

SYNTHESIS OF ETHYNYLAZASPIROCYCLOUNDECENE: APPROACH TO SIDE CHAIN UNSATURATED HISTRIONICOTOXINS

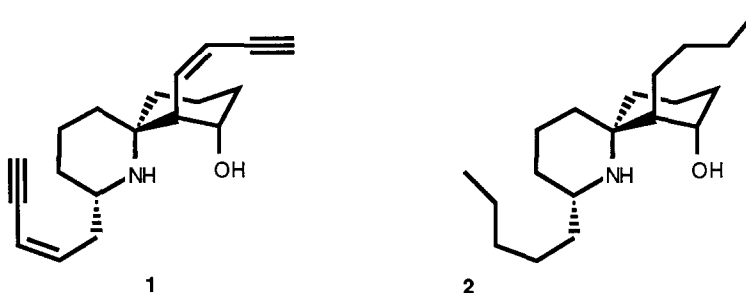
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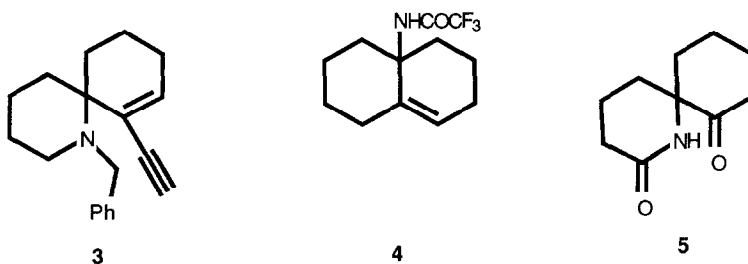
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Summary A short synthesis of *N*-benzyl-7-ethynyl-1-azaspirocyclo[5.5]undec-7-ene (3) from the spirocyclic ketolactam (5) via the vinyl iodide (8) is described. An unusual allylic rearrangement of the lactam (9) occurred during reaction with LiAlH₄.

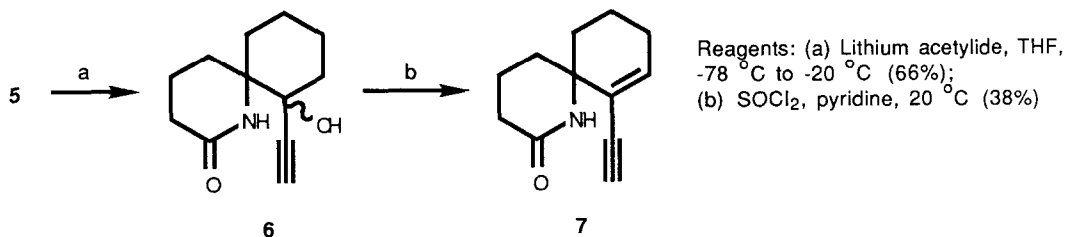
The Colombian frog *Dendrobates histrionicus* is the source of a variety of structurally distinct alkaloids.^{1,2} The histrionicotoxins, isolated from the skin extracts of this species, are a family of 1-azaspiro[5.5]undecan-8-ols possessing unsaturated side-chains at the 2- and 7-positions. Interest in these compounds stems from their unusual neurophysiological properties. Both histrionicotoxin (1) and the fully reduced derivative (2) are potent non-competitive antagonists of acetylcholine.¹⁻³ The search for synthetic routes to (1) and (2) has resulted in many interesting approaches and several syntheses of (2)⁴ as well as Kishi's single route to the more challenging target (1).⁵ We report a synthesis of the azaspirocyclic enyne (3) which formed the basis of our approach to (1).



The known azaspirocyclic ketolactam (5) was prepared either by ruthenium tetroxide-mediated cleavage⁶ of the alkene (4) or from ethyl cyclohexanone-2-carboxylate by the method of Bond.⁷ Treatment of the ketolactam with lithium acetylide in THF at -78 °C gave the acetylenic carbinol (6) as a single isomer. This tertiary alcohol

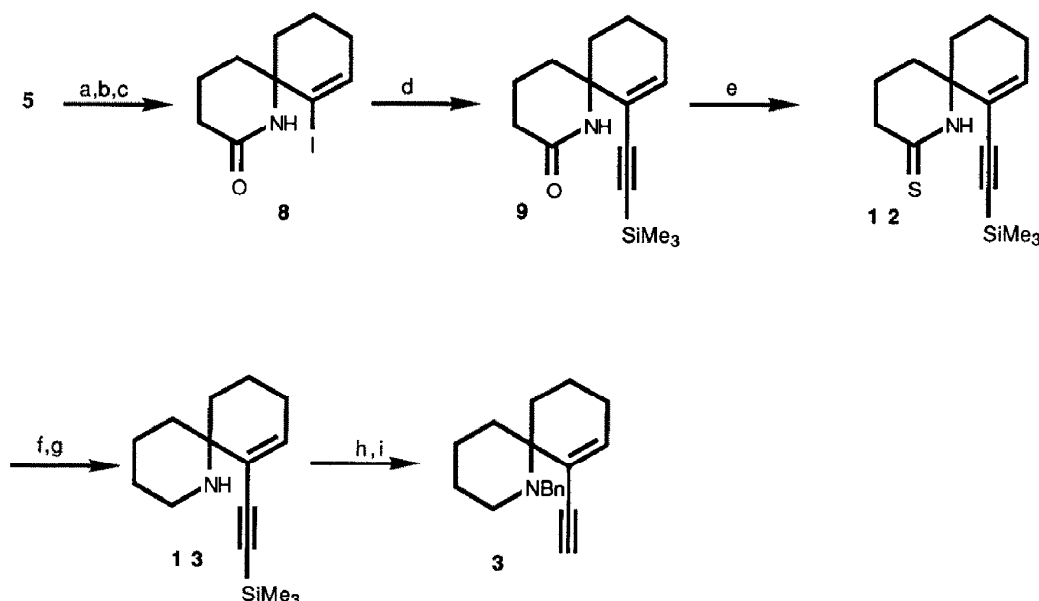


was surprisingly resistant to a wide spectrum of dehydrating conditions, the best of which proved to be thionyl chloride in pyridine (Scheme 1) to give (7) in 38% yield.



SCHEME 1

A more successful alternative route to this family of compounds is shown in Scheme 2. Treatment of (5) with hydrazine gave the corresponding hydrazone (m.p. 113.5-115.5 °C) which was converted into the vinyl iodide (8) (m.p. 139-140.5 °C) by oxidation with iodine in the presence of tetramethylguanidine (TMG).⁸ Palladium(0)-mediated coupling⁹ of (8) with ethynyltrimethylsilane gave the enyne (9) (m.p. 138.5-140 °C) which could be desilylated to give (7) (m.p. 173-174 °C).



SCHEME 2

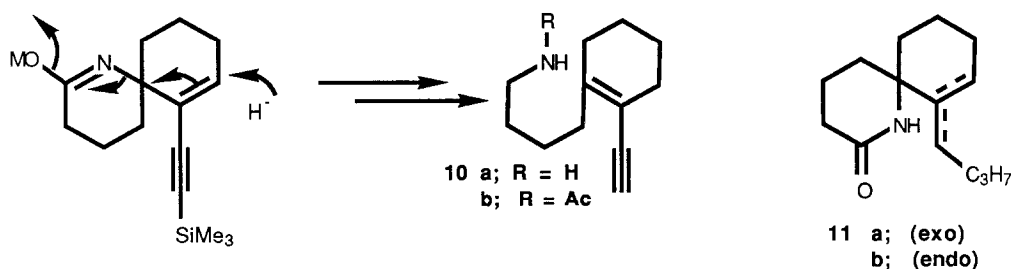
Reagents: (a) $\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$, 100°C ; (b) I_2 , TMG, THF, 20°C ; (c) TMG, 90°C (65%); (d) Ethynyltrimethylsilane, $\text{Pd}(\text{PPh}_3)_4$, CuI, $n\text{BuNH}_2$, benzene (87%); (e) $(p\text{-MeOC}_6\text{H}_4\text{PS}_2)_2$, toluene, 100°C ; (f) $\text{Me}_3\text{O}^+\text{BF}_4^-$, CH_2Cl_2 ; (g) NaBH_4 , MeOH, -78°C to -20°C ; (h) Na_2CO_3 , MeOH; (i) BnBr , NaI, MeCN, $i\text{Pr}_2\text{NEt}$ [58% from (9)]

Reduction of the lactam (9) with LiAlH_4 in refluxing THF gave an unexpected product (10a) in which desilylation of the acetylene probably occurred during aqueous work-up. This compound was characterised as the amide (10b). The formation of (10a) involves reductive allylic rearrangement for which a possible mechanism is shown in Scheme 3. Such allylic rearrangements with LiAlH_4 are relatively rare.¹⁰ All attempts to carry out direct reductions of (9) or the desilylated compound (7) were unsuccessful in complete contrast to the results for LiAlH_4 reduction of the mixture of exo- (11a) and endocyclic allylic amides (11b) where high yields of spirocyclic amine with no rearranged product were observed.⁶

An alternative less direct procedure was then used to reduce the lactam (9) under very mild conditions. Treatment of (9) with Lawesson's reagent¹¹ gave the thiolactam (12) (m.p. $152\text{--}153^\circ\text{C}$) which was converted into the corresponding thioimino ether with Meerwein's reagent (20°C , 2h). Subsequent reduction with NaBH_4 in methanol¹² gave the spirocyclic amine (13) in high yield and with no evidence of allylic rearrangement. Under these conditions partial desilylation of the enyne occurred. This was taken to completion with Na_2CO_3 in methanol (20°C , 1h) to give 7-ethynyl-1-azaspirocyclo[5.5]undec-7-ene which was purified as the *N*-benzyl derivative (3).¹³ In view of the extensive studies on the conversion of the corresponding 7-

butyl compound to perhydrohistrionicotoxin (2) which has been in much demand in isotopically labelled form for biological studies,^{1,2} the ready availability of compound (3) offers a concise approach to highly tritiated perhydrohistrionicotoxin analogues and derivatives.

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SCHEME 3

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13. All new compounds are racemic and exhibited spectroscopic and analytical data consistent with the proposed structure: (3): ¹H δ 7.52-7.14 (5H), 6.35 (1H, t, 4.2Hz), 3.94 & 3.09 (2H, ABq, 13.6 Hz), 2.85 (1H, s), 2.52 & 2.26 (2H, m), 2.11 - 1.47 (12H, m); ¹³C δ 141.2, 139.7, 129.8, 128.8, 127.8, 126.3, 84.4, 75.9, 55.6, 45.3, 35.0, 26.0, 21.7, 20.3, 19.8; (9): ¹H δ 6.24 (1H, t, 4.2 Hz), 5.56 (1H, br s), 2.25 (2H, m), 2.10 (2H, m), 2.07 - 1.6 (8H, m), 0.15 (9H,s); ¹³C δ 171.9, 138.1, 127.6, 103.7, 95.3, 55.5, 37.2, 33.0, 33.1, 25.5, 18.5, 17.5, -0.12.

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