

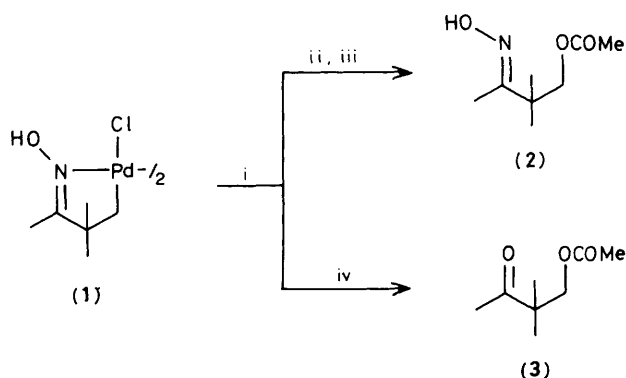
Oxidation of Unactivated Methyl Groups via a Cyclopalladation Reaction

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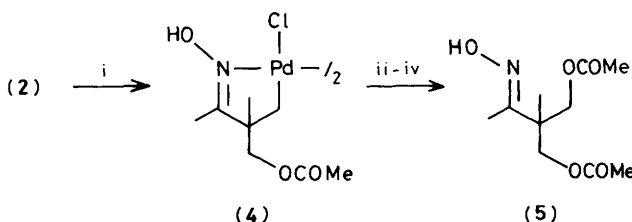
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The successive cyclopalladation–oxidation of *E*-pinacolone, *E*-2,2-dimethylcyclohexanone, 2,2,6,6-tetramethylcyclohexanone, and *E*-lupanone oxime led to the corresponding β -acetoxy derivatives; the second palladation in the case of the acetoxy compounds (2) and (8) takes place in one of the remaining methyl groups, being a regiospecific process, and in the case of the cyclic oximes the palladation occurs in the equatorial methyl group.

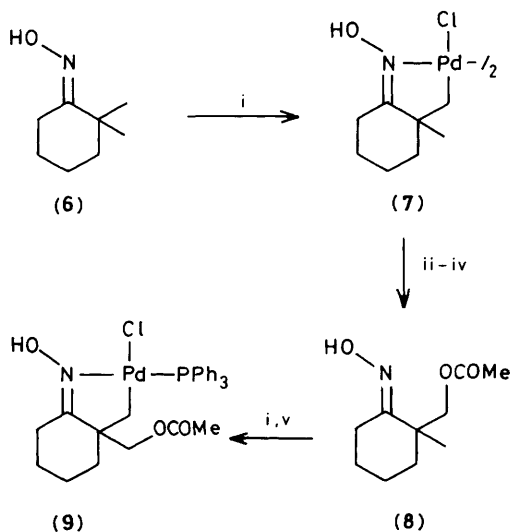
The cyclopalladation of *E*-pinacolone oxime leading to the dimeric organopalladium compound (1) has been described.¹



Scheme 1. Reagents: i, pyridine; ii, Pb(OAc)₄; iii, NaBH₄; iv, 2 equiv. Pb(OAc)₄.



Scheme 2. Reagents: i, Na₂PdCl₄–NaOAc; ii, pyridine; iii, Pb(OAc)₄; iv, NaBH₄.



Scheme 3. Reagents: i, Na₂PdCl₄; ii, pyridine; iii, Pb(OAc)₄; iv, NaBH₄; v, PPh₃.

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Very recently, the application of this reaction to the cyclopalladation of lanost-8-en-3-one oxime and its further deuteration or iodination has been published.² These results prompt us to report our independent findings in this field.

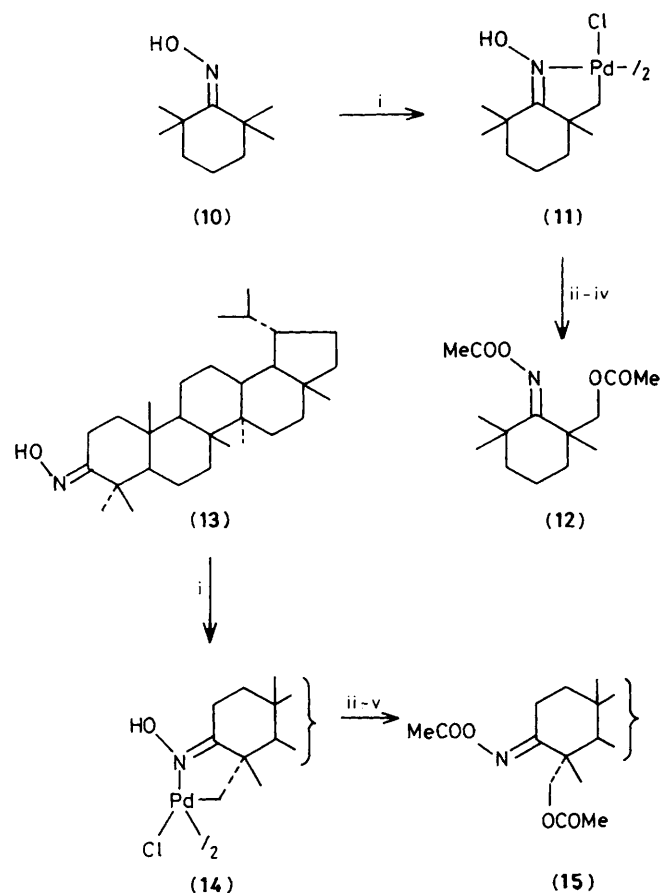
The oxidation of the pyridine complex derived from the compound (1) with lead tetra-acetate (1 equiv.) in acetic acid at room temperature led after reduction with sodium borohydride to the acetoxy oxime (2) (ca. 100% yield).§ In the absence of pyridine the reaction did not work. If the same process was carried out with double the stoichiometric amount of lead tetra-acetate the corresponding compound (3) was isolated (64% yield)§ (Scheme 1).

A second palladation–oxidation of compound (2) was done in order to study the regiochemistry of this process. Thus, when compound (2) was palladated the organopalladium dimer (4) was isolated (83% yield),§ and subsequent oxidation of this compound led to the diacetoxy oxime (5) (86% yield)§ (Scheme 2).

The palladation of *E*-2,2-dimethylcyclohexanone oxime (6) yielded the corresponding dimer (7) (85% yield)§ which was oxidized through its pyridine complex to give the compound (8) (96% yield).§ When this compound was again palladated the metallation took place on the remaining methyl group³ affording the organopalladium compound (9) after treatment with triphenylphosphine (82% yield)§ (Scheme 3).

‡ When the reaction product bears the oxime group the reduction of Pd^{II} to Pd⁰ was carried out to avoid the trapping of the Pd^{II} by the oxime.

§ All new compounds have yielded satisfactory analytical and/or spectral data. Selected data: (2), m.p. 85–86°C; ν 1730 cm^{−1} (C=O); δ_{H} 4.07 (CH₂O); δ_{C} 70.01 (CH₂O) and 170.98 (C=O); m/z 174 (M^+ + 1). (3), b.p. 100°C (0.1 mmHg, Kugelrohr); ν 1750 and 1710 cm^{−1} (C=O); δ_{H} 2.16 (MeCO) and 4.10 (CH₂O); δ_{C} 69.82 (CH₂O), 170.66 (CO₂), and 210.89 (C=O); m/z 159 (M^+ + 1). (4), m.p. 135–138°C; δ_{H} 2.34, 2.63 (2 d, J 8.4 Hz, CH₂Pd), 3.96, and 4.01 (2d, J 11.3 Hz, CH₂O); δ_{C} 33.67 (CH₂Pd) and 68.84 (CH₂O). (5), m.p. 45–47°C; δ_{H} 4.18 (CH₂O); δ_{C} 65.70 (CH₂O). (7), m.p. 182–185°C (decomp.); δ_{H} 2.51 and 2.57 (2d, J 8.2 Hz, CH₂Pd) and δ_{C} 26.48 (Me) and 40.40 (CH₂Pd); m/z 562 (M^+). (8), m.p. 125°C (0.1 mmHg, Kugelrohr); ν 1740 cm^{−1} (C=O); δ_{H} 4.08 and 4.20 (2d, J 11.0 Hz, CH₂O); δ_{C} 69.10 (CH₂O) and 171.07 (C=O); m/z 199 (M^+). (9), m.p. 156–158°C (decomp.); δ_{H} 1.40, 1.87 (2d, J 10.0 Hz, CH₂Pd), 3.81, 4.28 (2d, J 10.8 Hz, CH₂O), and 10.67 (d, J_{HP} 3.0 Hz, OH); δ_{C} 42.30 (CH₂Pd) and 68.29 (CH₂O); δ_{P} 30.49 p.p.m. (11), m.p. 177–179°C (decomp.); δ_{H} 2.33 and 2.46 (2d, J 7.1 Hz, CH₂Pd); δ_{C} 41.22 (CH₂Pd). (12), oil; ν 1745 cm^{−1} (C=O); δ_{H} 2.06, 2.19 (2 \times MeCO), 4.07, and 4.31 (2d, J 10.7 Hz, CH₂O); δ_{C} 20.07, 20.87 (2 \times CH₃CO), 71.34 (CH₂O), 169.16, 170.78, and 172.85 (C=N and 2 \times C=O). (14), m.p. > 210°C; δ_{H} 2.50 and 2.61 (br. s and d, J 7.5 Hz, CH₂Pd); δ_{C} 43.15 (CH₂Pd). (15), oil; ν 1740 cm^{−1} (C=O); δ_{H} 2.06, 2.17 (2 \times MeCO), 4.09, and 4.19 (2d, J 11.0 Hz, CH₂O); δ_{C} 68.29 (CH₂O), 169.65, and 170.84 (C=N and 2 \times C=O); m/z 542 (M^+ + 1).



Scheme 4. Reagents: i, $\text{Na}_2\text{PdCl}_4\text{-NaOAc}$; ii, Ac_2O ; iii, pyridine; iv, $\text{Pb}(\text{OAc})_4$; v, NaBH_4 .

The palladation of 2,2-dimethylcyclopentanone oxime fails owing to the difficulty in achieving planarity of the oxime and methyl groups necessary for palladation within the constraints applied by the five-membered ring.³

We have also studied the palladation-oxidation in a hindered ketone. Thus, when 2,2,6,6-tetramethylcyclohexanone oxime (**(10)**) was palladated the expected organopalladium dimer (**(11)**) was obtained (86% yield).§ In this case the oxime group must be acetylated in order to avoid its isomerization during the oxidation; the acetylation and further oxidation of compound (**(11)**) led to the product (**(12)**) (76% yield)§ (Scheme 4).

Finally, we have palladated lupanone oxime (**(13)**), isolating the expected dimer (**(14)**) (76% yield),§ which after acetylation and oxidation led to the expected 23-acetoxy derivative (**(15)**) (90% yield)§ (Scheme 4). The stereochemistry of (**(14)**), and hence the acetoxy derivative (**(15)**), was established by nuclear Overhauser effect (n.O.e.) studies at 500 MHz.

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