

# Dynamic Protection of Amines using 18-Crown-6

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**Summary** The regioselectivity of diamine monoacylation has been controlled by selective complexation with 18-crown-6 and a proton source

RECENTLY we reported a convenient method for the selective acylation of secondary amines in the presence of primary amines<sup>1</sup> 18-Crown-6 forms complexes with alkylammonium salts *via* three hydrogen bonds and pole-dipole interactions in the 2.7 Å cavity<sup>2</sup> We expected that rapid selective complexation of one (or more) ammonium function(s) in a polyamine substrate should permit selective functionalisation of uncomplexed sites Since dialkylammonium salts form less stable complexes owing to a reduction in hydrogen bonding, selective acylation of a secondary amino function in the presence of a primary is possible

selective monobenzoylation of ethylenediamine and homologues was also improved in the presence of 18-crown-6

The decrease in stability of crown-primary alkylammonium salt complexes with increasing steric congestion<sup>3</sup> should permit the selective acylation of a hindered primary amine in the presence of a non-hindered function Such selection is relevant to aminoglycoside chemistry As model systems, competition in the acylation and toluene-4-sulphonylation of mixtures of benzylamine and benzhydrylamine or 3 $\alpha$ -(axial) and 3 $\beta$ -(equatorial) amino-5 $\alpha$ -cholestanes<sup>4</sup> were studied (Table 2) Without crown ether the less hindered (benzyl- or 3 $\beta$ - respectively) amine was principally functionalised In the presence of 18-crown-6 the ratio of hindered non hindered amides was increased Consistent with sterically selective complexation<sup>3</sup> dicyclohexyl 18-crown-6 (entries 10, 11) was superior to 18-crown-6 In the

TABLE 1 Selective acylation of diamines RNH[CH<sub>2</sub>]<sub>n</sub>NH<sub>2</sub><sup>a</sup>

Entry	Equiv of 18-crown-6	RN(Ts)[CH <sub>2</sub> ] <sub>n</sub> -NHTs	Percentage yields of products RN(COAr)[CH <sub>2</sub> ] <sub>n</sub> -NHTs	RN(Ts)[CH <sub>2</sub> ] <sub>n</sub> -NHCOAr	RN(COAr)[CH <sub>2</sub> ] <sub>n</sub> -NHCOAr
1	0 <sup>b</sup>	30	16	0.2	37
2	1	12	63	traces	4
3	2	traces	79	traces	15
4	0 <sup>b</sup>	40	1	4	42
5	1	12	51	1	11
6	2	6	61	0.8	8
7	2	0	69	0	22
8	0 <sup>b</sup>	43	8	—	27
9	1	20	56	—	17
10	2	4	76	—	3
11	0 <sup>b</sup>	40	6	—	34
12	1	11	41	—	26
13	2	5	79	—	10
14	0 <sup>b</sup>	42	1	—	47
15	1	18	49	—	23
16	2	5	64	—	24
17	0 <sup>b</sup>	37	0	—	45
18	1 <sup>b</sup>	15	31	—	29
19	2 <sup>b</sup>	9	32	—	32

<sup>a</sup> R = Me (entries 1—7) and H (8—19) Ar = Ph (1—3, 7—19) and C<sub>6</sub>H<sub>4</sub>-4-NO<sub>2</sub> (4—6), *n* = 2 (1—6, 8—10), 3 (7, 11—13), 4 (14—16) and 8 (17—19) Typically benzoyl chloride and triethylamine were added in sequence to *N*-methylethylenediammonium di(toluen-4-sulphonate) and 18-crown 6 (1 mmol each) in dichloromethane (10 ml) When reactions were complete (t l c) toluene-4-sulphonyl chloride (1 mmol), triethylamine (4 mmol), and an excess of potassium chloride were added Yields refer to pure compounds isolated by direct chromatography on Merck Kieselgel H <sup>b</sup> Heterogeneous reactions It must be assumed that the high yield of RN-(COAr)[CH<sub>2</sub>]<sub>n</sub>NHCOAr in the blank reactions followed in part from the low solubility of the RNH<sub>2</sub><sup>+</sup>[CH<sub>2</sub>]<sub>n</sub>NH<sub>3</sub><sup>+</sup>2TsO<sup>-</sup> salts However, the increase in the yield of RN(COAr)[CH<sub>2</sub>]<sub>n</sub>NH 1s with increase in crown ether from 1 to 2 equiv is consistent only with selective complexation

Herein we report dramatic improvements in diamine monoacylation using dynamic protection (Table 1) For example the reaction of *N*-methylethylenediamine with benzoyl and toluene-4-sulphonyl chlorides in sequence† gave *N*-benzoyl-*N*-methyl-*N'*-toluene-4-sulphonylethylenediamine (16%) In the presence of 18-crown-6 the yield was increased to 79% Surprisingly (entries 8—19) the

steroid examples exclusive axial substitution was observed in the presence of *N*-benzylmono-aza-18-crown-6<sup>5</sup>

The advantage of the aza-crown was emphasised by competition experiments between benzylamine and *N*-benzyl-iso-propylamine Since the rate of tosylation of the latter was slow, a 62% yield of *N*-benzyl-*N*-isopropyl-toluene-4-sulphonamide was only obtained when the

† The diamine was used as its di-toluene-4-sulphonate and, to facilitate chromatographic separation, after acylation remaining amino functions were toluene-4-sulphonylated

TABLE 2. Selective acylation and sulphonylation of amines<sup>a</sup>

Entry	Equiv of crown ether	Amine	Ammonium salt	% Amides <sup>b</sup>	Hindered amide <sup>b</sup> mol fraction
1	0 <sup>c</sup>	PhCH <sub>2</sub> NH <sub>2</sub>	Ph <sub>2</sub> CHNH <sub>3</sub> <sup>+</sup> TsO <sup>-</sup>	96	0.61
2	1	"	"	85	0.78
3	2	"	"	79	1.00
4	0 <sup>c</sup>	"	"	93	0.31
5	1	"	"	92	0.47
6	2	"	"	95	0.52
7	0 <sup>c</sup>	"	"	95	0.04
8	1	"	"	97	0.30
9	2	"	"	91	0.44
10	1	"	"	98	0.59
11	2	"	"	98	0.71
12	0 <sup>c</sup>	PhCH <sub>2</sub> NH <sub>2</sub>	PhCH <sub>2</sub> Pr <sup>+</sup> NH <sub>2</sub> +TsO <sup>-</sup>	100	≤ 0.02
13	1	"	"	98	0.31
14	2	"	"	99	0.46
15	2	"	"	96	0.60
16	1	"	"	97	0.55
17	2	"	"	97	0.62
18	2	"	"	95	0.65
19	0	3β- : 3α-Amino-5α-cholestanes	CF <sub>3</sub> CO <sub>2</sub> H 1 : 1 : 1	83	0.12
20	1	"	"	85	0.40
21	2	"	"	81	0.59
22	0	"	"	96	0.26
23	1	"	"	88	0.47
24	2	"	"	88	0.70
25	1	"	"	84	1.0
26	2	"	"	87	1.0

<sup>a</sup> Reactions were carried out using 18-crown-6 (entries 2, 3, 5, 6, 8, 9, 13—15, 20, 21, 23, and 24), dicyclohexyl-18-crown-6 (Fluka AG) (10, 11), and *N*-benzylmono-aza-18-crown-6 (16—18, 25, 26) with (CF<sub>3</sub>CO)<sub>2</sub>O (1—3), PhCOCl (4—6), TsCl (7—18, 22—26), or Ac<sub>2</sub>O (19—21) as electrophile. <sup>b</sup> The ratios of amides were determined by n.m.r. spectroscopy ( $\pm 0.02$ ) (entries 12—18); all other ratios refer to pure isolated compounds. Typically toluene-4-sulphonyl chloride and then, over 5 min, triethylamine (1 mmol each) were added to a solution prepared from 18-crown-6, benzylamine, and benzhydrylammonium toluene-4-sulphonate (1 mmol each) in dichloromethane (10 ml) [or (entries 19—26) from 3α- and 3β-amino-5α-cholestanes and CF<sub>3</sub>CO<sub>2</sub>H (1:1:1)]. Chromatography on Merck Kieselgel H gave *N*-benzyl (0.68 mmol) and *N*-benzhydryl- (0.29 mmol) toluene-4-sulphonamides. In entries 15 and 18 the triethylamine was added over 1 week. <sup>c</sup> Heterogeneous reactions.

triethylamine was added slowly (1 week rather than 5 min) after the toluene-4-sulphonyl chloride (Table 2, entry 18).

Clearly dynamic protection provides a more convenient simple alternative to classical protection group methodologies.

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