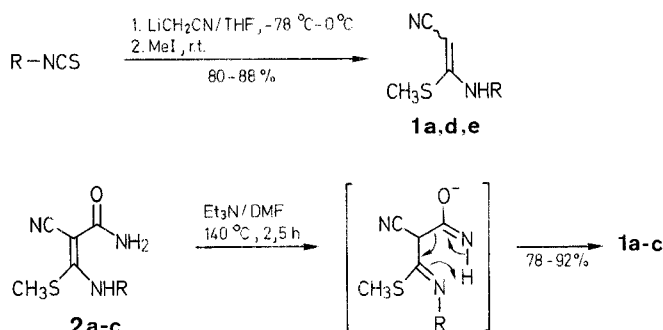


have mainly  $\alpha$ -alkyl and  $\beta$ -carbethoxy groups. However the yield reported are generally low due to the side reactions.<sup>2</sup> In order to increase the yield of the physiologically important 5-hydroxyindoles, we have now used  $\alpha$ -methylthio- $\beta$ -cyanoenamines which can act as effective enamines in this reaction owing to the electronic effect of cyano and methylthio groups.

$\alpha$ -Methylthio- $\beta$ -cyanoenamines [3-(*N*-substituted) 3-(methylthio)acrylonitriles] **1** were prepared by two procedures:

1. The reaction of some appropriate isothiocyanates with acetonitrile lithium anion followed by methylation; and
2. Decarboxyamidation of 3-(*N*-aryl substituted amino)-2-carbamoyl-3-(methylthio)acrylonitriles **2**.

Although there are several reports<sup>3</sup> on decarboxyamidation, the reaction given under 2 is unprecedented. The reaction can be reasonably interpreted by the formation of a six-membered chelate intermediate involving 3-(*N*-substituted) imino group followed by loss of cyanate.



As the reaction medium acetic acid gave the best result even though an effective use of nitromethane had been reported.<sup>4</sup>

Compounds **1** thus obtained (a mixture of *E*-, *Z*-, and imino forms) were allowed to react with 1,4-benzoquinone in acetic acid at room temperature for 4 h to afford *N*-substituted 3-cyano-5-hydroxy-2-(methylthio)indoles **3** in good yield, except **3e** (Table 2). In the preparation of **3e**, 3-cyano-5-hydroxy-2-(methylamino)benzofuran (**5e**) was accompanied in 10% yield as a by-product.

The reaction of **3** with Raney nickel gave *N*-substituted 5-hydroxy-3-methylindoles **4** in moderate yield. Although the direct transformation cyano into methyl groups by means of catalytic hydrogenation has appeared in the literature,<sup>5</sup> the present reaction accompanied by desulfurization is noteworthy.

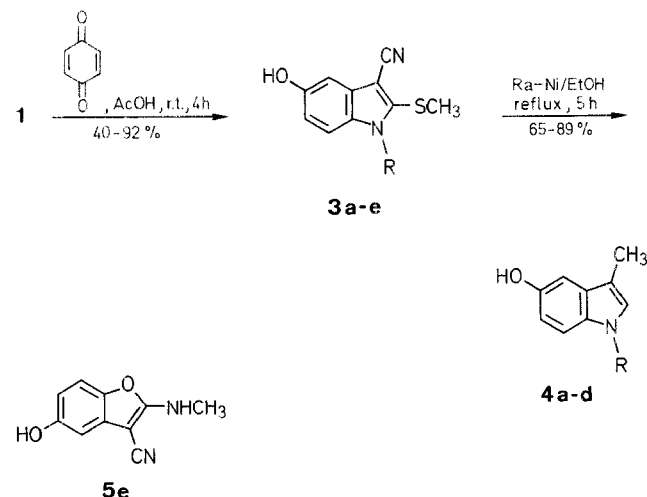
### Synthesis of *N*-Substituted 5-Hydroxyindoles

Masataka Yokoyama,\* Satoshi Watanabe, Hidekatsu Hatanaka

Department of Chemistry, Faculty of Science, Chiba University, Yayoi-cho, Chiba City 260, Japan

$\alpha$ -Methylthio- $\beta$ -cyanoenamines are found to be useful starting compounds for the synthesis of *N*-substituted 5-hydroxyindoles. In connection with the preparation of the starting compounds, a novel decarboxyamidation is presented.

The Nenitzescu reaction has been well known as an important synthetic method of 5-hydroxyindoles.<sup>1</sup> The reaction involves the condensation of 1,4-benzoquinones with enamines which



**Table 1.** Acrylonitriles **1** Prepared

Product	R	Yield (%)	E/Z/Imino Form <sup>a</sup>	Molecular Formula <sup>b</sup>	MS <i>m/e</i>	IR (KBr) $\nu$ (cm <sup>-1</sup> )
<b>1a</b>	C <sub>6</sub> H <sub>5</sub>	92 <sup>c</sup> , 88 <sup>d</sup>	1:1:1	C <sub>10</sub> H <sub>10</sub> N <sub>2</sub> S (190.3)	190, 143	3250, 3220, 2180
<b>1b</b>	4-ClC <sub>6</sub> H <sub>4</sub>	91 <sup>c</sup>	1:1:3	C <sub>10</sub> H <sub>8</sub> ClN <sub>2</sub> S (224.7)	226, 224	3280, 3070, 2180
<b>1c</b>	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	78 <sup>c</sup>	1:1:1	C <sub>11</sub> H <sub>12</sub> N <sub>2</sub> OS (220.3)	220, 173	3210, 2950, 2200
<b>1d</b>	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub>	80 <sup>d</sup>	2:1	C <sub>11</sub> H <sub>12</sub> N <sub>2</sub> S (204.3)	204, 91	3320, 3010, 2900, 2180
<b>1e</b>	CH <sub>3</sub>	86 <sup>d</sup>	6:1	C <sub>5</sub> H <sub>8</sub> N <sub>2</sub> S (128.2)	128, 81	3300, 3040, 2980, 2180

<sup>a</sup> Estimated by <sup>1</sup>H-NMR spectra.<sup>b</sup> Satisfactory microanalyses obtained: C  $\pm$  0.27, H  $\pm$  0.21, N  $\pm$  0.18.<sup>c</sup> Prepared from compound **2**.<sup>d</sup> Prepared from alkyl isothiocyanate.**Table 2.** <sup>1</sup>H-NMR Data of **1a-e**

Compound	<sup>1</sup> H-NMR (CDCl <sub>3</sub> ) $\delta$ , J (Hz)
<b>1a</b>	2.10 (s, 1H); 2.34 (s, 1H); 2.36 (s, 1H); 3.22 (br, $\frac{2}{3}$ H); 4.00 (s, $\frac{1}{3}$ H); 4.44 (s, $\frac{1}{3}$ H); 6.60 (br, $\frac{2}{3}$ H); 7.00 (m, 5H)
<b>1b</b>	2.36 (s, $\frac{3}{5}$ H); 2.56 (s, $\frac{2}{5}$ H); 2.64 (s, $\frac{3}{5}$ H); 3.44 (br, $\frac{2}{5}$ H); 4.20 (s, $\frac{1}{5}$ H); 4.75 (s, $\frac{1}{5}$ H); 6.00 (br, $\frac{1}{5}$ H); 6.60–7.40 (m, 2 $\frac{1}{2}$ H)
<b>1c</b>	2.16 (s, 1H); 2.40 (s, 1H); 2.48 (s, 1H); 3.28 (s, $\frac{2}{3}$ H); 3.68 (s, $\frac{10}{3}$ H); 3.84 (s, $\frac{1}{3}$ H); 4.30 (s, $\frac{1}{3}$ H); 5.96 (br, $\frac{1}{3}$ H); 6.44–7.08 (m, 4H)
<b>1d</b>	2.28 (s, 1H); 2.44 (s, 2H); 3.70 (s, $\frac{1}{3}$ H); 4.10 (d, $\frac{2}{3}$ H, <i>J</i> = 6); 4.16 (s, $\frac{2}{3}$ H); 4.44 (d, $\frac{2}{3}$ H, <i>J</i> = 6); 4.85 (br, $\frac{2}{3}$ H); 5.30 (br, $\frac{1}{3}$ H); 7.20 (s, 5H)
<b>1e</b>	2.25 (s, $\frac{3}{7}$ H); 2.36 (s, $\frac{18}{7}$ H); 2.68 (d, $\frac{18}{7}$ H, <i>J</i> = 5); 2.92 (d, $\frac{3}{7}$ H, <i>J</i> = 5); 3.54 (s, $\frac{1}{7}$ H); 3.80 (s, $\frac{9}{7}$ H); 5.50 (br, 1H)

**Table 3.** 5-Hydroxyindoles **3** and **4** Prepared

Product	R	Yield (%) <sup>a</sup>	m. p. (°C) or b. p. (°C)/mbar	Molecular Formula <sup>b</sup>	MS <i>m/e</i>
<b>3a</b>	C <sub>6</sub> H <sub>5</sub>	79	231–232	C <sub>16</sub> H <sub>12</sub> N <sub>2</sub> OS (280.3)	280, 265
<b>3b</b>	4-ClC <sub>6</sub> H <sub>4</sub>	92	187–188	C <sub>16</sub> H <sub>11</sub> ClN <sub>2</sub> O (314.8)	316, 314, 264
<b>3c</b>	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	87	204–205	C <sub>17</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S (310.4)	310, 295
<b>3d</b>	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub>	89	221–222	C <sub>17</sub> H <sub>14</sub> N <sub>2</sub> OS (294.4)	294, 279
<b>3e</b>	CH <sub>3</sub>	40	163–164	C <sub>11</sub> H <sub>10</sub> N <sub>2</sub> OS (218.3)	218, 203
<b>4a</b>	C <sub>6</sub> H <sub>5</sub>	89	oil, 175/0.07	C <sub>15</sub> H <sub>13</sub> NO (223.3)	223
<b>4b</b>	4-ClC <sub>6</sub> H <sub>4</sub>	83	oil, 200/0.2	C <sub>15</sub> H <sub>12</sub> ClNO (257.7)	258, 256, 221
<b>4c</b>	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	85	oil, 250/0.05	C <sub>16</sub> H <sub>15</sub> NO <sub>2</sub> (253.3)	253, 238
<b>4d</b>	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub>	65	142–143	C <sub>16</sub> H <sub>15</sub> NO (237.3)	237, 146

<sup>a</sup> Yields of compounds **3** and **4** are isolated yields, based on compounds **1** and **3**, respectively.<sup>b</sup> Satisfactory microanalyses obtained: C  $\pm$  0.23, H  $\pm$  0.09, N  $\pm$  0.11.**Table 4.** Spectral Data of **3a-e** and **4a-d**

Compound	IR (KBr or neat) $\nu$ (cm <sup>-1</sup> )	<sup>1</sup> H-NMR (CDCl <sub>3</sub> -DMSO- <i>d</i> <sub>6</sub> /TMS) $\delta$ , J (Hz)
<b>3a</b>	3300, 2220, 1615, 1490	2.28 (s, 3H); 6.40 (s, 1H); 6.76 (d, 1H, <i>J</i> = 8); 7.20–7.60 (m, 7H)
<b>3b</b>	3300, 2210, 1620, 1490	2.36 (s, 1H); 6.48 (s, 1H); 6.84 (d, 1H, <i>J</i> = 8); 7.28 (d, 2H, A part of AB system, <i>J</i> = 8); 7.40–7.56 (m, 3H)
<b>3c</b>	3300, 2210, 1615, 1510, 1490	2.30 (s, 3H); 3.80 (s, 3H); 6.38 (s, 1H); 6.72 (d, 1H, <i>J</i> = 8); 6.90 (d, 2H, B part of AB system, <i>J</i> = 8); 7.12 (d, 2H, A part of AB system, <i>J</i> = 8); 7.25 (d, 1H, <i>J</i> = 4)
<b>3d</b>	3250, 2210, 1610, 1490	2.32 (s, 3H); 5.28 (s, 2H); 7.00 (s, 5H); 6.40–7.40 (m, 3H)
<b>3e</b>	3310, 2220, 1630, 1580	2.40 (s, 1H); 3.62 (s, 3H); 6.60 (s, 1H); 6.64 (d, 1H, <i>J</i> = 8); 7.20 (d, 1H, <i>J</i> = 8)
<b>4a</b>	3300, 2900, 1700, 1610, 1590	2.28 (s, 3H); 5.40 (br, 1H); 6.68 (d, 1H, <i>J</i> = 8); 6.80–7.20 (m, 3H); 7.28 (s, 5H)
<b>4b</b>	3300, 2900, 1620, 1595	2.20 (s, 3H); 6.40–7.30 (m, 8H)
<b>4c</b>	3350, 2900, 1620	2.26 (s, 3H); 3.78 (s, 3H); 6.40–7.30 (m, 8H)
<b>4d</b>	3350, 2900, 1620	2.16 (s, 3H); 4.92 (s, 2H); 6.50 (m, 2H); 6.80–7.15 (m, 6H)

**3-(Benzylamino)-3-(methylthio)acrylonitrile (1d); Typical Procedure:**

To a stirred mixture of CH<sub>3</sub>CN (1.23 g, 30 mmol) and THF, (40 mL) is added an 1.65 molar hexane solution of BuLi (17 mL, 28 mmol) at –78°C. After stirring for 0.5 h at the same temperature benzyl isothiocyanate (1.49 g, 10 mmol) is added to the mixture. The resulting mixture is stirred for 0.5 h at 0°C and then extracted with water (3  $\times$  40 mL). The extract is washed with benzene (50 mL) and then treated with CH<sub>3</sub>I (4.26 g, 30 mmol). The mixture is stirred at room temperature for 1 h and then extracted with EtOAc (3  $\times$  50 mL). The extract is concentrated *in vacuo* to give a yellow material. Recrystallization from ethanol gives white prisms of **1d**; yield: 4.9 g (80%) (Table 1).

**(E)-3-(Phenylamino)-2-cyano-3-(methylthio)acrylamide (2a); Typical Procedure:**

A mixture of 2-cyano-3,3-bis(methylthio)acrylamide<sup>6</sup> (3.8 g, 20 mmol), aniline (2 mL, 22 mmol), and ethanol (40 mL) is refluxed for 15 h. The white crystals formed are collected, washed with ethanol, and recrystallized from ethanol to give **2a** as white needles; yield: 4.6 g (99%); m. p. 147–149°C (Lit.<sup>7</sup> m. p. 147–149°C). By the same method as mentioned above compounds **2b** and **2c** were prepared starting from 4-chloroaniline and 4-methoxyaniline, respectively.

**2b:** White prisms; yield: 98%; m. p. 198–199°C.C<sub>11</sub>H<sub>10</sub>ClN<sub>3</sub>OS calc. C 49.35 H 3.76 N 15.69 (267.7) found 49.05 3.75 15.67IR (KBr):  $\nu$  = 3360, 3310, 3250, 2200, 1680, 1640, 1600, 1580, 1560 cm<sup>-1</sup>.<sup>1</sup>H-NMR (CDCl<sub>3</sub>/DMSO-*d*<sub>6</sub>):  $\delta$  = 2.15 (s, 3H); 6.0 (br, 2H); 7.28 (s, 4H); 12.44 (br, 1H).

IR (KBr):  $\nu = 3360, 3310, 3250, 2200, 1680, 1640, 1600, 1580, 1560 \text{ cm}^{-1}$ .

$^1\text{H-NMR}$  ( $\text{CDCl}_3/\text{DMSO}-d_6$ ):  $\delta = 2.15$  (s, 3 H); 6.0 (br, 2 H); 7.28 (s, 4 H); 12.44 (br, 1 H).

MS:  $m/e = 267$  ( $\text{M}^+$ ).

**2c:** White prisms; yield 99%; m.p. 159–160°C.

$\text{C}_{12}\text{H}_{13}\text{N}_3\text{O}_2\text{S}$  calc. C 54.74 H 4.98 N 15.96  
(263.3) found 54.72 4.99 15.93

IR (KBr):  $\nu = 3390, 3300, 3180, 3000, 2950, 2200, 1640, 1600, 1540 \text{ cm}^{-1}$ .

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta = 2.80$  (s, 3 H); 3.76 (s, 3 H); 5.70 (br, 2 H); 6.80 (d, 2 H,  $J = 8 \text{ Hz}$ , B part of AB system); 7.08 (d, 2 H,  $J = 8 \text{ Hz}$ , A part of AB system); 12.15 (br, 1 H).

MS:  $m/e = 263$  ( $\text{M}^+$ ).

#### Decarboxyamidation of 2a: Typical Procedure:

A mixture of **2a** (233 mg, 1 mmol), triethylamine (323 mg, 3.2 mmol), and DMF (3 mL) is stirred at 140°C under nitrogen for 2.5 h. The mixture is quenched with aqueous  $\text{NH}_4\text{Cl}$  solution and then extracted with benzene ( $3 \times 25 \text{ mL}$ ). The benzene extract is washed with water (30 mL), dried ( $\text{Na}_2\text{SO}_4$ ), and evaporated *in vacuo* to give a yellow oil, which is subjected to preparative TLC on silica gel (EtOAc/hexane, 1:2) to give **1a** in 92% yield as white needles (recrystallization from ether or hexane).

#### 3-Cyano-5-hydroxy-2-methylthio-1-phenylindole (3a): Typical Procedure:

A mixture of **1a** (1.9 g, 10 mmol), 1,4-benzoquinone (1.1 g, 10 mmol), and acetic acid (30 mL) is stirred at room temperature for 4 h. The resulting precipitate is collected and washed with ethanol. Recrystallization from ethanol gives white needles; yield: 2.2 g (79%); m.p. 231–232°C.

In the case of **3e**, it did not precipitate after the reaction is completed. The reaction mixture is chromatographed on silica gel column using EtOAc/hexane (1:1) as eluent and then purified by preparative TLC on silica gel using  $\text{CHCl}_3/\text{EtOAc}$  (9:1) as eluent (**3e** moves faster than **5e**).

**5e:** White crystals; m.p. 165°C (dec).

$\text{C}_{10}\text{H}_8\text{N}_2\text{O}_2$  calc. C 63.83 H 4.28 N 14.89  
(188.2) found 63.80 4.27 14.90

IR (KBr):  $\nu = 3350, 3300, 2200, 1640 \text{ cm}^{-1}$ .

$^1\text{H-NMR}$  ( $\text{CDCl}_3/\text{DMSO}-d_6$ ):  $\delta = 3.0$  (s, 3 H); 4.44 (br, 1 H, appears together with  $\text{H}_2\text{O}$  peak); 6.3–6.8 (m, 3 H); 7.08 (br, 1 H).

MS:  $m/e = 188$  ( $\text{M}^+$ ).

#### 5-Hydroxy-3-methyl-1-phenylindole (4a): Typical Procedure:

A mixture of **3a** (1.4 g, 5 mmol), activated Raney nickel<sup>8</sup> (ca 10 g), and ethanol (80 mL) is refluxed for 5 h. The mixture is separated from nickel by decantation and centrifugation. The nickel is washed with ethanol and the washings are also centrifuged. The combined alcoholic solution is concentrated *in vacuo* to give a yellow oil, which is purified by preparative TLC on silica gel (EtOAc); colorless oil; yield: 0.99 g (89%); b.p. 175°/0.07 mbar.

Received: 5 January 1987; revised: 17 March 1987

- (1) Sundberg, R.J., in: *Comprehensive Heterocyclic Chemistry*, Katritzky, A.R., Rees, C.W. (eds.), Vol. 4, Pergamon Press, Great Britain, 1984, p. 340.
- (2) Allen, G.R., Jr., in: *Org. React.*, Dauben, W.G. (ed.), Vol. 20, John Wiley & Sons, New York, 1973, p. 337, and references cited therein.
- (3) Orchin, M., Reggel, L. *J. Am. Chem. Soc.* **1951**, 73, 436.  
Campagne, E., Bulbenko, G.F. *J. Org. Chem.* **1961**, 26, 4702.  
Nakagawa, H., Nakaminami, G., Ogura, F., Ono, H. *Bull. Chem. Soc. Jpn.* **1962**, 35, 1488.  
Craig, C., Moyle, M. *J. Chem. Soc.* **1963**, 4402.  
Smith, H.A., Hauser, C.R. *J. Am. Chem. Soc.* **1969**, 91, 7774.
- (4) Patrick, J.B., Saunders, E.K. *Tetrahedron Lett.* **1979**, 4009.
- (5) Andrade, J.G., Maier, W.F., Zapf, L., Schleyer, P.v.R. *Synthesis*, **1980**, 802, and references cited therein.
- (6) Takeshima, T., Yokoyama, M., Fukada, N., Akano, M. *J. Org. Chem.* **1970**, 35, 2438.

- (7) Yokoyama, M., Hatanaka, H., Sasaki, A., Shiraishi, T., Kumata, K., Sakamoto, K., Ogata, K. *J. Chem. Soc., Perkin Trans. 1* **1986**, 1187.
- (8) Mozingo, R., *Org. Synth. Coll. Vol. III* **1967**, 181.