



## DETERMINATION OF THE ABSOLUTE CONFIGURATIONS AND TOTAL SYNTHESIS OF NEW IMMUNOSUPPRESSANTS, MYCESTERICINS E AND G

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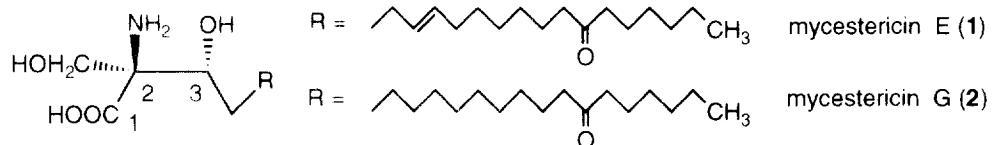
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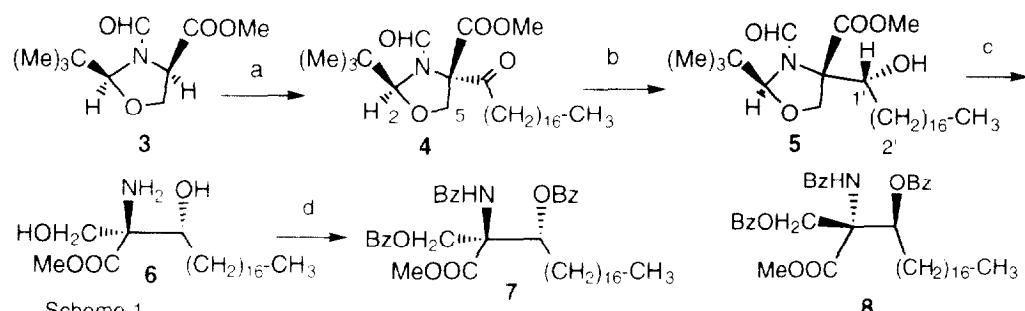
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**Abstract:** The absolute configurations of two new immunosuppressants, mycestericins E (**1**) and G (**2**), were established and both compounds were synthesized starting from 1,8-octanediol and a fully protected methyl (2S,4R)-oxazolidine-4-carboxylate **3**. Compounds **5** and **22**, with a (1'R)-hydroxyl group, were prepared from the  $\beta$ -keto esters **4** and **21**, respectively, by stereoselective reduction with metal borohydrides.

Recently, we reported the isolation of mycestericin E (**1**) as a new immunosuppressant from the culture broth of the fungus *Mycelia sterilia* ATCC 20349, which produces the immunosuppressant ISP-I, and elucidated the planar structure from spectroscopic and chemical evidence.<sup>1</sup> The dihydro derivative, mycestericin G (**2**), has also been isolated from the same broth. Compounds **1** and **2** showed immunosuppressive activities with IC<sub>50</sub> values of 1.3 nM and 370 nM, respectively, against mouse allogeneic mixed lymphocyte reaction (MLR).<sup>2</sup> Here we report determination of the absolute configurations of mycestericins E (**1**) and G (**2**), and the first total synthesis of **1** and **2**, starting from 1,8-octanediol and methyl (2S,4R)-2-*tert*-butyl-3-formyloxazolidine-4-carboxylate (**3**).<sup>3a</sup>



Determination of the absolute configurations of **1** and **2** was performed by comparison of their CD spectra with those of synthetic compounds **7** and **8**. The synthesis of **7** is outlined in Scheme 1. Stereoselective acylation<sup>3b</sup> of the oxazolidine **3** with stearoyl chloride afforded the  $\beta$ -keto ester **4**, syrup; IR (CHCl<sub>3</sub>): 1750 (CO), 1725 (CO), 1680 (CO) cm<sup>-1</sup>, after purification by silica gel flash column chromatography.<sup>4</sup> The relation between the ketone and the methyl ester was confirmed to be *s-cis* based on NOE experiments (Figure 1). Therefore, the stereoselective reduction of the  $\beta$ -keto ester **4** with metal borohydrides should give predominantly the (1'R)-hydroxy compound **5**<sup>5</sup> by hydride attack from the less hindered side because of the bulky formyl group at the *re*-face of the ketone. The results are summarized in the table. Treatment of (1'R)-**5** with 6 N HCl afforded the ester **6**, solid; mp 60-61 °C, which was benzoylated to give **7**.<sup>6</sup> The CD spectra of **7**, **9**,<sup>7</sup> and **10**,<sup>7</sup> exhibited similar negative Cotton effects at 241 nm, while **8** exhibited a positive Cotton



Reagents: a)  $\text{CH}_3(\text{CH}_2)_{16}\text{COCl}$  (1.2 eq.), LDA (1.2 eq.), THF, -100°C, 46%; b)  $\text{NaBH}_4$  (1.2 eq.), MeOH, 70 °C (1'R-5); c) 6 N HCl, MeOH, 80 °C, 1 h, 61%; d)  $\text{Bz}_2\text{O}$ ,  $\text{Et}_3\text{N}$ , 40°C, 2 h, 77%

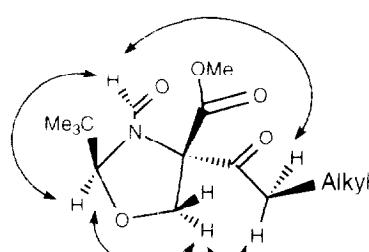


Figure 1.  $\longleftrightarrow$  NOE

Table. Reduction of 4.

Reagents / Solvent	Temp( C)	Time(h)	5(1'R)/(1'S) <sup>a</sup>
$\text{NaBH}_4$ / MeOH	0	0.5	85 / 15
$\text{LiBH}_4$ / $\text{Et}_2\text{O}$	0	2	80 / 20
$\text{Zn}(\text{BH}_4)_2$ / $\text{Et}_2\text{O}$	0	2	85 / 15

a: Ratio of 1'R/1'S determined by HPLC.

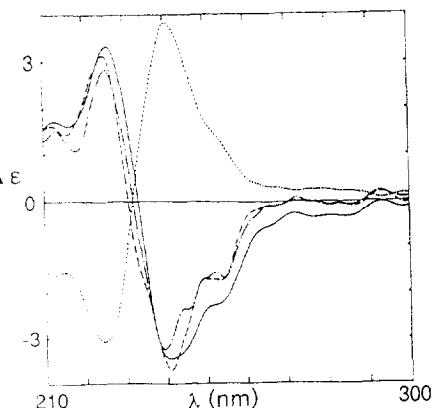


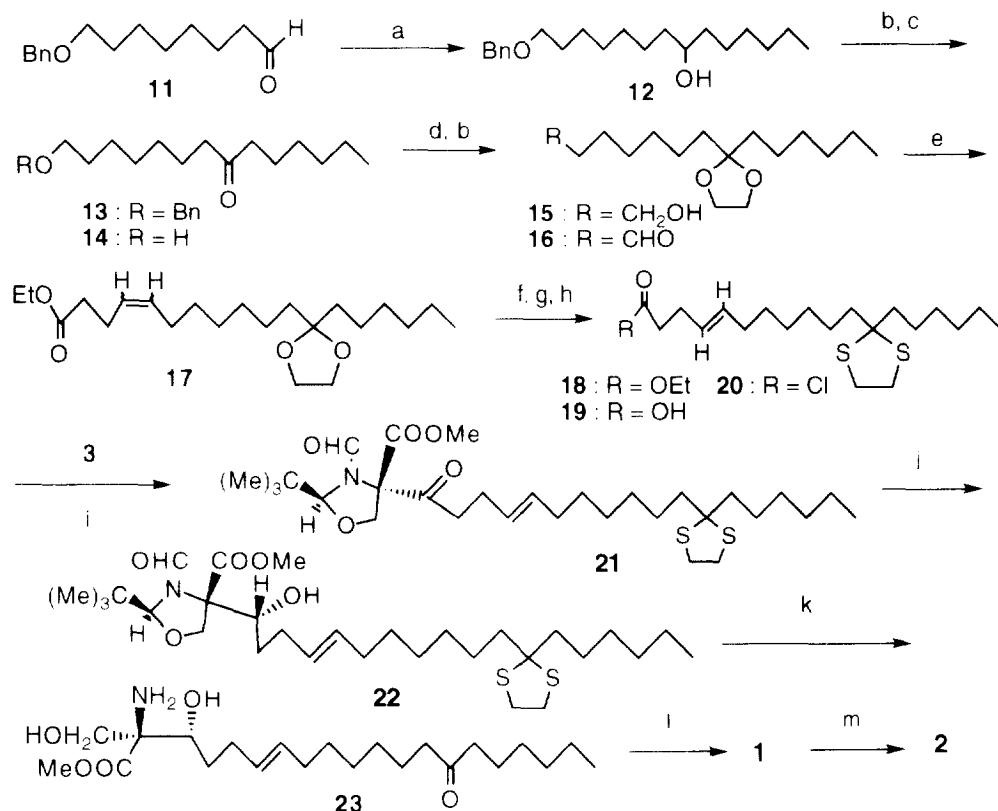
Figure 2. CD spectra of 7, 8, 9 and 10 (in MeOH).

7: ——; 9: ——  
8: ·····; 10: - - -

effect at 241 nm (Figure 2).<sup>8</sup> On this basis, the configurations of **1** and **2** were assigned as 2S, 3R.

Syntheses of **1** and **2** are summarized in Scheme 2. 8-(Benzylxy)octanal (**11**) was prepared in two steps (benzylation/NaH, THF, 60 °C and Swern oxidation) from 1,8-octanediol in 41 % overall yield by an improved method based on that of Grayshan *et al.*<sup>9</sup> Reaction of **11** with *n* hexylmagnesium bromide provided the hydroxy derivative **12**, which was converted to the ketone **13**, syrup; IR ( $\text{CHCl}_3$ ): 1705 (CO)  $\text{cm}^{-1}$ . Deprotection of the benzyl group of **13** with  $\text{FeCl}_3$ <sup>10</sup> gave the alcohol **14**, mp 52–53 °C,<sup>11</sup> and the carbonyl group was protected with ethylene glycol to give the 1,3-dioxolane **15**, syrup.<sup>12</sup> Alternatively, **15** was prepared from **13** in 83 % yield by dioxolanation, followed by debenylation with  $\text{H}_2/10\% \text{ Pd-C}$ . Wittig reaction of **16**, which was prepared from **15** by Swern oxidation, with (3-ethoxycarbonylpropyl)triphenylphosphonium bromide<sup>13</sup> provided **17**, syrup, IR ( $\text{CHCl}_3$ ): 2910, 1725 (CO), 950 (C=CH, E), 675 (C=CH, Z)  $\text{cm}^{-1}$ , as an 89 : 11 mixture of *E* and *Z* isomers.<sup>14</sup> The isomerization of **17** was performed under sun light at room temperature. After purification on an  $\text{AgNO}_3$ -silica gel column,<sup>15</sup> the major product was thioketalized to give **18**, syrup.<sup>16</sup> The configuration of the olefin of **18** was assigned as *E*-form based on the

<sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra. The acid chloride **20** was prepared from **18** by hydrolysis, followed by heating of **19** and SOCl<sub>2</sub>. Stereoselective acylation of **3** with **20** afforded the  $\beta$ -keto ester **21**, syrup; IR (CHCl<sub>3</sub>): 1750 (CO), 1720 (CO), 1680 (CO), 970 (C=C, E) cm<sup>-1</sup>, after purification by silica gel flash column chromatography.<sup>17</sup> Reduction of **21** with NaBH<sub>4</sub> provided **22**,<sup>18</sup> the thioketal group of which was removed with NCS/AgNO<sub>3</sub><sup>19</sup> to give the ketone. Without purification, the ketone was treated with acid to give the methyl ester **23** after purification by silica gel preparative tlc.<sup>20</sup> Hydrolysis of **23** afforded mycestericin E (**1**), white powder; mp 183–185°C, [α]<sub>D</sub><sup>24</sup> -8.5° (c 0.06, MeOH).<sup>21</sup> Mycestericin G<sup>22</sup> (**2**) also was obtained from **1** by hydrogenation. These products gave identical spectroscopic data with those of authentic mycestericins E and G.



Scheme 2

Reagents: a) BrMgC<sub>6</sub>H<sub>5</sub> (1.2 eq.), THF, 0°C, 83%; b) (COCl)<sub>2</sub> (1.5 eq.), DMSO (1.8 eq.), CH<sub>2</sub>Cl<sub>2</sub>, r.t., 89%; c) FeCl<sub>3</sub> (3 eq.), CH<sub>2</sub>Cl<sub>2</sub>, r.t., 90%; d) HOCH<sub>2</sub>CH<sub>2</sub>OH (7.3 eq.), p-TsOH, 92%; e) EtO<sub>2</sub>C-(CH<sub>2</sub>)<sub>3</sub>P<sup>+</sup>(Ph)<sub>3</sub>Br<sup>-</sup> (1.8 eq.), KN(SiMe<sub>3</sub>)<sub>2</sub> (1.5 eq.), THF, 0°C, 80%; f) hν, PhSSPh, hexane; ii) HSCH<sub>2</sub>CH<sub>2</sub>SH (2.7 eq.), BF<sub>3</sub>·Et<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>, r.t., 68% from **17**; g) 1 N NaOH, MeOH; h) SOCl<sub>2</sub> (6 eq.), benzene, r.f.; i) LiN(SiMe<sub>3</sub>)<sub>2</sub> (1.5 eq.), THF, -100°C, 70%; j) NaBH<sub>4</sub> (1.2 eq.), MeOH, 0°C, 97% (R/S mixture); k) i) NCS, AgNO<sub>3</sub>; ii) 10% MeOH in CF<sub>3</sub>COOH, 60°C, 38% from **22**; l) 2 N NaOH, MeOH, r.t., 68%; m) H<sub>2</sub>/10% Pd-C, MeOH.

### References and Notes

1. Sasaki, S.; Hashimoto, R.; Kiuchi, M.; Inoue, K.; Ikumoto, T.; Hirose, R.; Chiba, K.; Hoshino, Y.; Okumoto, T.; Fujita, T. *J. Antibiotics*, **1994**, *47*, 420.
2. Fujita, T.; Inoue, K.; Yamamoto, S.; Ikumoto, T.; Sasaki, S.; Toyama, R.; Chiba, K.; Hoshino, Y.; Okumoto, T. *J. Antibiotics*, **1994**, *47*, 208.
3. (a) Seebach, D.; Aeby, D. *J. Tetrahedron Lett.* **1984**, *25*, 2545; (b) Singh, N. P.; Giannis, A.; Henk, E.; Kolter, T.; Sandhoff, K.; Schmidt, R. R. *J. Carbohydrate Chemistry*, **1990**, *9*, 543.
4. **4:**  $[\alpha]_D^{24} +3.0^\circ$  ( $c=1.2$ ,  $\text{CHCl}_3$ ).  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 0.94, 0.98 (9 H, 2s), 1.25 (28 H, m), 1.60 (2 H, m), 2.49 (1 H, m), 2.68 (1 H, dt,  $J = 18$  Hz,  $J = 8$  Hz), 3.85, 3.88 (3 H, 2s), 4.08, 4.16 (1 H, 2d,  $J = 10$  Hz,  $J = 9$  Hz), 4.61, 4.71 (1 H, 2d,  $J = 10$  Hz,  $J = 9$  Hz), 5.08, 5.48 (1 H, 2s), 8.25, 8.32 (1 H, 2s, CHO). All new compounds gave satisfactory high-resolution CIMS and FABMS data.
5. The absolute configuration of the (*1R*)-hydroxyl group was confirmed by the modified Mosher's method.<sup>23</sup>  $\Delta\delta = \delta_S - \delta_R$  in ppm (270 MHz): COOMe (-0.148),  $5\alpha$ -H (+0.010),  $5\beta$ -H (-0.039),  $2'$ -H<sub>2</sub> (+0.076, +0.059).
6. (*2S,3R*)-7: syrup;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 1.25 (30 H, m), 1.75 (1 H, m), 1.91 (1 H, m), 3.86 (3 H, s), 5.01 (1 H, d,  $J = 12$  Hz), 5.39 (1 H, d,  $J = 12$  Hz), 5.84 (1 H, dd,  $J = 11$  Hz,  $J = 2$  Hz), 7.19-7.99 (16 H, m). The CD spectrum of the (*2S,3S*)-isomer of 7 exhibited a positive Cotton effect at 238 nm.
7. Tribenzoyl derivatives **9** and **10** were prepared from **1** and **2**, respectively, by esterification with  $\text{CH}_2\text{N}_2$ , followed by benzoylation with benzoic anhydride and  $\text{Et}_3\text{N}$ .
8. (*2R, 3S*)-**8** was prepared from methyl (*2S, 4S*)-2-*tert*-butyl-3-formyl-oxazolidine-4-carboxylate<sup>3</sup> and stearoyl chloride, and the diastereoisomers, (*3S*)- and (*3R*)-hydroxy derivatives, were purified by HPLC after reduction of the  $\beta$ -keto ester with  $\text{NaBH}_4$ . The CD spectrum of the (*2R, 3R*)-isomer of **8** showed a negative Cotton effect at 238 nm, which was not identical with that of tribenzoyl derivatives (**9** and **10**).
9. Grayshan, R.; Keal, C. A.; Sackville, M. A. *J. Chem. Research*, **1986**, 282; **11:** syrup;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  1.33 (6 H, m), 1.61 (4 H, m), 2.40 (2 H, td,  $J = 7$  Hz,  $J = 2$  Hz), 3.46 (2 H, t,  $J = 6$  Hz), 4.50 (2 H, s), 7.26-7.35 (5 H, m, PhH), 9.76 (1 H, d,  $J = 2$  Hz, CHO).
10. Park, M. H.; Takeda, R.; Nakanishi, K. *Tetrahedron Lett.* **1987**, *28*, 3823.
11. Just, G.; Payette, D. R. *Tetrahedron Lett.* **1980**, *21*, 3219; **14:**  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 1.26-1.32 (12 H, m), 1.53-1.60 (6 H, m), 2.38 (4 H, t,  $J = 7$  Hz), 3.64 (2 H, t,  $J = 7$  Hz).
12. **15:**  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 1.28-1.32 (2 H, t,  $J = 6$  Hz), 1.56-1.59 (6 H, m), 3.64 (2 H, t,  $J = 6$  Hz), 3.93 (4 H, s).
13. Thomas, E. J.; Whitehead, J. N. F. *J. Chem. Soc. Perkin Trans I*, **1989**, 499.
14. **17:**  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 1.23-1.28 (19 H, m), 1.55-1.59 (4 H, m), 2.00-2.04 (2 H, dt), 2.34-2.35 (4 H, m), 3.92 (4 H, s), 4.12 (2 H, q,  $J = 7$  Hz), 5.34 (1 H, m, C=H), 5.42 (1 H, dt,  $J = 10$  Hz,  $J = 5$  Hz, C=CH);  $^{13}\text{C-NMR}$  for the olefins (*E, Z*) of **17**,  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  127.3 (C=, Z), 127.9 (C=, *E*), 131.4 (C=, Z), 131.7 (C=, *E*).
15. 5%  $\text{AgNO}_3$ -silica gel was prepared by the method of Stowell; Stowell, J. C. *J. Org. Chem.* **1970**, *35*, 244.
16. **18:**  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 1.23-1.47 (16 H, m), 1.87-1.98 (6 H, m), 2.29-2.35 (4 H, m), 3.26 (4 H, s), 4.13 (2 H, q,  $J = 7$  Hz), 5.39 (1 H, dt,  $J = 15$  Hz,  $J = 5$  Hz, C=CH, *E*), 5.47 (1 H, dt,  $J = 15$  Hz,  $J = 5$  Hz, C=CH, *E*);  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  14.0, 14.2, 22.5, 26.8, 27.8, 28.9, 29.4, 29.6, 31.6, 32.4, 34.3, 39.3, 43.4, 60.1, 71.6, 77.2, 127.9 (C=, *E*), 131.6 (C=, *E*), 173.1.
17. **21:**  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$  0.88 (3 H, t,  $J = 7$  Hz), 0.94, 0.98 (9 H, 2s), 1.26-1.30 (16 H, m), 1.87-1.97 (6 H, m), 2.29-2.34 (2 H, m), 2.52-2.74 (2 H, m), 3.26 (4 H, s), 3.85, 3.88 (3 H, 2s), 4.09, 4.17 (1 H, 2d), 4.60, 4.69 (1 H, 2d), 5.08, 5.48 (1 H, 2s), 5.25-5.44 (2 H, m), 8.24, 8.32 (1 H, 2s).
18. (*R*)-and (*S*)-Hydroxy **22** were unstable during silica gel column chromatography or preparative tlc. Especially, the (*S*)-compound was readily decomposed, and could not be purified.
19. Corey, E. J.; Erickson, B. W. *J. Org. Chem.* **1971**, *36*, 3553.
20. **23:** syrup;  $[\alpha]_D^{24} +14.3$  ( $c=1.46$ ,  $\text{CHCl}_3$ ).  $^1\text{H-NMR}$  ( $\text{CD}_3\text{OD}$ )  $\delta$  0.89 (3 H, t,  $J = 7$  Hz), 1.23-1.36 (12 H, m), 1.41-1.56 (6 H, m), 1.93-2.02 (3 H, m), 2.13-2.25 (1 H, m), 2.43 (4 H, t,  $J = 7$  Hz), 3.47 (1 H, d,  $J = 11$  Hz), 3.83 (1 H, d,  $J = 11$  Hz), 3.74 (3 H, s, COOMe), 5.40 (1 H, dt,  $J = 15$  Hz,  $J = 6$  Hz, C=H), 5.46 (1 H, dt,  $J = 15$  Hz,  $J = 6$  Hz, C=CH)
21. Natural product, mycestericin E: white powder; mp 184-186 °C;  $[\alpha]_D^{24} -8.6^\circ$  ( $c=0.06$ , MeOH).
22. Natural product, mycestericin G: white powder; mp 187-189 °C;  $^1\text{H-NMR}$  ( $\text{CD}_3\text{OD}$ )  $\delta$  0.90 (3 H, t,  $J = 7$  Hz), 1.30 (20 H, m), 1.36 (1 H, m), 1.54 (4 H, quint,  $J = 7$  Hz), 1.62 (1 H, m), 2.44 (4 H, t,  $J = 7$  Hz), 3.81 (1 H, d,  $J = 11$  Hz), 3.83 (1 H, m), 3.93 (1 H, d,  $J = 11$  Hz).
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