FREE RADICAL REARRANGEMENT OF BICYCLO[2.2.2]-AND BICYCLO[4.2.0]OCTENONE SYSTEMS

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Abstract: Intramolecular cyclization of carbon radicals to carbonyl groups was carried out on bicyclo[2.2.2]- and bicyclo[4.2.0]octenones. An unexpected ring expansion product 19, produed by β -scission of the intermediate alkoxyl radical 22, was observed. The β , γ -situated double bond of the bicyclic ketones appears to have a dampening effect on the carbon radical-ketone cyclization and the allylic cleavage of the alkoxyl radical.

Introduction

Radical cyclization has been developed in this laboratory as a new method for ring expansion.^{1,2} The method involves the addition of carbon radicals to carbonyl groups³ and leads to intermediate oxy radicals. Recently, we have been exploring an extension of this strategy to bicyclo[2.2.2]- and bicyclo[4.2.0]octenones, intended to lead to stereoselective syntheses of the diastereomeric natural products (\pm)-trichodiene (1) and (\pm)-bazzanene (2). Numerous synthetic approaches to these substances have been reported in the literature.⁴



Our intent was to explore free radical addition to the carbonyl group of the bicyclic ketone 3 (eq 1). We anticipated that radical addition might be followed by ring cleavage of the oxy radical



4, yielding the allylic radical 5, then the enone precursor 6 to (\pm) -bazzanene (2). We wanted to learn whether the incipient allyl radical would provide sufficient driving force for ring opening, as had the ester group in our earlier ring expansion studies.^{1f-k} Further interest attached to this question because Snowden *et al.*⁵ had suggested a free radical path to explain the formation of a minor product 7 (eq 2), formed in the course of work leading to (\pm) -trichodiene (1). We have



confirmed the observation of 7, and we have proposed a new mechanism for its formation based on single and double electron transfer sequences.⁶

Results and Discussion

The exo diastereomer 3-exo, leading to (\pm) -bazzanene (2), is readily prepared by alkylating the bicyclic ketone 8 (eq 3a),⁷ first with allyl bromide then with methyl iodide.⁵ The propene



sidechain in 10-exo is then converted to the bromopropyl group yielding 3-exo. When tin hydride-promoted free radical addition to the carbonyl group of 3-exo was attempted, the reduction product 14-exo was produced (eq 4a). In this instance, hydrogen transfer from tin



hydride is apparently faster than addition to the ketone, even under conditions of slow addition of the reagent with a syringe pump. Alternatively, addition of the sidechain radical to the carbonyl group might occur yielding the oxy radical 4, but be reversible and not lead to the desired product because of slow allylic carbon-carbon bond cleavage (vide infra).

The endo diastereomer 3-endo was prepared by changing the alkylation sequence. Radical generation from 3-endo permitted us to study the competition for cyclization between the carbon-carbon double bond and the ketone (eq 4b). Failure to observe 7 and an 85% isolated yield of olefin addition product 15 indicate that radical addition to the carbonyl group in 3-endo is not

competitive with cyclization to the carbon-carbon double bond.⁸ Accordingly, this result is not consistent with the free radical mechanism shown in eq 2.5

In pursuit of an alternative strategy, the bicyclo[4.2.0]octenones 17-endo and 17-exo were prepared stereospecifically by photochemical 1,3-acyl shift⁹ of the bicyclo[2.2.2]octenones 3-exo and 3-endo (eq 5a and 5b). The four-membered ketone in 17-exo provides a substantially more



reactive carbonyl group than that in 3-exo and should provide a better opportunity to observe radical addition to the ketone followed by the desired fragmentation.

When the new bromide, 17-exo, was treated with Bu₃SnH, very little (11%) of the expected opening to 20 occurred. Far more interesting, the major product of the reaction, formed in 80% relative yield (isolated yield 69%), was the seven-membered ring product 19 resulting from cleavage of the central carbon-carbon bond and leading to ring expansion (eq 6a). This was not



the expected result. In the alkoxy radical intermediate 22 leading to ring expansion, it was surprising to find that bond b cleaved in preference to bond a (Scheme 1). Both bond scissions

Scheme 1



result in relief of strain in the cyclobutane ring, but bond a is allylic, and it was expected to be the main site of bond cleavage. An MM2 calculation¹⁰ reveals that **19** and **20** are about equally strained, with **19** being approximately 1 kcal/mole the less stable of the two. Tin hydride treatment of **17-endo** yielded only direct reduction product **21-endo** (eq 6b), a consequence of steric hindrance to ring closure.



Examination of models^{10a} reveals two conformations of the presumed intermediate oxy radical 22 (Scheme 2). In the more extended, and presumably more stable intermediate,



Scheme 2. Conformations of Oxy Radical 22

A, the allylic carbon-carbon bond **a** is canted at a less favorable angle (116.5°) with respect to the π -bond than it is in conformation **B** (81.2°). The obliqueness of the angle in conformation **A** may be such as to allow cleavage of bond **b** in preference to bond **a** (Scheme 1).

When the double bond in 3-exo is removed to give 23-exo, treatment of the latter with Bu₃SnH yields more bond a cleavage product (eq 7, relative yields are shown). This is also true



in the bicyclo[2.2.2] series. Hydrogenation of the double bond of 11 yields 27. Free radical generation by treatment of 27 with Bu_3SnH leads to radical addition to the carbonyl group affording 53% cleavage (relative yield) of ring bond c (eq 8 and Scheme 3). In both 3-exo and 17-exo the double bond has an apparent dampening effect on the reactivity of the carbonyl group.

Scheme 3





Perhaps this is an inductive deactivation. With an angle of 58.5° the stereoelectronic situation in the oxy radical 4 (Scheme 4), a possible intermediate from radical reaction of bicyclo[2.2.2]-

Scheme 4



octenone 3-exo, can be judged to be not very favorable for ring cleavage toward the allylic radical.

In the reaction of 27 an unexpected minor product 29 (eq 8) was observed. The α -methyl ketone 29 is apparently derived by fragmentation of the initial primary radical 31 leading to the



acyl substituted tertiary radical 32, while a three carbon piece is lost (cyclopropane or propene) (eq 9). We know of no precedent for such a transformation. No analogous product is detected from the reaction of 33 (eq 10).

We also examined the effect of removing the angular methyl group in 17-exo so that ring opening of 18-exo would yield a secondary radical in competition with cleavage to the allylic radical. Once again we observed only ring expansion and direct reduction, with no cleavage toward the allylic center (eq 11a). This result shows that the methyl group has little impact on



the cleavage of bond a. However, when the double bond is removed in 38-exo, 14% of bond a cleavage product was detected (eq 11b). This reinforces the notion that the double bond may have a deactivating effect on the cleavage of bond a.

MacDonald examined the β -scission of alkoxyl radicals^{11,12} from thermolysis of *in situ*generated hypohalites of fused-ring tertiary hydroxyl systems (eqs 12 and 13), finding that



cleavage also occurred away from the allylic carbon-carbon bond (Scheme 5). This is more easily



understood than is the small extent of allylic cleavage in 18-exo. Examination of models shows that in both conformations of the oxy radical 41 the allylic carbon-carbon σ -bonds f are oriented at an unfavorable angle (14.8° for conformation A, 13.8° for conformation B) with respect to the

 π -bond (Scheme 6).



Scheme 6. Conformations of Oxy Radical 41

Conclusion

Exploration of bicyclo[2.2.2]- and bicyclo[4.2.0] octenone systems has led to the discovery of an unusual β -scission of an alkoxyl radical. The carbon-carbon double bond of the β , γ unsaturated ketone appears to have a negative impact on both the radical-ketone cyclization and subsequent allylic cleavage of the alkoxyl radical. Nonetheless, this unexpected outcome is consistent with our ring expansion goals, and that inquiry now forms an important part of our research into the ring expansion of cyclobutanone-containing systems.^{1a-e}

Experimental Section

Materials and Methods. All reactions were performed under a nitrogen atmosphere. Tetrahydrofuran (THF), benzene, and diethyl ether were distilled from blue or purple solutions of sodium benzophenone ketyl under nitrogen immediately prior to use. A 1.5 M solution of lithium diisopropylamide (LDA) in THF and a 0.5 M solution of 9-borabicyclo[3.3.1]nonane (9-BBN) in THF (Aldrich) were used without further purification.

¹H and ¹³C NMR spectra were obtained on Bruker AC-300, or IBM AF-300 spectrometers (300 MHz for ¹H NMR, 75 MHz for ¹³C NMR). J values are given in Hz. Infrared (IR) spectra were obtained on an IBM IR/32 FTIR or a Mattson Cygnus 100 FTIR spectrometer. Gas chromatography and low resolution mass spectra (GC-MS) were obtained from a Hewlett-Packard (5890 series II) gas chromatograph equipped with a Hewlett-Packard (5970 series) mass spectrometer. The GC column was a 12 m x 0.2 mm I.D. x 0.33 μ m film thickness fused silica capillary column with 100% dimethyl polysiloxane (HP-1, Hewlett-Packard). Relative yields and ratios of diastereomers were determined on GC. High-Resolution mass spectra were obtained on a Varian MAT CH-5DF, or a VG-70G spectrometer. The purity of products isolated as single compounds was determined by GC or ¹H NMR.

1,4-Dimethylbicyclo[2.2.2]oct-5-en-2-one (8). A simple and efficient procedure for the preparation of 8 was developed.⁷ It relies on Diels-Alder trapping of 1,3-cyclohexadiene formed by acid-catalized isomeriation of 1,4-cyclohexadiene and avoids the tedious isolation of the former from the equilibrium mixture of the 1,3- and 1,4-isomer. Thus, crude 1,4-dimethyl-1,4-cyclohexadiene (9.6 g, 79% pure), prepared by Birch reduction of p-xylene, and α -chloro-acrylonitrile (12.4 g, 140 mmol), in 15 ml of p-xylene containing hydroquinone (0.1 g), was mixed with a solution of 36% HCl (5 mL) in water (36 mL). The mixture was heated at 100°C over a period of 5 days. Flash chromatography (10:1 hexanes-ether) after ether workup gave the Diels-

Alder adduct as an 8:1 diastereomeric mixture (11.2 g, 81%). A portion of Diels-Alder adduct (4.0 g, 20 mmol) and NaS·9H₂O (24.6 g, 100 mmol) in EtOH (50 mL) was stirred at reflux for 40 h. Flash chromatography (30:1 hexanes-ether) after ether workup gave 8 as a colorless liquid (2.5 g, 83%). The ¹H NMR, IR, and MS spectra of 8 are in good agreement with the reported values.^{5b}

Compounds 9-endo, 10-exo, 10-endo, 11-exo, 11-endo, and 12-endo were prepared following literature methods.^{5b}

General procedures for the bromination of alcohols: (1SR,3RS,4RS)-3-(3'-Bromopropy)-1,3,4-trimethylbicyclo[2.2.2]oct-5-en-2-one (3-exo). A 0.1 M solution of Br2 in pyridine was added dropwise to a solution of 11-exo (100 mg, 0.45 mmol) and triphenylphosphine (200 mg, 0.76 mmol) in pyridine (5 mL) at 0°C. The addition of Br2 was continued until the mixture was an orange tint, then 10% aq. HCl (8 mL) was added to the reaction mixture. Flash chromatography (20:1 hexanes-ether) of the crude product after ether workup gave 3-exo as a colorless oil (102 mg, 80%). ¹H NMR (CDCl₃) δ 1.01 (s, 3 H), 1.16 (s, 3 H), 1.17 (s, 3 H), 1.20-1.55 (3 H), 1.62 (m, 2 H), 1.88 (m, 2 H), 2.17 (M, 1 H), 3.38 (m, 2 H), 5.17 (d, J=8.1, 1 H), 6.13 (d, J=8.1, 1 H). ¹³C NMR (CDCl₃) δ 18.2 (q, J=127), 19.4 (q, J=126), 22.1 (q, J=128), 27.8 (t, J=128), 29.9 (t, J=131), 30.7 (t, J=129), 32.8 (t J=128), 35.2 (t, J=147), 43.8 (s), 47.7 (s), 49.0 (s), 131.0 (d, J=168), 144.3 (d, J=163), 216.0 (s). IR (neat) 1712 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 286 and 284 (<1, M⁺), 161 (4), 135 (6), 108 (100, M⁺-C₆H9⁷⁹BrO), 93 (36). MS (CI) *m/e* (rel. intensity): 287 and 285 (100, M⁺+H), 269 and 267 (3), 259 and 257 (3), 205 (25, M⁺-Br), 177 (20), 108 (65, M⁺-C₆H9⁷⁹BrO), 93 (10). The purity of 3-exo was 100% according to analysis by GC.

(1SR,3SR,4RS)-3-(3'-Bromopropyl)-1,3,4-trimethylbicyclo[2.2.2]oct-5-en-2-one (3-endo). Bromination of 11-endo (4.9 g, 22 mmol) gave 3-endo as a colorless oil (3.89 g, 62%) after flash chromatography. The ¹H NMR, ¹³C NMR, IR, and HRMS spectra of 3-endo are in good agreement with the reported values.^{5b}

(1SR,3SR,4RS)-3-(3'-Bromopropyl)-1,4-Dimethylbicyclo[2.2.2]oct-5-en-2-one (13-endo). Bromination of 12-endo (46 mg, 0.22 mmol) gave 13-endo as a light yellow oil (51 mg, 85%) after flash chromatography. ¹H NMR (CDCl₃) δ 1.12 (s, 3 H), 1.24 (s, 3 H), 1.40-2.00 (9 H), 3.35 (t, J=6.8, 2 H), 5.75 (d, J=8.1, 1 H), 6.07 (d, J=8.1, 1 H). ¹³C NMR (CDCl₃) δ 17.8 (q, J=127), 22.2 (q, J=126), 28.0 (t, J=125), 31.2 (t, J=130, 2 C), 34 2 (t, J=152), 35.5 (t, J=134), 40.2 (s), 49.0 (s), 52.5 (d, J=125), 131.7 (d, J=166), 140.1 (d, J=164), 214.7 (s). IR (neat) 1717 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 272 and 270 (<1, M⁺), 191 (<1, M⁺-⁷⁹Br), 162 (3), 108 (100, M⁺-C₅H₇⁷⁹BrO), 93 (46). MS (CI) *m/e* (rel. intensity) 273 and 271(100, M⁺+H), 191 (28, M⁺-⁷⁹Br), 163 (18), 108 (78, M⁺-C₅H₇⁷⁹BrO), 93 (12). The purity of 13-endo was >95% according to analysis by ¹H NMR.

General procedures for the preparation of bicyclo[4.2.0]oct-2-en-8-ones. (1RS,6SR,7RS)-7-(3'-Bromopropyl)-3,6,7-trimethylbicyclo[4.2.0]oct-2-en-8-one (17-exo). A solution of 3-endo (682 mg, 2.4 mmol) in ether (15 mL) was irradiated in a pyrex vessel placed in a photochemical reactor (Rayonet, Southern New England Ultraviolet Co.) using RPR 300-nm lamps for 43 h (checking by GC for extent of completion). The solvent was evaporated. Flash chromatography (30:1 hexanes-ether) of the crude product gave 17-exo as a clear colorless oil (457 mg, 67%). ¹H NMR (CDCl₃) δ 1.01 (s, 3 H), 1.22 (s, 3 H), 1.71 (s, 3 H), 1.40-2.10 (8 H), 3.42 (m, 2 H), 3.56 (br s, 1 H), 5.37 (br s, 1 H). ¹³C NMR (CDCl₃) δ 13.4 (q, J=126), 18.0 (q, J=125), 23.7 (q, J=123), 26.8 (t, J=123), 28.7 t, J=146), 31.0 (t, J=129), 33.5 (s), 34.1 (t, J=151), 34.3 (t, J=110), 62.1 (d, J=134), 64.2 (s), 114.5 (d, J=158), 137.7 (s), 213.2 (s). IR (neat) 1767 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 284 and 286 (0, M⁺), 135 (11), 108 (100, M⁺-C₆H9⁷⁹BrO), 77 (10). HRMS calcd for C₁₄H₂₁O (M⁺⁻⁷⁹Br): 205.1592. Found: 205.1592. The purity of 17-exo was 100% according to analysis by GC.

(1RS,6RS,7SR)-7-(3'-Bromopropyl)-3,6-dimethylbicyclo[4.2.0]oct-2-en-8-one (18-exo). Irradiation of 13-endo (180 mg, 0.87 mmol) gave 18-exo a clear colorless oil (97 mg, 54%) after flash chromatography. ¹H NMR (CDCl₃) δ 1.14 (s, 3 H), 1.40-2.15 (8 H), 1.74 (s, 3 H), 2.97 (td, J=7.6 and 2.1, 1 H), 3.08 (br s, 1 H), 3.42 (m, 2 H), 5.33 (br s, 1 H). ¹³C NMR (CDCl₃) δ 22.7 (q, J=128), 23.4, 24.1, 26.8, 29.5, 30.9, 31.3, 33.5 (t, J=165), 59.5 (d, J=128), 64.4 (d, J=136), 114.8 (d, J=166), 136.1 (s), 209.2 (s). IR (neat) 1771 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 272 and 270 (0, M⁺), 121 (4), 108 (100, M⁺-C₅H₇⁷⁹BrO), 93 (69). The purity of 18-exo was >98% according to analysis by ¹H NMR.

Free Radical Reaction of 3-exo. A solution of Bu₃SnH (26.8 μL, 0.10 mmol) and AIBN (2 mg) in benzene (3 mL) was added over 5 h to a stirring, refluxing solution of 3-exo (17.0 mg, 0.06 mmol) in benzene (4 mL). Flash chromatography (20:1 hexanes-ether) of the crude product after DBU workup¹³ afforded 14-exo (9.5 mg, 77%) as a clear colorless oil. ¹H NMR (CDCl₃) δ 0.88 (t, J=7.1, 3 H), 1.03 (s, 3 H), 1.18 (s, 3 H), 1.20 (s, 3 H), 1.10-1.95 (8 H), 5.72 (d, J=8.0, 1 H), 6.14 (d, J=8.0, 1 H). ¹³C NMR (CDCl₃) δ 15.1 (q, J=125), 17.8 (q, J=127), 18.2 (t, J=128), 19.5 (q, J=125), 22 0 (q, J=128), 30.0 (t, J=131), 30.7 (t, J=131), 36.8 (t, J=129), 43.8 (s), 48.4 (s), 49.0 (s), 130.9 (d, J=168), 144.6 (d, J=163), 216.3 (s). IR (neat) 1714 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity): 206 (3, M⁺), 164 (11), 135 (10), 108 (100, M⁺-C₆H₁₀O), 93 (27), 77 (6). HRMS calcd for C₁₄H₂₂O: 206.1671. Found: 206.1680. The purity of 14-exo was >95% according to analysis by ¹H NMR.

Free Radical Reaction of 3-endo. A solution of 3-endo (26.8 mg, 0.094 mmol), Bu₃SnH (32.8 μ L, 0.12 mmol) and AIBN (3 mg) in benzene (24 mL) was heated for 3 h at reflux. Flash chromatography (10:1 hexanes-ether) of the crude product after DBU workup gave 15 (16.4 mg, 85%) as a clear colorless oil. ¹H NMR (CDCl₃) δ 0.93 (s, 3 H), 0.93 (s, 3 H), 0.95 (s, 3 H), 1.17-1.30 (m, 4 H), 1.30-1.48 (m, 4 H), 1.48-1.96 (5 H). IR (neat) 1715 (vs, C=O) cm⁻¹. ¹³C NMR (CDCl₃) δ 18.5(q, J=124), 20.1 (q, J=124), 20.1 (t, J=125), 20.7 (q, J=124), 28.0 (t, J=127), 31.1 (t, J=126), 31.6 (t, J=125), 32.9 (t, J=126), 36.6 (s), 36.7 (t, J=129), 37.9 (d, J=132), 43.2(s), 49.4 (s), 224.1 (s). MS *m/e* (rel. intensity) 206 (85, M⁺), 188 (12), 173 (17), 163 (23), 132 (100), 122 (31), 107 (69), 93 (46), 81 (65), 67 (62). HRMS calcd for C₁₄H₂₂O: 206.1671. Found: 206.1702. The purity of 15 was >96% according to analysis by ¹H NMR.

Free radical Reaction of 13-endo. A solution of Bu₃SnH (19.8 μL, 0.074 mmol) and AIBN (2 mg) in benzene (3 mL) was added over 5.6 h to a stirring, refluxing solution of 13-endo (11.4 mg, 0.042 mmol) in benzene (3 mL). Flash chromatography (40:1 hexanes-ether) of the crude product after DBU workup afforded 16 as a colorless oil (7.0 mg, 87%). ¹H NMR (CDCl₃) δ 0.94 (s, 3 H), 1.00 (s, 3 H) 1.20-1.95 (14 H). ¹³C NMR (CDCl₃) δ 16.7, 20.3, 23.0, 24 3, 27.9, 29.8, 32.2, 34.1, 35.6, 37.3, 52.0, 222.0. IR (neat) 1711 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 192 (100, M⁺), 174 (25), 159 (67), 145 (33), 132 (35), 119 (63), 107 (92), 93 (57). HRMS calcd for C₁₃H₂₀O: 192.1514. Found: 192.1511. The purity of 16 was 100% according to analysis by GC.

Free Radical Reaction of 17-exo. A solution of Bu₃SnH (0.33 mL, 1.2 mmol) and AIBN (10 mg) in benzene (10 mL) was added over 9 h to a refluxing solution of 17-exo (151 mg, 0.53 mmol) in benzene (60 mL). GC of the reaction mixture showed 19 (2 diastereomers), 20, and 21-exo in the ratio of 80:11:9. Flash chromatography (40:1 hexanes-ether) of the crude product after DBU workup afforded the fast moving diastereomer of 19 (13.8 mg, 13%), the slow moving diastereomer of 19 (61.4 mg, 56%), 20 (8.2 mg, 8%), and 21-exo (7.2 mg, 7%).

Data for the fast moving diastereomer of 19: ¹H NMR (CDCl₃) δ 0.91 (d, J=6.8, 3 H), 0.96 (s, 3 H), 1.27 (m, 1 H), 1.40-1.90 (6 H), 1.70 (s, 3 H), 1.98 (m, 2 H), 2.30 (m, 1 H), 2.69 (m, 1 H), 2.81 (br s,

1 H), 5.13 (br.s, 1 H). ¹³C NMR (CDCl₃) δ 18.2, 23.8, 24.3, 25.3, 27.3, 27.3, 31.4, 34.0, 41.1, 42.3, 21.4, 119.3, 136.7, 214.1. IR (neat) 1701 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 206 (23, M⁺), 191 (17), 163 (19), 148 (8, M⁺-C₄H₁₀), 135 (53), 126 (89), 107 (100), 93 (91), 81 (57). HRMS calcd for C₁₄H₂₂O: 206.1671. Found: 206.1671. The purity of fast moving diastereomer of **19** was 100% according to analysis by GC.

Data for the slow moving diastereomer of 19: ¹H NMR (CDCl₃) δ 0.85 (s, 3 H), 0.87 (d, J=8.9, 3 H), 1.24-1.80 (6 H), 1.72 (s, 3 H), 1.85-2.05 (m, 3 H), 2.46 (t, J=6.6, 2 H), 3.22 (br s, 1 H), 5.10 (br s, 1 H). ¹³C NMR (CDCl₃) δ 17.1 (q, J=123), 18.6 (q, J=124), 23.3 (t, J=130), 23.5 (q, J=125), 27.3 (t, J=123), 30.1 (t, J=127), 30.9 (t, J=127), 36.0 (s), 40.8 (d, J=120), 43.3 (t, J=126), 57.4 (d, J=126), 117.7 (d, J=156), 136.3 (s), 214 6 (s). IR (neat) 1700 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 206 (67, M⁺), 191 (43), 163 (31), 148 (23), 135 (55), 126 (100), 107 (96), 93 (76), 81 (57). HRMS calcd for C₁₄H₂₂O: 206.1671. Found: 206.1671. The purity of slow moving diastereomer of **19** was 100% according to analysis by GC.

The ¹H NMR, IR, and MS spectra of 20 are identical with the literature values.^{5b}

Data for **21-exo**: ¹H NMR (CDCl₃) δ 0.91 (t, J=7.3, 3 H), 0.99 (s, 3 H), 1.18 (s, 3 H), 1.15-2.10 (8 H), 1.70 (s, 3 H), 3.35 (br s, 1 H), 5.38 (br s, 1 H). IR (neat) 1770 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 206 (3, M⁺), 188 (2), 162 (3), 135 (16), 108 (100, M⁺-C₆H₁₀O), 93 (66), 77 (15). The purity of **21-exo** was 100% according to analysis by GC.

Free Radical Reaction of 18-exo. A solution of Bu_3SnH (47.2 µL, 0.13 mmol) and AIBN (4 mg) in benzene (5 mL) was added during 5.6 h to a refluxing solution of 18-exo (28.6 mg, 0.11 mmol) in benzene (8 mL). GC analysis of the reaction mixture showed 36 and 37-exo in the ratio of 62:38. Flash chromatography (40:1 hexanes-ether) of the crude product after DBU workup afforded 36 (11.2 mg, 53%) and 37-exo (7.4 mg, 35%).

Data for 36: ¹H NMR (CDCl₃) δ 0.98 (s, 3 H), 1.12-2.10 (10 H), 1.73 (s, 3 H), 2.44 (t, J=6.0, 2 H), 3.15 (br s, 1 H), 5,12 (br s, 1 H). ¹³C NMR (CDCl₃) δ 21.9, 23.7, 24.2, 25.0, 27.2, 32.7, 33.4, 40.1, 42.9, 56.7, 117.6, 136.5, 214.2. IR (neat) 1701 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 192 (28, M⁺), 177 (12), 159 (8), 149 (44), 134 (12), 121 (28), 112 (73), 107 (100), 93 (75), 81 (43). HRMS calcd for C₁₃H₂₀O: 192.1514. Found: 192.1511. The purity of 36 was 100% according to analysis by GC.

Data for 37-exo: ¹H NMR (CDCl₃) δ 0.90 (t, J=7.2, 3 H), 1.12 (s, 3 H), 1.15-1.80 (6 H), 1.74 (s, 3 H), 2.11 (br s, 2 H), 2.98 (br t, J=7.0, 1 H), 3.07 (br s, 1 H), 5.34 (br s, 1 H). IR (neat) 1771 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 192 (0, M⁺), 180 (1), 164 (1), 149 (3), 135 (3), 121 (10), 108 (100, M⁺-C₅H₈O), 93 (100, M⁺-C₆H₁₁O), 91 (56), 77(44). The purity of **37-exo** was >97% according to analysis by GC.

(3SR)-3-(3'-Bromopropyl)-1,3,4-trimethylbicyclo[2.2.2]octan-2-one (27). A mixture of 3endo (220 mg, 0.77 mmol) and 10% Pd/C (15 mg) in EtOH (6 mL) was stirred under a hydrogen balloon at room temperature for 2.5 h. The catalyst was removed by filtration. The filtrate was concentrated to give 27 as a colorless oil (185 mg, purity: >98% by GC, yield: 84%). ¹H NMR (CDCl₃) δ 0.84 (s, 3 H), 0.89 (s, 3 H), 1.01 (s, 3 H), 1.3-2.12 (12 H), 3.37 (m, 2 H). ¹³C NMR (CDCl₃) δ 16.5 (q, J=128), 20.7 (q, J=126), 21.5 (q, J=125), 27.8 (t, J=129), 30.1 (2 C), 30.9, 31.6, 31.9, 34.9 (t, J=151), 37.6 (s), 42.2 (s), 50.1(s), 220.8 (s). IR (neat) 1709 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 288 and 286 (0, M⁺), 217 (3), 207 (100, M⁺-⁷⁹Br), 189 (21), 163 (5), 138 (12), 119 (19), 109 (42), 95 (31), 81 (27). HRMS calcd for C₁₄H₂₃O (M⁺-⁷⁹Br): 207.1725. Found: 207.1725.

(3SR)-3-(3'-Bromopropyl)-1,4-dimethylbicyclo[2.2.2]octan-2-one (33). Hydrogenation of 13endo (149 mg, 0.55 mmol) using Pd/C (10 mg) as catalyst gave 33 as a colorless oil (98 mg, purity: >98% by GC, yield: 66%). ¹H NMR (CDCl₃) δ 0.91 (s, 3 H), 0.94 (s, 3 H), 1.40-1.80 (11 H), 2.01 (m, 1 H), 2.26 (m, 1 H), 3.46 (m, 2 H). ¹³C NMR (CDCl₃) δ 20.3 (q, J=127), 24.9 (q, J=125), 25.7 (t, J=125), 28.2 (t, J=121), 30.8, 31.3, 32.0, 34.2, 35.0, 35.2 (s), 42.5 (S), 55.5 (d, J=122), 219.8 (s). IR (neat) 1714 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 274 and 272 (0, M⁺), 246 (1), 193 (100, M⁺-⁷⁹Br), 175 (18), 164 (51), 149 (45), 135 (11), 123 (11), 109 (51), 95 (52). HRMS calcd for C₁₃H₂₁O (M⁺-⁷⁹Br): 193.1592. Found: 193.1587.

(1SR,6RS,8RS)-8-(3'-Bromopropyl)-1,4,8-trimethylbicyclo[4.2.0]-7-one (23-exo). Hydrogenation of 17-exo (140 mg, 0.50 mmol) using Pd/C (30 mg) as catalyst gave 23-exo as a 40.5:59.5 mixture of diastereomers A and B (107 mg, 75%).

¹H NMR (CDCl₃) of the diastereomeric mixture of **23-exo**: δ 0.86 (d, J=6.6, 3 H for diastereomer A), 0.89 (d, J=6.5, 3 H for diastereomer B), 0.85-2.00 (m), 0.99 (s, 3 H for diastereomer A), 1.00 (s, 3 H for diastereomer B), 1.23 (s, 3 H for diastereomer B), 1.24 (s, 3 H for diastereomer A), 3.19 (t, J=7.8), 3.25-3.50 (m). IR (neat) 1763 (vs, C=O) cm⁻¹. MS of diastereomer A of **23-exo** *m/e* (rel. intensity) 288 and 286 (<1, M⁺), 273 and 271 (15, M⁺-CH₃), 259 and 257 (9, M⁺-C₂H₅), 245 and 243 (17, M⁺-C₃H₇), 207 (8, M⁺⁻⁷⁹Br), 178 and 176 (62), 165 (65), 121 (17), 109 (23), 95 (58), 81 (38), 69 (100). MS for diastereomer B of **23-exo** *m/e* (rel. intensity): 288 and 286 (2, M⁺), 273 and 271 (13, M⁺-CH₃), 259 and 257 (10, M⁺-C₂H₅), 245 and 243 (11, M⁺-C₃H₇), 207 (12), 178 and 176 (60), 165 (61), 121 (19), 109 (19), 95 (52), 81 (28), 69 (100).

(1SR,6RS,8RS)-8-(3'-Bromopropyl)-1,4-dimethylbicyclo[4.2.0]octan-7-one (38-exo). Hydrogenation of 18-exo (105 mg, 0.39 mmol) using PtO₂ (10 mg) as catalyst gave 38-exo as the single diastereomer (75 mg, purity: >97% by GC, yield: 71%). ¹H NMR (CDCl₃) δ 0.93 (d, J=6.4, 3 H), 1.60-2.10 (11 H), 1.07 (s, 3 H), 2.56 (td, J=10.0 and 1.9, 1 H), 3.3 (td, J=5.0 and 1.9, 1 H), 3.14 (m, 2 H). ¹³C NMR (CDCl₃) δ 22.3 (q, J=125), 23.1 (t, J=127), 24.6 (q, J=125), 29.2, 30.0, 30.2, 31.1, 31.5, 33.1, 33.5, 59.0 (d, J=118), 60.7 (d, J=128), 211.3 (s). IR (neat) 1765 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 274 and 272 (1, M⁺), 259 and 257 (1, M⁺-CH₃), 151 (5), 110 (100), 95 (85), 81 (29), 68 (38). HRMS calcd for C₁₃H₂₁⁷⁹BrO: 272.0776. Found: 272.0791.

Free Radical Reaction of 27. A solution of Bu_3SnH (258 µL, 0.96 mmol) and AIBN (10 mg) in benzene (15 mL) was added over 5.4 h to a stirring, refluxing solution of 27 (170 mg, 0.59 mmol) in benzene (45 mL). GC analysis of the reaction mixture showed 25, 28, and 29 in the ratio of 53:24:11. Flash chromatography (40:1 hexanes-ether) of the crude product after DBU workup afforded 25 (50 mg, 41%), 28 (22 mg, 18%), and 29 (7.8 mg, 8%).

Data for 25 (single diastereomer): ¹H NMR (CDCl₃) δ 0.85 (d, J=6.2, 3 H), 0.80-1.30 (5 H), 0.92 (s, 3 H), 1.01 (s, 3 H), 1.46 (m, 2 H), 1.57 (m, 2 H), 1.69 (m, 2 H), 1.88 (m, 1 H), 2.09 (m, 2 H), 2.26 (m, 1 H). ¹³C NMR (CDCl₃) δ 17.4 (q, J=125), 18.0 (q, J=126), 18.7 (t, J=115), 22.6 (q, J=124), 30.5, 30.7, 30.9, 32.2, 32.6 (t, J=109), 33.6 (t, J=115), 37.4 (s), 41.2 (t, J=126), 54.1 (s), 224.3 (s). IR (neat) 1730 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 208 (<1, M⁺), 137 (1), 111 (15), 98 (100), 83 (12), 69 (38). HRMS calcd for C₁₁H₁₈O (M⁺-C₃H₆): 166.1358. Found: 166.1358. The purity of 25 (single diastereomer) was >96% according to analysis by ¹H NMR.

Data for 28: ¹H NMR (CDCl₃) δ 0.83 (s, 3 H), 0.85 (t, 3 H), 0.90 (s, 3 H), 1.01 (s, 3 H), 1.35-1.90 (12 H). ¹³C NMR (CDCl₃) δ 12.5, 16.6, 18.0, 21.0, 21.8, 30.4 (2 C), 31.2, 32.1, 36.1, 37.8, 42.5, 51.1, 221.7. IR (neat) 1711 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 208 (4, M⁺), 190 (2), 166 (38), 138 (100), 126 (8), 109 (15), 95 (14), 81 (12), 69 (20). HRMS calcd for C₁₄H₂₄O: 208.1827. Found: 208.1827. The purity of 28 was >95% according to analysis by ¹H NMR.

The IR, MS spectra, and GC retention time of 29 agree well with those of an authentic

sample.

Preparation of an Authentic Sample of 29. Hydrogenation of (1SR,3SR,4RS)-1,4-dimethyl-3-methylbicyclo[2.2.2]oct-5-en-2-one (55.1 mg, 0.34 mmol) using Pd/C (5 mg) as catalyst gave 29 as a colorless oil (50.4 mg, purity: >98%, yield: 89%). ¹H NMR (CDCl₃) δ 0.89 (s, 3 H), 0.90 (s, 3 H), 1.05 (d, J=7.3, 3 H), 1.30-1.70 (8 H), 1.90 (q, J=7.4, 1 H). ¹³C NMR (CDCl₃) δ 10.9 (q, J=127), 20.4 (q, J=126), 24.8 (q, J=125), 27.5 (t, J=110), 27.6 (s), 30.5 (t, J=126), 31.7 (t, J=127), 34.7 (t, J=128), 42.0 (s), 51.5 (d, J=124), 220.6 (s). IR (neat) 1716 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 166 (13, M⁺), 151(1, M⁺-CH₃), 137 (1, M⁺-C₂H₅), 123 (2, M⁺-C₃H₇), 108 (100, M⁺-C₃H₄O), 93 (25, M⁺-C₄H₇O), 81 (10, M⁺-C₅H₁₀O). HRMS calcd for C₁₁H₁₈O: 166.1359. Found: 166.1359. The purity of **29** was 100% according to analysis by GC.

Free Radical Reaction of 33. A solution of Bu_3SnH (61 µL, 0.23 mmol) and AIBN (5 mg) in benzene (8 mL) was added over 5.6 h to a stirring, refluxing solution of 33 (35 mg, 0.13 mmol) in benzene (10 mL). GC analysis of the reaction mixture showed 34 as a single diastereomer and 35 in the ratio of 53:47. Flash chromatography (40:1 hexanes-ether) of the crude product after DBU workup afforded 34 (10.6 mg, 42%), 35 (8.8 mg, 35%).

Data for 34 (single diastereomer): ¹H NMR (CDCl₃) δ 0.87 (d, J=6.4, 3 H), 0.92 (s, 3 H), 1.11 (m, 2 H), 1.15-2.15 (13 H), 2.26 (m, 1 H). ¹³C NMR (CDCl₃) δ 18.9, 20.4, 22.6, 25.6, 30.3, 30.5, 32.6, 35.1, 35.2, 35.7, 40.9, 59.7, 220.9. IR (neat) 1732 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 194 (1, M⁺), 110 (17), 95 (18), 84 (100), 69 (19). The purity of 34 was >95% according to analysis by GC.

Data for 35: ¹H NMR (CDCl₃) δ 0.92 (s, 3 H), 0.92 (s, 3 H), 0.92 (t, J=6.8, 3 H), 1.40-1.80 (13 H). ¹³C NMR (CDCl₃) δ 14.4, 20.4, 22.8, 25.0, 28.3, 29.6, 31.0, 31.3, 35.1, 35.2, 42.5, 56.7, 220.4. IR (neat) 1715 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 194 (5, M⁺), 152 (30), 137 (3), 124 (100), 108 (29), 93 (18), 81 (15). HRMS calcd for C₁₃H₂₂O: 194.1671. Found: 194.1657. The purity of 35 was >95% according to analysis by GC.

Free Radical Reaction of 23-exo (mixture of two diastereomers). A solution of Bu₃SnH (151 μ L, 0.56 mmol) and AIBN (10 mg) in benzene (11 mL) was added over 6 h to a stirring, refluxing solution of 23-exo (107 mg, 0.37 mmol) in benzene (25 mL). GC analysis of the reaction mixture showed 24 (four diastereomers: A, B, C, and D), 25 (two diastereomers: A and B), and 26-exo (two diastereomers: A and B) in the ratio of 68 (11:6:24:27) : 22 (9:13) : 10 (7:3). Flash chromatography (45:1 hexanes-ether) of the crude product after DBU workup afforded diastereomer D of 24 (15.3 mg, 20%), diastereomer C of 24 (18.4 mg, 24%), a mixture of diastereomers A and B of 25 (11.6 mg, 15%), and a mixture of diastereomers A and B of 26-exo (6.2 mg, 8%).

Data for diastereomer D of 24: ¹H NMR (CDCl₃) δ 0.80 (s, 3 H), 0.85 (d, J=6.9, 3 H), 0.95 (d, J=6.1, 3 H), 0.80-1.90 (11 H), 2.20 (m, 1 H), 2.36 (m, 2 H), 2.59 (m, 1 H). ¹³C NMR (CDCl₃) δ 17.9, 21.9, 22.4, 22.6, 29.6, 29.8, 33.0 (2 C), 33.5, 34.5, 38.9, 42.1, 66.5, 216.6. IR (neat) 1684 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 208 (18, M⁺), 193 (9, M⁺-CH₃), 165 (9), 137 (24), 126 (26), 110 (53), 95 (100), 81 (67). HRMS calcd for C₁₄H₂₄O: 208.1827. Found: 208.1774. The purity of diastereomer D of **24** was 100% according to analysis by GC.

IR (neat) of the mixture of diastereomers A and B of 25: 1732 (vs, C=O) cm⁻¹. MS of diastereomers A of 25m/e (rel. intensity): 208 (<1, M⁺), 137 (2), 111 (37), 98 (100), 83 (37), 69 (53).

IR (neat) of the mixture of diastereomers A and B of 26-exo: 1763 (vs, C=O) cm⁻¹. MS of diastereomer A of 26-exo m/e (rel. intensity): 208 (11, M⁺), 193(4, M⁺-CH₃), 179 (12), 166 (29), 137 (53), 109 (30), 98 (49), 81 (26), 69 (100).

Free Radical Reaction of 38-exo. A solution of Bu_3SnH (63 µL, 0.2 mmol) and AIBN (5 mg) in benzene (5 mL) was added over 8 h to a stirring, refluxing solution of 38-exo (30 mg, 0.1 mmol) in benzene (10 mL). GC of the reaction mixture showed 39, 34 (single diastereomer), and 40-exo in the ratio of 49:14:37. Flash chromatography (40:1 hexanes-ether) of the crude product after DBU workup afforded 39 (8.1 mg, 42%) and a mixture of 34 and 40-exo (7.9 mg).

Data for 39: ¹H NMR (CDCl₃) δ 0.94 (s, 3 H), 0.95 (d, J=5.7, 3 H), 1.20-1.80 (12 H), 2.15 (br t, J=14.1, 1 H), 2.26-2.22 (2 H), 2.62 (m, 1 H). ¹³C NMR (CDCl₃): δ 22.5, 23.5, 24.4, 27.9, 29.8, 30.2, 32.9, 33.5, 35.9, 42.4, 43.0, 64.5, 216.2. IR (neat) 1688 (vs, C=O) cm⁻¹. MS *m/e* (rel. intensity) 194 (10, M⁺), 179(13, M⁺-CH₃), 161(7), 137 (17), 112 (100), 95 (44), 81 (73), 67 (56). HRMS calcd for C₁₃H₂₂O: 194.1671. Found: 194.1671. The purity of **39** was 100% according to analysis by GC.

IR (neat) of the mixture of 34 and 40-exo: 1771 (vs, C=O for 40-exo), 1732 (vs, C=O for 34) cm⁻¹. MS of 34 m/e (rel. intensity): 194 (<1, M⁺), 110 (21), 95, (16), 84 (100), 69 (16). MS of 40-exo m/e (rel. intensity): 194 (5, M⁺), 151 (4), 137 (3), 123 (5), 110 (71), 95 (100), 81 (31), 68 (62).

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References and Notes

- (1) (a) Dowd, P.; Zhang, W. J. Org. Chem. in press. (b) Dowd, P.; Zhang, W. J. Am. Chem. Soc. 1992, 114, 10084. (c) Zhang, W; Dowd, P. Tetrahedron Lett. 1992, 33, 7307. (d) Zhang, W; Dowd, P. Tetrahedron Lett. 1992, 3285. (e) Dowd, P.; Zhang, W. J. Am. Chem. Soc. 1991, 113, 9875. (f) Dowd, P.; Choi, S.-C. Tetrahedron , 1992, 48, 4773. (g) Dowd, P.; Choi, S.-C. Tetrahedron 1991, 47, 4847. (h) Dowd, P.; Choi, S.-C. Tetrahedron 1989, 45, 77. (i) Dowd, P.; Choi, S.-C. Tetrahedron Lett., 1989, 30, 6129. (j) Dowd, P.; Choi, S.-C. J. Am. Chem. Soc. 1987, 109, 6548. (k) Dowd, P.; Choi, S.-C. J. Am. Chem. Soc., 1987, 109, 3493. See also (l) Beckwith, A. L. J.; O'Shea, D. M.; Westwood, S. W. J. Am. Chem. Soc., 1988, 110, 2565. (m) Beckwith, A. L. J.; O'Shea, D. M.; Westwood, S. W. J. Chem. Soc., Chem. Commun. 1987, 666. (n) Bowman, W. R.; Westlake, P. J. Tetrahedron, 1992, 48, 4027. (o) Crimmins, M.T.; Dudek, C. M.; Cheung, A. W.-H. Tetrahedron Lett. 1992, 33, 181. (p) Ellwood, C. W.; Pattenden, G. J. Org. Chem. 1991, 56, 6447. (q) Boger, D. L.; Mathrink, R. T. J. Org. Chem. 1990, 55, 5442. (r) Nishida, A.; Takahashi, H.; Takeda, H.; Tekada, N.; Yonemitsu, O. J. Am. Chem. Soc. 1990, 112, 902.
- (2) For reviews on ring expansion, see: (a) Hesse, H. Ring Enlargement in Organic Chemistry;
 VCH: Weinheim, 1991. (b) Stach, H.; Hesse, M. Tetrahedron 1988, 44, 1573.
- (3) For intramolecular addition of alkyl radicals to aldehydes, see: (a) Walton, R.; Fraser-Reid, B. J. Am. Chem. Soc. 1991, 113, 5791. (b) Beckwith, A. L. J.; Hay, B. P. J. Am. Chem. Soc. 1989, 111, 2674. (c) Beckwith, A. L. J.; Hay, B. P. J. Am. Chem. Soc. 1989, 111, 230. (d) Tsang, R.; Dickson, Jr, J. K.; Pak, H.; Walton, R.; Fraser-Ried, B. J. Am. Chem. Soc. 1987, 109, 3484.
- (4) Harding, K. E.; Clement, K. S.; Tseng, C.-Y. J. Org. Chem. 1990, 55, 4403 and references there.
- (5) (a) Snowden, R. L.; Sonnay, P. J. Org. Chem. Soc. 1984, 49, 1465. (b) Snowden, R. L.; Brauchli,

R.; Sonnay, P. Helv. Chim. Acta. 1989, 72, 570.

- (6) Zhang, W; Dowd, P. submitted for publication.
- (7) We modified the reported procedures^{5b} and developed a simple and effective method for the preparation of 8. Birch reduction of *p*-xylene yielded 1,4-dimethyl-1,4-cyclohexadiene. This substance underwent one-pot isomerization to the 1,4-dimethyl-1,3-cyclohexadiene and subsequent Diels-Alder reaction with α -chloroacrylonitrile in the presence of 5% aqueous hydrochloride containing hydroquinone at 90° C over a period of five days (eq i). Hydrolysis of the intermediate cycloadducts gave product 8.



(8) An analogous reaction (eq ii) was reported by Stork, G. and Baine, N. H. Tetrahedron Lett. 1985, 26, 5927.



- (9) Reviews (a) Schaffner, K. Demuth, M. In Modern Synthetic Methods; Scheffold, R. Ed., Springer-Verlag: Berlin Heidelberg, 1986, pp 61. (b) Dauben, W.G.; Lodder, G.; Ipatschki, J. Top. Curr. Chem. 1975, 54, 73. (c) Houk, K. N. Chem. Rev. 1976, 76,1.
- (10) (a) For the MM2 calculations,^{10b} conformational analyses of 4, 22, and 41 were carried out using -OH instead of -O· with the Chem 3D Plus program. (b) Burkert, U.; Allinger, N. L. In Molecular Mechanics, ACS Monogragh 177, American Chemical Society: Washington, D. C., 1982.
- (11) (a) O'Dell, D. E.; Loper, J. T.; MacDonald, T. L. J. Org. Chem. 1988, 53, 5125. (b) MacDonald, T. L.; O'Dell, D. E. J. Org. Chem. 1981, 46, 1501. See also (c) Beckwith, A. L. J.; Kazlauskas, R.; Syner-Lyons, M. R. J. Org. Chem. 1983, 48, 4718.
- (12) For mechanistic studies of β-scission of oxy radicals, see: (a) Greene, F. D.; Savitz, M. L.; Osterholtz, F. D.; Lau, H. H.; Smith, W. N.; Zanet, P. M. J. Org. Chem. 1963, 28, 55. (b) Janjatovic, J.; Majerski, Z. J. Org. Chem. 1980, 45, 4892. (c) Bensadoun, N.; Brun, P.; Casanova, J; Waegell, B. J. Chem. Res., Synp. 1981, 236. (d) Mariano, P. S.; Bay, E. J. Org. Chem. 1980, 45, 1763.
- (13) Curran, D. P.; Chang, C.-T. J. Org. Chem. 1989, 54, 3140.