

**Synthesis of *N,N*-Disubstituted 3-Oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazoles: Oxidative Debenzylation and Cyclization of 1,1,5-Trisubstituted *S*-Benzyl-2-thioisobiurets**

A. K. PANDEY, R. SINGH, V. K. VERMA\*

Applied Chemistry Section, School of Applied Sciences, Institute of Technology, Banaras Hindu University, Varanasi-221005, India

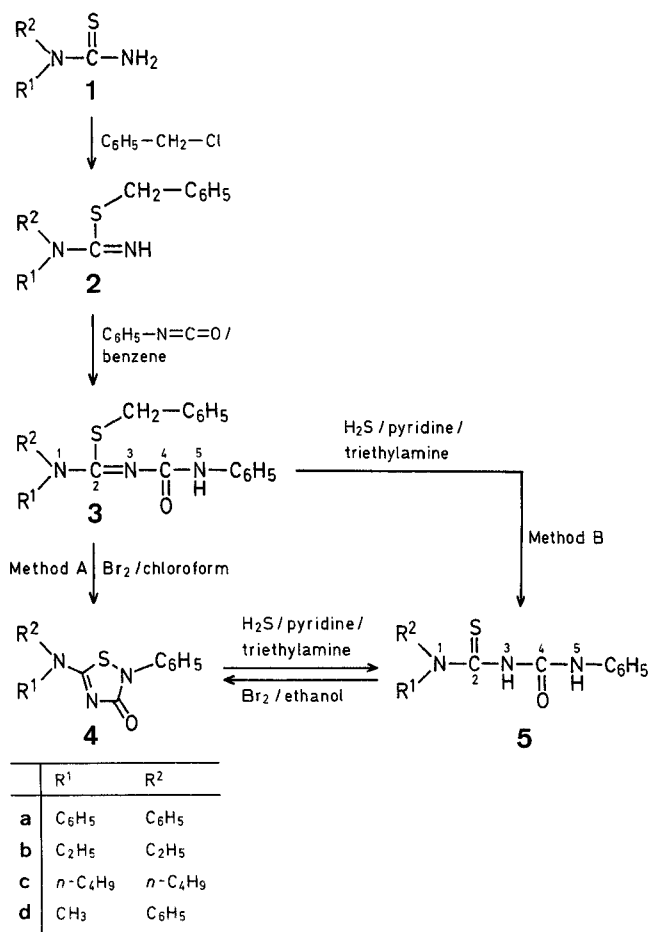
The chemistry of 1,2,4-thiadiazoles and their hydro derivatives has been extensively studied and reviewed<sup>1</sup>. 3-Oxo-2-phenyl-5-phenyliminotetrahydro-1,2,4-thiadiazole has been prepared by oxidation of a 1,5-diphenyl-2-thiobiuret<sup>2</sup>, and other 2- or 4-substituted 3-oxo-5-aryliminotetrahydro-1,2,4-thiadiazoles have been obtained from the reaction of chloro-(phenylimino)-methanesulfonyl chloride with urea or substituted ureas<sup>3</sup>, or from *S*-allyl-2-thioisobiurets<sup>4</sup>. However, there is no record of the synthesis of the related *N,N*-disubstituted 5-amino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazoles.

Our interest in the oxidative dealkylation and cyclization of N- and S-containing systems<sup>5</sup> led us to develop a method for

0039-7881/82/1232-1068 \$ 03.00

© 1982 Georg Thieme Verlag · Stuttgart · New York

the synthesis of the hitherto unknown *N,N*-dialkyl-, *N*-alkyl-*N*-aryl-, and *N,N*-diaryl-5-amino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazoles (**4**) by the oxidative debenzoylation and cyclization of *S*-benzyl-2-thioisobiurets (**3**) with bromine (Method A). We describe here the preparation of the starting materials **3** and their conversion into compounds **4**. The structure of compounds **4** was confirmed by microanalyses and I.R.-spectral data as well as by their alternative synthesis from compounds **3** via reductive debenzoylation to the thioisbiurets **5** and oxidative cyclization of compounds **5** (Method B) and by the reductive cleavage of compounds **4** to the thioisbiurets **5** with hydrogen sulfide/pyridine/triethylamine.



Melting points are uncorrected. I.R. spectra were recorded on a Perkin-Elmer spectrophotometer, model 720.

#### Benzyl *N,N*-Diphenylcarbamimidodithioate (**2a**); Typical Procedure<sup>6</sup>:

A solution of *N,N*-diphenylthiourea (**1a**; 10 g, 0.04 mol) and benzyl chloride (5.5 g, 0.04 mol) in ethanol (50 ml) is heated at 100 °C for 90 min and is then evaporated in vacuo to leave crude **2a** hydrochloride which on basification afforded the free base; yield: 8 g (80%); m.p. 80–81 °C (Ref.<sup>10</sup>, m.p. 81–82 °C).

#### *S*-Benzyl-1,1,5-triphenyl-2-thioisobiuret (**3a**); Typical Procedure:

A solution of benzyl *N,N*-diphenylcarbamimidodithioate (**2a**; 8 g, 0.025 mol) and phenyl isocyanate (3 g, 0.025 mol) in benzene (40 ml) is refluxed for 4 h, and then evaporated in vacuo. The semisolid product is stirred with petroleum ether (10 ml) and triturated with ethanol (10 ml) affording **3a**; yield: 6 g (75%); m.p. 234–235 °C.

#### *N,N*-Disubstituted 5-Amino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazoles (**4**); Typical and General Procedures:

##### Method A, Typical Procedure:

*5-Diphenylamino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazole* (**4a**): Bromine (1 ml) is gradually added to a well stirred suspension of *S*-benzyl-1,1,5-triphenyl-2-thioisobiuret (**3a**; 3 g, 0.006 mol) in chloro-

form (8 ml) until the color or bromine persists, benzyl bromide (lacrimatory fumes!) being developed in the reaction. Stirring is continued for 1 h at room temperature. The semisolid product is isolated by decantation of the solvent, washed with ether (10 ml), and triturated with ethanol. The resultant product **4a** is isolated by filtration and recrystallized from ethanol; yield: 2.2 g (75%); m.p. 284–285 °C.

##### Method B, General Procedure:

*1,1-Disubstituted 5-Phenyl-2-thioisbiurets* (**5**): The respective *S*-benzyl-5-phenyl-2-thioisobiuret **3** (0.009 mol) is dissolved in pyridine/triethylamine (6/1; 30 ml). A stream of dry hydrogen sulfide is passed through the solution for 4 h<sup>7</sup> and the mixture then poured onto ice (20 g) and acidified with dilute hydrochloric acid. The precipitated product **5** is isolated by suction and recrystallized from ethanol.

*5-Amino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazoles* (**4**): The 2-thioisbiuret **5** (0.01 mol) is dissolved in ethanol (15 ml) and bromine (2 ml, 10 mol) is added dropwise with stirring. Then, ether (0.025 ml) is added. The precipitated product **4** is isolated by suction and recrystallized from ethanol. [The products **4** thus obtained were identical (mixture m.p., superimposable I.R. spectra) with the corresponding products **4** obtained by Method A].

#### 1,1-Disubstituted 5-Phenyl-2-thioisbiurets (**5**) from the Reductive Cleavage of Compounds **4**; General Procedure:

The *N,N*-disubstituted 5-amino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazole (**4**; 0.01 mol) is dissolved in pyridine/triethylamine (6/1; 30 ml). A stream of dry hydrogen sulfide is passed through the solution for 5 h whereafter the mixture is cooled to 5 °C and acidified with dilute hydrochloric acid. The precipitated product **5** is isolated by suction and recrystallized from ethanol. [The products **5** thus obtained were found to be identical with the corresponding products **5** obtained from compounds **3** with hydrogen sulfide].

**Table 1.** 1,1,5-Trisubstituted *S*-Benzyl-2-thioisobiurets (**3**)

3	Yield [%]	m.p. [°C]	Molecular formula <sup>a</sup>	I.R. (nujol), $\nu$ [cm <sup>-1</sup> ]		
				C=O	C=N	NH
a	75	234–235°	C <sub>27</sub> H <sub>23</sub> N <sub>3</sub> OS (437.5)	1660	1600	3310
b	70	210–212°	C <sub>19</sub> H <sub>23</sub> N <sub>3</sub> OS (341.4)	1620	1580	3300
c	70	208–209°	C <sub>23</sub> H <sub>31</sub> N <sub>3</sub> OS (397.5)	1615	1590	3320
d	78	239–240°	C <sub>22</sub> H <sub>21</sub> N <sub>3</sub> OS (375.4)	1650	1598	3305

<sup>a</sup> The microanalyses were in good agreement with the calculated values: C,  $\pm 0.04$ ; H,  $\pm 0.00$ . Exception: **3b**, C – 0.51.

**Table 2.** *N,N*-Disubstituted 5-Amino-3-oxo-2-phenyl-2,3-dihydro-1,2,4-thiadiazoles (**4**)

4	Yield [%]	m.p. [°C]	Molecular formula <sup>a</sup>	I.R. (nujol), $\nu$ [cm <sup>-1</sup> ]	
				C=O	C=N
a	75	283–285°	C <sub>20</sub> H <sub>15</sub> N <sub>3</sub> OS (345.4)	1660	1610
b	70	270–272°	C <sub>12</sub> H <sub>15</sub> N <sub>3</sub> OS (249.3)	1620	1580
c	71	270–271°	C <sub>16</sub> H <sub>23</sub> N <sub>3</sub> OS (305.4)	1615	1570
d	78	243–245°	C <sub>15</sub> H <sub>13</sub> N <sub>3</sub> OS (283.3)	1650	1600

<sup>a</sup> The microanalyses were in good agreement with the calculated values: C,  $\pm 0.03$ ; H,  $\pm 0.03$ .

**Table 3.** 1,1,5-Trisubstituted 2-Thiobiurets (**5**)

<b>5</b>	Yield [%]	m.p. [°C]	Molecular formula <sup>a</sup>	I.R. (nujol), $\nu$ [cm <sup>-1</sup> ]	
				C=O	C=S
<b>a</b>	58	244–245°	C <sub>20</sub> H <sub>17</sub> N <sub>3</sub> OS (347.5)	1662	1290
<b>b</b>		<sup>b</sup>	C <sub>12</sub> H <sub>17</sub> N <sub>3</sub> OS (251.3)	1618	1300
<b>c</b>	55	217–218°	C <sub>16</sub> H <sub>25</sub> N <sub>3</sub> OS (307.4)	1615	1280
<b>d</b>	52	231–232°	C <sub>15</sub> H <sub>15</sub> N <sub>3</sub> OS (285.3)	1655	1250

<sup>a</sup> The microanalyses were in good agreement with the calculated values: C,  $\pm 0.01$ ; H,  $\pm 0.01$ .

<sup>b</sup> Semisolid.

Received: May 20, 1982

\* Address for correspondence.

<sup>1</sup> F. Kurzer, *Adv. Heterocycl. Chem.* **5**, 119 (1965).

<sup>2</sup> C. P. Joshua, *Indian J. Chem.* **1**, 391 (1963).

<sup>3</sup> N. Nimdeskar, M. G. Pranjpe, *J. Indian Chem. Soc.* **57**, 1123 (1980).

<sup>4</sup> A. K. Pandey, R. Singh, V. K. Verma, *Indian J. Chem. [B]* **21**, 150 (1982).

<sup>5</sup> R. Singh, V. K. Verma, *Indian J. Chem. [B]* **15**, 192 (1977); *J. Indian Chem. Soc.* **54**, 908 (1977).

R. Rai, V. K. Verma, *Indian J. Chem.* **18**, 284 (1979); *J. Indian Chem. Soc.* **55**, 1263 (1978).

<sup>6</sup> E. A. Werner, *J. Chem. Soc.* **57**, 295 (1890).

<sup>7</sup> A. E. S. Fairful, D. A. Peak, *J. Chem. Soc.* **1955**, 796.