Isolation and Direct Observation of Intramolecular Hydroacylation of a *cis*-Hydridopent-4-enoylrhodium(III) Complex

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A stable *cis*-hydridopentenoylrhodium(\mathfrak{m}) trimethylphosphine complex, isolated from oxidative addition of pent-4-enal to RhCl(PMe₃)₃, undergoes intramolecular hydroacylation to cyclopentanone.

Rhodium-catalysed aldehyde hydroacylation of olefins,^{1–4} although still in its initial development stages, is a useful method, especially the intramolecular version, for synthesis of cyclopentanones.^{1–3} Attempts to isolate a postulated key intermediate, a pent-4-enoylhydridorhodium complex, have failed hitherto.¹ We now report the isolation of such an intermediate and the direct observation of its intramolecular cyclization.

Following our recent observation that cis-hydridoalkylrhodium(III) trimethylphosphine complexes are relatively stable owing to the very slow rate of PMe₃ dissociation from them.⁵ we have studied the reaction of $RhCl(PMe_3)_3^6$ (1) with pent-4-enal. Addition of excess of pent-4-enal to an orange solution of (1) in toluene at 25 °C under N_2 resulted in a colour change to yellow after 30 min. Removal of the solvent under high vacuum and crystallization of the residue from toluene–pentane at -40 °C afforded the complex (2) as light yellow crystals [81% yield, i.r. (Nujol) 1940 (s, ν_{Rh-H}) and 1625 cm⁻¹ (s, $\nu_{C=0}$); ¹H n.m.r. (C₆D₆) δ 1.23 (d, J 7.1 Hz, 9 H, PMe₃), 1.32 (t, J 3.0 Hz, 18 H, 2 PMe₃), 2.32 (superimposed dt, J₁ 7.1, J₂ 7.2 Hz, 2 H, =CH-CH₂), 2.79 (t, J 7.2 Hz, 2 H, CH₂CO), 5.00 (d, J 9.9 Hz, 1 H, HCH=), 5.10 (d, J 17.7 Hz, 1 H, HCH=), 5.95 (m, 1 H, =CH-), and -8.26 (ddt, $J_{H-P,trans}$ 189.3, $J_{\text{H-Rh}}$ 18.3, $J_{\text{H-P},cis}$ 14.8 Hz); ³¹P{¹H} n.m.r. $(C_6D_6) \delta = -6.89$ (dd, J_{P-Rh} 114.3, J_{P-P} 31.3 Hz, 2 P), and -23.74 p.p.m. (dt, J_{P-Rh} 93.2, J_{P-P} 31.2 Hz, 1P)]. Since the ¹H and ³¹P n.m.r. spectra are consistent only with a structure containing a PMe_a ligand trans to the hydride ligand and having two identical trans PMe3 ligands, the hydrido and acyl ligands have to be in mutually cis positions, as shown in structure (2).

Although stable at room temperature, complex (2) undergoes complete decomposition at 50 °C (t_{\pm} ca. 1 h), resulting in intramolecular hydroacylation to form cyclopentanone (72% yield) and (1) (71%), in addition to a competing migra-



Scheme 1. $L = PMe_3$.

tory elimination process which yields RhCl(CO)(PMe₃)₂ (3) (27%), but-1-ene (27%), and PMe₃ (25%). Since (1) is being regenerated, this process completes a catalytic cycle for pent-4-enal intramolecular hydroacylation. Indeed, heating pent-4-enal with a catalytic amount of (1) at 50 °C for 2 h affords cyclopentanone (300% yield based on Rh) and but-1-ene (40%). Although Rh-promoted cyclization of pent-4-enal to cyclopentanone is well known,¹⁻³ catalytic activity was observed before only in the presence of ethylene.

In order to provide for double bond co-ordination, PMe₃ dissociation from (2) is expected. To test this, and to determine which ligand dissociates, (2) was partially decomposed (40 °C; 1 h) in the presence of excess of $P(CD_3)_3$.† The ¹H n.m.r. doublet at δ 1.23 completely disappeared in recovered (2), whereas the triplet at δ 1.32 was not affected. Thus, reversible dissociation of the PMe₃ ligand *trans* to the hydride ligand takes place, whereas the PMe₃ ligands *cis* to the hydride do not dissociate. This is probably a result of the relatively large *trans* effect exerted by the hydride ligand.

Based on these observations, a plausible mechanism for pent-4-enal intramolecular hydroacylation includes fast oxidative addition of pent-4-enal to (1) to yield complex (2),

† Prepared from CD₃MgI and P(OPh)₃.

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followed by slow, reversible PMe₃ dissociation from (2) to form a common unsaturated square-pyramidal intermediate (4). A trigonal bipyramidal intermediate is less likely than (4) because Berry pseudorotations in such a complex may lead to $P(CD_3)_3$ incoporation in positions *cis* to the hydride as well. Intramolecular insertion into the Rh–H bond of (4) followed by reductive elimination leads to cyclopentanone, whereas competing migratory elimination leads to the decarbonylation products (3) and but-1-ene (Scheme 1).

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