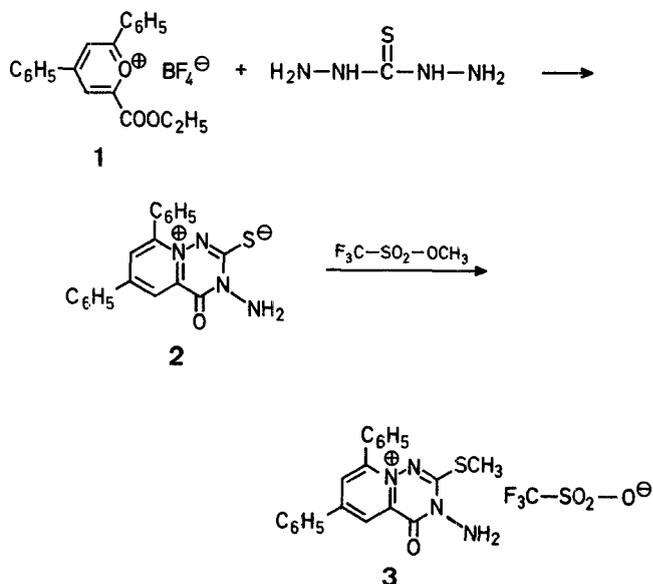


amines has been shown to give the corresponding pyridinium salts which undergo deethoxycarbonylation on treatment with *t*-butylamine^{3,4}. Similarly, compound **1** reacts with aldehyde and ketone hydrazones to yield the corresponding alkylidenaminopyridinium salts; the ketone derivatives cyclise under basic conditions to give oxidopyrido[1,2-*b*]pyridazinium betaines⁵. The reaction of **1** with phenylhydrazine and semicarbazide leads to dihydropyrazoloquinoline and 4-oxopyrido[2,1-*f*][1,2,4]triazinium-2-oxide derivatives, respectively⁵. Recently, we have reported the reaction of compound **1** with unsubstituted and *N*³-arylamidrazones to give 4-oxopyrido[2,1-*f*][1,2,4]triazinium betaines and 4-oxo-3,4-dihydropyrido[2,1-*f*][1,2,4]triazinium salts, respectively⁶.

The present paper records work which is part of our attempts to prepare bridgehead-nitrogen heterocycles from pyrylium (**1**) and amino compounds.

The pyrylium salt **1**, readily available from ethyl pyruvate, benzylideneacetophenone, and boron trifluoride-diethyl ether³, reacts with thiocarbohydrazide at room temperature in methanol to give 3-amino-4-oxo-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium-2-thiolate (**2**) as a crystalline solid in 84% yield. Compound **2** undergoes *S*-methylation on reaction with methyl triflate at room temperature in dry dichloromethane to give 3-amino-2-methylthio-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium triflate (**3**) as a crystalline solid in 82% yield.



Nitrogen-Bridgehead Heterocycles from 2-Ethoxycarbonyl-4,6-diphenylpyrylium Cation¹

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In spite of much work on the reaction of pyrylium cations with nitrogen nucleophiles², only few studies have been devoted to the reaction of 2-functionalised pyrylium cations, in particular, 2-ethoxycarbonyl pyrylium salts, with ambident nitrogen nucleophiles. The reaction of 2-ethoxycarbonyl-4,6-diphenylpyrylium tetrafluoroborate (**1**) with primary

Structures **2** and **3** are based on microanalytical data and spectral evidence. Compound **2** displays a clear I.R. absorption at $\nu = 1660 \text{ cm}^{-1}$ which is assigned to the carbonyl stretching vibration; the presence of a primary amino group is confirmed by the presence of two absorption bands at $\nu = 3270$ and 3160 cm^{-1} , respectively. The ¹H-N.M.R. spectrum shows two doublets for the pyridinium ring protons at $\delta = 9.25$ and 8.7 ppm ($J = 2 \text{ Hz}$), and a singlet at $\delta = 1.97 \text{ ppm}$ for the amino group. The ¹H-N.M.R. spectrum of compound **3** shows a singlet at $\delta = 2.15 \text{ ppm}$ attributable to the *S*-methyl group.

Compound **2** reacts with phenacyl bromides at reflux temperature in methanol (12 h) to give 3-aryl-6-oxo-8,10-diphenyl-2 *H*, 6 *H*-pyrido[2,1-*f*][1,3,4]thiadiazino[2,3-*c*][1,2,4]triazin-11-ium bromides (**4**) as yellow crystalline solids in high yields.

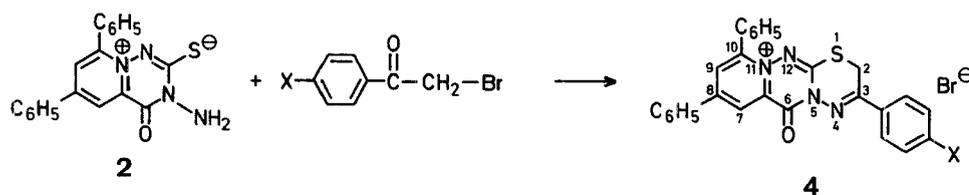


Table 1. 3-Aryl-6-oxo-8,10-diphenyl-2*H*,6*H*-pyrido[2,1-*f*][1,3,4]thiadiazino[2,3-*c*][1,2,4]triazin-11-ium Bromides (**4**) prepared

4 X	Yield ^a [%]	m.p. ^b [°C] (solvent), appearance	Molecular Formula ^c	M.S. ^d <i>m/e</i>	I.R. (Nujol) ^e ν [cm ⁻¹]	¹ H-N.M.R. (DMSO- <i>d</i> ₆ / TMS _{in}) ^f δ [ppm]
a H	76	200–202° (ethanol), yellow prisms	C ₂₇ H ₁₉ BrN ₄ OS (527.2)	447 (4), 415 (30), 388 (17), 298 (49), 230 (100), 203 (12), 202 (10), 135 (4), 116 (4), 115 (10), 105 (38), 82 (30), 81 (22), 80 (30), 79 (23), 77 (32)	1720, 1620, 1600, 1530, 1410, 1290, 1280, 1230, 1010, 910, 780, 770, 725, 720, 705	4.45 (s, 2H); 7.2–8.41 (m, 15H); 8.7 (d, 1H, <i>J</i> = 2 Hz); 9 (d, 1H, <i>J</i> = 2 Hz)
b Br	66	230–232° (ethanol), yellow prisms	C ₂₇ H ₁₈ Br ₂ N ₄ OS (606.1)	526 (2), 494 (19), 298 (39), 230 (68), 215 (2), 213 (2), 203 (32), 202 (33), 196 (2), 194 (3), 185 (96), 183 (100), 156 (36), 155 (36), 115 (8), 82 (42), 81 (24), 80 (42), 79 (22), 77 (23)	1730, 1620, 1590, 1530, 1410, 1295, 1230, 1210, 1080, 1025, 915, 910, 850, 785, 780, 720, 710	4.5 (s, 2H); 7.1–8 (m, 14H); 8.6 (d, 1H, <i>J</i> = 2 Hz); 8.9 (d, 1H, <i>J</i> = 2 Hz)
c Cl	70	233–235° (ethanol), yellow prisms	C ₂₇ H ₁₈ BrClN ₄ OS (561.7)	481 (3), 449 (22), 298 (51), 230 (100), 203 (20), 202 (21), 171 (4), 169 (4), 152 (4), 150 (4), 140 (45), 139 (45), 115 (5), 113 (12), 111 (13), 82 (96), 81 (28), 80 (97), 79 (28), 77 (8)	1720, 1620, 1590, 1530, 1490, 1400, 1350, 1290, 1230, 1090, 1070, 1000, 910, 905, 845, 820, 780, 730, 720, 705	4.45 (s, 2H); 7.4–8.5 (m, 14H); 8.8 (d, 1H, <i>J</i> = 2 Hz); 9 (d, 1H, <i>J</i> = 2 Hz)
d NO ₂	87	249–251° (ethanol/ether), orange prisms	C ₂₇ H ₁₈ BrN ₅ O ₃ S (572.2)	492 (4), 460 (17), 298 (15), 230 (14), 203 (13), 202 (14), 180 (11), 161 (13), 150 (14), 122 (11), 115 (12), 82 (97), 81 (45), 80 (100), 79 (45), 77 (14)	1730, 1620, 1570, 1520, 1400, 1350, 1290, 1270, 1200, 1070, 1025, 860, 820, 770, 750, 720, 690	4.45 (s, 2H); 7.2–8.5 (m, 14H); 8.75 (d, 1H, <i>J</i> = 2 Hz); 9 (d, 1H, <i>J</i> = 2 Hz)
e C ₆ H ₅	75	218–219° (ethanol/ether), green plates	C ₃₃ H ₂₃ BrN ₄ OS (603.3)	523 (5), 491 (21), 298 (15), 230 (37), 211 (11), 203 (13), 202 (14), 192 (13), 181 (100), 153 (32), 115 (5), 82 (67), 81 (30), 80 (65), 79 (30), 77 (9)	1710, 1620, 1590, 1520, 1435, 1400, 1290, 1270, 1230, 1215, 1070, 1030, 1005, 900, 840, 830, 785, 780, 770, 730, 715, 710, 645	4.45 (s, 2H); 7.2–8.4 (m, 19H); 8.75 (d, 1H, <i>J</i> = 2 Hz); 9 (d, 1H, <i>J</i> = 2 Hz)
f OCH ₃	75	211–213° (ethanol/ether), yellow plates	C ₂₈ H ₂₁ BrN ₄ O ₂ S (557.2)	477 (4), 445 (15), 298 (15), 230 (35), 203 (15), 202 (16), 165 (12), 146 (13), 135 (10), 115 (17), 107 (7), 82 (75), 81 (29), 80 (75), 79 (29), 77 (10)	1710, 1620, 1590, 1560, 1520, 1430, 1400, 1250, 1220, 1200, 1140, 1110, 1080, 1030, 1010, 1000, 955, 895, 845, 820, 775, 770, 755, 720, 685, 640	3.85 (s, 3H); 4.45 (s, 2H); 6.95–8.45 (m, 14H); 8.75 (d, 1H, <i>J</i> = 2 Hz); 9 (d, 1H, <i>J</i> = 2 Hz)

^a Yield of isolated pure product.

^b Uncorrected.

^c The microanalyses were in good agreement with the calculated values: C, ± 0.16 ; H, ± 0.21 ; N, ± 0.15 .

^d Recorded at 70 eV. on a Hewlett-Packard 5993 C instrument.

^e Recorded on a Nicolet FT 5DX spectrometer.

^f Recorded at 60 MHz on a Varian EM-360 A spectrometer.

Table 2. 1-Substituted 5-Oxo-7,9-diphenyl-1,5-dihydro[1,2,4]triazolo[5,1-*c*]pyrido[2,1-*f*][1,2,4]triazin-10-ium-2-thiolates (**5**) prepared

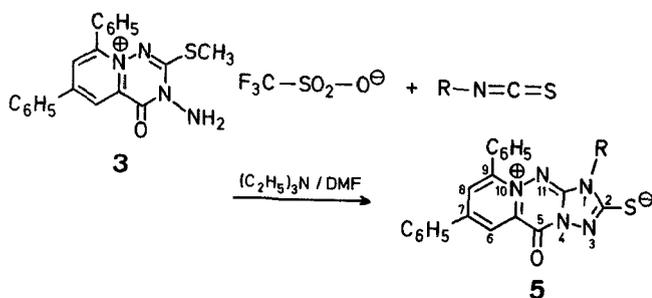
5 R	Yield ^a [%]	m.p. ^b [°C] (solvent), appearance	Molecular Formula ^c	M.S. ^d <i>m/e</i>	I. R. (Nujol) ^e ν [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃ /TFA/ (TMS) _{int}) ^f δ [ppm]
a C ₆ H ₅	58	274–275° (AcOH), orange prisms	C ₂₆ H ₁₇ N ₅ OS (447.3)	415 (19), 414 (37), 389 (27), 298 (19), 258 (7), 230 (91), 203 (66), 202 (66), 91 (16), 77 (66), 64 (100), 58 (26)	1710, 1620, 1540, 1500, 1435, 1200, 1185, 890, 805, 770, 750, 725, 695	7.15–8.2 (m, 15H); 8.5 (d, 1H, <i>J</i> = 2 Hz); 9.2 (d, 1H, <i>J</i> = 2 Hz)
b 4-Br–C ₆ H ₄	88	296–298° (AcOH), brown prisms	C ₂₆ H ₁₆ BrN ₅ OS (526.2)	494 (2), 467 (16), 298 (3), 258 (4), 230 (9), 203 (4), 202 (5), 77 (4), 64, 100, 58 (4)	1710, 1630, 1500, 1200, 1180, 1140, 990, 920, 830, 770, 755, 750, 710, 700	7.25–8.3 (m, 14H); 8.55 (d, 1H, <i>J</i> = 2 Hz); 9.25 (d, 1H, <i>J</i> = 2 Hz)
c 4-Cl–C ₆ H ₄	60	286–287° (AcOH), orange prisms	C ₂₆ H ₁₆ ClN ₅ OS (481.8)	450 (13), 424 (19), 298 (15), 258 (12), 230 (63), 203 (28), 77 (45), 64 (100), 58 (8)	1710, 1630, 1550, 1500, 1440, 1180, 1140, 1100, 920, 900, 830, 770, 750, 745, 710, 700	7.2–8.1 (m, 1H); 8.3 (d, 1H, <i>J</i> = 2 Hz); 9.05 (d, 1H, <i>J</i> = 2 Hz)
d 4-H ₃ C–C ₆ H ₄	60	300–302° (AcOH), orange prisms	C ₂₇ H ₁₉ N ₅ OS (461.3)	429 (10), 403 (7), 298 (5), 258 (5), 230 (20), 203 (13), 115 (8), 105 (10), 91 (16), 77 (40), 64 (100), 58 (21)	1720, 1630, 1520, 1480, 1440, 1340, 1190, 1150, 920, 890, 820, 770, 760, 750, 720, 700	2.45 (s, 3H); 7.2–8.2 (m, 14H); 8.55 (d, 1H, <i>J</i> = 2 Hz); 9.3 (d, 1H, <i>J</i> = 2 Hz)
e <i>c</i> -C ₆ H ₁₁	50	256–258° (AcOH), brown prisms	C ₂₆ H ₂₃ N ₅ OS (453.3)	497 (11), 471 (23), 298 (9), 258 (12), 230 (71), 203 (13), 202 (13), 77 (15), 64 (100), 58 (11)	1690, 1620, 1530, 1440, 1400, 1280, 1200, 1030, 1000, 925, 885, 840, 810, 770, 740, 695	1.4–2.3 (m, 11H); 7.2– 8.1 (m, 10H); 8.3 (d, 1H, <i>J</i> = 2 Hz); 9.1 (d, 1H, <i>J</i> = 2 Hz)

^{a, b} see Table 1.^c The microanalyses were in good agreement with the calculated values: C, ±0.22; H, ±0.22; N, ±0.17.^{d, e, f} see Table 1.

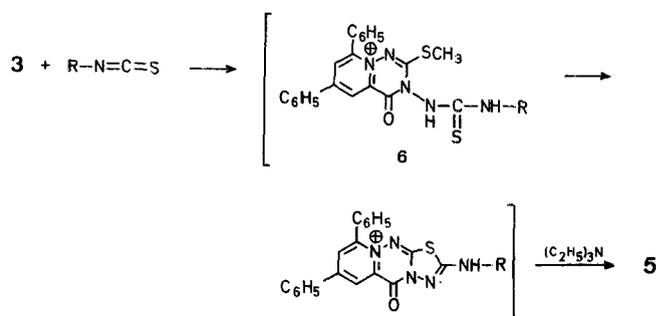
Compounds **4** display a clear I.R. absorption at $\nu = 1730\text{--}1710\text{ cm}^{-1}$ which is assigned to the carbonyl stretching vibration. The ¹H-N.M.R. spectra of **4**, show among others, signals for $-\text{CH}_2-$ at $\delta = 4.5\text{--}4.4\text{ ppm}$. Compounds **4** are converted into the corresponding tetrafluoroborates by treatment with tetrafluoroboric acid.

Attempts to promote the sulfur extrusion from compounds **4** by treatment with triethylamine were unsuccessful.

Compound **3** reacts with isothiocyanates in dimethylformamide in the presence of triethylamine at room temperature (24h) to give 3-substituted 5-oxo-7,9-diphenyl-1,5-dihydro[1,2,4]triazolo[5,1-*c*]pyrido[2,1-*f*][1,2,4]triazin-10-ium-2-thiolates (**5**) as crystalline solids.



We assume that this reaction involves initial addition of the amino group to the isothiocyanate to give the thiourea derivative **6** the S-atom of which induces the intramolecular nucleophilic displacement of the methylthio group to form the 1,3,4-thiadiazole ring which, under the reaction conditions, undergoes rearrangement to the zwitterionic compound **5**. This mechanism is similar to that suggested⁷ for the reaction of 1-amino-2-methylthio-4,6-diphenyl-pyridinium iodides with isothiocyanates under the same conditions.



The structural elucidation of compounds **5** was accomplished on the basis of spectral and microanalytical data. The mass spectra show the characteristic fragment ions ($M^+ - \text{NCs}$) and $M^+ - 32$, and thus are in good agreement with the presence of an exocyclic S-atom. On the other hand, compound **5a** undergoes methylation with methyl iodide to give the S-methyl derivative the ¹H-N.M.R. spectrum of which shows a singlet at $\delta = 3.6\text{ ppm}$ due to the S-CH₃ group.

3-Amino-4-oxo-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium-2-thiolate (**2**):

To a solution of 2-ethoxycarbonyl-4,6-diphenylpyridium tetrafluoroborate (**1**; 3.7 g, 9.4 mmol) in methanol (25 ml), thiocarbonylhydrazide (1.3 g, 9.4 mmol) is added. A deep red coloration immediately develops and after 4 h at room temperature, the precipitated solid is separated by filtration and recrystallised from acetic acid to give **2**; yield: 2.72 g (84%); yellow needles, m.p. 225–226°C.

C₁₉H₁₄N₄O₂ calc. C 65.90 H 4.04 N 16.17
(346.2) found 65.70 4.09 16.29

M.S.: *m/e* = 314 ($M^+ - \text{S}$, 8%), 298 (9), 284 (5), 258 (5), 230 (100), 115 (6), 77 (10).

I.R. (Nujol): $\nu = 3270, 3160, 1660, 1625, 1485, 1400, 1340, 1255, 1180, 910, 770, 760, 700 \text{ cm}^{-1}$.

$^1\text{H-N.M.R. (TFA/TMS}_{\text{ext}})$: $\delta = 1.97$ (s, 2H); 8.3–7.15 (m, 10H); 8.7 (d, 1H), $J = 2 \text{ Hz}$; 9.25 ppm (d, 1H, $J = 2 \text{ Hz}$).

3-Amino-2-methylthio-4-oxo-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium Trifluoromethanesulfonate (3):

Methyl triflate (0.40 g, 2.4 mmol) is added to a stirred mixture of 3-amino-4-oxo-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium-2-thiolate (**2**; 0.69 g, 2 mmol) and dry dichloromethane (25 ml). The resultant solution is stirred at room temperature for 5 h. The precipitated solid is separated by filtration and recrystallised from ethanol/ether (1/4) to give **3**; yield: 0.83 g (82%); colorless needles; m.p. 259–261 °C.

$\text{C}_{21}\text{H}_{17}\text{F}_3\text{N}_4\text{O}_4\text{S}_2$ calc. C 49.42 H 3.33 N 10.97
(510.3) found 49.51 3.27 10.87

M.S.: $m/e = 314$ ($\text{M}^+ - \text{F}_3\text{C}-\text{SO}_3^\ominus - \text{SCH}_3$, 18%), 312 (63), 298 (100), 230 (73), 258 (35), 203 (64), 115 (18), 77 (23).

I.R. (Nujol): $\nu = 3340, 3260, 1720, 1630, 1530, 1270, 1230, 1150, 1040, 900, 770, 755, 700, 640 \text{ cm}^{-1}$.

$^1\text{H-N.M.R. (DMSO-}d_6/\text{TMS}_{\text{int}})$: $\delta = 2.15$ (s, 3H); 6.27 (s, 2H); 7.3–8.45 (m, 10H), 8.75 (d, 1H, $J = 2 \text{ Hz}$); 8.9 ppm (d, 1H, $J = 2 \text{ Hz}$).

6-Oxo-3-aryl-8,10-diphenyl-2*H*,6*H*-pyrido[2,1-*f*][1,3,4]thiadiazino[2,3-*c*][1,2,4]triazin-11-ium Bromides (4); General Procedure:

The appropriate phenacyl bromide (1 mmol) is added to a stirred solution of 3-amino-4-oxo-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium-2-thiolate (**2**; 0.346 g, 1 mmol) in methanol (20 ml). The mixture is stirred at reflux temperature for 13 h, and then allowed to cool. The solvent is removed under reduced pressure and the residue is treated with ether (10 ml). The solid formed is separated by filtration and recrystallised from the appropriate solvent.

1-Substituted 5-Oxo-7,9-diphenyl-1,5-dihydro[1,2,4]triazolo[5,1-*c*]pyrido[2,1-*f*][1,2,4]triazin-10-ium-2-thiolates (5); General Procedure:

To a stirred solution of 3-amino-2-methylthio-4-oxo-6,8-diphenyl-3,4-dihydropyrido[2,1-*f*][1,2,4]triazin-9-ium triflate (**3**; 0.51 g, 0.98 mmol) in dry dimethylformamide (30 ml), triethylamine (0.98 mmol) and the appropriate isothiocyanate (0.98 mmol) are added. The resultant solution is stirred at room temperature for 24 h, then poured into ice/water, and the precipitated solid isolated by suction, washed with cold methanol (20 ml), and recrystallised from acetic acid.

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¹ Part IX of the series: Bridgehead Nitrogen Heterocycles; Part VIII: P. Molina, A. Tárraga, E. Romero, M. Lorenzo Peña, *Synthesis* **1984**, 71.

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