

1,3-Disubstituted-2-carboxy Quinolones: Highly Potent and Selective Endothelin A Receptor Antagonists

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Abstract—The design, synthesis, and in vitro biological activity of a series of 2-carboxy quinolone antagonists selective for the endothelin A receptor are presented. Introduction of a second acid group in position 3 of the quinolone ring increases dramatically the selectivity for ET_A. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

Endothelin 1 (ET-1), a 21 amino-acid peptide, belongs to a family of highly potent endogenous vasoconstrictor agents along with ET-2 and ET-3.¹ It exerts its action by interacting with at least two specific G protein coupled receptors, ET_A and ET_B.²

ET_A, which is selective for ET-1 and ET-2 over ET-3, is mainly found in vascular smooth muscle cells (VSMC) and mediates vasoconstriction³ and VSMC proliferation.⁴ ET_B, which is non-selective for the three isoforms, mediates either vasoconstriction or vasorelaxation, depending on its tissue location.^{5,6}

Endothelin has been implicated in a number of human diseases, including hypertension, congestive heart failure, acute renal failure, pulmonary hypertension, restenosis, stroke and cerebral vasospasm.⁷

Several peptidic as well as non-peptidic endothelin antagonists have been reported in the literature, being either ET_A selective (i.e., BQ 123,⁸ BMS 182874,⁹ A-127772,¹⁰) ET_B selective (BQ 788¹¹) or non selective (i.e., PD 142893,¹² Bosentan,¹³ SB 209670¹⁴).

During the course of our screening of G-protein coupled receptor antagonists, we discovered that 6-substituted quinolones, which were fairly good antagonists of the

AT1 receptor,¹⁵ also showed moderate affinity for both ET_A and ET_B receptors (Fig. 1).

Struck by the high affinity level of SB 209670 (**1**) for both ET_A and ET_B receptors, and by the major contribution of a methylenedioxyphenyl group and an acid function, we superimposed the indane ring of SB 209670 with the quinolone ring. A good overlap could be obtained with 1,3-disubstituted-2-carboxylic quinolones (**2**) (Fig. 2).

Herein we describe the synthesis and structure activity relationships of this series, yielding highly potent and selective ET_A antagonists.

Chemistry

As no synthesis of 1,3-disubstituted-2-carboxy quinolones had been described, we envisaged to create the quinolone ring by cyclisation of the corresponding oxamate **6**. Thus, *o*-nitroacetophenone was reacted with the appropriate aldehyde at room temperature overnight, in the presence of 2N sodium hydroxide in methanol.¹⁶ As few as 0.1 to 0.5 equiv of sodium hydroxide are sufficient for the reaction to occur. For base sensitive aldehydes, the reaction was performed in the presence of a catalytic amount of piperidine and acetic acid in refluxing toluene, yielding the corresponding chalcones in 80 to 95% yield. Both the double bond and the nitro group were cleanly reduced over platinum oxide, giving compound **4** (Scheme 1).

The resulting aminoketones **4** were then submitted to reductive amination with the corresponding aldehyde

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yielding compounds **5** in 50 to 70% yield. Condensation of **5** with ethyl oxalyl chloride proceeded smoothly to give compound **6** quantitatively.

The key step of the synthesis was the cyclization of **6** into the corresponding quinolone ring. This could in fact be achieved in high yield under a variety of conditions (K_2CO_3 , MeOH; DBU, EtOH; EtONa, EtOH; *t*BuOK, THF), the two first leading to cleaner reaction mixtures. Finally, the ester group was saponified with 2 equiv of 6N KOH in EtOH under heating, and the compounds tested as their potassium salts.

Most of the derivatives showed good aqueous solubility (>20 mg/mL).

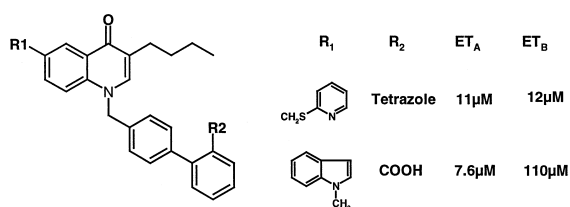


Figure 1.

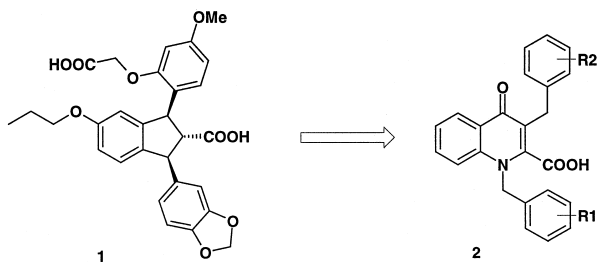


Figure 2.

Biological results

All the compounds were tested for their ability to inhibit ^{125}I -ET1 binding to rat ventricle (ET_A) or to rat cerebellum (ET_B).¹⁷

From the results in Table 1, it appears that the position of the methylenedioxyphenyl group is very important in this series, **9** being 25 times more potent than **10** both on ET_A and ET_B. We then investigated the effect of various substituents on the methylenedioxyphenyl group in R₁ (cpds **11a–16**); of these, derivatives with a substituent (Cl, vinyl, Et) next to the methylene group showed the highest activity, five to ten times more potent than compound **9**, giving rise to low nanomolar derivatives. In addition, all these compounds showed a good selectivity for ET_A, **11a** being the most dissociated.

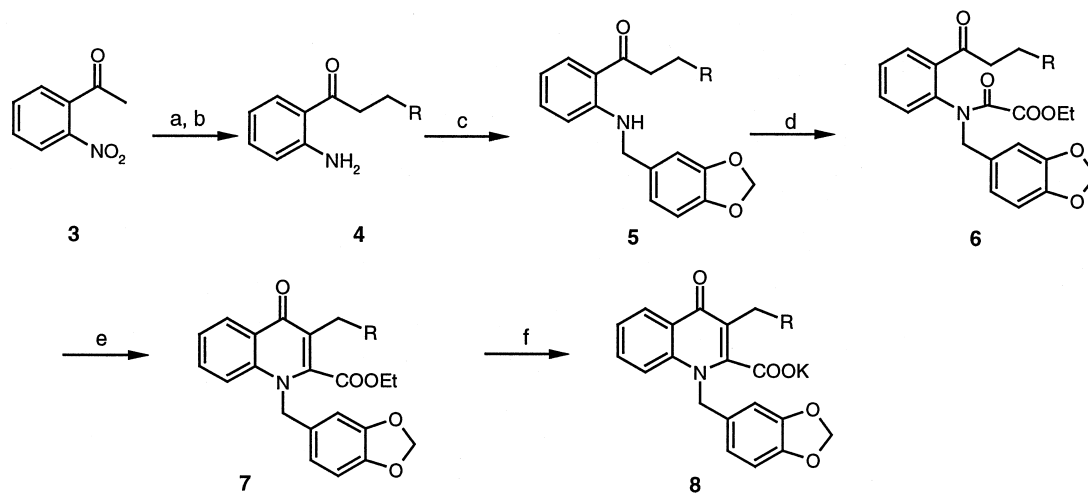
Therefore this group was retained in position 1 of the quinolone ring and the influence of R₂ was investigated (Table 2).

Several types of substituents were evaluated as R₂: aromatic (**11c–11k**), heteroaromatic (**11l–11n**) and non-aromatic (**11o–11r**).

Clearly, the deletion of CH₂R₂ leads to a loss of activity (**11b**). All the aromatic analogues tested are in the same range of activity, except compound **11e**, which incorporates the side chain of SB 209670 and is surprisingly far less active. The selectivity on ET_A which is always good, increases dramatically through the introduction of an acid function in **11f** making this compound 4000 times more potent on ET_A than on ET_B.¹⁸

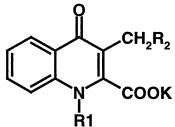
The same phenomena is observed for non-aromatic compounds: introduction of an acid or an ester (**11p**, **11q** and **11r**) function increases the selectivity (>1000).

However, introduction of an acid function in the *meta* position does not lead to such a boosting in selectivity (**11i**).



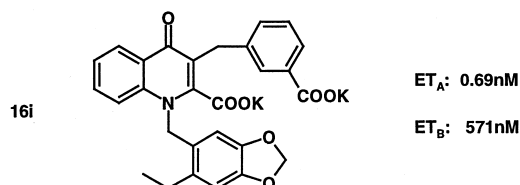
Scheme 1. Reagents and conditions: (a) RCHO, 0.1–0.5 equiv 2N NaOH, MeOH or cat. Piper; AcOH, Toluene, reflux, 80–95%; (b) H₂, PtO₂, EtOAc, 100%; (c) 3,4-methylenedioxybenzaldehyde, NaBH₃CN, AcOH, MeOH, 50–70%; (d) ClCOCOOEt, THF, NEt₃, rt, 100%; (e) K₂CO₃, EtOH or DBU, EtOH, 80–90%; (f) 6N KOH, EtOH, reflux, 90%.

Table 1. Endothelin receptor binding affinity (nM)

				
Compound	R ₁	CH ₂ R ₂	ET _A	ET _B
9			10.7	606
10			255	16000
11a			1.6	216
12			7.8	244
13			18.9	187
14			9.5	244
15			1.8	70
16			1	69

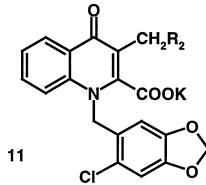
At the receptor level, there is no difference in activity or selectivity while going from the acid to the ester (**11i** versus **11k**, **11q** versus **11r**). When introducing an acid function, there is no improvement on ET_A activity (**11c** versus **11f**) but only a strong drop in ET_B activity, suggesting a significant difference between ET_A and ET_B receptor structure in this region.

In vivo, compounds of type **11** and **16** gave the best results in the pithed rat assay¹⁹ (ED₅₀ = 1 mg/kg IV), therefore the best substituents found in **11** as R₂ were introduced in **16**. This led to compound **16i** which shows an ED₅₀ of 0.1mg/kg IV in the pithed rat, together with a sub-nanomolar affinity for ET_A. Moreover, this compound is orally active in the ET-1 induced death in mice²⁰ (ED₅₀ = 10 mg/kg po).



In conclusion, we have found a new series of highly potent and selective ET_A antagonists, which are valuable tools for further understanding the physiological and pathophysiological role of endothelin.

Table 2. Endothelin receptor binding affinity for compounds **11**

			
Compound	CH ₂ R ₂	ET _A	ET _B
11b	H	8700	29,000
11c		4.1	157
11d		10.9	570
11e		54.3	549
11f		4.6	17,100
11g		1.9	789
11h		5.7	166
11i		3	500
11j		5.6	190
11k		8.2	747
11l		20.4	1603
11m		5.7	1890
11n		29	1900
11o		1.1	100
11p		93	9500
11q		3.4	6630
11r		4.7	5000

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19. Antagonists were injected as a bolus (1 mL/kg), ten min before endothelin was injected in a cumulative dose manner every two min (0.1–10 µg/kg, 0.1 mL/dose/rat) in pithed rats (male Sprague-Dawley).
20. ET-1 (10 nmol/kg IV) was used to induce sudden death in female Swiss mice (*n*=10). Drugs were administered 10 min before ET-1 for iv administration and 30 min for po administration.