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(N,N-dialkylamino)-pyrimidines are usually obtained by displacement of chlorine from chloropyrimidines by amines2. The required chloropyrimidines are usually prepared in poor yields by treating the oxopyrimidines with phosphorus halides, using N,N-dimethylaniline as catalyst. In the latter transformation, experimental difficulties have been reported, when the pKb of the chloropyrimidines are close to that of N,N-dimethylaniline². Recently, direct formation of aminopyrimidines has been reported by heating the corresponding oxopyrimidines with phosphorus triamide³ or phosphorus pentoxide/amine mixture⁴. A method for direct synthesis of the N-substituted aminopyrimidines is desirable. We now report a novel procedure for 4-Nmono- and -disubstituted pyrimidines by reaction of guanidine with ketene S, N-acetals, generated in situ from the ketene S,S-acetals or from ketones and aryl or alkyl isothiocyanates.

When ketene S,S-acetal 1 was reacted with aniline (2a) in boiling ethanol, the corresponding S,N-acetal (3a) was obtained in nearly quantitative yield (characterized by spectral and analytical data). Treatment of 3a (generated in situ) with guanidine nitrate in the presence of sodium ethoxide followed by refluxing in ethanol, yielded 2-amino-4-anilino-5-cyano-1,6-dihydro-6-oxopyrimidine (4a) in 57% yield. The pyrimidines 4b-f (Table 1) were similarly prepared in 47-57% overall yields (Scheme A).

Scheme A

The reaction was next extended to ketene S,S-acetals 5 derived from ketones, which are generally less reactive than 1. Thus the reaction of 5a and aniline in refluxing ethanol, always yielded a mixture of unreacted 5a, S,N-acetal 6a, and N,N-acetal 7a even after prolonged heating. However, when 5a was heated at 160 °C for 15 h, with a slight excess of aniline (1.2 equiv), 5a completely disappeared, though the mixture of 6a and 7a was always formed (4:1 ratio) (Scheme B).

$$A_{r} \xrightarrow{S-CH_{3}} \xrightarrow{RNH_{2}} A_{r} \xrightarrow{S-CH_{3}} A_{r} \xrightarrow{NHR} A_{r} \xrightarrow{$$

Scheme B

When this mixture was reacted with guanidine nitrate in the presence of sodium ethoxide, the pyrimidine 8a was formed in 18% yield and the corresponding N,N-acetal 7a remained unreacted. Attempts to prepare S,N-acetals ex-

A Novel and Convenient Synthesis of 2-Amino-4-(*N*-alkyl-*N*-arylamino)-pyrimidines using Polarized Ketene *S*, *S*- and *S*, *N*-Acetals¹

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While 2,4-diaminopyrimidines have been synthesized by reacting guanidine with cyano-substituted 3-carbon fragments, the corresponding 4-(N-alkyl-N-arylamino)- and 4-

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clusively from 5a and amines failed, and the N,N-acetal 7a was always formed as impurity, which could only be separated by column chromatography. Thus, the direct preparation of ketene S,N-acetals 6a-e from ketones was developed by extending the earlier reported method^{5.6} (Method B) (Scheme C). The ketene S,N-acetals 6a-e were prepared in 75-85% yield (Table 2).

Ar
$$\leftarrow$$
 CH₃ + R-N=C=S $\xrightarrow{NaH/DMF/CH_3 J \text{ (1 equiv)}}$ Ar \xrightarrow{O} S-CH₃
Scheme C \rightarrow 6 a-e

The reaction of pure S,N-acetal 6a with guanidine under identical conditions yielded pyrimidine 8a in 35% yield. The pyrimidines 8b-e were similarly prepared in 34-50% yield (Scheme D and Table 3). Reaction of S,N-acetal 6f, reported earlier⁷, gave the aminopyrimidine 8f in 30% yield.

Scheme D

Reaction of ketene S,S-acetal 5a with morpholine yielded N,N-acetal 9a exclusively under varying conditions and attempts to prepare S,N-acetal 10 were unsuccessful. However, when 9a (generated either in situ, or after isolation) was reacted with guanidine nitrate and sodium ethoxide in refluxing ethanol, 4-morpholinopyrimidine 8g was obtained in 28% yield. Pyrimidines 8h-i were similarly prepared in 30% yields (Scheme E and Table 3).

Scheme E

2-Amino-4-arylamino (or -morpholino)-5-cyano-6-oxo-1,6-dihydro-pyrimidines 4; General Procedure:

A solution of 1 (4.34 g, 0.01 mol) and amine (0.01 mol) in absolute ethanol (35 ml) is heated under reflux for 4 h (checked by T.L.C.), the mixture is then added to a solution of guanidine nitrate (0.01 mol) and sodium ethoxide [prepared by dissolving sodium (0.02 mol) in refluxing ethanol (35 ml)] in absolute ethanol (35 ml) and the combined reaction mixture heated under reflux for 12 h. The ethanol is removed under reduced pressure, the residue diluted with ice-cold water (40 ml), and acidified with 4 normal hydrochloric acid (15 ml). The crude, white oxopyrimidines 4 are filtered and purified by crystallization from acetic acid (Table 1).

1-Aryl-3-arylamino-3-methylthio-1-oxo-2-propenes 6:

Method A: A mixture of ketene S,S-acetal 5 (0.02 mol) and amine (0.022 mol) is heated at 160 °C for 15 h in an oil bath (monitored by T.L.C.). The mixture of S,N-(6) and N,N-(7) acetals is separated by column chromatography on silica gel using benzene/hexane (1:1) as eluent; S,N-acetals 6a-d were prepared by this procedure (Table 2).

Table 1. 2-Amino-4-arylamino-(or -morpholino)-5-cyano-6-oxo-1,6-dihydropyrimidines 4a-f

Product Yield			m.p.	Molecular	I.R. (KBr)	¹ H-N.M.R. (CF ₃ COOH)		
No.	R¹	R ²	[%]	[°C]	formula ^a	ν [cm 1]	δ [ppm]	
4a	C ₆ H ₅	Н	57	340°	C ₁₁ H ₉ N ₅ O (227.2)	3420, 3270, 3155, 2200, 1660, 1635, 1585	6.8-7.3 (m, 5 H _{arom}); 8.6 (br. s, 1 H, NH)	
4b	4-H ₃ C—C ₆ H ₄	Н	51	360° (dec.)	$C_{12}H_{11}N_5O$ (241.3)	3280, 3220, 3140, 2215, 1650, 1600, 1540	insoluble	
4c	4-H ₃ CO—C ₆ H ₄	Н	54	326-328°	$C_{12}H_{11}N_5O_2$ (257.3)	3425, 3265, 3150, 2215, 1662, 1600, 1560	3.51 (s, 3 H, OCH ₃); 6.80 (A ₂ B ₂ q, 4 H _{arom}); 8.56 (br. s, 1 H, NH)	
4d	4-Cl—C ₆ H ₄	Н	57	364–366°	C ₁₁ H ₈ ClN ₅ O (261.7)	3480, 3310, 3280, 3200, 3130, 2200, 1668, 1640, 1600	insoluble	
4e	$4-F-C_6H_4$	Н	53	368-370°	$C_{11}H_8FN_5O$ (245.2)	3475, 3292, 3200, 3123, 2204, 1670, 1624, 1600	6.6-7.1 (m, 4H _{arom}); 8.43 (br. s, 1H, NH)	
4f	(CH ₂) ₂ —O—(Cl	$H_2)_2$	47	327° (dec)	$C_9H_{11}N_5O_2$ (221.2)	3280, 3175, 3100, 2180, 1670, 1655, 1580	3.66 (s, 8 H, morpholine CH ₂)	

The microanalyses were in satisfactory agreement with the calculated values (C ±0.38, H ±0.33, N ±0.39); exceptions: 4b H, −0.47, N, −0.44; 4c, N +0.57; 3e, 3f, C −0.49.

Table 2. 1-Aryl-3-aryl-(or alkyl)-amino-3-methylthio-1-oxo-2-propenes 6 and 1-Aryl-3,3-bis[arylamino]-1-oxo-2-propenes 7

Product			Yield [%]		m.p.	Molecular	•		¹ H-N,M.R. (CDCl ₃)
No.	Ar	R	by Mo	ethod B	[°C]	formula ^a	$\nu_{ m NH}$	$ u_{\mathbf{C}=\mathbf{c}_{\mathbf{O}}}$	δ [ppm]
6a	Н	C₀H₅	40	82	56-57°	C ₁₆ H ₁₅ NOS (269.3)	3230	1575	2.50 (s, 3 H, SCH ₃); 6.11 (s, 1 H, H-2); 7.4-7.8 (m, 8 H _{arom}); 8.0-8.3 (m, 2 H _{arom})
6b	4-H ₃ CO—C ₆ H ₄	C ₆ H ₅	32	74	66:-67°	C ₁₇ H ₁₇ NO ₂ S (299.3)	3230	1595	OCH ₃); 6.23 (s, 1H, H-2); 7.2–7.7 (m, 3H _{4rom}); 7.8 (m, 4H _{arom}); 8.33 (dd, 2H _{arom})
6с	4-Br-C ₆ H ₄	C ₆ H ₅	41	84	85–86°	C ₁₆ H ₁₄ BrNOS (348.3)	3330	1570	2.48 (s, 3 H, SCH ₃); 6.01 (s, 1 H H-2); 7.3-7.7 (m, 7 H _{arom}); 7.8-8.2 (m, 2 H _{arom})
6d	C ₆ H ₅	4-Cl—C ₆ H ₄	31	76	116117°	C ₁₆ H ₁₄ ClNOS (303.8)	3400	1608	2.51 (s, 3 H, SCH ₃); 6.18 (s, 1 H H-2); 7.6–7.9 (m, 7 H _{arom}); 8.2–8.4 (m, 2 H _{arom})
6e	C_6H_5	C ₂ H ₅	A. Area.	70	oil	C ₁₂ H ₁₅ NOS (221.3)	3350	1540	1.33 (t, 3H, CH ₂ CH ₃); 2.48 (s 3H, SCH ₃); 3.48 (q, 2H CH ₂ CH ₃); 5.66 (s, 1H, H-2); 7.3- 7.6 (m, 3H _{arom}); 7.7-8.0 (m, 2 H _{arom}); 11.7-12.1 (br, ½H, NH)
7a	C ₆ H ₅	I_5 C_6H_5			132-133	C ₂₁ H ₁₈ N ₂ O (314.4)	3335, 3 1608	258, 3200,	5.65 (s, 1 H, H-2); 6.5 (br, 1 H, NH); 6.7 (br, 1 H, NH); 7.0-7.6 (m, 13 H _{arom}); 7.6-7.9 (m, 2 H _{arom})
7b	4-H ₃ CO—C ₆ H ₄	C ₆ H ₅	12	# 1000a.cd	126-127°	$C_{22}H_{20}N_2O_2$ (344.4)	3385, 3 1590	190, 3100,	
7c	4-BrC ₆ H ₄	C ₆ H ₅	12	representation of the contract	109–110°	C ₂₁ H ₁₇ BrN ₂ O (393.3)	3270, 3 1618	210, 3100,	5.56 (s, 1 H, H-2); 6.5 (br. s, 1 H NH); 6.6–7.0 (br, 1 H, NH); 7.0– 7.5 (m, 10 H _{arom}); 7.5–8.1 (m 4 H _{arom})
7d	C ₆ H ₅	4-Cl—C ₆ H ₄	11		194°	C ₂₁ H ₁₆ Cl ₂ N ₂ O (383.3)	3300, 3 1590	250, 3175,	5.66 (s, 1 H, H-2); 6.5 (br. s, 1 H NH); 7.2-7.7 (m, 11 H _{arom}); 7.8- 8.0 (m, 2 H _{arom}); 13.7 (br, 1 H NH)

^a The microanalyses were in satisfactory agreement with the calculated values (C \pm 0.26, H \pm 0.41, N \pm 0.37); exception 6c, C \pm 0.53.

Table 3. 2-Amino-6-aryl-4-aryl-(or -alkyl)-amino- (8a-f) or -4-morpholinopyrimidines (8g-i)

Produ No.	act Ar	R	Yield [%]	m.p. [°C]	Molecular formula*	1.R. (KBr) ν [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃) δ [ppm]
8a	C ₆ H ₅	C ₆ H ₅	35 (18) ^b	159° °	C ₁₆ H ₁₄ N ₄ (262.3)	3400, 3270, 3130, 1620, 1590	5.22 (s, 2H, NH ₂); 6.61 (s, 1H, H-5); 7.0–7.7 (m, 8 H _{470m}); 7.7–8.0 (m, 2H _{470m})
8b	4-H ₃ CO—C ₆ H ₄	C ₆ H ₅	34	167°	C ₁₇ H ₁₆ N ₄ O (292.3)	3450, 3380, 3170, 1628, 1600	3.71 (s, 3 H, OCH ₃); 5.1 (br. s, 2 H, NH ₂); 6.35 (s, 1 H, H-5); 6.7-7.3 (m, 7 H _{arom}); 7.76 (dd,
8c	4-BrC ₆ H ₅	C_6H_5	45	147-148°	C ₁₆ H ₁₃ BrN ₄ (341.2)	3370, 3320, 3280, 3120, 1618, 1590	2H _{arom}) 5.0 (br. s, 2H, NH ₂); 6.38 (s, 1H, H-5); 7.2–7.8 (m, 9 H _{arom})
8d ^d	C_6H_5	4-ClC ₆ H ₄	50	138°	C ₁₆ H ₁₃ ClN ₄ (296.8)	3400, 3220, 3150, 1638, 1590	5.21 (br. s, 2H, NH ₂); 6.26 (s, 1H, H-5); 7.1–7.4 (m, 7 H _{arom}); 7.8–7.9 (m, 2 H _{arom})
8e	C ₆ H ₅	C ₂ H ₅	40	148-149°	C ₁₂ H ₁₄ N ₄ (214.3)	3320, 3280, 1645, 1590	1.30 (t, 3 H, CH ₂ CH ₃); 3.55 (q, 2H, CH ₂ CH ₃); 5.2 (br, 2H, NH ₂); 6.50 (s, 1H, H-5); 7.7–8.0 (m, 3 H _{arom}); 8.2–8.5 (m, 2 H _{arom})

Table 3. (Continued)

Product No. Ar R		R	Yield [%]	m.p. [°C]	Molecular formula ^a	I.R. (KBr) ν [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃) δ [ppm]
8f	4-ClC ₆ H ₄	CH ₂ CH(OC ₂ H ₅) ₂	30	118-119°	C ₁₆ H ₂₁ ClN ₄ O ₂ (336.8)	3300, 3180, 3090, 1640, 1610	1.18 (t, 6H, OCH ₂ CH ₃); 3.4–3.9 (m, 6H, NCH ₂ + OCH ₂ CH ₃); 4.63 (t, 1H, —CH=:); 5.0 (br, 2H, NH ₂); 5.2 (br. t, 1H, NH); 6.11 (s, 1H, H-5); 7.36 (dd, 2H _{arom}); 7.83 (dd, 2H _{arom})
8g	C ₆ H ₅		28	170°	C ₁₄ H ₁₆ N ₄ O (256.3)	3470, 3275, 3135, 1620, 1575	3.5–3.9 (m, 8 H, morpholine CH ₂); 4.9 (br. s, 2 H, NH ₂); 6.26 (s, 1 H, H-5); 7.3–7.5 (m,
8h	4-H ₃ CO—C ₆ H ₄	American	31	214-215°	C ₁₅ H ₁₈ N ₄ O ₂ (286.3)	3328, 3170, 1630, 1580	3 H _{arom}); 7.8-8.0 (m, 2 H _{arom}) 3.5-3.8 (m, 8 H, morpholine CH ₂); 3.81 (s, 3 H, H ₃ CO); 5.0 (br. s, 2 H, NH ₂); 6.20 (s, 1 H, H-5); 6.90 (dd, 2 H _{arom}); 7.83
8i	4-Cl—C ₆ H ₄	***************************************	30	184–185°	C ₁₄ H ₁₅ ClN ₄ O (290.8)	3405, 3270, 3135, 1622, 1580	(dd, 2 H _{arom}) 3.63 (s, 8 H, morpholine CH ₂); 6.11 (s, 1 H, H-5); 7.10 (s, 4 H _{arom}) ^e

The microanalyses were in satisfactory agreement with the calculated values (C ± 0.44 , H ± 0.42 , N ± 0.39); exceptions: **8b**, C -0.46; **8h**, C -0.49.

Method B: The S,N-acetals 6a-e are prepared from the corresponding ketones, aryl or alkyl isothiocyanates, and sodium hydride according to Ref.^{5,6} (Scheme C, Table 2).

2-Amino-4-aryl-(or -alkyl)-amino-6-arylpyrimidines 8a-f; General Procedure:

A solution of ketene S,N-acetal 6a-f (0.01 mol) in absolute ethanol (10 ml) is added to a suspension of guanidine nitrate (1.22 g, 0.01 mol) and sodium ethoxide (from 0.01 mol sodium) in absolute ethanol (50 ml). The reaction mixture is then heated under reflux for 24 h and after removal of solvent under reduced pressure (water bath), the residue is diluted with ice-cold water (50 ml) and extracted with chloroform (2 × 100 ml). The chloroform layer is dried with sodium sulfate, evaporated, and the crude pyrimidines 8a-e purified by crystallization (dichloromethane/hexane) (Table 3). The pyrimidine 8f is obtained by column chromatography on silica gel with benzene/ethyl acetate (7:3) as eluent of the residue obtained after evaporation of chloroform.

2-Amino-6-aryl-4-morpholinopyrimidines 8g-i; General Procedure: N,N-Morpholinoacetals 9a-c are generated in situ by refluxing a solution of ketene S,S-acetal 5 (0.01 mol) and morpholine (0.022 mol) for 4 h in absolute ethanol (35 ml). The reaction mixture is then added to a suspension of guanidine nitrate (0.01 mol) and sodium ethoxide (from 0.01 mol sodium) in absolute ethanol (35 ml). The reaction mixture, after being refluxed for 24 h, is worked up as for 8a-f. The crude pyrimidines are crystallized from ethanol (Table 3).

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b From mixture of 6a and 6b.

[°] M.S.: m/e = 262 (M⁺); Lit.⁸ m.p. 305 °C.

^d M.S.: m/e = 296 (M⁺, ³⁵Cl); Lit. ⁸ m.p. 304 °C.

^c In CF₃COOH solution.

Part 13, For part 12 see A. Kumar, H. Ila, and H. Junjappa, J. Chem. Res. (S) 1979, 268.

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