

**Research Article** 

### Te(II)-induced heterocyclization of 1,2-alkadienephosphonates

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<b>Received:</b> 12.10.2012 • Accepted: 26.10.2013	٠	Published Online: 14.04.2014	٠	<b>Printed:</b> 12.05.2014
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**Abstract:** The reactivity of some 1,2-alkadienephosphonates towards phenyltelluryl halides was investigated. A plausible mechanism of the reaction is discussed.

Key words: 1,2-Alkadienephosphonates, electrophilic addition, phosphorus heterocycles

#### 1. Introduction

The applications of organophosphorus compounds as pharmaceutical, agricultural, and chemical agents are well documented.<sup>1,2</sup> Among them, oxaphosphole derivatives, which have structures similar to those of phosphosugars, have received particular interest.<sup>3,4</sup> Consequently, many attempts for their synthesis have been made. One of the easiest and most fruitful methods for the synthesis of these derivatives is electrophile-induced heterocyclization of 1,2-alkadienephosphonates.<sup>5</sup>

Keeping in mind that the scope of applications of organotellurides has been known for years because of their ready transformation to other compounds via reactions with organometallic reagents,  $^{6-10}$  here we wish to report the results of our study on the electrophilic addition of organotellurides to some 1,2alkadienephosphonates.

#### 2. Experimental

#### 2.1. Analytical methods

The <sup>1</sup>H NMR and <sup>31</sup>P NMR spectra were measured at normal probe temperature on a Bruker Avance DRX 250 MHz spectrometer using tetramethylsilane (TMS) (<sup>1</sup>H) and 85%  $H_3PO_4$  (<sup>31</sup>P) as internal references in CDCl3 solution.

Chemical shifts are given in parts per million (ppm) and are downfield from the internal standard. The infrared (IR) spectra were run on a Shimadzu IRAffinity-1 spectrophotometer. Elemental analyses were carried out by the University of Shumen Microanalytical Service Laboratory. Phenyltelluryl chloride was synthesized as described previously.<sup>11-15</sup>

Compounds 1, 3, 4, 7, and 9 were synthesized according to the literature.<sup>16-18</sup>

The solvents were purified by standard methods. All reactions were carried out in oven-dried glassware under an argon atmosphere and with exclusion of moisture. All compounds were checked for their purity on TLC plates. Melting points are uncorrected.

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### 2.2. Synthesis of 2-alkoxy-5-alkyl-5-alkyl-4-phenyltellanyl-5*H*-[1,2]-oxaphosphole 2-oxides and of 2-alkoxy-4-phenyltellanyl-1-oxa-2-phospha-[4,5]-dec-3-ene 2-oxide 2a–d

#### 2.2.1. General procedure

To a solution of **1** (5 mmol) in methylene chloride (10 mL) was added a solution of phenyltelluryl chloride (1.24 g, 5.2 mmol) in 5 mL of methylene chloride under stirring and cooling (-10 to -12 °C). After 1 h of stirring at the same conditions, the reaction mixture stood overnight, and was concentrated and recrystallized in heptane/benzene (2:1).

**2a**, cryst. colorless needles; 1.59 g (87%), mp °C (147–149), IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1540, 1235, 960 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.46 (d, J<sub>HP</sub> 26.0 Hz, 1H), 3.70 (d, J<sub>HP</sub> 12.2 Hz, 3H), 1.59 (s, 3H), 1.55 (s, 3H). <sup>31</sup> P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 33.09; Anal., Calcd. for  $C_{12}H_{15}O_3PTe$  (M<sub>r</sub> = 365.81): P 8.47; Found P 8.43; **2b**, cryst. colorless needles; 1.38 g (73%), mp °C (150–152), IR (KBr)  $\nu_{max}$  /cm<sup>-1</sup> 2980, 2677, 1580, 1235, 1000 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.49 (d, J<sub>HP</sub> 26.1 Hz, 1H), 4.17 (m, J<sub>HP</sub> 10.0 Hz, 2H), 1.36 (t, J<sub>HH</sub> 7.0 Hz, 3H) 1.52 (s, 3H), 1.57 (s, 3H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 32.0; Anal., Calcd. for C<sub>13</sub>H<sub>17</sub>O<sub>3</sub>PTe  $(M_r = 379.836)$ : P 8.15; Found P 8.11; **2c**, cryst. colorless needles; 1.46 g (77%), mp °C (149–150); IR (KBr)  $\nu_{max}$  /cm<sup>-1</sup> 2980, 2677, 1545, 1235, 980 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.55, 6.59\* (d, J<sub>HP</sub> 26.0 Hz, 1H), 3.80, 3.82\* (d, J<sub>HP</sub> 11.6 Hz, 2H), 1.51, 154\* (s, 3H), 1.89 (m, 2H), 0.92 (t, 3H). <sup>31</sup> P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 33.12; Anal., Calcd. for  $C_{13}H_{17}O_3PTe$  (M<sub>r</sub> = 379.836): P 8.15; Found P 8.10; (\*Additional signals for diastereomers); 2d, cryst. colorless needles; 1.44 g (71%), mp °C (155–157); IR (KBr)  $\nu_{max}$  /cm<sup>-1</sup> 2980, 2677, 1540, 1235, 990 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.42 (d, J<sub>HP</sub> 25.8 Hz, 1H), 3.80 (d, J<sub>HP</sub> 11.2 Hz, 2H), 1.68 (m, 10H). <sup>31</sup> P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 33.23; Anal., Calcd. for  $C_{15}H_{19}O_3PTe$  ( $M_r = 405.872$ ): P 7.63; Found P 7.60.

# 2.3. Synthesis of (5-alkyl-5-alkyl-2-oxo-4-phenyltellanyl-2,5-dihydro- $2\lambda^5$ -[1,2]-oxaphosphol-2-yl) dialkylamines 5a-c and of dialkyl-(2-oxo-4-phenyltellanyl-1-oxa- $2\lambda^5$ phospha-spiro[4,5]-dec-3-ene 2-yl)amines 6a–c

#### 2.3.1. General procedure

To a solution of **3** or **4** (5 mmol) in methylene chloride (10 mL) was added a solution of phenyltelluryl chloride (1.24 g, 5.2 mmol) in 5 mL of methylene chloride under stirring and cooling (-10 to -12 °C). After 1 h of stirring at the same conditions, the reaction mixture stood overnight, and was concentrated and recrystallized in heptane/benzene (2:1).

**5a**, cryst. colorless needles; 1.67 g (82%), mp °C (147–149); IR (KBr)  $\nu_{max}$ /cm<sup>-1</sup> 2980, 2677, 1589, 1225, 1004 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 5.88 (d, J<sub>HP</sub> 24.2 Hz, 1H), 1.40 (s, 3H), 1.58 (s, 3H), 1.00 (t, J<sub>HH</sub> 7.0 Hz, 3H), 2.93 (m, J<sub>HP</sub> 13.6 Hz, 2H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 28.3; Anal., Calcd. for C<sub>15</sub>H<sub>22</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 406.896): P 7.61, N 3.44; Found P 7.59, N 3.41; **5b**, cryst. colorless needles; 1.62 g (77%), mp °C (149–150); IR (KBr)  $\nu_{max}$ /cm<sup>-1</sup> 2980, 2677, 1590, 1235, 980 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.55, 6.59\* (d, J<sub>HP</sub> 22.4 Hz, 1H), 1.51, 154\* (s, 3H), 1.89 (m, 2H), 0.92 (t, 3H), 1.04 (t, J<sub>HH</sub> 7.0 Hz, 3H), 3.00 (m, J<sub>HP</sub> 12.1 Hz, 2H).

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<sup>31</sup> P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 27.9; *Anal.*, Calcd. for C<sub>16</sub>H<sub>24</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 420.922): P 7.36, N 3.32; Found P 7.33, N 3.29 (\*Additional signals for diastereomers); **5c**, cryst. colorless needles; 1.81 g (81%), mp °C (155–157); IR (KBr)  $\nu_{max}$ /cm<sup>-1</sup> 2980, 2677, 1588, 1225, 1000 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 5.87 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 0.98 (t, J<sub>HH</sub> 7.0 Hz, 3H), 2.92 (m, J<sub>HP</sub> 12.4 Hz, 2H). <sup>31</sup> P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 32.3; *Anal.*, Calcd. for C<sub>18</sub>H<sub>26</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 446.958): P 6.93, N 3.13; Found P 6.90, N 3.10.

**6a**, cryst. colorless needles; 1.89 g (87%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1580, 1230, 1000 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.08 (d, J<sub>HP</sub> 24.2 Hz, 1H), 1.40 (s, 3H), 1.58 (s, 3H), 1.24 (ss, 6H), 2.93 (m, 1H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 28.3; Anal., Calcd. for C<sub>17</sub>H<sub>26</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 434.948): P 7.12, N 3.22; Found P 7.10, N 3.19; **6b**, cryst. colorless needles; 1.66 g (74%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1597, 1235, 900 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.55, 6.59\* (d, J<sub>HP</sub> 26.0 Hz, 1H), 1.51, 154\* (s, 3H), 1.89 (m, 2H), 0.92 (t, 3H), 1.24 (ss, 6H), 2.93 (m, 1H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 28.3; Anal., Calcd. for C<sub>18</sub>H<sub>28</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 450.974): P 6.89, N 3.12; Found P 6.86, N 3.10; **6c**, cryst. colorless needles; 1.99 g (84%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1590, 1228, 1004 cm<sup>-1</sup>; <sup>-1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 5.87 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 1.24 (s, 6H), 2.93 (m, 1H); <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 5.87 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 1.24 (s, 6H), 2.93 (m, 1H); <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 5.87 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 1.24 (s, 6H), 2.93 (m, 1H); <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 28.3; Anal., Calcd. for C<sub>20</sub> H<sub>30</sub> O<sub>2</sub> NPTe (M<sub>r</sub> = 475.01): P 6.52, N 2.95; Found P 6.50, N 2.91.

# 2.4. Synthesis of (5-alkyl-5-alkyl-2-oxo-4-phenyltellanyl-2,5-dihydro- $2\lambda^5$ -[1,2]-oxaphosphol-2-yl) alkylamines 8a,b and of alkyl-(2-oxo-4-phenyltellanyl-1-oxa- $2\lambda^5$ phospha-spiro[4,5]-dec-3-ene 2-yl)amine 8c

#### 2.4.1. General procedure

To a solution of 7 (5 mmol) in methylene chloride (10 mL) was added a solution of phenyltelluryl chloride (1.24 g, 5.2 mmol) in 5 mL of the same solvent under stirring and cooling (-10 to -12 °C). After 1 h of stirring at the same conditions, the reaction mixture stood overnight, and was concentrated and recrystallized in heptane/benzene (2:1).

**8a**, cryst. colorless needles; 1.61 g (82%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1580, 1245, 1004 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.56–7.46 (m, 2H); 7.29–7.23 (m, 3H); 5.35 (d, J<sub>HP</sub> 27.75 Hz, 1H); 2.54 (m, 2H); 1.46 (s, 3H); 1.51 (s, 3H); 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H); 0.91 (t, 3H); <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 29.0; *Anal.*, Calcd. for C<sub>14</sub>H<sub>20</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 392.87): P 7.88, N 3.56; Found P 7.83, N 3.51; **8b**, cryst. colorless needles; 1.52 g (75%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1589, 1230, 960 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.55, 6.59\* (d, J<sub>HP</sub> 26.0 Hz, 1H), 1.51, 154\* (s, 3H), 1.89 (m, 2H), 0.92 (t, 3H), 2.54 (m, 2H), 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H); 0.91 (t, 3H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 28.3; *Anal.*, Calcd. for C<sub>15</sub>H<sub>22</sub>O<sub>2</sub>NPTe (M<sub>r</sub> = 406.896): P 7.61, N 3.44; Found P 7.58, N 3.40 (\*Additional signals for diastereomers); **8c**, cryst. colorless needles; 1.71 g (79%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1587, 1253, 1004 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 3H), 5.87 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 2.54 (m, 2H), 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H), 2.00 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 2.54 (m, 2H), 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H), 2.00 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 2.54 (m, 2H), 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H), 2.00 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 2.54 (m, 2H), 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H), 2.00 (d, J<sub>HP</sub> 10.00 Hz, 1H); 1.28–1.19 (m, 2H); 0.91 (t, 3H); <sup>31</sup>P NMR (250

MHz, CDCl<sub>3</sub>) ppm: 28.3; Anal., Calcd. for C $_{17}$ H $_{24}$ O $_2$ NPTe (M $_r$  = 432.932): P 7.15, N 3.23; Found P 7.11, N 3.20.

# 2.5. Synthesis of 4-(5-alkyl-5-alkyl-2-oxo-4-phenyltellanyl-2,5-dihydro- $2\lambda^5$ -[1,2]-oxaphosphol-2-yl) morpholines 10a,b and of 4-(2-oxo-4-phenyltellanyl-1-oxa- $2\lambda^5$ phospha-spiro[4,5]-dec-3-ene 2-yl)morpholine 10c

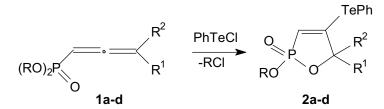
#### 2.5.1. General procedure

To a solution of **9** (5 mmol) in methylene chloride (10 mL) was added a solution of phenyltelluryl chloride (1.24 g, 5.2 mmol) in 5 mL of methylene chloride under stirring and cooling (-10 to -12 °C). After 1 h of stirring at the same conditions, the reaction mixture stood overnight, and was concentrated and recrystallized in heptane/benzene (2:1).

**10a**, cryst. colorless needles; 1.30 g (62%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1589, 1225, 1004 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.56–7.46 (m, 2H); 7.29–7.23 (m, 3H); 6.34 (d, J<sub>HP</sub> 23.0 Hz, 1H); 1.46 (s, 3H); 1.51 (s, 3H), 2.87, 3.76 (m, 8H); <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 33.42; Anal., Calcd. for C<sub>15</sub>H<sub>20</sub>O<sub>3</sub>NPTe (M<sub>r</sub> = 420.88): P 7.36, N 3.33; Found P 7.32, N 3.30; **10b**, cryst. colorless needles; 1.45 g (67%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1595, 1225, 1000 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 6.55, 6.59\* (d, J<sub>HP</sub> 26.0 Hz, 1H), 1.51, 154\* (s, 3H), 1.89 (m, 2H), 0.92 (t, 3H), 2.87, 3.76 (m, 8H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 34.12; Anal., Calcd. for C<sub>16</sub>H<sub>22</sub>O<sub>3</sub>NPTe (M<sub>r</sub> = 434.906): P 7.12, N 3.22; Found P 7.09, N 3.18 (\*Additional signals for diastereomers); **10c**, cryst. colorless needles; 1.40 g (61%), mp °C (147–149); IR (KBr)  $\nu_{max}/cm^{-1}$  2980, 2677, 1589, 1273, 998 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) ppm: 7.86–7.87 (m, 2H), 7.30–7.51 (m, 3H), 5.87 (d, J<sub>HP</sub> 23.5 Hz, 1H), 1.68 (m, 10H), 2.87, 3.76 (m, 8H). <sup>31</sup>P NMR (250 MHz, CDCl<sub>3</sub>) ppm: 33.22; Anal., Calcd. for C<sub>18</sub>H<sub>24</sub>O<sub>3</sub>NPTe (M<sub>r</sub> = 460.942): P 6.72, N 3.04; Found P 6.69, N 2.99.

#### 3. Results and discussion

In our first report on this subject<sup>19</sup> we demonstrated that the reaction of dialkyl esters of 1,2-alkadienephosphonic acids with phenyltelluryl chloride leads to the formation of 4-phenyltelluro-2,5-dihydro-1,2-oxaphosphole 2-oxide derivatives (Figure 1).



 $2a, R, R^1, R^2 = Me, 2b, R = Et, R^1 = R^2 = Me, 2c, R = R^1 = Me, R^2 = Et, 2d, R = Me, R^1 + R^2 = cyclohexyl$ 

Figure 1. Reaction of dialkyl esters of 1,2-alkadienephosphonic acids with phenyltelluryl chloride.

In 2007, Yuan and co-workers reported the same results using different synthetic conditions.<sup>20</sup>

Continuing our investigations on this reaction, we studied the reaction of N,N-dialkylamido-O-alkyl-1,2alkadienephosphonates previously described by us,<sup>17</sup> with the same reagent, and established that in all cases with good yields the oxaphosphole derivatives 5a-c and 6a-c were obtained (Figure 2):

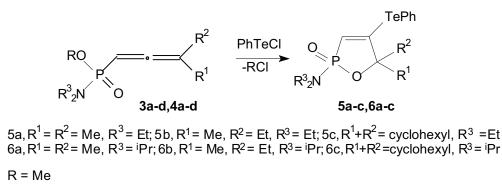
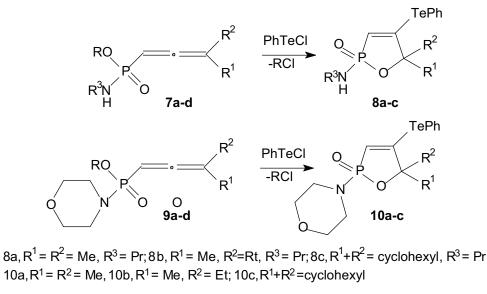


Figure 2. Reaction of N,N-dialkylamido-O-alkyl-1,2-alkadienephosphonates with phenyltelluryl chloride.

The results reported above encourage us to investigate the reactivity of N-alkylamido-O-alkyl-1,2-alkadienephosphonates as well as the reactivity of N-morpholino-O-alkyl-1,2-alkadienephosphonates also previously reported by us.<sup>18</sup> We expected both substrates to react with phenyltelluryl chloride with formation of the corresponding 2,5-dihydro-1,2-oxaphosphole 2-oxide derivatives (Figure 3).



R = Me

Figure 3. Reaction of N-alkylamido-O-alkyl-1,2-alkadienephosphonates and of N-morpholino-O-alkyl-1,2-alkadienephosphonates with phenyltelluryl chloride.

All the synthetic results obtained as well as our previous experience<sup>5</sup> give us reason to suggest the following plausible mechanism of the telluro-induced cyclization of 1,2-alkadienephosphonates (Figure 4):

The attack of the reagent affecting the  $C^2-C^3$  double bond of the allenephosphonate system leads to the formation of "onium" intermediate **A**, which is in equilibrium with carbocation **B**. The latter can be transformed to quaziphosphonium intermediate **C**, which undergoes dealkylation (Michalis–Arbuzov reaction – second stage) to afford the final 2,5-dihydro-1,2-oxaphosphole 2-oxide derivatives **2**, **5**, **6**, **8**, and **10**.

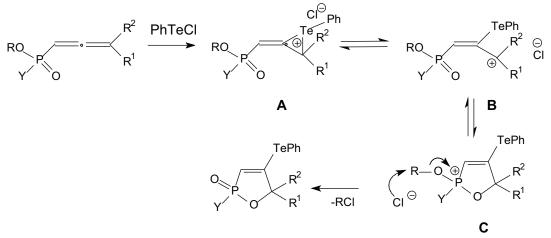


Figure 4. Plausible mechanism of the telluro-induced cyclization of 1,2-alkadienephosphonates.

#### References

- Cupta, H. C. L. In *Insecticides: Toxicology and Uses*; Agrotech Publishing Academy Press: Udaipur, India, 1999, p. 51211.
- 2. Matsumara, F. In Toxicology of Insecticides, 2nd ed.; Plenum Press: New York, NY, USA, 1985; pp 62–90.
- Racha, S.; Vageese, C.; Vemishetti, P.; El-Subbagh, H. I.; Abushanab, E.; Panzica, R. P. J. Med. Chem. 1996, 39, 1130–1135.
- Reist, E. J.; Sturm, P. A.; Pong, R. Y.; Tanga, M. J.; Sidwell, R. W. In Synthesis of Acylonucleoside Phosphonates for Evaluation as Antiviral Agents in Nucleotide Analogues as Antiviral Agents; Martin, J. C., Ed., American Chemical Society: Washington D.C., USA, 1989; pp. 17–34
- For review see: Enchev, D. D. Topics in Heterocyclic Chemistry, Springer: Berlin/Heidelberg, Germany, 2010, 21, 23–63.
- 6. Comasseto, J. V.; Ling, W. L.; Petragnani, N.; Stefani, H. A. Synthesis 1997, 373-404.
- 7. Dabdoub, M. J.; Justino, A.; Guerro, P. G. Jr. Organometallics 1998, 17, 1901–1903.
- 8. Dabdoub, M. J.; Baroni, C. M. J. Org. Chem. 2000, 65, 54–60.
- 9. Dabdoub, M. J.; Begnini, M. L.; Guerro, P. G. Jr.; Baroni, C. M. J. Org. Chem. 2000, 65, 61–67.
- 10. Petragnani, N.; Stefani, H. A. Tetrahedron 2005, 61, 1613-1679.
- 11. Schultz, P.; Klar, G. Z. Naturforsch. 1975, 30B, 40-43.
- 12. Klapotke, T. M.; Krumm, B.; Schwab, I. Z Kryst. NCS 2005, 220, 594–596.
- 13. Alcock, N. W.; Harrison, W. D. J. Chem. Soc. Dalton Trans. 1984, 869-875.
- 14. Maksimenko, A. A.; Zaharov, A. V.; Sadekov, I. D. Russ. Chem. Rev. 2000, 69, 861–882.
- Klapotke, T. M.; Krumm, B.; Mager, P.; Piotrowski, H.; Schwab, I.; Vogt, M. Eur. J. Inorg. Chem. 2002, 2701– 2709.
- Mark, V. In Selective Organic Transformations, Thyaraian, B. S., Ed., John Wiley & Sons, New York NY, USA, 1970, p. 319.
- 17. Angelov, Ch. M.; Enchev, D. D. Phosphorus Sulfur Silicon and the Related Elem. 1987, 34, 163-168.
- 18. Enchev, D. D. Phosphorus Sulfur Silicon and the Related Elem. 2005, 180, 2131–2135.
- Stankolov, S. P.; Enchev, D. D. Proc. of the 5th International Conference of the Chemical Societies of the South-East European Countries, September 10–14, 2006, Ohrid, FIROM, OCH-54, 478.
- 20. Yuan, J.; Ruan, X.; Yang, Y.; Huang, X. Synlett 2007, 2871–2875.

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