

## A New Synthetic Route to 1,3-Benzoxazepines

Kee-Jung Lee,\* Dae Ock Choi, Seongkon Kim, Jae Uk Jeong, Hokoon Park

Division of Chemistry, Korea Institute of Science and Technology, P.O. Box 131 Cheongryang, Seoul, Korea

A novel method for the synthesis of 5-hydroxy-1,3-benzoxazepines **5** utilizing the Staudinger reaction followed by an intramolecular aza-Wittig reaction, from *o*-acyloxyphenacyl azides **2** is reported.

In recent years, there has been a significant interest in the chemistry of iminophosphoranes (phosphazenes).<sup>1</sup> These compounds can be obtained by reaction of tertiary phosphines with organic azides as originally reported by Staudinger.<sup>2</sup> Iminophosphoranes are known to react with carbonyl compounds to give linear<sup>3</sup> or cyclic<sup>4</sup> imines, and this method has been used for the preparation of quinolines,<sup>5</sup> isoquinolines,<sup>6</sup> benzoxazoles,<sup>7</sup> pyrazines,<sup>8</sup> and triazoles.<sup>9</sup>

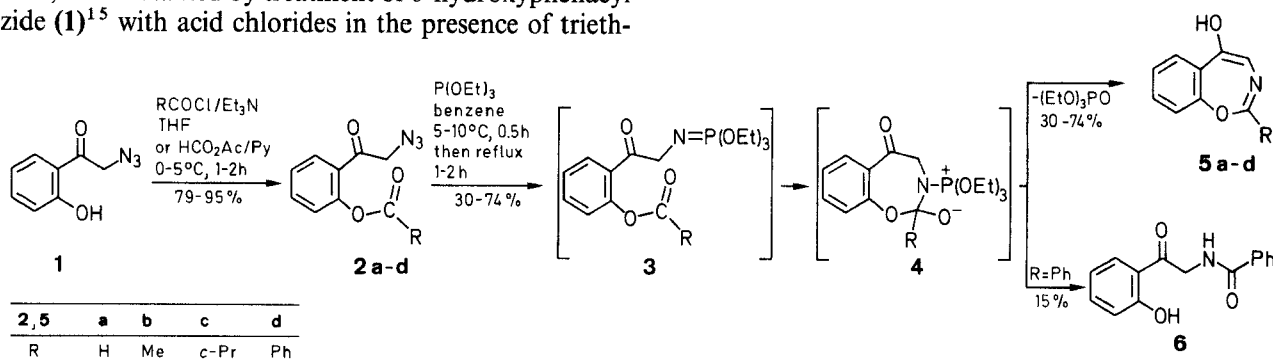
Continuing our studies on the synthesis of heterocyclic compounds,<sup>10</sup> we considered the reaction of *o*-acyloxyphenacyl azides **2** with triethyl phosphite as an entry to a new type of iminophosphoranes of potential utility in heterocyclic synthesis. We now describe a new method for the synthesis of the hitherto unknown 5-hydroxy-1,3-benzoxazepines **5** based upon the Staudinger reaction followed by an intramolecular aza-Wittig reaction of *o*-acyloxyphenacyl azides **2**. Two general methods for the preparation of 1,3-benzoxazepine ring system have been reported. The first method is based on the photo rearrangement of isoquinoline *N*-oxides,<sup>11,12</sup> whereas the second one involves the thermal rearrangement of azido chromenes.<sup>13,14</sup>

The starting compounds, *o*-acyloxyphenacyl azides **2b–d**, were obtained by treatment of *o*-hydroxyphenacyl azide (**1**)<sup>15</sup> with acid chlorides in the presence of trieth-

ylamine in tetrahydrofuran. On the other hand, **2a** was prepared by treatment of **1** with excess acetic formic anhydride in the presence of pyridine. The reaction of *o*-acyloxyphenacyl azides **2** with an equimolecular amount of triethyl phosphite in benzene at 5–10°C for 0.5 h, and then heating to reflux temperature gave the desired 5-hydroxy-1,3-benzoxazepines **5** in yields ranging from 30 to 74%. The isolation of iminophosphoranes **3** was unsuccessful<sup>8</sup> under various reaction conditions.

The formation of the cyclization products **5** can be assumed to proceed via the expected phosphorane intermediate **3** with the evolution of nitrogen followed by an intramolecular aza-Wittig reaction between phosphorane and ester carbonyl group, followed by spontaneous tautomerization. This assumption may be supported by the isolation of the rearrangement product, *N*-(*o*-hydroxyphenacyl)benzamide (**6**), when the reaction of *o*-benzoyloxyphenacyl azide (**2d**) with triethyl phosphite was carried out. Similar rearrangement pathway has been reported<sup>16</sup> in the reaction of *o*-azidophenyl benzoate with hexamethylphosphoric triamide.

Structural elucidation of **5** is accomplished on the basis of spectral data<sup>11</sup> and microanalyses. The IR spectra show absorption in the OH stretching region of  $\nu = 3047\text{--}3085\text{ cm}^{-1}$ . The <sup>1</sup>H-NMR spectra of **5** are also consistent with the structure assigned. These values are in



**Table 1.** *o*-Acyloxyphenacyl Azides **2** Prepared

Product	Reaction Time (h)	Yield <sup>a</sup> (%)	mp (°C) <sup>b</sup>	Molecular Formula <sup>c</sup>	<sup>1</sup> H-NMR (CDCl <sub>3</sub> /TMS) $\delta$ , J (Hz)
<b>2a</b>	2	91	oil	C <sub>9</sub> H <sub>7</sub> N <sub>3</sub> O <sub>3</sub> (205.2)	4.43 (s, 2H, CH <sub>2</sub> ), 7.10–7.93 (m, 4H <sub>arom</sub> ), 8.33 (s, 1H, CHO)
<b>2b</b>	1	95	54–55	C <sub>10</sub> H <sub>9</sub> N <sub>3</sub> O <sub>3</sub> (219.2)	2.33 (s, 3H, CH <sub>3</sub> ), 4.40 (s, 2H, CH <sub>2</sub> ), 7.07–7.87 (m, 4H <sub>arom</sub> )
<b>2c</b>	1	79	50–51	C <sub>12</sub> H <sub>11</sub> N <sub>3</sub> O <sub>3</sub> (245.2)	1.17 (m, 4H, CH <sub>2</sub> -cyclopropyl), 1.82 (m, 1H, CH-cyclopropyl), 4.36 (s, 2H, CH <sub>2</sub> ), 7.03–7.83 (m, 4H <sub>arom</sub> )
<b>2d</b>	1.5	90	59–60	C <sub>15</sub> H <sub>11</sub> N <sub>3</sub> O <sub>3</sub> (281.3)	4.36 (s, 2H, CH <sub>2</sub> ), 7.17–8.33 (m, 9H <sub>arom</sub> )

<sup>a</sup> Yield of pure isolated product, except **2a**.

<sup>b</sup> Recrystallized from Et<sub>2</sub>O/petroleum ether.

<sup>c</sup> Satisfactory microanalyses obtained: C  $\pm 0.28$ , H  $\pm 0.07$ , N  $\pm 0.30$ , except for **2a**, which could not be purified.

**Table 2.** 5-Hydroxy-1,3-benzoxazepines **5** Prepared

Prod- uct	Reaction Time (h)	Yield <sup>a</sup> (%)	mp (°C) (solvent)	Molecular Formula <sup>b</sup>	IR (KBr) $\nu_{OH}$ (cm <sup>-1</sup> )	<sup>1</sup> H-NMR (DMSO- <i>d</i> <sub>6</sub> + CDCl <sub>3</sub> /TMS) $\delta$ , <i>J</i> (Hz)	MS (70 eV) <i>m/z</i> (%)
<b>5a</b>	2	30	157–158 (Et <sub>2</sub> O)	C <sub>9</sub> H <sub>7</sub> NO <sub>2</sub> (161.2)	3047	6.86–7.19 (m, 3H <sub>arom</sub> ), 7.60 (s, 1H, H-4), 7.68 (m, 1H <sub>arom</sub> ), 7.98 (s, 1H, H-2), 9.92 (s, 1H, OH)	161 (M <sup>+</sup> , 21), 134 (29), 133 (22), 106 (13), 105 (100), 104 (29), 78 (42), 77 (45), 76 (23), 65 (19), 63 (22)
<b>5b</b>	2	74	160–161 (Et <sub>2</sub> O)	C <sub>10</sub> H <sub>9</sub> NO <sub>2</sub> (175.2)	3050	2.47 (s, 3H, CH <sub>3</sub> ), 6.86–7.29 (m, 3H <sub>arom</sub> ), 7.45 (s, 1H, H-4), 7.57 (m, 1H <sub>arom</sub> ), 10.33 (s, 1H, OH)	175 (M <sup>+</sup> , 43), 134 (29), 133 (15), 105 (100), 104 (15), 91 (13), 78 (19), 77 (31), 76 (13), 65 (15), 63 (12)
<b>5c</b>	1	68	181–182 (Et <sub>2</sub> O)	C <sub>12</sub> H <sub>11</sub> NO <sub>2</sub> (201.2)	3059	1.05 (m, 4H, CH <sub>2</sub> -cyclopropyl), 2.11 (m, 1H, CH-cyclopropyl), 6.83–7.14 (m, 3H <sub>arom</sub> ), 7.41 (s, 1H, H-4), 7.59 (m, 1H <sub>arom</sub> ), 9.94 (s, 1H, OH)	201 (M <sup>+</sup> , 67), 172 (14), 134 (38), 133 (14), 131 (37), 121 (18), 105 (100), 104 (18), 78 (13), 77 (28), 76 (10), 65 (20), 63 (10)
<b>5d</b>	1	39	251–252 (Et <sub>2</sub> O/ EtOAc)	C <sub>15</sub> H <sub>11</sub> NO <sub>2</sub> (237.3)	3085	6.90–7.21, 7.48–7.54 (m, 6H <sub>arom</sub> ), 7.67 (s, 1H, H-4), 7.80, 8.10 (m, 3H <sub>arom</sub> ), 10.19 (s, 1H, OH)	237 (M <sup>+</sup> , 58), 181 (37), 134 (49), 105 (100), 104 (14), 89 (27), 78 (13), 77 (54), 76 (23), 65 (21), 63 (27)

<sup>a</sup> Yield of isolated pure product. Low yield **5a** presumably due to the instability of **2a**.

<sup>b</sup> Satisfactory microanalyses obtained: C  $\pm$  0.32, H  $\pm$  0.27, N  $\pm$  0.31.

good agreement with those reported for the similar system.<sup>13</sup> In the NMR spectrum of **5b** the signal at  $\delta$  = 10.33 (OH) disappeared by addition of deuterium oxide. All compounds show a molecular ion and peaks due to the loss of RCN from the molecular ion, then further the expulsion of carbon monoxide. The most intense peaks in the spectra of **5** occur at *m/z* = 105 and corresponds to the benzoyl radical ion.

We have thus worked out a useful and simple method for the synthesis of 5-hydroxy-1,3-benzoxazepines **5**, which are not easily obtainable by other routes.

Dry N<sub>2</sub> gas was routinely employed as the reaction atmosphere in all reactions. Benzene and THF were dried and distilled from Na and LiAlH<sub>4</sub>, respectively. Silica gel EM 7747 for column chromatography was used throughout for product separation. Melting points were taken using an Electrothermal melting point apparatus and are uncorrected. Microanalyses were obtained using a Perkin-Elmer 240 DS element analyzer. Mass spectra were obtained using a Hewlett Packard model 5985 B spectrometer. IR spectra were recorded on a Analect FX 6160 IR spectrophotometer. <sup>1</sup>H-NMR spectra were measured on either a Bruker AM-200 (**5a–d**) or a Varian EM-360 A spectrometer (**2a–d**).

$\alpha$ -Hydroxyphenacyl azide (**1**)<sup>15</sup> and acetic formic anhydride<sup>17</sup> were prepared following literature procedures.

#### *o*-Formyloxyphenacyl Azide (**2a**):

To a stirred solution of *o*-hydroxyphenacyl azide (**1**; 1.77 g, 10 mmol) in excess acetic formic anhydride (30 mL) is added dropwise pyridine (0.97 g, 12.3 mmol) at 0–5°C. After stirring for 2 h at r.t. the mixture is poured into cold water (200 mL), and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3  $\times$  50 mL). The extract is dried (MgSO<sub>4</sub>), and the solvent is evaporated under reduced pressure to give **2a** as

a light yellow oil, which is subjected to the next reaction without further purification;<sup>18</sup> yield: 1.86 g (91 %).

#### *o*-Acyloxyphenacyl Azides **2b–d**; General Procedure:

To a stirred solution of *o*-hydroxyphenacyl azide (**1**; 1.77 g, 10 mmol) in THF (20 mL) is added dropwise Et<sub>3</sub>N (1.11 g, 11 mmol) and the acid chloride (10.5 mmol) at 0–5°C. After stirring for the time indicated in Table 1 at r.t., the precipitated solid (Et<sub>3</sub>N  $\cdot$  HCl) is filtered, and the filtrate is concentrated under reduced pressure. The residual oil is purified by short-column chromatography on silica gel (hexane/EtOAc, 10:1) to give **2b–d**.

#### 5-Hydroxy-1,3-benzoxazepines **5**; General Procedure:

To a stirred solution of the appropriate *o*-acyloxyphenacyl azide **2** (10 mmol) in benzene (30 mL) is added slowly P(OEt)<sub>3</sub> (1.66 g, 10 mmol) at 5–10°C. The mixture is stirred for 0.5 h at r.t., and then refluxed for the time indicated in Table 2. The mixture is concentrated under reduced pressure, and the residual material is chromatographed on a silica gel column (hexane/EtOAc, 1:1) to give **5** as colorless crystals. An analytical sample is prepared by recrystallization from the appropriate solvent (Table 2).

In the case of the reaction of **2d** with P(OEt)<sub>3</sub>, the rearrangement product, *N*-(2-hydroxyphenacyl)benzamide (**6**) is also isolated; yield: 15%; mp 148–149°C (EtOAc/Et<sub>2</sub>O).

#### **6:**

C<sub>15</sub>H<sub>13</sub>NO<sub>3</sub> calc. C 70.58 H 5.13 N 5.49  
(255.3) found 70.86 5.23 5.15

<sup>1</sup>H-NMR (DMSO-*d*<sub>6</sub>/TMS):  $\delta$  = 4.81 (d, 2H, *J* = 5.66 Hz, CH<sub>2</sub>), 7.01, 7.52, 7.94 (m, 9H<sub>arom</sub>), 8.85 (t, 1H, *J* = 5.62 Hz, NH), 11.39 (s, 1H, OH).

MS (70 eV): *m/z* = 255 (M<sup>+</sup>), 121, 105, 93, 77, 65, 51.

The authors are indebted to the Ministry of Science and Technology for financial support, project number NO6545.

Received: 21 November 1989

- (1) Gololobov, Yu.G.; Zhmurova, I.N.; Kasukhin, L.F. *Tetrahedron* **1981**, 37, 437.
- (2) Staudinger, H.; Meyer, J. *Helv. Chim. Acta* **1919**, 2, 635.
- (3) Wong, S.C.K.; Johnson, A.W. *J. Org. Chem.* **1972**, 37, 1850.
- (4) Pailer, M.; Haslinger, E. *Monatsh. Chem.* **1970**, 101, 508.  
Lambert, P.H.; Vaultier, M.; Carrie, R. *J. Chem. Soc., Chem. Commun.* **1982**, 1224.
- (5) Foster, S.A.; Leyshon, L.J.; Saunders, D.G. *J. Chem. Soc., Chem. Commun.* **1973**, 29.
- (6) Aubert, T.; Farnier, M.; Hanquet, B.; Guillard, R. *Synth. Commun.* **1987**, 17, 1831.
- (7) Leyshon, L.J.; Saunders, D.G. *J. Chem. Soc., Chem. Commun.* **1971**, 1608.
- (8) Zbiral, E.; Stroh, J. *Liebigs Ann. Chem.* **1969**, 727, 231.
- (9) Bruche, L.; Garanti, L.; Zecchi, G. *Synthesis* **1985**, 304.
- (10) Lee, K.-J.; Kim, S.; Um, H.; Park, H. *Synthesis* **1989**, 638.
- (11) Buchardt, O.; Lohse, C.; Duffield, A.M.; Djerassi, C. *Tetrahedron Lett.* **1967**, 2741.
- (12) Albin, A.; Fasani, E.; Dacrema, L.M. *J. Chem. Soc., Perkin Trans. 1* **1980**, 2738.
- (13) Desbene, P.-L.; Cherton, J.-C. *Tetrahedron* **1984**, 40, 3567.
- (14) Le Roux, J.P.; Cherton, J.-C.; Desbene, P.-L. *C.R. Acad. Sci. Ser. C* **1974**, 278, 1389; *C.A.* **1974**, 81 105473.
- (15) Boyer, J.H.; Straw, D. *J. Am. Chem. Soc.* **1953**, 75, 2683; Buu-Hoi, Ng.Ph.; Lavit, D. *J. Chem. Soc.* **1955**, 18.
- (16) Cadogan, J.I.G.; Stewart, N.J.; Tweddle, N.J. *J. Chem. Soc., Chem. Commun.* **1978**, 182.
- (17) Krimen, L.I. *Org. Synth.* **1970**, 50, 1.
- (18) Purification by column chromatography on silica gel was unsuccessful due to decomposition.