

## Enantioselective Construction and Utilization of 2-(Cyclohex-2-enyl)phenols

Hiroyuki Konno and Kunio Ogasawara\*

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980-8578, Japan

Fax +81-22-217-6845

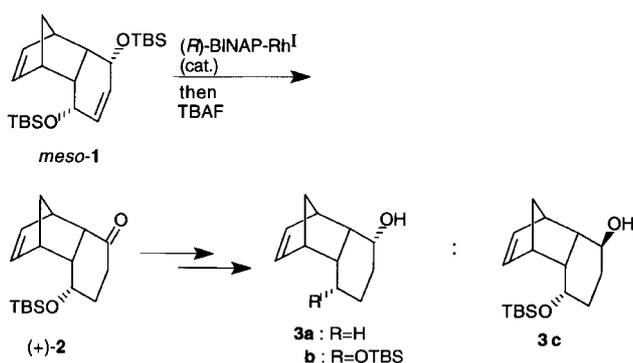
E-mail konol@mail.cc.tohoku.ac.jp

Received 22 May 1998

**Abstract:** Enantioselective construction of 2-(cyclohex-2-enyl)phenols from chiral equivalents of cyclohex-2-enols has been investigated by employing a concurrent retro-Diels-Alder reaction and Claisen rearrangement protocol. Utilizing the enantiomerically pure product obtained, the first enantiocontrolled synthesis of a phenolic natural sesquiterpene (+)-curcudiol, isolated from the marine sponge *Didiscus flavus*, has been demonstrated.

We have encountered<sup>1</sup> difficulties in preservation of the original chiral integrity of chiral allyl alcohols in their conversion into chiral 2-allylphenols<sup>2</sup> through a sequential Mitsunobu aryl ether formation<sup>3,4</sup> and Claisen rearrangement.<sup>5,6</sup> The enantiomeric purities of the products were diminished considerably in both the ether formation<sup>4,7</sup> and the rearrangement steps. In relation to our recent development<sup>8</sup> of a chiral building block serving as chiral cyclohexenone, we examined the synthesis of chiral 2-(cyclohex-2-enyl)phenols using the chiral cyclohexenone synthon to extend its synthetic utility as the chiral cyclohexenol equivalent. We wish to report here an enantio- and stereo-selective construction of 2-(6-*tert*-butyldimethylsiloxy-2-cyclohexenyl)phenol and its conversion into a natural phenolic sesquiterpene (+)-curcudiol isolated from the marine sponge *Didiscus flavus*.<sup>9</sup>

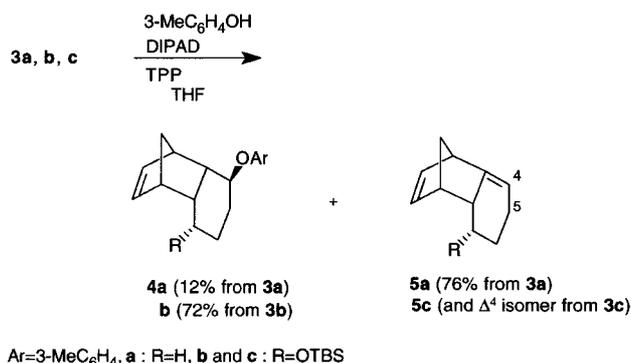
Since one major factor leading to loss of the original enantiomeric purity was due to the racemization in the Mitsunobu coupling using allylic alcohols,<sup>4</sup> we first examined the coupling between 3-methylphenol and each of the three non-allylic chiral cyclohexanols serving as chiral cyclohex-2-enols to avoid the allylic racemization. Thus, the enantiomerically pure ketone (+)-**2**, obtained by a chiral BINAP-Rh<sup>I</sup>-catalyzed asymmetric<sup>8</sup> of the *meso*-endiol bis-silyl ether **1**, was first transformed into the three substrates, chiral cyclohex-2-enol equivalent **3a** and *syn*-1,4 and *anti*-1,4 substituted chiral cyclohex-2-enol equivalents, **3b** and **3c** (Scheme 1).



Scheme 1

The Mitsunobu reaction of these non-allylic alcohols with 3-methylphenol was carried out in the presence of two equivalents each of diisopropyl azodicarboxylate (DIPAD) and triphenylphosphine (TPP) in THF. Only the *endo*-alcohol **3b**, m.p. 128–130 °C,  $[\alpha]_D^{29} -16.9$  (*c* 1.1, CHCl<sub>3</sub>), carrying a *syn*-4-substituent furnished the expected *exo*-aryl ether **4b**,  $[\alpha]_D^{34} +33.6$  (*c* 1.2, CHCl<sub>3</sub>), carrying an *anti*-4-substituent in

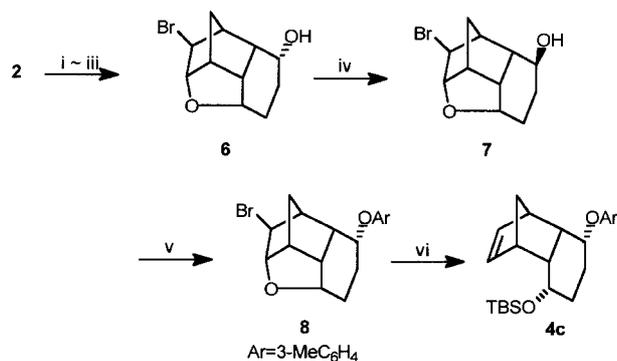
satisfactory yield (72%) as the single product. Dehydration prevailed in the reaction of the *endo*-alcohol **3a**,  $[\alpha]_D^{32} -43.7$  (*c* 0.6, CHCl<sub>3</sub>), not carrying a 4-substituent to give the elimination product **5a** in 76% yield as a volatile oil and only 12% yield of the desired *exo*-ether **4a**,  $[\alpha]_D^{26} +7.2$  (*c* 2.5, CHCl<sub>3</sub>), was generated. In the reaction of the *exo*-alcohol **3c** carrying an *anti*-4-substituent, dehydration occurred exclusively to afford a mixture of the two regioisomeric olefins, **5c** and its  $\Delta^4$ -isomer, without generation of the aryl ether **4c** having the *exo-syn*-1,4-stereochemistry. The observed specific ether formation in the reaction of the *endo*-alcohol **3b** may be due to the steric repulsion between the *endo-syn*-1,4-substituents which alleviates the antiperiplanar hydrogen and oxygen-phosphorus disposition appropriate for the E2-elimination. Meanwhile, with **3a** and **3c** which do not have the 1,4-steric interaction, the elimination rather than the substitution occurs to give the olefins as observed (Scheme 2).



Scheme 2

We next examined the thermolysis of the aryl ethers. For comparison, we prepared the *endo*-aryl ether **4c**, which could not be obtained directly from **3c**, by employing an alternative way from the same keto-ether **2**. Thus, **2**, after desilylation, was exposed to NBS to form the bromo ketone having an ether linkage which was then reduced from the convex face to give the bromohydrin **6**, m.p. 133–134 °C,  $[\alpha]_D^{31} +103.0$  (*c* 1.1, CHCl<sub>3</sub>). The Mitsunobu reaction of **6** with 4-nitrobenzoic acid,<sup>10</sup> followed by the hydrolysis of the resulting benzoate gave the *exo*-alcohol **7**,  $[\alpha]_D^{29} +67.2$  (*c* 1.4, CHCl<sub>3</sub>), which on the second Mitsunobu reaction furnished the *endo*-aryl ether **8** as the single stereoisomer. During these two-fold Mitsunobu inversions, dehydration did not take place at all, supporting the above-mentioned 1,4-repulsion explanation. On reductive cleavage of the bromo-ether functionality followed by silylation of the secondary hydroxy group, **8** yielded the *endo*-aryl ether **4c**,  $[\alpha]_D^{30} +10.0$  (*c* 1.4, CHCl<sub>3</sub>), bearing a *syn*-4-substituent (Scheme 3).

With the three substrates in hand, thermolysis was carried out in boiling diphenyl ether to initiate the concurrent retro-Diels-Alder reaction and Claisen rearrangement.<sup>11</sup> Thermolysis of **4a** was completed within 30 min to give the 2-(cyclohex-2-enyl)phenol **10a**,  $[\alpha]_D^{28} +102.1$  (*c* 0.2, CHCl<sub>3</sub>), as the single product in 58% yield, whose enantiomeric purity was determined to be 97% ee by hplc using a chiral column

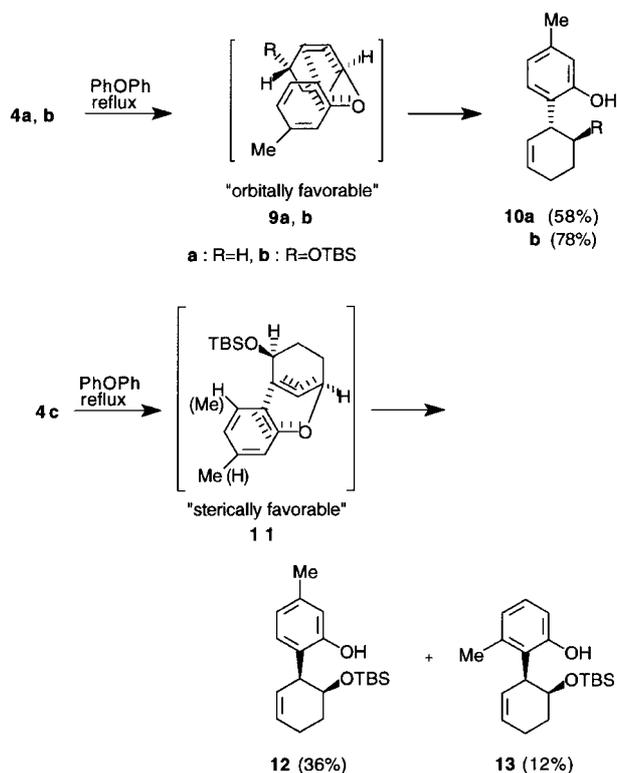


Reagents and conditions: i)  $\text{Bu}_4\text{NF}$ . ii) NBS. iii)  $\text{NaBH}_4$  (69%). iv)  $4\text{-NO}_2\text{C}_6\text{H}_4\text{CO}_2\text{H}$ , DIPAD, TPP then dil.  $\text{NaOH}$  (86%). v)  $3\text{-MeC}_6\text{H}_4\text{OH}$ , DIPAD, TPP. vi) (a)  $\text{Zn}$ ,  $\text{AcOH}$  (cat.),  $\text{EtOH}$ , reflux, (b)  $\text{TBS-Cl}$ , imidazole, (82% from 7)

Scheme 3

(CHIRALCEL OJ, 5%  $\text{Pr}^i\text{OH}$ -hexane) indicating about 1.5% loss of the original chiral integrity of the starting keto-ether **2**. The reaction was found to proceed most easily with **4b**, having no significant 1,4-interaction, to give 5-methyl-2-(6-siloxycyclohex-2-enyl)phenol **10b**,  $[\alpha]_{\text{D}}^{31} -24.6$  ( $c$  0.5,  $\text{CHCl}_3$ ), stereo- and regioselectively, in 78% yield as the single product within 30 min. In a shorter reaction time ( $\sim 5$  min), a separable mixture of **10b** and the aryl cyclohex-2-enyl ether **9b**,  $[\alpha]_{\text{D}}^{33} +89.1$  ( $c$  0.7,  $\text{CHCl}_3$ ), was obtained though the reaction could not be terminated definitely at the retro-Diels-Alder stage. Since none of the stereoisomers could be detected by hplc [after desilylation: Microsorb(Si), 1%  $\text{Pr}^i\text{OH}$ -hexane] in the reaction of **9b**, it was presumed that the original chiral integrity originated from the keto-ether **2** was preserved in the product **10b**. On the other hand, the thermolysis of the *endo*-aryl ether **4c** having a considerable 1,4-interaction took a longer reaction time to give a separable 6:1 mixture of 5-methyl-2-(6-siloxycyclohex-2-enyl)phenol **12** and 3-methyl-2-(6-siloxycyclohex-2-enyl)phenol **13** in 45% total yield after 2.5 h. Although the reaction proceeded stereoselectively as none of the stereoisomers of **12** and **13** were detected by hplc analysis [after desilylation: Microsorb(Si), 1%  $\text{Pr}^i\text{OH}$ -hexane], it lost regioselectivity to some extent. The observed differences in the regioselectivity among the allyl aryl ethers **4a**, **b** and **4c** were presumed to be due to the stereochemistry of the 4-substituent: **4a**, **b**, without having 1,4-interaction, were allowed to take the orbitally favorable chair-like transition states **9a**, **b** in the Claisen rearrangement to give the single products **10a**, **b**, respectively, while **4c**, having a considerable 1,4-interaction, was forced to take sterically favorable two boat-like transition states **11** to give rise to the mixture of the two regioisomers **12** and **13** (Scheme 4).

Having concluded the cyclohexenylphenol **10b** to be the only compound accessible practically from the keto-ether **2** via the *syn*-4-substituted cyclohex-2-enol equivalent **3b** through the Mitsunobu reaction and the concurrent retro-Diels-Alder and Claisen reaction, we explored its utilization as a chiral building block for the enantiocontrolled construction of a natural phenolic sesquiterpene (+)-curcudiol **19** isolated from the marine sponge *Didiscus flavus*.<sup>9</sup> Thus, the phenol **10b** was first transformed into the methyl ether **14**,  $[\alpha]_{\text{D}}^{27} +131.7$  ( $c$  0.3,  $\text{CHCl}_3$ ), in 72% yield, which was sequentially transformed into the monopivalate **15**,  $[\alpha]_{\text{D}}^{29} -18.4$  ( $c$  0.9,  $\text{CHCl}_3$ ), in 73% yield through one-flask ozonolysis and reduction followed by specific monoacylation. Reductive removal<sup>13</sup> of the primary hydroxy functionality of **15** via the tosylate afforded **16**,  $[\alpha]_{\text{D}}^{27} -12.3$  ( $c$  0.3,  $\text{CHCl}_3$ ), whose secondary oxygen functionality was removed via the



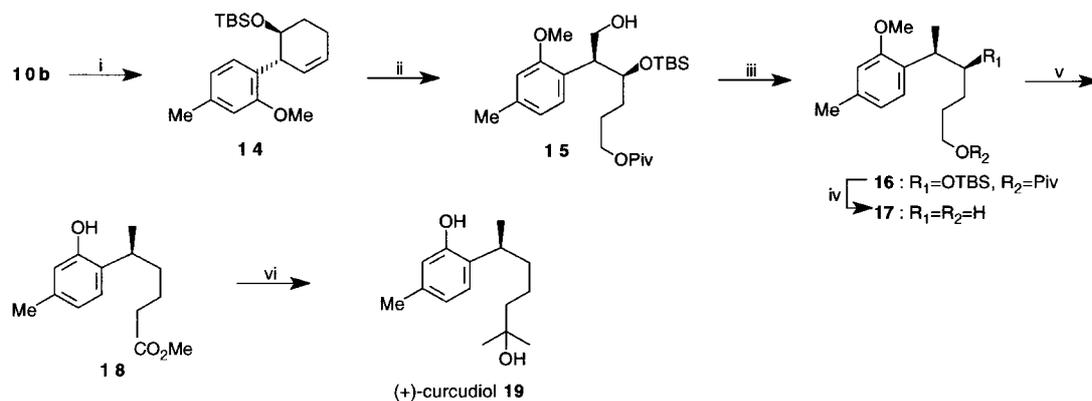
Scheme 4

xantate<sup>14</sup> to give the primary alcohol **17**,  $[\alpha]_{\text{D}}^{31} +5.6$  ( $c$  0.5,  $\text{CHCl}_3$ ), in 62% overall yield after reductive deacylation. Enantiomeric purity of **17** was determined to be  $>99\%$  ee by hplc using a chiral column (CHIRALCEL OJ, 3%  $\text{Pr}^i\text{OH}$ -hexane). By employing a sequence of reactions involving oxidation, ether cleavage and esterification, **17** was converted to the penultimate ester **18**,  $[\alpha]_{\text{D}}^{32} +20.7$  ( $c$  0.2,  $\text{CHCl}_3$ ), which was treated with excess methylmagnesium iodide to accomplish the first enantiocontrolled synthesis<sup>15</sup> of (+)-curcudiol **19**,  $[\alpha]_{\text{D}}^{30} +9.9$  ( $c$  0.2,  $\text{CHCl}_3$ ), [natural:  $[\alpha]_{\text{D}}^{22} +9.2$  ( $c$  10.8,  $\text{CHCl}_3$ )]. Overall yield of **19** from **17** was 44% (Scheme 5).

In summary, it was concluded that only the chiral cyclohex-2-enol equivalent **3b** bearing a *syn*-4-substituent was tolerable under the Mitsunobu reaction and the concurrent retro-Diels-Alder and Claisen rearrangement reaction protocol to give the 2-(cyclohex-2-enyl)phenol **10b** satisfactorily without losing the original chiral integrity<sup>16</sup> of the chiral starting material **2**.

## References and Notes

- (1) Takano, S.; Akiyama, M.; Ogasawara, K. *J. Chem. Soc., Perkin Trans. 1* **1985**, 2447.
- (2) Goering, H. L.; Kimoto, W. I. *J. Am. Chem. Soc.*, **1965**, *87*, 1748.
- (3) Pertinent reviews, see: Mitsunobu, O. *Synthesis* **1981**, 1. Hughes, D. L. *Org. React.* **1992**, *42*, pp. 335-656. Hughes, D. L. *Org. Prep. Proc. Int.* **1996**, *28*, 127.
- (4) A discussion on the Mitsunobu reaction of cyclic allylic alcohols, see: Shull, B. K.; Sakai, T.; Nichols, J. B.; Koreeda, M. *J. Org. Chem.* **1997**, *62*, 8294.
- (5) Pertinent reviews, see: Rhoads, S. J.; Raulins, N. R. *Org. React.* **1975**, *22*, 1. Wipf, P. In "Comprehensive Organic Synthesis" Trost, B. M.; Fleming, I.; Paquette, L. A. Eds. Pergamon Press, Oxford, **1991**, *5*, pp. 827-874.



*Reagents and conditions:* i) MeI, NaH, THF (72%). ii) (a) O<sub>3</sub>, then NaBH<sub>4</sub>, MeOH-CH<sub>2</sub>Cl<sub>2</sub>, (b) Piv-Cl, pyridine (73%). iii) (a) TsCl, pyridine, (b) NaBH<sub>4</sub>, DMSO (93%). iv) (a) HF, MeCN, (b) CS<sub>2</sub>, MeI, NaH, (c) Bu<sub>3</sub>SnH, AIBN (cat.), (d) LiAlH<sub>4</sub> (62%). v) (a) PDC, DMF, (b) BBr<sub>3</sub>, then CH<sub>2</sub>N<sub>2</sub> (59%). vi) MeMgI, THF (75%).

### Scheme 5

- (6) Quite recently, a catalytic asymmetric protocol for the formation of allyl aryl ethers and their Claisen rearrangements has been disclosed, see: Trost, B. M.; Toste, F. D. *J. Am. Chem. Soc.* **1998**, *120*, 815.
- (7) Sakagami, H.; Samizu, K.; Kamikubo, T.; Ogasawara, K. *Synlett* **1996**, 163.
- (8) Hiroya, K.; Kurihara, Y.; Ogasawara, K. *Angew. Chem., Int. Ed. Engl.* **1995**, *34*, 2287.
- (9) Wright, A. E.; Pomponi, S. A.; McConnell, O. J.; Kohmoto, S.; McCarthy, P. J. *J. Nat. Prod.* **1987**, *50*, 976.
- (10) Martin, S. F.; Dodge, J. A. *Tetrahedron Lett.* **1991**, *32*, 3017.
- (11) Sugahara, T.; Ogasawara, K. *Synlett* **1996**, 319.
- (12) Fleming, I. In *Frontier Orbitals and Organic Chemical Reactions*, J. Wiley & Sons, London, **1976**.
- (13) Guanti, G.; Narisano, E.; Podgorski, T.; Thea, S.; Williams, A. *Tetrahedron* **1990**, *46*, 7081.
- (14) Barton, D. H. R.; McCombie, S. W. *J. Chem. Soc., Perkin Trans 1*, **1975**, 1574. Barton, D. H. R.; Crich, D.; Löbbberding, A.; Zard, S. Z. *Tetrahedron* **1986**, *42*, 2329.
- (15) Racemic synthesis of curcudiol, see: Ono, M.; Yamamoto, Y.; Todoroki, R.; Akita, H. *Heterocycles* **1994**, *37*, 181.
- (16) The corresponding cyclopent-2-enyl ether lost some of the original chiral integrity under the same concurrent retro-Diels-Alder and Claisen conditions, see: Sugahara, T.; Ogasawara, K. *Tetrahedron: Asymmetry* in press.