

An Enantioconvergent Route to (-)-Kainic Acid

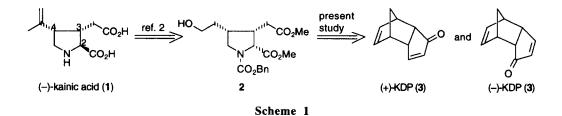
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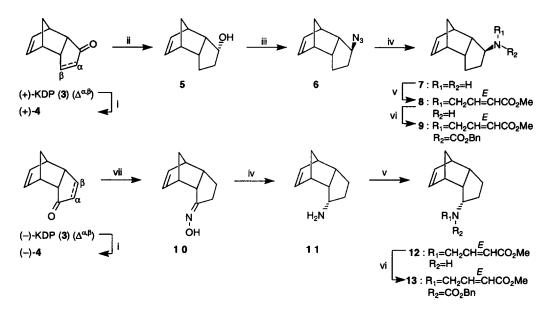
Abstract: A new stereocontrolled route to (-)-kainic acid, the representative of the kainoid amino acids and exhibiting insecticidal, anthelmintic and neuroexcitatory properties, has been devised in an enantioconvergent way using either (+)- or (-)-ketodicyclopentadiene as the starting material by employing a concurrent retro-Diels-Alder reaction and intramolecular ene reaction as the key step. © 1997, Elsevier Science Ltd. All rights reserved.

The kainoid amino acids have attracted considerable interest because of their pronounced insecticidal, anthelmintic and neuroexcitatory properties.¹ Among these properties, the neuroexcitatory activity is attributed to their *trans*-C-2/C-3:*cis*-C-3/C-4 structure and the functionality on the C-4 center besides 2-carboxy-3-carboxymethyl functionalities. Accordingly, an efficient enantiocontrolled procedure being capable of producing the appropriate 2,3,4-trisubstituted pyrrolidines carrying a versatile C-4 functionality has been sought. We report here a new stereocontrolled route to (–)-kainic acid (1) in a formal sense by the synthesis of the key intermediate² 2 carrying a versatile 4-(2-hydroxyethyl) group starting either from the (+)- or (–)-enantiomer of optically pure ketodicyclopentadiene^{3,4} (KDP) (3) by employing a concurrent one-pot retro-Diels-Alder reaction and intramolecular ene reaction as the key step (Scheme 1).



(+)-KDP [(+)-3] (>99% ee) was transformed stereoselectively into the *endo*-alcohol⁵ 5, mp 96 °C, $[\alpha]_D^{28}$ -13.1 (c 0.5, CHCl₃), in 88% yield via the (+)-ketone⁶ (+)-4. The Mitsunobu reaction⁷ of 5 with diphenylphosphoryl azide (DPPA) gave the *exo*-azide 6, $[\alpha]_D^{27}$ +70.4 (c 1.0, CHCl₃), with inversion which was sequentially transformed into the *exo*-carbamate 9, $[\alpha]_D^{25}$ +49.1 (c 1.0, CHCl₃), having the *N*-(*E*-3ethoxycarbonyl-2-propenyl) group via the exo-primary amine 7 and the exo-secondary amine 8. Overall yield of 9 from 5 was 44% in four steps.

On the other hand, (-)-KDP [(-)-3] (>99% ee) was transformed into the single oxime 10, $[\alpha]_D^{27}$ -279.5 (c 1.3, CHCl₃), via the (-)-ketone⁶ (-)-4. On sequential stereoselective reduction, N-alkylation, and carbamoylation, 10 furnished the *endo*-carbamate 13, $[\alpha]_D^{27}$ -83.8 (c 1.6, CHCl₃), via the *endo*-primary amine 11 and the *endo*-secondary amine 12. Overall yield of 13 from (-)-3 was 49% in five steps (Scheme 2).



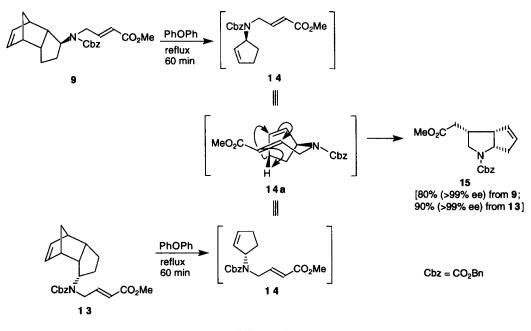
Scheme 2

Reagents and conditions: i, Zn, AcOH-EtOH (1:4), reflux [97% for (-)-4]; ii, NaBH₄, MeOH, -78 °C [88% from (+)-3]; iii, DPPA, (NCO₂Et)₂, PPh₃, THF, 0 °C ~ room temp. (94%); iv, LiAlH₄, THF, room temp.; v, BrCH₂CH=CHCO₂Me, Et₃N, DMF, 0 °C; vi, ClCO₂Bn, NaH, DMF, -20 °C [47% for 9 from 6 and 49% for 13 from (-)-3]; vii, NH₂OH·HCl, pyridine, room temp. (97%).

Having obtained the exo-9 and the endo-13 diastereomers, we next examined the thermolysis of these compounds in hot diphenyl ether. Since we had learned that the retro-Diels-Alder reaction of certain substrates having a bicyclo[2.2.1]heptene system relating to 9 and 13 was best carried out in diphenyl ether at boiling point⁸ (~260 °C), we expected both of these would form the same monocyclic carbamate 14 having a 1,6-diene system which further would undergo the intramolecular ene reaction[°] under the same conditions to give the bicyclo product 15 having a trisubstituted pyrrolidine framework.

When the *exo-9* was heated in diphenyl ether at refluxing temperature, the reaction terminated within 60 min to give rise to the bicyclic product **15**, $[\alpha]_{D}^{27}$ +95.8 (*c* 1.0, CHCl₃), in 80% yield as the single isomer as expected. On the same treatment, the *endo*-diastereomer **13** afforded the same bicyclic product **15**, $[\alpha]_{D}^{30}$ +94.7 (*c* 1.1, CHCl₃), in 90% yield as a single isomer. The optical purities of the product **15** from both **9** and **13** were determined to be >99% ee by hplc using a chiral column (CHIRALCEL OD, elution with 5% *i*-

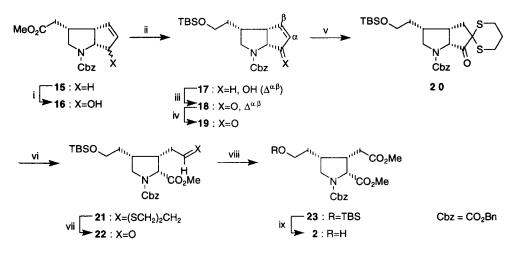
PrOH/hexane), indicating that the original chiral integrity of the precursor carbamates was preserved under these thermal conditions. The relative stereochemistry of 15 was determined to be the all-*cis* configuration as shown by facile iodolactone formation (single product in 73% yield) when the carboxylate generated from 15 by saponification was treated with iodine in acetonitrile¹⁰; its absolute stereochemistry, however, was determined after converting it into the key intermediate² 2 (see below). These findings led us to conclude that the intramolecular ene reaction of the transient 1,6-diene 14 took place in a stereospecific manner taking the most unstrained transition state 14a to afford the all-*cis* product 15 (Scheme 3).





In order to obtain the key intermediate 2 of (-)-kainic acid (1), the ene product 15 was first treated with selenium(IV) oxide in the presence of formic acid¹¹ to give the allylic alcohol 16 in 68% yield as a diastereomeric mixture. On sequential hydride reduction, selective *O*-silylation, manganese(IV) dioxide oxidation, and the cuprate-mediated 1,4-reduction,¹² 16 furnished the bicyclic ketone 19, $[\alpha]_D^{27}$ +106.6 (*c* 1.3, CHCl₃), in 58% overall yield *via* the silyl ether 17 and the enone 18, $[\alpha]_D^{27}$ +124.0 (*c* 1.6, CHCl₃). The ketone 19 was transformed into the α -diketone monothioketal¹³ 20, $[\alpha]_D^{21}$ -21.8 (*c* 1.1, CHCl₃), in 85% yield which was cleaved under basic conditions¹³ to give the dithiane 21, $[\alpha]_D^{28}$ -7.1 (*c* 1.0, CHCl₃), in 60% yield after esterification. Hydrolysis of the dithiane group¹³ of 21 afforded the aldehyde 22, $[\alpha]_D^{27}$ +19.2 (*c* 0.7, CHCl₃), in 90% yield which, on oxidation with a mixture of iodine and potassium hydroxide in methanol,¹⁴ furnished the key intermediate 2, $[\alpha]_D^{31}$ +13.1 (*c* 0.3, CHCl₃) [lit.^{2b}: $[\alpha]_D^{27}$ +11.7 (*c* 0.5, CHCl₃)], in 61% yield after desilylation¹⁵ of the resulting monocyclic silyl ether 23, $[\alpha]_D^{28}$ +32.8 (*c* 0.1, CHCl₄) (Scheme 4).

As we have obtained^{2a} (-)-kainic acid (1) from 2, the present synthesis constitutes a formal enantioconvergent synthesis starting from both (+)- and (-)-enantiomers of KDP (3). However, the present synthesis



Scheme 4

Reagents and conditions: i, SeO₂ (2 equiv.), HCO₂H (2 equiv.), dioxane, 90 °C, 4 h (68%); ii, LiAlH₄, THF (87%) then *t*-BuMe₂SiCl, Et₃N, CH₂Cl₂, room temp. (89%); iii, MnO₂, benzene-CH₂Cl₂ (1:1), room temp. (81%); iv, *i*-Bu₂AlH, CuI, HMPA-THF (1:4), -78 °C (92%); v, pyrrolidine, benzene, reflux then (TsSCH₂)₂CH₂, Et₃N, MeCN, reflux (85%); vi, *t*-BuOK (1.5 equiv.), *t*-BuOH, trace H₂O, \sim 50 °C, acid workup then CH₂N₂ (60%); vii, MeI, CaCO₃, H₂O-MeCN (1:4), reflux (90%); viii, l₂, KOH, MeOH, -10 °C; ix, HF·pyridine, THF, room temp. (61% from **22**).

has more potential value for the stereocontrolled construction of a variety of modified kainoid amino acids for biological investigation based on the functionality of the tricyclic starting material 3 which allows various stereocontrolled modifications⁴ not only on the side chain moieties, but also on the core pyrrolidine framework.

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