(l)catalyzed asymmetric aldol reactions with polyfluorinated benzaldehydes was found to improve both diastereo- and enantioselectivity in the formation of *trans*-oxazolines.

Recently we have reported the unusual stereochemical outcome in the gold(I)-catalyzed asymmetric aldol reaction² of fluorinated benzaldehydes 1 with methyl α -isocyanoacetate (2a).³ Thus, the successive substitution of hydrogen atoms by fluorine in the phenyl ring of benzaldehyde brought about gradual increase of both the yield of *cis*-oxazolines 4 and their enantiomeric purity, whilst the ee's of *trans*-oxazolines 5 significantly decreased. The extremes of the trend were observed in the aldol reaction of penta- and tetrafluorobenzaldehydes 1a and 1b with 2a where corresponding *cis*-oxazolines 4 were formed with higher than 60% *cis*-selectivity and with up to 90% enantioselectivity (eq 1). In this communication we report the asymmetric aldol reaction of fluorobenzaldehydes 1a-f with N,N-dimethyl- α -isocyanoacetamide⁴ (2b) which improves both *trans*-selectivity in the oxazoline ring formation and % ee of the dominant *trans*-oxazolines 6 (eq 2).



Ar_f: C₆F₅ 2,3,5,6-F₄-C₆H 2,4,6-F₃-C₆H₂ 2,6-F₂-C₆H₃ 2-F-C₆H₄ 4-F-C₆H₄ C₆H₅

ent	ry Ar _f in	conditions ^a		yield ^t	ratio ^c	% eed	
	aldehyde	temp (°C)	time (h)	(%)	trans-6/cis-7	trans-6	cis-7
1	C ₆ F ₅ (1a)	15 ^e	48	82	77/23 (57/43)	80 (36)	20
2	C ₆ F ₅ (1a)	25	24 ^f	87	81/19	68	24
3	2,3,5,6-F4-C6H (1b)) 20	20	88	89/11	77	28
4	2,3,5,6-F4-C6H (1b)) 15 ^e	50	78	84/16 (47/53)	84 (48)	26
5	2,4,6-F ₃ -C ₆ H ₂ (1c)	10	48	83	85/15 (67/33)	91 (73)	48
6	2,6-F ₂ -C ₆ H ₃ (1d)	20	25	84	77/23 (75/25)	93 (86)	64
7	2-F-C ₆ H ₄ (1e)	22	25	87	83/13 (89/11)	93 (90)	-
8	4-F-C ₆ H ₄ (1f)	20	24	89	92/8 (93/7)	94 (96)	-
98	$C_{6}H_{5}(1g)$	25	25	74 ^h	94/6 (95/5)	94 (95)	-

 Table 1. Gold(I)-Catalyzed Asymmetric Aldol Reactions of Fluorobenzaldehydes 1a-f with

 N,N-Dimethyl-α-isocyanoacetamide (2b)

^a The reactions were carried out in dichloromethane under argon atmosphere. The gold catalyst was prepared in situ from $[Au(c-HexNC)_2]BF_4$ and chiral ligand 3. Ratio of 1/2b/Au(I)/3 =1.3/1/0.02/0.022 unless otherwise noted. b Isolated yield after passing reaction mixture through a short silica gel column (3 x 1.5 cm) using ethyl acetate as an eluent. ^c Determined by ¹H NMR analysis. See also ref. 8. Data previously reported on the Au(I)-catalyzed aldol reaction of corresponding fluorobenzaldehydes³ and benzaldehyde² with isocyanoacetate 2a are given in the parentheses. ^d Determined by HPLC analysis of N,N-dimethyl-2-(N-benzoylamino)-3-hydroxy-3-(fluorophenyl)propanamides 9a-f9 with a chiral stationary phase column (SUMICHIRAL OA-2000, 2000I, 4500, or 4900), hexane/1,2-dichloroethane/ethanol = 50/15/1, 100/20/1, or 250/20/1. The absolute configuration of **6a** is (4S,5R) (see text). All other *trans*-oxazolines **6b**f are assumed to have the same (45,5R) configuration by similarity in the order of elution in the condition of chiral HPLC analysis. The absolute configuration of the cis-oxazolines 7 was not determined. The highest % ee of trans-(45,5R)-oxazoline, previously reported on the Au(I)catalyzed aldol reaction of corresponding fluorobenzaldehydes³ and benzaldehyde² with isocyanoacetate 2a are given in the parentheses. e Reaction was started at 0 °C and then reaction temperature was allowed to rise to the indicated temperature. f One mol % of the catalyst was used. 8 Previously reported data, see ref. 4. h Yield of trans-oxazoline 6h, see ref. 4.

We started our experiments with the reaction of isocyanoacetamide $2b^5$ with pentafluorobenzaldehyde (1a) which was, in the light of the previous work,³ expected to display the most significant difference in the reactivity and selectivity from fluorine-free benzaldehyde. Thus, to a stirred solution of chiral ferrocenylphosphine ligand 3 (30.6 mg, 0.044 mmol), bis(cyclohexyl isocyanide)gold(I) tetrafluoroborate (20 mg, 0.040 mmol), and *N*,*N*-dimethyl- α -isocyanoacetamide (2b) (224 mg, 2.0 mmol) in 4 mL of freshly distilled dichloromethane at 0 °C was added under argon pentafluorobenzaldehyde (1a) (500 mg, 2.56 mmol). After mixing, the reaction mixture was stirred at 15 °C for 48 h. After removing the solvent under reduced pressure the crude reaction mixture was subjected to ¹H NMR analysis that has shown that the ratio of *trans*-oxazoline **6a** to *cis*-isomer **7a** is 77/23 (Table 1, entry 1). Silica gel column (3 x 1.5 cm) chromatography of the reaction mixture with ethyl acetate as eluent gave 503 mg (82%) of the oxazolines. The oxazolines **6a** and **7a** were converted into *N*,*N*-dimethyl-2-(*N*-benzoylamino)-3-hydroxy-3-(pentafluorophenyl)propanamides **9a** by hydrolysis with conc. HCl in methanol at 50 °C (2 h) followed by benzoylation of the resulting amine with

benzoyl chloride in the presence of triethylamine in dichloromethane. Chiral HPLC analysis of N-benzoyl derivatives 9a has shown that the enantiomeric purity of syn-isomer (derived from trans-oxazoline 6a) is 80% ee and that of anti-isomer is 20% ee (entry 1). Hydrolysis of trans-6a with 6 N HCl gave after dehydrochlorination⁶ with propylene oxide in methanol β -(pentafluorophenyl)serine 8a. Comparison of $[\alpha]_D$ value of amino acid 8a obtained ($[\alpha]_D^{25} + 12.1$ (c 0.5, 6 N HCl)) with the literature value (lit.⁷ for (2S,3R)-8a: $[\alpha]_D^{25} + 13.03$) establishes its (2S,3R)-absolute configuration and, consequently, the (4S,5R)-absolute configuration of trans-oxazoline 6a. At higher temperature (25 °C) (entry 2) the aldol reaction of isocyanoacetamide 2b with pentafluorobenzaldehyde (1a) proceeded with higher trans-selectivity (81%) but the % ee was lower (68%). Similar reactivity and stereochemical outcome were observed in the aldol reaction of 2,3,5,6-tetrafluorobenzaldehyde (1b) with amide 2b (entries 3 and 4). Thus, the highest trans-selectivity (89%) with 77% ee of oxazoline 6b was observed at 20 °C, whilst low temperature reaction gave trans-oxazoline 6b with 84% of diastereoselectivity and 84% ee.

In the aldol reactions of trifluoro- and difluorobenzaldehydes 1c and 1d with isocyanoacetamide 2b (entries 5 and 6), the desired *trans*-oxazolines 6c and 6d were obtained with higher enantioselectivity (91% ee for 6c and 93% ee for 6d) than in the reaction with isocyanoacetate 2a. These % ee's are in line with the highest levels of asymmetric induction which previously have been achieved in the gold(I)-catalyzed asymmetric aldol reaction.² Finally, the aldol reactions of isocyanoacetamide 2b with monofluoro-substituted benzaldehydes 1e,f gave expectedly high diastereo- and enantioselectivities (entries 7 and 8). However, the use of isocyanoacetamide 2b for the synthesis of corresponding β -(monofluorophenyl)serines 8, as well as fluorine free one (entry 9), through oxazolines 6e-g, seems to have no advantages over the application of isocyanoacetate 2a, which provides a rather better stereochemical outcome.^{2,3}

The present results have revealed that the stereochemical outcome of the gold(I)-catalyzed asymmetric aldol reactions of N,N-dimethyl- α -isocyanoacetamide (2b) with polyfluorobenzaldehydes 1a and 1b stands in marked contrast to that of isocyanoacetate 2a. Thus, amide 2b provides preferential formation and high % ee of *trans*-oxazolines, while ester 2a favors *cis*-diastereoselectivity and high % ee of *cis*-isomers. It follows that both *threo*- and *erythro*-(polyfluorophenyl)serines can be prepared selectively in high enantiomeric purity *via* gold(I)-catalyzed asymmetric aldol reaction by use of amide 2b and ester 2a, respectively. In the reaction of both amide 2b and ester 2a, the same *si*-face of enolate derived from isonitrile 2a or 2b is favourable for electrophilic attack of aldehyde. The rationale of the influence of the fluorine atoms on the stereochemistry of the aldol reaction requires futher investigation.

Acknowledgment. We thank the JSPS Foundation, Ministry of Education, Japan, for a Grant-in-Aid for Scientific Research, and Uehara Memorial Foundation for partial financial support of this work.

References and Notes

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- 8 Owing to an effect of aromatic ring, the methyl protons of dimethylamino group in cis-oxazolines 7 shifted upfield in comparison with those of the *trans*-oxazolines 6. ¹H NMR (δ , CDCl₃) for oxazolines 6 and 7: trans-6a, 3.02 (s, 3 H), 3.26 (s, 3 H), 4.86 (dd, J = 7.6 Hz, 2.1 Hz, 1 H), 6.59 (d, J = 7.6 Hz, 1 H), 6.89 (d, J = 2.1 Hz, 1 H). cis-7a, 2.77 (s, 3 H), 3.13 (s, 3 H), 5.43 (dd, J = 11.2 Hz, 2.1 Hz, 1 H), 5.98 (d, J = 11.2 Hz, 1 H), 7.08 (d, J = 2.1 Hz, 1 H). trans-6b, 3.01 (s, 3 H), 3.26 (s, 3 H), 4.89 (dd, J = 7.8 Hz, 2.1 Hz, 1 H), 6.62 (d, J = 7.8 Hz, 1 H), 6.88 (d, J = 2.1 Hz, 1 H), 7.02-7.19 (m, 1 H). cis-**7b**, 2.75 (s, 3 H), 3.11 (s, 3 H), 5.42 (dd, J = 11.1 Hz, 2.1 Hz, 1 H), 6.05 (d, J = 11.1 Hz, 1 H), 7.01 (d, J = 2.1 Hz, 1 H), 7.00-7.15 (m, 1 H). trans-6c, 2.99 (s, 3 H), 3.24 (s, 3 H), 4.84 (dd, J = 7.6 Hz, 1.00 Hz)2.0 Hz, 1 H), 6.54 (d, J = 7.6 Hz, 1 H), 6.88 (d, J = 2.0 Hz, 1 H), 6.60-6.75 (m, 2 H). cis-7c, 2.68 (s, 3 H), 3.15 (s, 3 H), 5.36 (dd, J = 11.2 Hz, 2.1 Hz, 1 H), 6.03 (d, J = 11.2 Hz, 1 H), 6.63-6.76 (m, 2 H), 7.10 (d, J = 2.1 Hz, 1 H). trans-6d, 2.94 (s, 3 H), 3.18 (s, 3 H), 4.83 (dd, J = 7.6 Hz, 2.3 Hz, 1 H), 6.53 (d, J = 7.6 Hz, 1 H), 6.85 (d, J = 2.3 Hz, 1 H), 7.44-7.58 (m, 3 H). cis-7d, 2.75 (s, 3 H), 2.91 (s, 3 H), 5.33 (dd, J = 11.3 Hz, 2.1 Hz, 1 H), 6.03 (d, J = 11.3 Hz, 1 H), 7.06 (d, J = 2.1 Hz, 1 H), 7.15-7.28 (m, 3 H). trans-6e, 3.00 (s, 3 H), 3.17 (s, 3 H), 4.71 (dd, J = 7.8 Hz, 2.1 Hz, 1 H), 6.22 (d, J = 7.8 Hz, 1 H), 6.93 (d, J = 2.1 Hz, 1 H), 7.05-7.30 (m, 4 H). cis-7e, 2.62 (s, 3 H), 2.88 (s, 3 H), 5.41 (dd, J = 10.2 Hz, 2.1 Hz, 1 H), 5.93 (d, J = 10.2 Hz, 1 H), 7.11-7.37 (m, 5 H). trans-6f, 2.90 (s, 3 H), 3.11 (s, 3 H), 4.53 (dd, J = 7.9 Hz, 2.1 Hz, 1 H), 6.06 (d, J = 7.9 Hz, 1 H), 6.88 (d, J = 2.1 Hz, ,1 H), 6.92-7.21 (m, 4 H). cis-7f, 2.63 (s, 3 H), 2.73 (s, 3 H), 5.20 (dd, J = 10.9 Hz, 2.0 Hz, 1 H), 5.49 (d, J = 10.9 Hz, 1 H), 6.91-7.20 (m, 5 H).
- 9 N-Benzoyl derivatives 9a-f were prepared by hydrolysis of corresponding oxazolines with conc. HCl in methanol followed by benzoylation with benzoyl chloride in the presence of triethylamine. ¹H NMR (δ , CDCl₃) for N-benzoyl derivatives 9a-g: syn-9a, 3.04 (s, 3 H), 3.28 (s, 3 H), 5.09 (d, J = 7.6 Hz, 1 H), 5.44 (dd, J = 7.6 Hz, 5.3 Hz, 1 H), 5.68 (dd, J = 7.3 Hz, 5.3 Hz, 1 H), 7.27 (d, J = 7.3 Hz, 1 H), 7.41-7.50 (m, 3 H), 7.71-7.75 (m, 2H). anti-9a, 3.03 (s, 3 H), 3.20 (s, 3 H), 4.61 (d, J = 4.4 Hz, 1 H), 5.43 (dd, J = 8.7 Hz, 3.3 Hz, 1 H), 5.57 (dd, J = 4.4 Hz, 3.3 Hz, 1 H), 7.33 (d, J = 8.7 Hz, 1 H), 7.44-7.56(m, 3 H), 7.76-7.79 (m, 2H). syn-9b, 2.95 (s, 3 H), 3.20 (s, 3 H), 4.96 (d, J = 7.6 Hz, 1 H), 5.40 (dd, J = 7.6 Hz, 5.7 Hz, 1 H), 5.62 (dd, J = 7.3 Hz, 5.7 Hz, 1 H), 6.91 (tt, J = 9.6 Hz, 7.6 Hz, 1 H), 7.19 (d, J = 7.3 Hz, 1 H), 7.33-7.45 (m, 3 H), 7.63-7.66 (m, 2H). anti-9b, 2.92 (s, 3 H), 3.12 (s, 3 H), 4.68 (bs, 1 H); 5.40 (m, 1 H), 5.51 (dd, J = 5.3 Hz, 3.6 Hz, 1 H), 6.95 (tt, J = 9.6 Hz, 7.6 Hz, 1 H), 7.31-7.47 (m, 4 H), 7.68-7.72 (m, 2H). syn-9c, 2.99 (s, 3 H), 3.21 (s, 3 H), 4.57 (d, J = 7.4 Hz, 1 H), 5.36 (dd, J = 7.4 Hz, 6.3 Hz, 1 H), 5.66 (dd, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (d, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (d, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (d, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (d, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (d, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (d, J = 7.9 Hz, 6.3 Hz, 1 H), 6.59 (t, J = 8.7 Hz, 2 H), 7.20 (t, J = 8.7 Hz, 2 H7.9 Hz, 1 H), 7.36-7.52 (m, 3 H), 7.67-7.70 (m, 2H). anti-9c, 2.95 (s, 3 H), 3.13 (s, 3 H), 4.41 (d, $J = 10^{-10}$ 5.0 Hz, 1 H), 5.42-5.46 (m, 2 H), 6.62 (t, J = 8.6 Hz, 2 H), 7.25-7.50 (m, 4 H), 7.76-7.79 (m, 2H). syn-9d, 2.96 (s, 3 H), 3.18 (s, 3 H), 4.60 (m, 1 H), 5.43 (dd, J = 7.6 Hz, 6.9 Hz, 1 H), 5.70 (dd, J =8.3 Hz, 6.9 Hz, 1 H), 6.77-6.88 (m, 2 H), 7.14-7.48 (m, 5 H), 7.62-7.71 (m, 2H). anti-9d, 2.91 (s, 3 H), 3.09 (s, 3 H), 4.22 (bs, 1 H), 5.50 (m, 2 H), 6.80 (m, 2 H), 7.05-7.21 (m, 1 H), 7.31-7.44 (m, 4 H), 7.71-7.74 (m, 2H). syn-9e, 2.95 (s, 3 H), 3.16 (s, 3 H), 4.75 (d, J = 1.6 Hz, 1 H), 5.33 (dd, J = 1.6 Hz, 1 H H H Hz, 1 H), 5.33 (dd, J = 1.6 Hz, 1 8.9 Hz, 2.0 Hz, 1 H), 5.37 (m, 1 H), 6.95-7.07 (m, 3 H), 7.12-7.21 (m, 1 H), 7.28-7.52 (m, 4 H), 7.55-7.63 (m, 2 H). syn-9f, 2.86 (s, 3 H), 2.90 (s, 3 H), 4.75 (d, J = 2.0 Hz, 1 H), 5.04 (dd, J = 4.3 Hz, 2.0 Hz, 1 H), 5.18 (dd, J = 8.7 Hz, 4.3 Hz, 1 H), 6.90-6.96 (m, 2 H), 7.09 (d, J = 8.7 Hz, 1 H) 7.31-7.44 (m, 5 H), 7.63-7.67 (m, 2 H).