

Syntheses of 3(5)-Substituted-4-(*N*-methylanilino)-5(3)-aminopyrazoles by Reaction of β -Hydroxy- α -cyanoenamines with Hydrazines

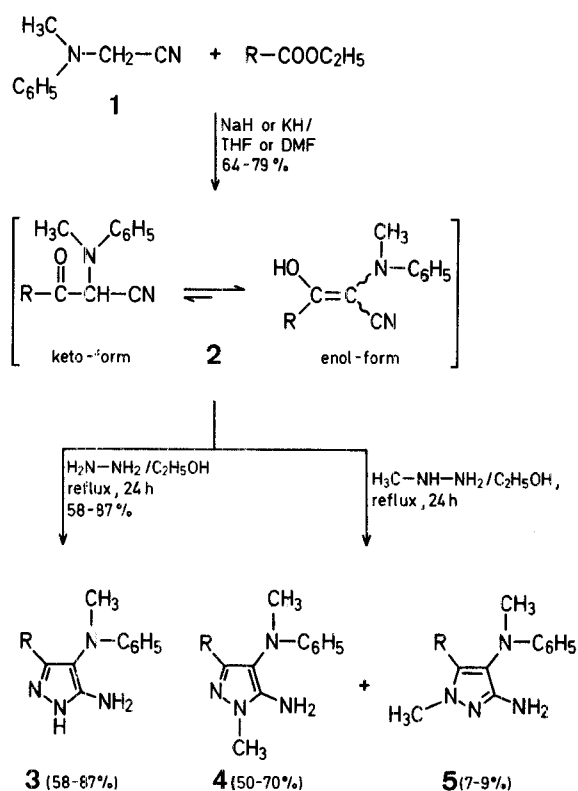
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

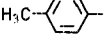
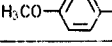
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Aminopyrazoles **3–5** are prepared by the reaction of β -hydroxy- α -cyanoenamines **2** with hydrazine hydrate and monomethylhydrazine.

In our previous paper¹, the reaction of α -(*N*-methylanilino)-acetonitrile (**1**) with esters has been reported to give new β -hydroxy- α -cyanoenamines **2** in good yields. We now report on the use of **2** for the synthesis of some new aminopyrazoles **3–5**.

Syntheses of pyrazoles by the reaction of β -ketonitriles with hydrazines have been reported^{2,3,4}. There is, however, no information on the synthesis of pyrazoles using β -hydroxy- α -cyanoenamines **2**. We describe here an efficient sequence for the syntheses of new pyrazoles. The method reported here is based on the reaction of β -hydroxy- α -cyanoenamines **2** with hydrazine hydrate and monomethylhydrazine.



2-5	R	2-5	R
a		d	
b		e	H
c		f	C ₂ H ₅

β -Hydroxy- α -cyanoenamines **2**¹ exist as a mixture of keto- and enol-forms. Refluxing a solution of **2** in ethanol with hydrazine hydrate or monomethylhydrazine gave **3**, or **4** and **5**, respectively. For example, when a mixture of **2a** and hydrazine hydrate was refluxed, 5(3)-amino-4-(*N*-

Table. Pyrazoles 3, 4 and 5 prepared

Prod- uct	Yield [%]	m. p. [°C]	Molecular Formula ^a	M.S. (<i>m/e</i> (M^+)) (rel. intensity %)	I.R. (KBr) ν_{NH} [cm^{-1}]	¹ H-N.M.R. (CDCl_3/TMS) δ [ppm]
3a	74	94–96°	C ₁₆ H ₁₆ N ₄ (264.3)	264 (100)	3500–3100	3.12 (s, 3H, NCH ₃); 6.30 (br., 2H, NH ₂); 6.5–7.6 (m, 11H, NH + H _{arom})
3b	82	123.5–125°	C ₁₇ H ₁₈ N ₄ (278.4)	278 (100)	3500 3100	2.29 (s, 3H, CH ₃); 3.13 (s, 3H, NCH ₃); 5.86 (br., 2H, NH ₂); 6.5–7.6 (m, 10H, NH + H _{arom})
3c	58	77–79°	C ₁₇ H ₁₈ N ₄ O (294.4)	294 (100)	3450 3100	3.11 (s, 3H, NCH ₃); 3.73 (s, 3H, OCH ₃); 6.06 (br., 2H, NH ₂); 6.5–7.7 (m, 10H, NH + H _{arom})
3d	79	92–94°	C ₁₆ H ₁₅ ClN ₄ (298.8)	300 ($M^+ + 1$, 41); 298 ($M^+ - 1$, 100)	3450 3100	3.12 (s, 3H, NCH ₃); 6.31 (br., 2H, NH ₂); 6.5–7.7 (m, 10H, NH + H _{arom})
3f	87	oil	C ₁₂ H ₁₆ N ₄ (216.3)	216 (100)	3450–3100 ^b	1.09 (t, 3H, $J = 7.2$ Hz, CH ₂ CH ₃); 2.41 (q, 2H, $J = 7.2$ Hz, CH ₂ CH ₃); 3.16 (s, 3H, NCH ₃); 6.19 (br., 3H, NH + NH ₂); 6.4–7.4 (m, 5H _{arom})
4a	59	144–146°	C ₁₇ H ₁₈ N ₄ (278.4)	278 (100)	3425, 3180	3.13 (s, 3H, NCH ₃); 3.31 (br., 2H, NH ₂); 3.73 (s, 3H, 1-NCH ₃); 6.5–7.8 (m, 10H _{arom})
4d	50	147.5–148.5°	C ₁₇ H ₁₇ ClN ₄ (312.9)	314 ($M^+ + 1$, 39); 312 ($M^+ - 1$, 100)	3350, 3190	3.11 (s, 3H, NCH ₃); 3.35 (br., 2H, NH ₂); 3.69 (s, 3H, 1-NCH ₃); 6.5–7.8 (m, 9H _{arom})
4f	70	90–91.5°	C ₁₃ H ₁₈ N ₄ (230.3)	230 (100)	3380, 3200	1.08 (t, 3H, $J = 7.6$ Hz, CH ₂ CH ₃); 2.36 (q, 2H, $J = 7.6$ Hz, CH ₂ CH ₃); 3.17 (s, 3H, NCH ₃); 3.41 (br., 2H, NH ₂); 3.60 (s, 3H, 1-NCH ₃); 6.4–7.4 (m, 5H _{arom})
5a	9	151–152°	C ₁₇ H ₁₈ N ₄ (278.4)	278 (100)	3425, 3280, 3190	2.98 (s, 3H, NCH ₃); 3.42 (br., 2H, NH ₂); 3.69 (s, 3H, 1-NCH ₃); 6.5–7.5 (m, 10H _{arom})
5d	9	175–177°	C ₁₇ H ₁₇ ClN ₄ (312.9)	314 ($M^+ + 1$, 39); 312 ($M^+ - 1$, 100)	3420, 3270, 3180	2.98 (s, 3H, NCH ₃); 3.33 (br., 2H, NH ₂); 3.68 (s, 3H, 1-NCH ₃); 6.5–7.5 (m, 9H _{arom})
5f	4	78–80°	C ₁₃ H ₁₈ N ₄ (230.3)	230 (100)	3450, 3310, 3210	1.04 (t, 3H, $J = 7.3$ Hz, CH ₂ CH ₃); 2.41 (q, 2H, $J = 7.3$ Hz, CH ₂ CH ₃); 3.15 (s, 3H, NCH ₃); 3.59 (s, 3H, 1-NCH ₃); 4.26 (br., 2H, NH ₂); 6.4–7.4 (m, 5H _{arom})

^a Satisfactory microanalyses obtained; C \pm 0.30, H \pm 0.27, N \pm 0.29.^b Measured as film.

methylanilino)-3(5)-phenylpyrazole (**3a**) was obtained in 74% yield. Likewise, the reaction of **2b**, **2c**, **2d**, and **2f** with hydrazine hydrate gave 3(5)-(*p*-tolyl)-, 3(5)-(*p*-anisyl)-, 3(5)-(*p*-chlorophenyl)- and 3(5)-ethyl-5(3)-amino-4-(*N*-methylanilino)-pyrazoles (**3b**, **3c**, **3d**, and **3f**), respectively. The reaction using **2e**, however, did not give the corresponding 5(3)-amino-4-(*N*-methylanilino)-pyrazole (**3e**), but a substantial amount of *N*-methylaniline and unknown products. Under similar conditions, treatment of **2a** with monomethylhydrazine led to 5-amino-4-(*N*-methylanilino)-3-phenyl-1-methylpyrazole (**4a**) and 3-amino-4-(*N*-methylanilino)-5-phenyl-1-methylpyrazole (**5a**); **4a** being predominant. Likewise, the reaction using **2d** and **2f** gave 3-(*p*-chlorophenyl)- and 3-ethyl-5-amino-4-(*N*-methylanilino)-1-methylpyrazoles (**4d** and **4f**), and 5-(*p*-chlorophenyl)- and 5-ethyl-3-amino-4-(*N*-methylanilino)-1-methylpyrazoles (**5d** and **5f**).

Definitive evidence for the structural assignments of **4** and **5** is obtained from the mass spectra: cleavage of **4** results in the fragment $[\text{H}_3\text{C}-\text{N}=\text{C}(\text{NH}_2)-\text{C}=\text{N}-\text{C}_6\text{H}_5]^+$ whereas cleavage of **5** results in the fragment $[\text{H}_3\text{C}-\text{N}=\text{C}(\text{R})-\text{C}=\text{N}-\text{C}_6\text{H}_5]^+$.

The melting points were determined with a Yanagimoto MP-32 melting point apparatus. I.R. spectra were taken on a JASCO A-202

spectrometer, and ¹H-N.M.R. spectra were recorded on a Hitachi R-24 or R-600 spectrometer (60 MHz). α -(*N*-Methylanilino)-acetonitrile (**1**) was prepared according to Ref.⁵ and compounds **2a–d** according to Ref.¹.

2-(*N*-Methylanilino)-3-hydroxyacrylonitrile (**2c**):

To a mixture of tetrahydrofuran (17 ml) and potassium hydride (0.553 g, 13.8 mmol) cooled in an ice bath, is added dropwise a mixture of **1** (0.644 g, 4.41 mmol) and ethyl formate (0.517 g, 6.98 mmol) dissolved in tetrahydrofuran (6 ml) under a nitrogen atmosphere. The mixture is stirred for 4 h, the ice bath is removed and the mixture is stirred for 1 h at room temperature. The mixture is then poured into saturated ammonium chloride solution (20 ml) and the product is extracted with dichloromethane (2 \times 50 ml). The organic layer is washed with water (2 \times 50 ml), dried with anhydrous magnesium sulfate and the solvent is removed. The residue is purified by column chromatography on silica gel using a 6:1 mixture of benzene and hexane as eluent; yield: 0.608 g (79%); m. p. 82.5–84 °C.

C₁₀H₁₀N₂O calc. C 68.95 H 5.79 N 16.08
(174.2) found 68.93 5.81 16.06

I.R. (KBr): $\nu = 3350$ (OH); 2220 cm^{-1} (CN).

M.S. (70 eV): $m/e = 174$ (M^+ , 100%); 157 (38%), 145 (92%).

¹H-N.M.R. (CDCl_3/TMS): $\delta = 2.89$ [s, N—CH₃ of (*Z*)-isomer]; 2.97 [s, N—CH₃ of (*E*)-isomer]; 6.60–7.50 ppm (m, =CH— + H_{arom}).

1-Cyano-2-hydroxy-1-(N-methylanilino)-1-butene (2f):

Analogously, the reaction of **1** (0.641 g, 4.38 mmol) with ethyl propanoate (0.528 g, 5.17 mmol) in the presence of sodium hydride (0.237 g, 9.88 mmol) in dimethylformamide gives **2f**; yield: 0.567 g (64%); m.p. 111–112 °C.

C₁₂H₁₄N₂O calc. C 71.26 H 6.98 N 13.85
(202.3) found 71.30 6.96 13.86

I.R. (KBr): ν = 3200 (OH); 2220 cm⁻¹ (CN).

M.S. (70eV): m/e = 202 (M⁺, 71%); 146 (49%); 145 (100%); 131 (32%); 118 (22%); 104 (30%).

¹H-N.M.R. (CDCl₃/TMS): δ = 1.25 (t, J = 7.0 Hz, CH₂—CH₃); 2.49 (q, J = 7.0 Hz, CH₂—CH₃); 2.94 (s, N—CH₃); 6.50–7.60 ppm (m, H_{arom} + OH).

3(5)-Amino-4-(N-methylanilino)-5(3)-phenylpyrazole (3); Typical Procedure:

A mixture of **2a** (0.502 g, 2.01 mmol) and hydrazine hydrate (1.39 g, 27.8 mmol) in ethyl alcohol (15 ml) is gently refluxed for about 24 h until the disappearance of **2a** is confirmed by means of T.L.C. The mixture is cooled, and poured into water (50 ml) and extracted with dichloromethane (2 × 50 ml). The organic layer is washed with water (2 × 50 ml) and dried with magnesium sulfate. After filtering off magnesium sulfate and distilling off dichloromethane, the residue is purified by means of column chromatography on silica gel using a 4 : 1 mixture of benzene and ethyl acetate as eluent; yield: 0.393 g (74%) (Table).

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