

gel permeation on Sephadex LH-20. Sugar [3,4] and methylation [5] analyses showed 3 and 4 to be glucopyranosides. The structure of 3 is evident from its elemental composition, spectral properties and the fact that it gave *p*-cresyl- $\beta$ -D-glucopyranoside upon catalytic hydrogenation. Parishin (4) is a neutral substance which on catalytic hydrogenation followed by treatment with diazomethane yielded *p*-cresyl- $\beta$ -D-glucopyranoside and trimethyl citrate in the molar ratio 3:1. The fission must be due to hydrogenolysis of benzyl esters, as no signals for aromatic methyl groups were observed in the NMR spectrum of 4. These results demonstrate that 4 is tris[4-( $\beta$ -D-glucopyranosyloxy)benzyl] citrate, a structure also consistent with the NMR spectrum.

#### EXPERIMENTAL

General conditions were the same as in an earlier communication [2].

**Plant material.** *Vanda parishii*. Rchbf. was delivered from Mr N. Prakash, Chandra Orchid and Bulb Nurseries, 8 1/2 miles P.O. Kalimpong, West Bengal, India.

**Isolation of 3 and 4.** Fresh plants of *V. parishii* (3 kg) were extracted with MeOH (10 l), and the solution was concentrated to 0.65 l. A part (100 ml) of this extract was diluted to 300 ml with water and washed with  $\text{CHCl}_3$  (4  $\times$  50 ml). The aqueous layer was saturated with butanol and extracted with butanol saturated with water (7  $\times$  50 ml). The butanolic phase was washed with water (25 ml) and evaporated to dryness. A part (2 g) of the residue (6.4 g) was chromatographed on a silica gel column (5  $\times$  8.5 cm) using  $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  (13:7:2, lower phase) as eluent. The fraction containing 3 (300 mg) was filtered through a column of Sephadex LH-20 (5  $\times$  70 cm) using EtOH- $\text{H}_2\text{O}$  (1:1) as eluent giving crude 3 (96 mg), which was crystallised from *iso*-PrOH- $\text{H}_2\text{O}$ . Recrystallisation from EtOAc-EtOH gave 3 (56 mg). A part (220 mg) of the fraction containing 4 (450 mg) was chromatographed on silica gel (2.6  $\times$  11 cm) using the same eluent as above. The fraction containing 4 (150 mg) was filtered through a column of Sephadex LH-20 (2.5  $\times$  83 cm) using EtOH- $\text{H}_2\text{O}$  (1:1) as eluent, giving 4 as a colourless amorphous solid (78 mg).

**Glucoside 3.** Needles (EtOAc-EtOH), mp 154–157°C;  $[\alpha]_D^{25} - 63^\circ$  (c 0.77, MeOH). (Found: C 54.6; H 6.3; O 39.1.  $\text{C}_{13}\text{H}_{18}\text{O}_7$  requires: C 54.5; H 6.3; O 39.1). IR:  $\nu_{\text{max}}^{\text{KBr}}$  3700–3000(s), 1615(m), 1590(m), 1510(s)  $\text{cm}^{-1}$ . UV:  $\lambda_{\text{max}}^{\text{MeOH}}$  (log  $\epsilon$ ) 277.5 (2.96), 271 (3.04) nm.  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ ):  $\delta$  3.36–4.12 7.19 and 7.42 (4 H,  $\text{A}_2\text{B}_2$  system,  $J$  9 Hz).

**Parishin (4).** Amorphous solid,  $[\alpha]_D^{25} - 59^\circ$  (c 0.80, MeOH). IR:  $\nu_{\text{max}}^{\text{KBr}}$  3700–3000(s), 1735(s), 1615(m), 1590(m), 1515(s)  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR ( $\text{D}_2\text{O}$ ):  $\delta$  2.76 and 2.94 (4 H, two AB systems,  $J$  15 Hz), 3.4–4.0 (18 H), 4.4–5.1 (the benzylic and the anomeric protons; the HOD signal partially overlapping), 6.9–7.4 (12 H).  $^1\text{H}$  NMR (Pyridine- $d_3$ ):  $\delta$  3.32 and 3.37 (4 H, two AB systems,  $J$  15 Hz), 3.80–4.66 (18 H), 5.09 (s, 4 H), 5.28 (s, 2 H), 5.58 (d, 3 H,  $J$  6 Hz), 7.1–7.5 (12 H).

**Hydrogenation of 3.** A soln of 3 (51 mg) in MeOH (9 ml) was hydrogenated over Pd (20 mg, 10% on carbon) at room temp. and atm. pres. After 7 hr the catalyst was filtered off and the soln was evaporated to dryness giving *p*-cresyl- $\beta$ -D-glucopyranoside, mp 180–182°C (*iso*-PrOH- $\text{H}_2\text{O}$ );  $[\alpha]_D^{25} - 67^\circ$  (c 0.76,  $\text{H}_2\text{O}$ ) (lit. [6] mp 178–179.5°C;  $[\alpha]_D^{20} - 67.7^\circ$  ( $\text{H}_2\text{O}$ )), further identified by NMR.

**Hydrogenation of 4.** Parishin (92 mg) was hydrogenated as described for 3. Catalyst was filtered off and the soln was treated with an excess of  $\text{CH}_2\text{N}_2$  in Et $_2\text{O}$  and evaporated to dryness. Residue was washed with  $\text{CHCl}_3$  (5  $\times$  1 ml) and the  $\text{CHCl}_3$  phase was evaporated to dryness. Residue was crystallised from Et $_2\text{O}$ -hexane at  $-20^\circ\text{C}$  giving trimethyl citrate (14 mg), mp 76–78°C (lit. [7] mp 78.5–79°C). The total amount of trimethyl citrate was found by GLC to be 17.5 mg. The residue insoluble in  $\text{CHCl}_3$  above was dissolved in  $\text{H}_2\text{O}$  and the soln washed with  $\text{CHCl}_3$ -MeOH (1:1). The aq phase was evaporated to dryness giving *p*-cresyl- $\beta$ -D-glucopyranoside (68 mg), mp 181–182.5°C;  $[\alpha]_D^{25} - 66^\circ$  (c 0.27,  $\text{H}_2\text{O}$ ).

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### NEOCAPILLEN, A NEW ACETYLENIC HYDROCARBON FROM *ARTEMISIA CAPILLARIS*

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**Key Word Index**—*Artemisia capillaris*; Compositae; acetylenic hydrocarbon; neocapillen.

**Abstract**—During an investigation of *Artemisia capillaris*, a new acetylenic hydrocarbon, neocapillen, was isolated as a minor component and its structure determined.

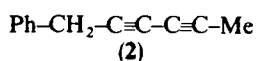
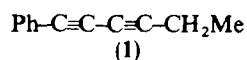
The structures of the new acetylenic compounds, 1-(2'-methoxyphenyl)-2,4-hexadiyne (*o*-methoxy capillen), capillanol and norcapillen, in the essential oil of the stalks and leaves of *Artemisia capillaris* was described previously [1-3]. As a continuation of our study on the components of *A. capillaris* extracts, which have been recognized to make an important contribution to the characteristic odour, we wish to report the identification of a new polyacetylenic hydrocarbon neocapillen (1), which seemed to be derived from capillen (2) [4].

Fresh roots of *A. capillaris* were chopped finely and extracted with ether. The extract was chromatographed on alumina and divided into 3 fractions from which compound 1, which had a Shiitake-like odour, was isolated by PLC.

By elementary analysis, the molecular formula of 1 was determined to be  $C_{12}H_{10}$  (Found: C, 93.44; H, 6.56; Calcd.: 94.46; H, 6.54%). The spectral data of 1 are given in the Experimental. The IR showed absorptions at 2210, 2200 and  $2120\text{ cm}^{-1}$  ( $\text{C}\equiv\text{C}$ ), at 1595 and  $1490\text{ cm}^{-1}$  (aromatic), and at 755 and  $690\text{ cm}^{-1}$  (aromatic H). These data indicated that the compound was an aromatic mono-substituted hydrocarbon with a  $C_6H_5$  unit. The IR absorptions at 1210 and  $1120\text{ cm}^{-1}$  and the intense UV absorption at 222 nm in *n*-hexane indicated the presence of a 1-phenyl-1,3-diyne function [5]. The NMR spectrum showed no olefinic proton, a signal at  $\delta$  2.16(2H) was the two protons of a methylene attached to a conjugated diacetylenic bond. The five protons in the benzene ring appeared as a broad singlet from  $\delta$  7.05 to 7.55. Consequently, the splitting pattern of the signals in the NMR spectrum were consistent with structure (1) for neocapillen. It had a UV spectrum almost super-imposable with that of norcapillen [3].

Besides the molecular ion peak at  $m/e$  154 the other significant MS peaks were at  $m/e$  153( $M^+-H$ , 82%), 152( $M^+-H_2$ , 47%), 139( $\text{Ph}-\text{C}\equiv\text{C}-\text{C}\equiv\text{C}-\text{CH}_2^+$ , 36%), 113( $\text{Ph}-\text{C}\equiv\text{C}-\text{C}^+$ , 7%).

For the confirmation of the identity, compound 1 was prepared from capillen. Good agreement of the IR, NMR and MS spectra was observed between neocapillen (1) and the synthetic material. 1 may be an artefact produced from 2 by UV irradiation, although no acetylenic substance has been detected in the upper parts of the plants. 1 was catalytically hydrogenated over  $\text{PtO}_2$  in ethanol to give octahydronorcapillen, which was found to be identical with *n*-hexylbenzene in all respects (IR, NMR, MS spectrum). Thus, the structure of neocapillen was confirmed to be 1-phenyl-1,3-hexadiyne. The synthesis of this phenylacetylene had been reported [6]. Many phenyl diacetylene ( $\text{Ph}-\text{C}\equiv\text{C}-\text{C}\equiv\text{C}-\text{R}$ ) derivatives which have  $C_7$  or  $C_{11}$ -side chains have been detected in the genus *Bidens*, *Coreopsis* and *Dahlia*.



The first isolation of neocapillen from a natural source is significant from biogenetic considerations as all the phytoacetylenes so far isolated, have a phenyl diacetylene skeleton or a modification thereof.

## EXPERIMENTAL

UV spectra were measured in EtOH. NMR spectra were determined at 60 MHz, in  $\text{CCl}_4$  and with TMS as internal standard.

**Plant material and oil removal.** Fresh roots (5 kg) of *A. capillaris*, collected in October at the suburbs of Osaka prefecture, were chopped finely and extracted with  $\text{Et}_2\text{O}$  for 5 weeks at room temp. Neocapillen was also obtained after only several hr extraction.

**Isolation of neocapillen (1).** The extract (15.8 g) was chromatographed on deactivated alumina with *n*-hexane,  $\text{C}_6\text{H}_6$  and  $\text{Et}_2\text{O}$  and divided into 3 fractions. Elution with  $\text{C}_6\text{H}_6$  gave neocapillen (1) which was then isolated by PLC ( $R_f$  0.60) developed with *n*-hexane-EtOAc (6:1). MS  $m/e$  (rel. int.): 154(100%)  $M^+$ , 153(82), 45(63), 152(47), 139(36), 73(29); 89(24), 88(22), 43(20), 151(18). IR  $\nu_{\text{max}}$ ,  $\text{cm}^{-1}$ : 3025, 3005, 2960, 2900, 2855, 2825, 2210, 2200, 2120, 1943, 1880, 1595, 1570, 1490, 1450, 1440, 1425, 1375, 1330, 1290, 1170, 1150, 1065, 1050, 1020, 950, 785, 755, 690. UV  $\lambda_{\text{EtOH}}^{\text{max}}$ , nm: (log  $\epsilon$ ) 211 nm (3.42), 222(3.55), 244(2.63), 257(3.13), 272(3.32), 289(3.20). NMR:  $\delta$  1.00(3H, t,  $J = 6.2$  Hz), 2.16(2H, q,  $J = 6.2$  Hz), 7.05 ~ 7.55(5H, m).

**Catalytic hydrogenation of 1.** Catalytic hydrogenation of 1 (50 mg) in EtOH (5 ml) over  $\text{PtO}_2$  (1.4 mg) was carried out at room temp for 3 hr. The product was purified by GLC (Carbowax-20 M 5%, 80 ~ 100 mesh, 4 mm  $\times$  3 m, 1.5 Kg/ $\text{cm}^2$ ), to give a colourless oil, NMR:  $\delta$  0.88(3H, t,  $-\text{CH}_2-\text{CH}_3$ ), 1.26(8H, m,  $-(\text{CH}_2)_4-$ ), 2.68(2H, t,  $\phi-\text{CH}_2-$ ), 7.23(5H, m,  $\phi-$ ). (Found: C, 88.80; H, 11.20; Calcd for  $\text{C}_{12}\text{H}_{18}$ : C, 88.81; H, 11.19%).

**Isolation of capillen (2)** The essential oil from stalk and leaves of a *A. capillaris* was chromatographed on activated alumina (60 g, 300 mesh) with *n*-hexane,  $\text{C}_6\text{H}_6$ ,  $\text{Et}_2\text{O}$ , EtOAc and EtOH, and divided into 5 fractions. Subsequent elution with  $\text{C}_6\text{H}_6$  gave capillen (2) which was then isolated by prep GLC, and identified (UV, IR, NMR) by comparison with an authentic specimen [4,7].

**Catalytic hydrogenation of 2.** Catalytic hydrogenation of 2 (55 mg) in EtOH (5 ml) over  $\text{PtO}_2$  (1.5 mg) was carried out at room temp for 3.5 hr. The product was purified by prep GLC. NMR:  $\delta$  0.88 (3H, t,  $-\text{CH}_2-\text{CH}_3$ ), 1.26(8H, m,  $-(\text{CH}_2)_4-$ ), 2.68(2H, t,  $\phi-\text{CH}_2-$ ), 7.23(5H, m,  $\phi-$ ). MS  $m/e$ : 162( $M^+$ , 26%).

**Irradiation of capillen (2).** 2 (50 mg) dissolved in a mixture of 70 ml MeOH and a trace of KOH were irradiated for about 24 hr in a Pyrex tube with light of wavelength 3500 Å. This soln was acidified and extracted with  $\text{Et}_2\text{O}$ ; the extract was washed, dried and evaporated to leave a viscous reddish oil. After separation by prep GLC 10 mg of neocapillen (1) were obtained as a pure compound.

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