

Synthesis of Some 1-Aryl-4-(2-hydroxybenzoyl)-pyrazol-5-one and 1-Aryl[1]benzopyrano[2,3-*c*]pyrazol-4(1*H*)-one Derivatives from 3-Acyl-4-hydroxycoumarins

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4-Hydroxycoumarins are versatile intermediates for the synthesis of various heterocyclic compounds. The chemistry of coumarins with 3:4 fused ring systems has recently been reviewed¹. The formation of phenylhydrazones **3** via the 3-acylcoumarins **1** and a phenylhydrazine **2** is well established^{2,3,4}. The cyclodehydration of **3** to **4** occurs in boiling xylene with a small amount of *p*-toluenesulfonic acid^{3,4}.

We now describe a new type of ring transformation of the phenylhydrazones **3** which opens a general route to new 1-

aryl-4-(2-hydroxybenzoyl)-pyrazol-5-ones **5**. 4-Aroylpyrazol-5-one derivatives^{5,6} are important compounds because of their antibacterial, herbicidal⁷, and chelating properties⁸. When the hydrazones **3** are refluxed in acetic acid, they are transformed into the pyrazol-5-one derivatives **5** by a nucleophilic attack at the C-2 lactone carbonyl group with ring opening. The reaction can be achieved in a one step synthesis without isolation of the hydrazone **3**, merely by refluxing the acylcoumarins **1** and the phenylhydrazine **2** in acetic acid.

Compounds **5** are easily converted into the substituted 1-aryl[1]benzopyrano[2,3-*c*]pyrazol-4(1*H*)-ones **6**. The melting points of compounds **6a** and **6f** are in agreement with those prepared by an alternate route⁹. The previously reported syntheses of this class of compounds involve the cyclization of substituted 1-phenyl-4-phenoxy-pyrazole-4-carboxylic acids^{10,11} or 1-aryl(or alkyl)-3-alkyl-4-(*o*-hydroxybenzoyl)-5-chloropyrazoles¹⁰ or 1-phenyl(or methyl)-3-methyl-4-(*o*-chlorobenzoyl)-pyrazol-5-ones¹² as precursors.

The structures of **5** are established by microanalysis, I.R., ¹H-N.M.R., and mass spectra (for **5a**), and by chemical evidence.

3-Acetyl-, 3-propanoyl-, 3-butanoyl-, and 3-benzoyl-4-hydroxycoumarins (**1a**, **b**, **c** and **d**) are prepared according to Ref.² and Ref.¹³; 6-methyl- and 6-chloro-3-acetyl-4-hydroxycoumarins (**1f**, **g**) are prepared according to Ref.¹⁴ and Ref.¹⁵.

3-Benzoyl-4-hydroxy-6-chlorocoumarin (**1h**) is prepared from ethyl benzoylacetate and 2-acetoxy-5-chlorobenzoyl chloride by the method of Anschütz¹⁶; yield: 30%; m.p. 174 °C.

C ₁₆ H ₉ ClO ₄	calc.	C 63.90	H 3.01	Cl 11.79
(300.7)	found	64.06	2.99	11.78

I.R. (KBr): $\nu = 1745 \text{ cm}^{-1}$ (C=O).

1-Aryl-3-alkyl-4-(2-hydroxybenzoyl)-pyrazol-5-ones (**5**); General Procedure:

Method A: 3-Acyl-4-hydroxycoumarin **1** (0.01 mol) and phenylhydrazine **2** (0.01 mol) in acetic acid (50 ml) are heated under reflux for 1 h (or 6 h in the case of **5e**). Acetic acid is evaporated in vacuo and the residue is dissolved in 10% aqueous potassium carbonate solution (200 ml). The solution is extracted with chloroform (2 × 50 ml) and the aqueous layer is acidified to pH 1 with 5 normal hydrochloric acid. The precipitate is collected by suction, washed with water (10 ml), and dried in the air to give **5**, which is recrystallized from ethanol (or acetic acid in the case of **5e**).

Method B: The phenylhydrazone **3a**, **b**, or **c** (prepared according to Ref.²; 0.01 mol) in acetic acid (50 ml) is heated under reflux for 1 h and worked up as described in Method A. The compounds **5a**, **b**, and **c** are obtained in 71%, 67%, and 65% overall yield, respectively.

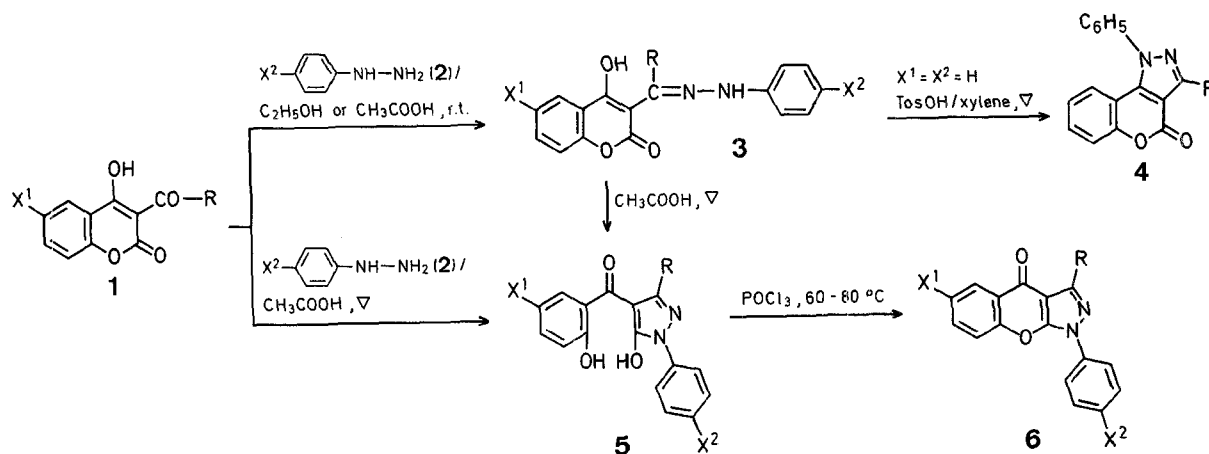


Table 1. 1-Aryl-3-alkyl-4-(2-hydroxybenzoyl)-pyrazol-5-ones **5**

Prod- uct	R	X ¹	X ²	Yield ^a [%]	m.p. ^b [°C]	Molecular Formula ^c	I.R. (KBr) $\nu_{\text{C=O}}$ [cm ⁻¹]	¹ H-N.M.R. (DMSO- <i>d</i> ₆) δ [ppm]
5a^d	CH ₃	H	H	68	198–199°	C ₁₇ H ₁₄ N ₂ O ₃ (294.3)	1630	2.05 (s, 3 H); 6.9–7.2 (m, 2 H); 7.3 (m, 7 H); 9.80 (br. s, 2 H)
5b	C ₂ H ₅	H	H	65	146–147°	C ₁₈ H ₁₆ N ₂ O ₃ (308.3)	1630	1.10 (t, 3 H, <i>J</i> = 8 Hz); 2.52 (q, 2 H, <i>J</i> = 8 Hz); 6.8–7.1 (m, 2 H); 7.1–8.2 (m, 7 H); 10.12 (br. s, 2 H)
5c	<i>n</i> -C ₃ H ₇	H	H	78	160–161°	C ₁₉ H ₁₈ N ₂ O ₃ (322.3)	1625	0.80 (t, 3 H, <i>J</i> = 8 Hz); 1.52 (sext, 2 H, <i>J</i> = 8 Hz); 2.50 (t, 2 H, <i>J</i> = 8 Hz); 6.7–7.1 (m, 2 H); 7.1–7.8 (m, 7 H); 10.77 (br. s, 2 H)
5d	C ₆ H ₅	H	H	60	125–126°	C ₂₂ H ₁₆ N ₂ O ₃ (356.4)	1630	6.6–7.0 (m, 2 H); 7.0–7.7 (m, 10 H); 7.7–7.95 (m, 2 H); 8.45 (br. s, 2 H)
5e	CH ₃	H	NO ₂	62	254–255°	C ₁₇ H ₁₃ N ₃ O ₅ (339.3)	1630	2.10 (s, 3 H); 6.8–7.05 (m, 2 H); 7.3–7.7 (m, 2 H); 8.14 (d, 2 H, <i>J</i> = 9 Hz); 8.38 (d, 2 H, <i>J</i> = 9 Hz); 10.62 (br. s, 2 H)
5f	CH ₃	CH ₃	H	67	164–165°	C ₁₈ H ₁₆ N ₂ O ₃ (308.3)	1635	2.12 (s, 3 H); 2.25 (s, 3 H); 6.85 (d, 1 H, <i>J</i> = 8 Hz); 7.1–7.7 (m, 5 H); 7.7–7.9 (m, 2 H); 8.32 (br. s, 2 H)
5g	CH ₃	Cl	H	70	192–193°	C ₁₇ H ₁₃ ClN ₂ O ₃ (328.7)	1635	6.18 (s, 3 H); 6.93 (d, 1 H, <i>J</i> = 8 Hz); 7.1–7.6 (m, 5 H); 7.6–7.85 (m, 2 H); 10.40 (br. s, 2 H)
5h	C ₆ H ₅	Cl	H	80	172–173°	C ₂₂ H ₁₅ ClN ₂ O ₃ (390.8)	1625	6.70 (d, 1 H, <i>J</i> = 9 Hz); 7.3–7.7 (m, 5 H); 7.7–7.9 (m, 2 H); 9.27 (br. s, 2 H)

^a Method A.^b Recrystallized from ethanol (acetic acid in the case of **5e**).^c The microanalyses were in satisfactory agreement with the calculated values (C ± 0.28, H ± 0.21, N ± 0.29, Cl ± 0.34).^d M.S.: *m/e* (relative intensity) = 294 (M⁺, 35); 200 (15); 121 (100).**Table 2.** 1-Aryl-3-alkyl[1]benzopyrano[2,3-*c*]pyrazol-4(1*H*)-ones **6**

Prod- uct	R	X ¹	X ²	Yield [%]	m.p. ^a [°C]	Molecular Formula ^b or Lit. m.p. [°C]	I.R. (KBr) $\nu_{\text{C=O}}$ [cm ⁻¹]	U.V. (C ₂ H ₅ OH) λ_{max} [nm] (ϵ)	¹ H-N.M.R. (CDCl ₃) δ [ppm]
6a	CH ₃	H	H	78	172–173°	168–170 ⁹⁹	1670	242 (28 400); 264 (16 800)	2.70 (s, 3 H); 7.2–7.7 (m, 6 H); 7.85 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.33 (dd, 1 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz)
6b	C ₂ H ₅	H	H	60	140–141°	C ₁₈ H ₁₄ N ₂ O ₂ (290.3)	1665	242 (27 000); 264 (15 900)	1.43 (t, 3 H, <i>J</i> = 8 Hz); 3.10 (q, 2 H, <i>J</i> = 8 Hz); 7.2–7.7 (m, 6 H); 7.20 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.35 (dd, 1 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz)
6c	<i>n</i> -C ₃ H ₇	H	H	67	125–126°	C ₁₉ H ₁₆ N ₂ O ₂ (304.3)	1670	242 (27 900); 264 (16 800)	1.07 (t, 3 H, <i>J</i> = 8 Hz); 1.92 (sext, 2 H, <i>J</i> = 8 Hz); 3.06 (t, 2 H, <i>J</i> = 8 Hz); 7.3–7.8 (m, 6 H); 7.90 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.34 (dd, 1 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz)
6d	C ₆ H ₅	H	H	63	176–177°	C ₂₂ H ₁₄ N ₂ O ₂ (338.3)	1670	242 (32 500); 262 (25 300)	7.2–7.8 (m, 9 H); 7.93 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.2–8.5 (m, 3 H)
6e	CH ₃	H	NO ₂	52	270–271°	C ₁₇ H ₁₁ N ₃ O ₄ (321.3)	1680	218 (36 200); 290 (22 400)	2.68 (s, 3 H); 7.3–7.8 (m, 3 H); 7.3–7.8 (m, 3 H); 8.13 (d, 1 H, <i>J</i> = 9 Hz); 8.2–8.4 (m, 1 H); 8.35 (d, 1 H, <i>J</i> = 9 Hz)
6f	CH ₃	CH ₃	H	63	152–153°	149–150 ⁹⁹	1685	240 (28 800); 264 (17 100)	2.47 (s, 3 H); 2.70 (s, 3 H); 7.4–7.7 (m, 5 H); 7.90 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.13 (s, 1 H)
6g	CH ₃	Cl	H	70	150–151°	C ₁₇ H ₁₁ ClN ₂ O ₂ (310.7)	1670	242 (25 200); 264 (14 500)	2.65 (s, 3 H); 7.3–7.7 (m, 5 H); 7.83 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.22 (d, 1 H, <i>J</i> _{1,2} = 2 Hz)
6h	C ₆ H ₅	Cl	H	80	215–216°	C ₂₂ H ₁₃ ClN ₂ O ₂ (372.8)	1665	242 (29 700); 260 (25 000)	7.3–7.7 (m, 8 H); 7.88 (dd, 2 H, <i>J</i> _{1,2} = 8 Hz, <i>J</i> _{1,3} = 2 Hz); 8.3–8.5 (m, 3 H)

^a Recrystallized from ethanol.^b The microanalyses were in satisfactory agreement with the calculated values (C ± 0.25, H ± 0.29, N ± 0.23, Cl ± 0.20); exception **6e** (C, –0.53).

1-Aryl-3-alkyl[1]benzopyrano[2,3-c]pyrazol-4-(1H)-ones (6); General Procedure:

A stirred mixture of **5** (0.01 mol) and phosphoryl chloride (15 ml) is heated in a water bath to 60–70 °C (80 °C in the case of **5e**) for 20 min. The mixture is poured into ice/water (400 ml) and extracted with chloroform (3 × 100 ml). The extracts are washed successively with 1 normal aqueous sodium hydroxide (3 × 100 ml), water (2 × 20 ml), and dried with sodium sulfate. The solvent is evaporated under reduced pressure and the residual crude compound **6** is recrystallized from ethanol.

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