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C₂₀-trifluoro-5-oxo-ETE: A metabolically stable 5-oxo-ETE derivative

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ABSTRACT

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The total synthesis of C₂₀-trifluoro-6(*E*),8(*Z*),11(*Z*),14(*Z*) 5-oxo-ETE is reported. This compound was designed as an ω -oxidation-resistant analog of 5-oxo-ETE that would be resistant to metabolism. The trifluoro derivative of 5-oxo-ETE stimulated calcium mobilization in neutrophils and desensitized these cells to subsequent exposure to 5-oxo-ETE. © 2011 Elsevier Ltd. All rights reserved.

Arachidonic acid (AA) 1 is converted to a large number of biologically active products (eicosanoids) that are important in a variety of pathological conditions, including inflammatory and allergic diseases.^{1–5} Because of the complex array of products with different chiral centers and double bond configurations, chemical synthesis of eicosanoids has been critical for their identification and determination of their physiological and pharmacological properties. Over the past several years, our work has focused on the synthesis and biological effects of a variety of AA-derived products formed both enzymatically by lipoxygenase (LO) and cyclooxygenase (COX) pathways⁶⁻⁹ and nonenzymatically by reactive oxygen species (ROS)-induced oxidation.¹⁰⁻¹² 5-LO, which catalyzes the 5-peroxidation of AA leads to 5-HPETE 4^{13-15} and leukotriene A (LTA₄).¹⁶ LTA₄ is transformed by LTA₄ hydrolase to LTB₄, a potent neutrophil chemoattractant that acts principally via the BLT₁ receptor.¹⁷ LTA₄ is also converted to the cysteinyl leukotriene LTC₄ by the addition of glutathione (GSH). LTC₄ is further metabolized to LTD₄. These substances, previously known as 'slow reacting substance of anaphylaxis', are potent stimulators of bronchoconstriction and vascular permeability and are critical mediators in asthma.

5-oxo-6,8,11,14-eicosatetraenoic acid (5-oxo-ETE) **6** is the most potent chemotactic agent for eosinophils among lipid mediators.¹⁸ It is formed from 5-HETE **5** by the action of a very specific and selective dehydrogenase and exerts its biological actions through a dedicated receptor. Because of its effects on eosinophils, 5-oxo-ETE may be an important mediator in diseases in which these

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cells are prominent, including asthma and other allergic conditions, and diseases of the gastrointestinal system.

5-oxo-ETE 6 is synthesized by 5-hydroxyeicosanoid dehydrogenase (5-HEDH), which is highly specific for eicosanoids containing a 5(S)-hydroxyl group followed by a 6,7-*trans*-double bond (Scheme 1). This enzyme is present in neutrophils,¹⁹ monocytes,²⁰ lymphocytes,²⁰ eosinophils,²¹ platelets,²² and some types of structural cells. 5-oxo-ETE 6 acts through a specific G_i-coupled receptor,^{23–25} which has been cloned^{26,27} and is highly expressed by eosinophils > neutrophils > monocytes as well as by certain tumor cell lines. This receptor has been designated as the OXE receptor (OXE-R). 5-oxo-ETE is active in vivo, eliciting infiltration of eosinophils into rat lung²⁸ and both eosinophils and neutrophils into human skin.²⁹ OXE-R is highly selective for 5-oxo-ETE **6** over a variety of its metabolites, LTs and other eicosanoids. Although 5-oxo-ETE 6 is a chemoattractant for both neutrophils²³ and monocytes,³⁰ it is less potent than LTB₄. In contrast, among lipid mediators, 5-oxo-ETE 6 is the strongest chemoattractant known for human eosinophils²¹ and also induces a variety of other responses in these cells,³¹⁻³³ some of which are markedly enhanced by the proinflammatory cytokines GM-CSF and TNFa.34

An obstacle in the biological evaluation of 5-oxo-ETE is its susceptibility to ω -oxidation²⁴ and incorporation into cellular lipids,²⁵ both of which result in dramatic losses in biological activity. One way to block metabolism by ω -oxidation would be to modify the ω -methyl group of 5-oxo-ETE. This could also potentially interfere with incorporation into lipids. However, alteration of the C₂₀ methyl group could also have a marked effect on biological activity, as 20-hydroxy-5-oxo-ETE is about 100 times less potent than 5-oxo-ETE.²⁴



Scheme 1. Eicosanoids.

Substitution of a trifluoromethyl group for the C₂₀ methyl group of 5-oxo-ETE is one choice for a metabolically stable 5-oxo-ETE agonist. This would prevent ω -oxidation, the major pathway for the metabolism of 5-oxo-ETE, and may also possibly reduce the rate of incorporation in lipids. Scheme 2 shows the synthesis of the CF₃ trifluoro derivative **23**.³⁵

The synthesis of trifluoro derivative **23** was achieved as described in detail in Scheme 2. The aldehyde **12** was obtained by oxidation of **11** which in turn was obtained by protecting triol **9** with dimethoxy propane. The α , β -unsaturated aldehyde **16** was prepared as previously described starting from D-arabinose.³⁶ The trifluoro phosphonium salt **19** was prepared readily from commercial 1,1,1-trifluoro-6-bromohexane³⁷ by refluxing with triphenyl phosphine in acetonitrile.

The use of D-arabinose (Scheme 3) for the preparation of **25** and ultimately **23**, although not chirally economical, is nevertheless a very convenient procedure. The starting material **24** is very cheap and **25** can be prepared in batches of 200 g at a time. Also, **16** can be used for the preparation of 5-HETE and other lipoxygenase products.

To determine the biological potency of **23**, we conducted preliminary experiments to examine its effects on intracellular calcium levels in human neutrophils. Neutrophils were prepared from blood from healthy subjects as previously described by removing erythrocytes with dextran 500 and mononuclear cells by centrifugation over Ficoll–Paque.³⁸ The neutrophils were suspended in phosphate-buffered saline (PBS) and loaded with indo-1 acetoxymethyl ester.²⁴ Five minutes prior to data acquisition, CaCl₂ (1.8 mM) and MgCl₂ (1 mM) were added. Fluorescence was measured using a spectrofluorometer with a temperature-controlled cuvette holder equipped with a magnetic stirrer. After stabilization of the baseline, various concentrations of **6** or **23** were added, followed 1.5 min later by addition of **6** (10 nM) to evaluate agonistinduced desensitization. After another 0.5 min, digitonin (0.1% final concentration) was added.

The response of neutrophils to 5-oxo-ETE (10 nM) following initial addition of vehicle is shown in Figure 1, top curve. There was a strong increase in fluorescence, which peaked a few seconds after addition of 5-oxo-ETE. Digitonin was added before fluorescence returned to baseline to permit measurement of the maximal fluorescence. Addition of C_{20} -trifluoro-5-oxo-ETE (F_3 -50) (100 nM) also resulted in a sharp increase in fluorescence, although somewhat less than that for 5-oxo-ETE and completely abolished the response to 5-oxo-ETE (Fig. 1, bottom curve).

The concentration-response curves for compounds **6** and **23** on calcium mobilization are shown in Figure 2A. 5-oxo-ETE was a potent inducer of calcium mobilization with an EC_{50} of 4 nM. C_{20} -trifluoro-5-oxo-ETE was somewhat less potent, with a maximal response about 50% of that for 5-oxo-ETE. Both of the above agonists desensitized neutrophils to subsequent exposure to



Scheme 2. Total synthesis of long lasting agonist C₂₀-trifluoro-5-oxo-ETE. Reagents and conditions: (a) 10-camphor sulphonic acid, CH₃CN, rt, 80%; (b) PCC, Al₂O₃, dry CH₂Cl₂, 0 °C-rt, 75%; (c) **13**, dry THF, LiHMDS, -78 °C-rt, 73%; (d) THF, TBAF, rt, 87%; (e) PPh₃, imidazole, l₂, dry CH₂Cl₂, rt, 90%; (f) PPh₃, CH₃CN, 50 °C, 2 days, 92%; (g) **15**, dry THF, 1.6 M BuLi, -78 °C-rt, 75%; (h) THF/H₂O, TFA, 78%; (i) dry CH₂Cl₂, Na₂CO₃, LTA, 74%; (j) **19**, dry THF, -78 °C-rt, n-BuLi, 84%; (k) LiOH·H₂O, isopropanol, rt, 93%; (l) TBAF, AcOH, THF, 50 °C, 12 h, 89%; (m) Dess-Martin periodinane, dry CH₂Cl₂, rt, 88%.



Scheme 3. Synthesis of α,β-unsaturated aldehyde 16. Reagents and Conditions: (a) TBDPS-Cl, imidazole, CH₂Cl₂, rt, 98%; (b) periodic acid, THF, ether, 0 °C, 83.3%; (c) 27, THF, rt, 91.5%; d) Pd/C H₂, EtOH, rt, 96.5%; e) periodic acid, THF, ether, rt, 77.5%; (f) 30, benzene, reflux, 76.3%.



Figure 1. Effects of 5-oxo-ETE and trifluoro-5-oxo-ETE on intracellular calcium levels in indo-1-labeled neutrophils. Either vehicle or trifluoro-5-oxo-ETE (100 nM) was added after 1 min, followed by 5-oxo-ETE (10 nM) at 2.5 min and digitonin (0.1%) at 3.5 min. Fluorescence due to the binding of Ca^{++} to indo-1 was monitored using excitation and emission wavelengths of 331 and 410 nm.

5-oxo-ETE, as illustrated by Figure 1. The IC_{50} for 5-oxo-ETE-induced desensitization was 3 nM, whereas that for C_{20} -trifluoro-5-oxo-ETE was 7.5 nM (Fig. 2B).

Although C₂₀-trifluoro-5-oxo-ETE is some what less potent an agonist than 5-oxo-ETE, it is considerably more potent than the 5-oxo-20-HETE **7**, a C₂₀-hydroxy metabolite of 5-oxo-ETE, indicating that addition of a substituent at this position does not necessarily dramatically reduce biological activity. The potent inhibitory effect of C₂₀-trifluoro-5-oxo-ETE on 5-oxo-ETE-induced calcium mobilization (Fig. 2B) suggests that the trifluoro derivative binds strongly to the 5-oxo-ETE receptor, whereas the limited maximal response suggests that it is a partial agonist, having some antagonist properties. These results suggest that further modification of the C₂₀ methyl group may lead to the identification of 5-oxo-ETE analogs with potent agonist or antagonist effects. Also, Figure 2B suggests that an appropriately radio-labeled C₂₀-trifluoro-5-oxo-ETE could be useful in generating an OXE-R binding assay and for long term in vivo experimentation.

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A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.bmcl.2011.02.021.

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Figure 2. Concentration-response relationships for the effects of 5-oxo-ETE and trifluoro-5-oxo-ETE on intracellular calcium levels in neutrophils. (A) Agonist effects of 5-oxo-ETE and trifluoro-5-oxo-ETE on intracellular Ca⁺⁺ levels in neutrophils. (B) Inhibition of 5-oxo-ETE (10 nM)-induced Ca⁺⁺ mobilization in neutrophils due to prior treatment with different concentrations of 5-oxo-ETE or trifluoro-5-oxo-ETE as shown in Figure 1.

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 ¹³C NMR CDCl₃: 199.1, 174.1, 140.09, 136.92, 129.66, 129.61, 129.51, 128.05, 126.99, 126.31, 39.78, 33.78, 33.50, 28.59, 26.80, 26.69, 25.69, 21.47, 19.04.
 ¹NMR CDCl₃: δ 7.52-7.47 (1H, dd, J = 15.2, 11.5), 6.04-6.14 (2H, m), 5.83-5.76 (1H, q), 5.41-5.23 (4H, m), 3.00-3.04 (2H, t, J = 7.2), 2.74-2.77 (2H, t), 2.57-2.61 (2H, t, J = 7.12), 2.35-2.38 (2H, m), 1.87-2.06 (6H, m) 1.46-1.54 (2H, m), 1.34-1.41(2H, m).
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