

Note

Disappearance of Differentiation-induction of Friend Leukemia Cells upon Racemization of Trichostatic Acid[†]

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Friend erythroleukemia cells have been used as a model of differentiation of normal erythroid cells since they can be induced *in vitro* to have various normal functions specific to erythroid cells such as hemoglobin biosynthesis.¹⁾ Although the mechanism of the differentiation is unknown, suppression of their proliferation is assumed to be a prerequisite for their differentiation.^{2~6)} Many chemicals such as dimethylsulfoxide (DMSO)⁷⁾ and hexamethylene bisacetamide (HMBAA)⁸⁾ are inducers.

We used the differentiation of Friend leukemia cells to search for new antitumor substances. Actinomycin V,⁹⁾ cosmomycins,^{10,11)} and trichostatic acid¹²⁾ were potent inducers of Friend cell differentiation, and may provide clues to elucidate the mechanism of Friend leukemia cell differentiation. We have been studying the relationship

between differentiation-inducing activity and the structure of these compounds. This paper reports that the racemate of trichostatic acid prepared from (+)-trichostatic acid lost the ability to induce Friend leukemia cell differentiation while retaining its cytotoxicity.

The properties and culture conditions of Friend leukemia cells, F5-5 were described in our previous paper.¹²⁾ Cell differentiation was evaluated by the frequency of hemoglobin-positive cells.¹³⁾

(+)-Trichostatic acid (Fig. 1) was racemized in 1 N NaOH at 30°C for 3 hr. The reaction product was purified by HPLC with a μ Bondapak C18 column using a solvent system of acetonitrile and 10 mM ammonium acetate buffer (35:65, pH 4.0). Circular dichroism (CD) and ¹H-NMR spectra of the product eluted at the same retention time as that of (+)-trichostatic acid were examined. The product showed the same spectra as those of (+)-trichostatic acid in ¹H-NMR. It, however, showed no significant peaks in the CD spectrum, while (+)-trichostatic acid had peaks at 243, 260, 295, 299, 307, and 349 nm (Fig. 2). These results indicate that trichostatic acid was successfully racemized.

We examined whether the (+) isomer of trichostatic acid is absolutely necessary for the differentiation-inducing activity. (+)-Trichostatic acid caused differentiation of F5-5 cells from 1.0 to 20 μ M (Fig. 3). Differentiation of F5-5 cells was first observed at a concentration that slightly inhibited the cell growth and reached a maximum at the concentration that inhibited the

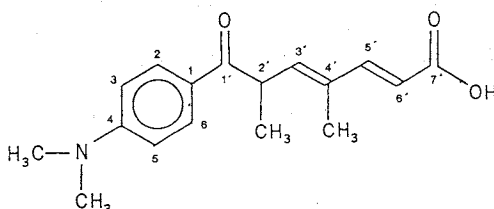


FIG. 1. Chemical Structure of Trichostatic Acid.

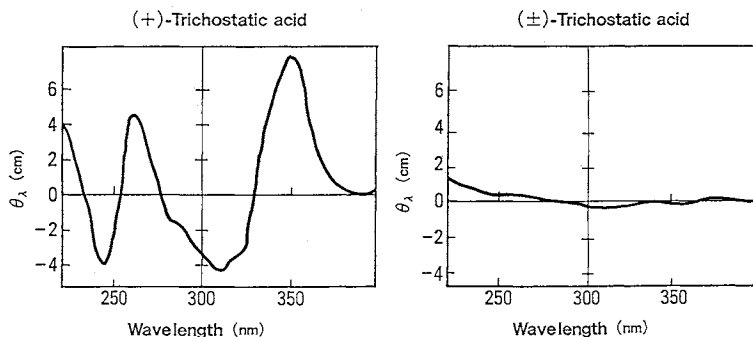


FIG. 2. CD Spectra of (+)- and (±)-Trichostatic Acid.

[†] This study was presented at the Annual Meeting of the Agricultural Chemical Society of Japan, April, 1985, p. 221.

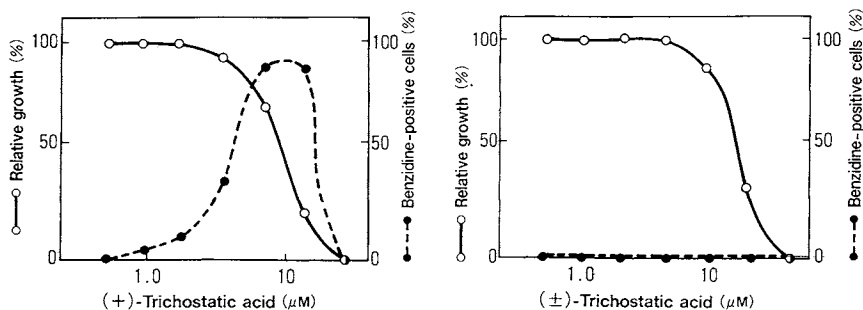


FIG. 3. Cytotoxicity and Differentiation-inducing Activity of (+)- and (±)-Trichostatic Acid.

A Friend cell line, F5-5, was grown at 2×10^4 cells/ml in the presence of various amounts of the compounds. On the 6th day, the percentage of benzidine-positive cells (---●---) and final cell number (—○—) were measured.

cell growth to half the level of the control. At the optimal concentration ($8 \sim 18 \mu\text{M}$), approximately 90% of the cells differentiated within 6 days. It inhibited cell proliferation and eventually became cytotoxic at concentrations above $36 \mu\text{M}$.

The racemate of trichostatic acid completely lost differentiation-inducing activity from 0.45 to $36 \mu\text{M}$, but retained cytotoxicity at concentrations at which (+)-trichostatic acid is cytotoxic (Fig. 3). The optical active center at $2'$ position is considered to be necessary for the differentiation-inducing activity but not for cytotoxicity. During suppression of proliferation, (+)-trichostatic acid may act on the differentiation site by interacting with the gene necessary for terminal differentiation. To verify this hypothesis, we are separating (–)-trichostatic acid from the racemate to analyze the effect of the (–) isomer on the differentiation of F5-5 cells.

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