

# GENERATION AND REARRANGEMENT OF A 1-NORBORNENE IN PYROLYSIS OF 1-SILA-2-OXYNORBORNANES

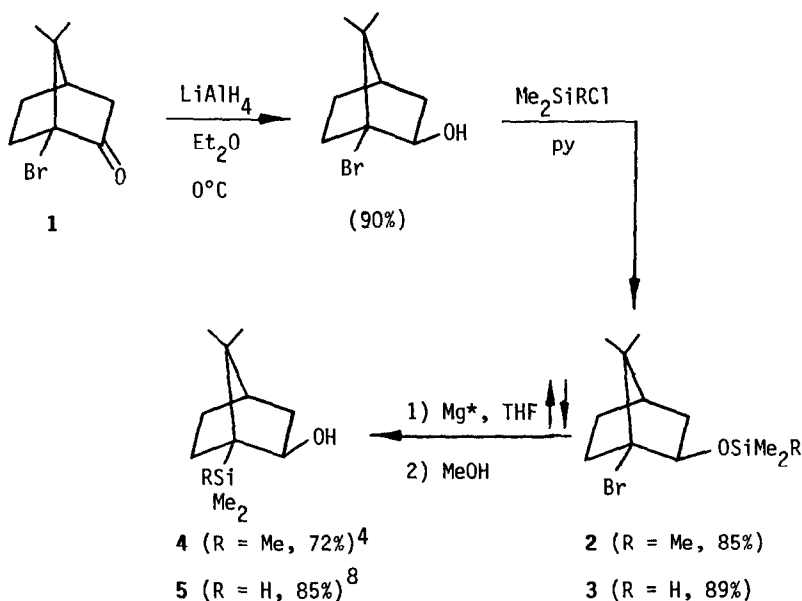
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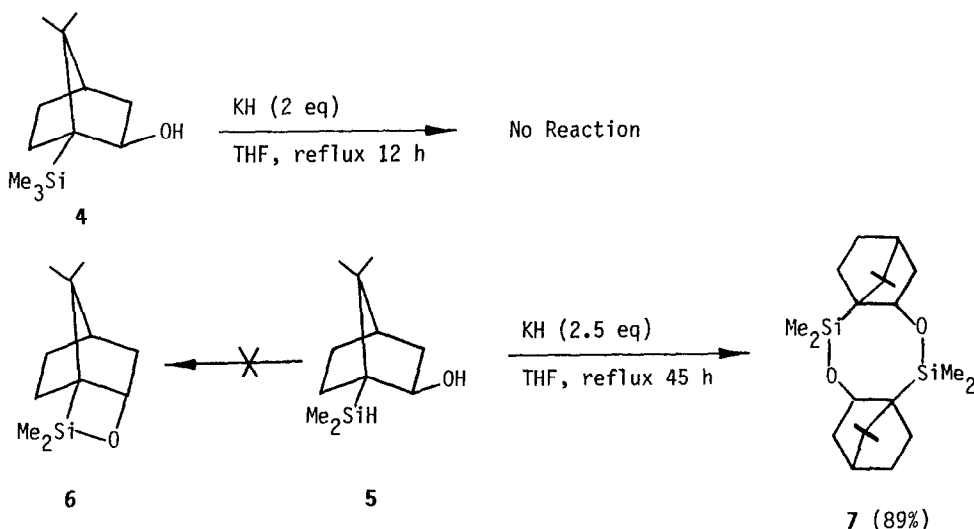
**Abstract:** Gas phase pyrolysis of 1-sila-2-oxybornenes yields a nortricyclene which is proposed to arise from a  $\beta$ -elimination to form a 1-norbornene, which isomerizes to a 2-norbornacylidene which undergoes  $\gamma$ -CH insertion.

Extension of Peterson Olefination to the conversion of  $\beta$ -hydroxysilanes into alkenes has proven to be an extremely useful procedure in organic synthesis.<sup>1</sup> We have begun to explore the possibility of using this procedure in the synthesis of anti-Bredt olefins. Toward this end, 1-trimethylsilyl-7,7-dimethylbicyclo[2.2.1]heptan-2-ol (**4**) was synthesized, starting with bromoketone **1**,<sup>2</sup> by the procedure outlined in Scheme 1.<sup>3</sup>

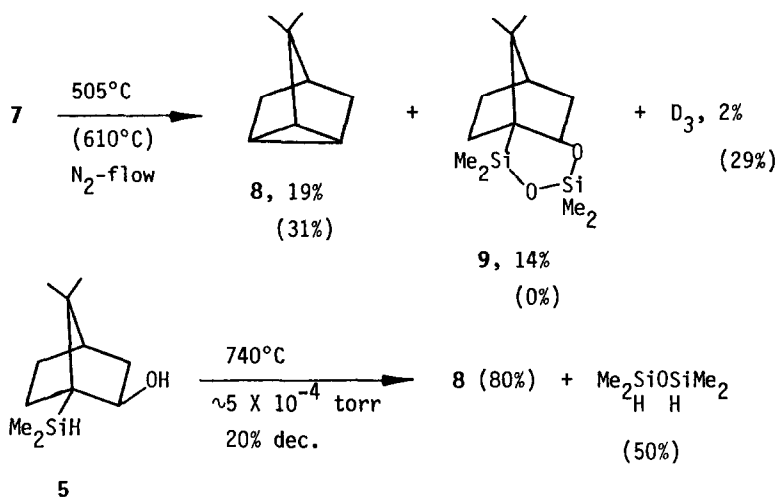
Scheme 1



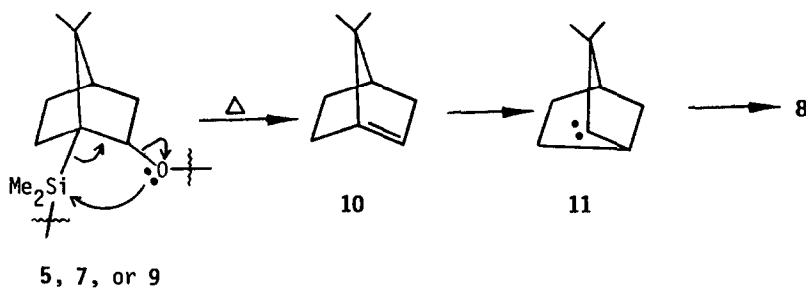
Since KH is well known to be an efficient reagent for the conversion of  $\beta$ -hydroxysilanes to olefins,<sup>7</sup> a THF mixture of **4** and a two-fold excess of KH was refluxed overnight. Analysis by GCMS revealed that virtually all of **4** remained intact, and thus a perhaps unsurprising limitation to this route to olefins was realized. In contrast, the reaction of KH and the dimethylsilyl analog **5**<sup>8</sup> appeared to begin upon mixing. It was hoped that **5** would undergo a base-induced intramolecular etherification to produce silaoxetane **6**. Instead, etherification occurred in a bimolecular fashion to afford the cyclic silyl ether **7** as an inseparable mixture of syn- and anti-isomers (4:5 by PMR) in 89% yield after refluxing for 45 hours.



Based on our earlier experience with cyclic ethers of this type,<sup>9</sup> we thought it possible that pyrolysis of **7** would result in olefin extrusion. Pyrolysis of **7** (30 mg) was conducted by slow, dropwise addition into a vertical, quartz chip packed tube at 505°C continuously swept with a 32 mL/min flow of nitrogen. Mass recovery was low due to the small sample size, but the GC trace of the pyrolysate was very clean, showing only two major products, nortri-cyclene **8** (19%) and cyclic disiloxane **9** (14%). Identification of **9** was accomplished by independent synthesis, while **8**<sup>10</sup> was spectrally compared with an authentic sample. Pyrolysis of **7** at 610°C raised the yield of **8** to 31%, completely eliminated **9** from the product mixture, and afforded hexamethylcyclodisiloxane, D<sub>3</sub>, in 29% yield along with a few percent of D<sub>4</sub>. Similar behavior was found in the flash vacuum pyrolysis (FVP) of **5** which at 740°C (ca.  $5 \times 10^{-4}$  torr) decomposed to the extent of 20% to afford **8** (16%, 80% corrected) and sym-tetra-methyldisiloxane (10%, 50% corrected). Likewise, FVP (760°C) of **4** produced **8** (17%) and hexamethyldisiloxane (28%), but this pyrolysis was much less clean than that of **5**.



As to the intriguing origin of nortricyclene 8, we can think of no alternative to intramolecular C-H insertion by 2-norbornanylidene 11. It is well established that such carbenes efficiently undergo  $\gamma$ -CH insertion to afford nortricyclenes.<sup>11</sup> The origin of 11 is less obvious, but we suggest that it arises from rearrangement of 1-norbornene 10 via a 1,2-alkyl shift. Although thermal isomerization of an olefin to a carbene is hardly a common reaction, we note a close precedent reported by Chan,<sup>12</sup> who has presented strong evidence that a bicyclo[2.2.2]oct-1-ene isomerized to a bicyclo[3.2.1]octanylidene by an analogous process. Thus, it is our current thinking that 5, 7, and 9 undergo thermally-induced elimination to produce 10<sup>14</sup> which owing to its considerable strain is prone to rearrangement to carbene 11 which cyclizes to the observed product. We are currently attempting to extend this route to other strained olefins.



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- The key step of this sequence is the conversion of the  $\beta$ -bromoether **2** to the desired  $\beta$ -silyl alcohol **4**.<sup>4</sup> Following the work of Speier,<sup>5</sup> this was accomplished in 72% yield by the action of activated magnesium<sup>6</sup> in refluxing THF.
- 4**: mp 35-9°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.83-3.86 (m, 1H), 1.19 (s, 3H), 0.81 (s, 3H), 1.8-0.9 (m, 7H), 0.06 (s, 9H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  82.05, 48.79, 43.87, 42.77, 29.79, 27.39, 23.29, 22.90, -0.49; calcd for C<sub>12</sub>H<sub>24</sub>SiO m/e 212.15965, found m/e 212.15988.
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- 5**: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.93 (sept, 1H, J = 3.9 Hz), 3.84 (m, 1H), 1.8-0.9 (m, 7H), 1.19 (s, 3H), 0.90 (s, 3H), 0.15 (d, 3H, J = 3.9 Hz), 0.11 (d, 3H, J = 3.9 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  81.43, 48.47, 47.90, 43.09, 42.51, 29.73, 27.34, 22.58, 22.39, -4.14, -4.68; IR (neat) 3492 cm<sup>-1</sup> (O-H), 2102 cm<sup>-1</sup> (Si-H); calcd for C<sub>11</sub>H<sub>21</sub>SiO (M<sup>+</sup> - 1) m/e 197.13616, found m/e 197.13575. Anal. Calcd for C<sub>11</sub>H<sub>22</sub>SiO: C, 66.60; H, 11.18. Found: C, 66.49; H, 11.53.
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