

CHIRAL SULFONYLMETHYL ISOCYANIDES. SYNTHESIS AND ASYMMETRIC INDUCTION¹

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We are engaged in the synthesis and application of chiral derivatives of methyl isocyanides. Chart 1 depicts 7 chiral analogues (2 - 6) of tosylmethyl isocyanide (TosMIC, 1)² presently under investigation. This prelim gives a brief outline of the synthesis of 5d and 6 (vide infra), and a comparison of asymmetric inductions achieved with 3 - 6.

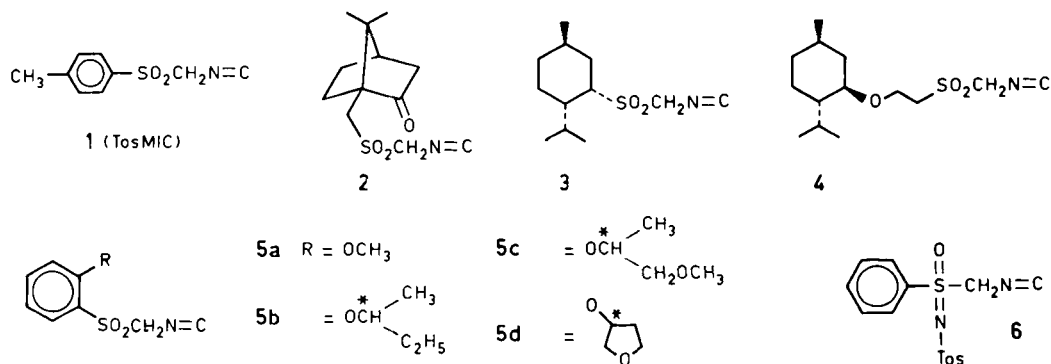
Asymmetric inductions were determined by reaction of 3 - 6 with trifluoroacetophenone (and with acetophenone) according to Eq. 1. Base-mediated cycloaddition of isocyanide 3³ to trifluoroacetophenone gave in 96% yield a mixture of diastereomeric oxazolines 7 (R = neomenthyl, X = O, Table, entry 2). Asymmetric induction determines the configuration at C(5) in 7 and, after acid hydrolysis,⁴ the configuration of hydroxy aldehyde 8. Thus, the e.e. (or o.p.)⁵ of 8 is a measure of the asymmetric induction exerted by 3. The d.e. of 7, which is easily determined by ¹H NMR (using the C(4)H signal) and by ¹⁹F NMR, also is a reflection of the asymmetric induction

power of $\underline{3}$. Unlike the e.e. of $\underline{8}$, the value of the d.e. of $\underline{7}$ is independent of the e.e. of $\underline{3}$.^{*} For example, the asymmetric induction obtained with $\underline{3}$, 18% as determined from the d.e. of $\underline{7}$, is reduced to an e.e. of 16% for $\underline{8}$, due to the use of $\underline{3}$ of 90% e.e. (entry 2).⁶ A somewhat higher asymmetric induction of 39% was obtained with isocyanide $\underline{5d}$, which carries potential chelating ether functions (entry 6). The best result, an asymmetric induction of 80% (80% d.e. of $\underline{7}$, entry 7), was obtained with sulfonimidoylmethyl isocyanide $\underline{6}$ which, unlike isocyanides $\underline{2}$ - $\underline{5}$, carries its chiral center next to the reactive methylene.

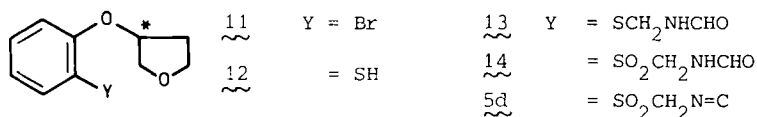
Absolute configuration and o.p. of **8** were determined by chemical correlation with Mosher's acid⁷ **9** (Eq. 1).

(R)-(-)-[o-(3-Tetrahydrofuranyloxy)phenylsulfonyl]-methyl isocyanide (5d, a crystalline solid, stable at room temperature) was prepared in 6 steps and

Chart 1



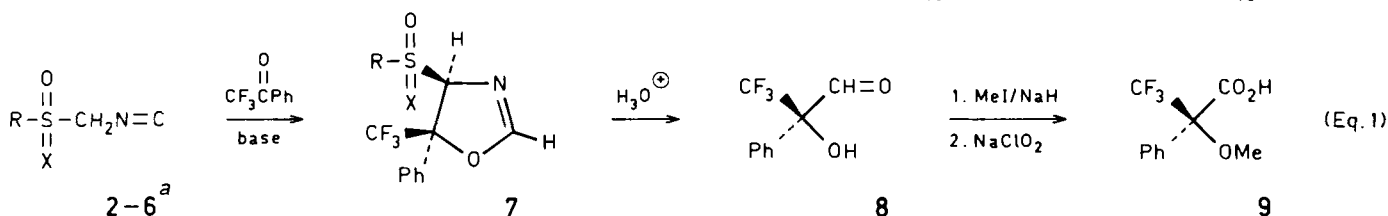
* Note added in proof: Changes in the e.e. of **3** only change the ratio of enantiomers of **7**, which is not reflected in NMR.



35% overall yield from optically pure (*S*)-(+)-3-hydroxytetrahydrofuran⁸ (**10**) as follows: (1) *o*-bromophenol and **10**-tosylate [mp 34.5–35.5°C, 80% from **10**] heated at 100°C (0.5 h) with powdered KOH gave **11** [69%, bp 115–7°C (0.15 mmHg), $[\alpha]_{\text{D}}^{20}$ -35.8° (CHCl₃)] with inversion and 50% racemisation; (2) **11** with BuLi and freshly sublimed sulfur in Et₂O (-78 to 15°C, 3 h) gave **12** [73%, bp 105–108°C (0.2 mmHg), $[\alpha]_{\text{D}}^{22}$ -30.6° (CHCl₃)]; (3) **12** with *N*-(tosylmethyl)formamide⁹ and *t*-BuOK analogously to a published procedure³ gave **13** (80%) which was

oxidized (mCPBA, 5 h, 20°C) to **14** (75%, mp 107.5–110°C, $[\alpha]_{\text{D}}^{22}$ -28.8° (CHCl₃)); (4) **14** was dehydrated with POCl₃ and *i*-Pr₂NH according to the method of Ugi¹⁰ to **5d** [66%, mp 107.4–108.2°C, $[\alpha]_{\text{D}}^{20}$ -40.8° (CHCl₃), e.e. 47%]. (-)-*S*-Phenyl-*N*-tosylsulfonyl-midoymethyl isocyanide (**6**) was prepared from partially resolved *S*-phenyl-*N*-tosylsulfonylimidoyl fluoride,¹¹ methyl isocyanide and 2.1 equiv. of BuLi in THF (-70°C, 1.0 h) in 35% yield, $[\alpha]_{\text{D}}^{21}$ -2.2° (CHCl₃), e.e. 34%.

TABLE. Reactions of Isocyanides **3** – **6** with Acetophenones to Oxazolines **7** and Hydroxy Aldehydes **8**



Entry	Isocyanides 2 – 6			Oxazolines 7 ^b			(R)-(+)- 8	
	mp (°C) ^c	c.y. (%) ^d	e.e. (%)	c.y. (%) ^e	d.e. (%) ^f	cond. ^g	c.y. (%) ^h	e.e. (%) ⁱ
1	2	oil	28 ^j	>98 ^k	–			
2	3	67.7–68.4	39 ^l	90 ^m	96	18 A (B)	60	16
3	4	oil	12 ⁿ	>98 ^k	[74]	33 C	47	31 ^o
4	5b	oil	13 ^p	50 ^q	[76]	40 D	50	15 ^o
5	5c	oil	29 ^r	rac.	[57]	33 D ^o		
6	5d	107.4–108.2	35	47 ^s	98	39 A (B)	43	18
7	6	93–96 (d)	35	34 ^s	91	80 B (A)	58	27

(a) For R and X, see Chart 1. (b) See footnote 6. (c) Oils are of limited stability at 20°C. (d) Overall chemical yields. (e) Chemical yields of crude **7**, which were hydrolyzed as such to **8**. (f) Determined by ¹H NMR [C(4)H doublets between δ 4.5–6.5] and by ¹⁹F NMR. (g) Reaction conditions: A = Ti(OEt)₄, *N*-Etpiperidine, CH₂Cl₂, 0°C; B = Triton B, THF, 20°C; C = PTC (NaOH-benzene), TEBA, 20°C; D = BuLi, THF, -78°C; conditions between brackets gave similar results. (h) Calculated on **7**. (i) Determined through **9**, ref. 5 and **7**. (j) In 4 steps from (+)-10-camphorsulfonic acid. (k) According to ¹³C NMR. (l) In 6 steps from (-)-menthol, ref. 3. (m) Determined via neomenthylthiol, ref. 3, and MeP(O)Cl₂, ref. 12. (n) In 5 steps from (-)-menthol. (o) The data between brackets refer to reactions with acetophenone, CH₃ instead of CF₃ in Eq. 1; also footnote 6. (p) In 5 steps from (R)-(+)-2-butanol. (q) Determined with Eu(dcm)₃. (r) In 5 steps from rac-1-methoxy-2-propanol. (s) Calculated from d.e. of **7** and e.e. of **8**.

References and Notes

- Chemistry of Sulfonylmethyl Isocyanides
31. For part 30, see: J. Moskal and A.M. van Leusen, *J. Org. Chem.* **51**, 4131 (1986).
- The chemistry of TosMIC has been reviewed:
(a) A.M. van Leusen, *Lect. Heterocycl. Chem.* **5**, S111 (1980); (b) A.M. van Leusen, in "Perspectives in the Organic Chemistry of Sulfur", B. Zwanenburg and A.J.H. Klunder, Eds., Elsevier, Amsterdam 1987, p. 119.
- D. van Leusen, P.H.F.M. Rouwette and A.M. van Leusen, *J. Org. Chem.* **46**, 5159 (1981).
- See ref. 2a.
- Throughout this paper, e.e. and o.p. are assumed to be equal.
- Further examples carried out with acetophenone, instead of trifluoroacetophenone, will be treated in a full paper.
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