Reaction of α-Ketoketene S, N-Acetals with Hydroxylamine: A Facile General Route to 5-Aryl-3-(N-arylamino, N-alkylamino, or N-azacycloalkyl)-isoxazoles<sup>1</sup>

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In the course of our synthetic studies on polarized ketene S,S-, S,N-, and N,N-acetals, we have developed simple and convenient routes for N-alkyl(or aryl)-aminopyrazoles<sup>2</sup>, -pyrimidines<sup>3</sup>, and the corresponding -pyridones<sup>4</sup>. As part of this investigation, we now report a facile general method for 5-aryl-3-N-arylamino-(or -alkylamino or -azacycloalkyl)-isoxazoles 2 by reacting 1 with hydroxylamine. Our survey of the literature revealed that the doubly activated ketene S,N-acetal derived from acetylacetone has been reacted<sup>5</sup> with hydroxylamine to give the corresponding 3-anilinoisoxazole. However, no attempts to develop a general synthetic route for 3-N-substituted-aminoisoxazoles from the easily available ketoketene S,N-acetals 1 have been made.

**SYNTHESIS** 

Table. 5-Aryl-3-(N-substituted-amino)-isoxazoles 2a-o prepared

| Prod<br>No. |   | R <sup>1</sup>  | $\mathbb{R}^2$ | Yield<br>[%] | m.p.<br>[°C]   | Molecular<br>formula <sup>a</sup><br>or Lit.<br>m.p. [°C]     | I.R. (KBr)<br>v [cm <sup>-1</sup> ] | <sup>1</sup> H-N.M.R. (CDCl <sub>3</sub> or CDCl <sub>3</sub> /DMSO- $d_6$ ) $\delta$ [ppm]  | M.S.<br>m/e<br>(M+) |
|-------------|---|---|----------------|--------------|----------------|---|-------------------------------------|--|---------------------|
| 2a          | C <sub>6</sub> H <sub>5</sub>                     | C₀H₅  | Н              | 88           | 143-144°       | 142-143°5   | 3340, 1620,<br>1600                 | 6.25 (s, 1 H, H-4); 6.7–7.85 (m, 10 H <sub>arom</sub> ); 8.0 (br. s, 1 H, exchangeable with D <sub>2</sub> O, NH)  | 236                 |
| 2b          | 4-H <sub>3</sub> C—C <sub>6</sub> H <sub>4</sub>  | $C_6H_5$  | Н              | 80           | 186°           | C <sub>16</sub> H <sub>14</sub> N <sub>2</sub> O (250.3)      | 3380, 1625,<br>1600                 | 2.31 (s, 3 H, CH <sub>3</sub> ); 6.28 (s, 1 H, H-<br>4); 6.7-7.7 (m, 4 H <sub>arom</sub> ); 8.3 (m,<br>5 H <sub>arom</sub> ); 8.30 (br. s, 1 H,<br>exchangeable with D <sub>2</sub> O, NH) | 250                 |
| 2c          | 4-H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub> | $C_6H_5$  | Н              | 75           | 174°           | $C_{16}H_{14}N_2O_2$ (266.3)                                  | 3410, 1624,<br>1602                 | 3.78 (s, 3 H, OCH <sub>3</sub> ); 6.22 (s, 1 H, H-4); 6.7-7.5 (m, 9 H <sub>arom</sub> + NH)  | -                   |
| 2d          | 4-Cl—C <sub>6</sub> H <sub>4</sub>                | $C_6H_5$  | Н              | 85           | 194°           | $C_{15}H_{11}CIN_2O$ (270.5)                                  | 3400, 1625,<br>1600                 | 6.30 (s, 1 H, H-4); 6.75–7.8 (m, 9 H <sub>arom</sub> ); 8.62 (br. s, 1 H, exchangeable with D <sub>2</sub> O, NH)  | 272,<br>270         |
| 2e          | C <sub>6</sub> H <sub>5</sub>                     | $C_2H_5$  | Н              | 75           | 101°           | $C_{11}H_{12}N_2O$ (188.2)                                    | 3265, 1625,<br>1600                 | 1.22 (t, 3 H, CH <sub>3</sub> ); 3.25 (br. q, 2 H, CH <sub>2</sub> ); 3.98 (br. s, 1 H, NH); 5.86 (s, 1 H, H-4); 7.15-7.5 (m, 3 H <sub>arom</sub> ); 7.5-7.8 (m, 2 H <sub>arom</sub> )     | _                   |
| 2f          | 4-H <sub>3</sub> CO—C <sub>6</sub> H <sub>4</sub> | $C_2H_5$  | Н              | 69           | <b>78</b> -80° | $C_{12}H_{14}N_2O_2$ (218.3)                                  | 3275, 1628,<br>1600                 | 1.21 (t, 3 H, CH <sub>3</sub> ); 3.22 (q, 2 H, CH <sub>2</sub> ); 3.70 (s, 3 H, OCH <sub>3</sub> ); 3.78 (br. s, 1 H, NH); 5.80 (s, 1 H, H-4); 6.6–7.7 (m, 4 H <sub>arom</sub> )           | an consu            |
| 2g          | 4-Cl—C <sub>6</sub> H <sub>4</sub>                | $C_2H_5$  | Н              | 92           | 141°           | $C_{11}H_{11}CIN_2O$ (222.5)                                  | 3280, 1628,<br>1600                 | 1.21 (t, 3 H, CH <sub>3</sub> ); 3.25 (br. q, 2 H, CH <sub>2</sub> ); 3.75 (br. s, 1 H, NH); 5.81 (s, 1 H, H-4); 7.2–7.7 (m, 4 H <sub>arom</sub> )   | -                   |
| 2h          | $C_6H_5$  | $C_6H_5CH_2$  | Н              | 56           | 136°           | $C_{16}H_{14}N_2O$ (250.3)                                    | 3300, 1622                          | 4.35 (br. s, 3 H, CH <sub>2</sub> + NH); 5.92 (s, 1 H, H-4); 7.1–7.45 (m, 8 H <sub>arom</sub> ); 7.4–7.75 (m, 2 H <sub>arom</sub> )  | 10000               |
| 2i          | 4-Cl—C <sub>6</sub> H <sub>4</sub>                | $C_6H_5CH_2$  | Н              | 86           | 116°           | C <sub>16</sub> H <sub>13</sub> CIN <sub>2</sub> O<br>(284.5) | 3320, 1630                          | 4.30 (br. s, 3 H, CH <sub>2</sub> + NH); 5.90 (s, 1 H, H-4); 7.1–7.5 (m, 7 H <sub>arom</sub> ); 7.5–7.7 (m, 2 H <sub>arom</sub> )  | 286,<br>284         |
| 2j          | 4-H <sub>3</sub> CC <sub>6</sub> H <sub>4</sub>   | C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub>                         | Н              | 92           | 121°           | C <sub>17</sub> H <sub>16</sub> N <sub>2</sub> O (264.3)      | 3320, 1630                          | 2.32 (s, 3 H, CH <sub>3</sub> ); 4.20 (br. s, 1 H, NH); 4.35 (br. s, 2 H, CH <sub>2</sub> ); 5.91 (s, 1 H, H-4); 7.05–7.4 (m, 7 H <sub>arrom</sub> ); 7.4–7.7 (m, 2 H <sub>arrom</sub> )   | * 300               |
| 2k          | C <sub>6</sub> H <sub>5</sub>                     | —(CH <sub>2</sub> ) <sub>4</sub> —                                    |                | 90           | 95°            | $C_{13}H_{14}N_2O$ (214.3)                                    | 1630, 1610,<br>1595                 |  | communication       |
| 21          | C <sub>6</sub> H <sub>5</sub>                     | (CH <sub>2</sub> ) <sub>5</sub>                                       |                | 88           | 85°            | C <sub>14</sub> H <sub>16</sub> N <sub>2</sub> O<br>(228.3)   | 1620, 1595,<br>1580                 | 1.5–1.75 (m, 6 H, CH <sub>2</sub> ); 3.1–3.3 (m,<br>4 H, CH <sub>2</sub> ); 5.98 (s, 1 H, H-4); 7.2–<br>7.4 (m, 3 H <sub>arom</sub> ); 7.5–7.7 (m,<br>2 H <sub>arom</sub> )                |                     |
| 2m          | $C_6H_5$  | (CH <sub>2</sub> ) <sub>2</sub> O(CH <sub>2</sub> ) <sub>2</sub>      |                | 87           | 155°           | $C_{13}H_{14}N_2O_2$ (230.3)                                  | 1625, 1595,<br>1550                 |  | _                   |
| 2n          | $C_6H_5$  | (CH <sub>2</sub> ) <sub>2</sub> N(CH<br>C <sub>6</sub> H <sub>5</sub> | 2)2-           | 88           | 127°           | C <sub>19</sub> H <sub>19</sub> N <sub>3</sub> O<br>(309.4)   | 1625, 1595,<br>1545                 |  |                     |

<sup>&</sup>quot; Satisfactory microanalyses obtained: C  $\pm 0.39$ , H  $\pm 0.33$ , N  $\pm 0.34$ .

When 1a was refluxed with hydroxylamine (generated in situ) in ethanol, corresponding 3-anilinoisoxazole 2a was obtained in 88% yield. The other 3-anilino- (2b-d), 3-ethylamino- (2e-g), and 3-benzylaminoisoxazoles (2h-j) were similarly obtained in 56-92% yields. The corresponding S.N-acetals 1k-n, derived from secondary amines, similarly yielded the corresponding isoxazoles 2k-n in 87-90% yields (Table).

Very few 3-N-substituted-aminoisoxazoles have been reported in the literature. The only known 5-phenyl-3-anilino-(or p-bromoanilino)-isoxazoles were prepared by reaction of hydroxyl-

amine with either phenylpropiolthioanilide<sup>6,7</sup> or benzoyl ketene O,N-acetal<sup>8</sup>. The starting materials are not easily available in both the methods and the yield of isoxazole is low in the former case. The present method therefore provides a simple and high yield method for 2 from the easily available S,N-acetals 1.

The S,N-acetals 1a-j were prepared by our earlier reported procedure<sup>3</sup>, while the cyclic S,N-acetals 1k-n were obtained by methylation of the corresponding thioamides in the presence of potassium carbonate in refluxing acetone<sup>9</sup>.

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## 5-Aryl-3-N-arylamino-(or -alkylamino or -azacycloalkyl)-isoxazoles; General Procedure:

A solution of S, N-acetal 1a-n (0.01 mol) and hydroxylamine (generated from 0.04 mol of hydroxylamine hydrochloride and 0.04 mol of potassium hydroxide in 5 ml of water, neutral to litmus) in ethanol (25 ml) is refluxed for 3-4 h. The ethanol is removed on a water bath and the concentrated reaction mixture is poured into ice-cold water (150 ml). The water layer is extracted with chloroform ( $2 \times 75$  ml), and the chloroform layer is dried with sodium sulfate, and evaporated to give isoxazoles 2a-n which are crystallised from ethanol (Table).

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